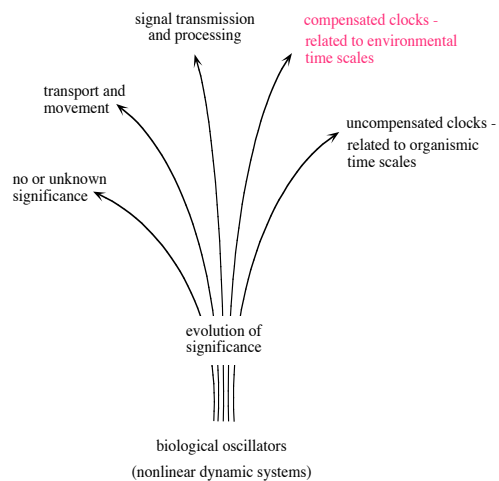


Dynamics, adaptation and fluctuations in bio-networks
March 24-27, Santa Barbara, California

Modelling circadian clocks and temperature-compensation

Peter Ruoff
Dartmouth Medical School, USA
Stavanger University Center, Norway

Biological Oscillators



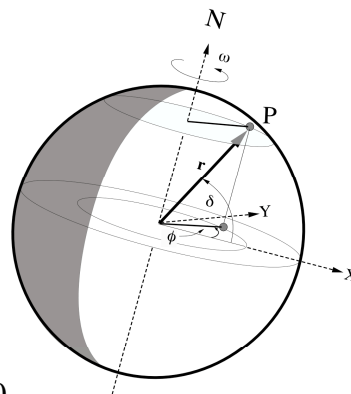
Circadian rhythms



- are important for the adaptation of organisms to their environment
 - for example to anticipate day/night changes as well as seasonal changes (important for many physiological processes).
- act as *clocks*, they have *temperature-compensation* as well as other *homeostatic regulation mechanisms* of their period against environmental influences.

Describing circadian clocks purely mathematically

The rotation/oscillation of point P along the coordinates X, Y describes an harmonic oscillator, which is used, for example, as the basis of the (harmonic) *Cosinor* or *Van der Pol* description of circadian rhythms.



$$\begin{aligned} \frac{dX}{dt} &= -\omega Y \\ \frac{dY}{dt} &= \omega X \end{aligned} \quad \Leftrightarrow \quad \ddot{X} + \omega^2 X = 0$$

$\omega = 1.157 \times 10^{-5} \text{ Hz}$

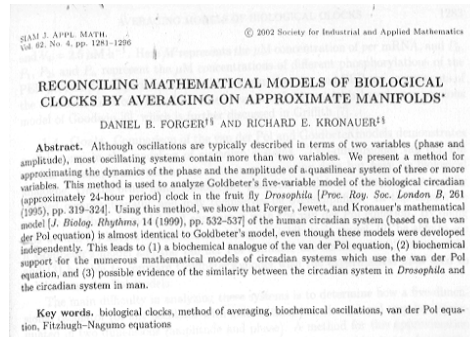
Describing circadian clocks purely mathematically

Van der Pol oscillator:

$$\frac{dX}{dt} = \epsilon [Y + \epsilon X - \epsilon X^3]$$

$$\frac{dY}{dt} = -\omega X$$

Normally ϵ is chosen to be small (0.1-0.25), which makes the van der Pol oscillator quasilinear and very similar to the harmonic oscillator.

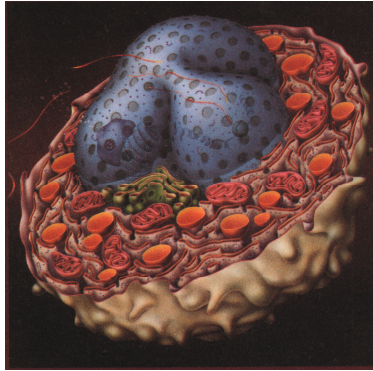


Physical chemical approach to model circadian rhythms

”..... All living things are physico-chemical machines.....How can you make progress if you do not know physical chemistry?”

Sinclair Lewis, *Arrowsmith*

Modelling circadian clocks in reaction kinetic terms



artistic view of an eukaryotic cell

All chemical change, no matter how complex, is believed to be due to chemical (*elementary*) processes.

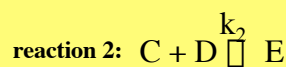
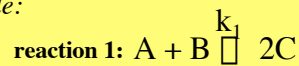
If we would know all processes and their kinetics in a single cell, we would be able to give a *quantitative* description of the cell's physiology through the chemical rate equations.

Chemical Rate Equations

Considering a set of reacting metabolites $\{S_i\}$, the reaction system is described by a set of coupled differential equations:

$$\frac{dS_i}{dt} = \sum_j n_{ij} \varpi_j \quad \begin{array}{l} n_{ij} \text{ denote stoichiometric coefficients} \\ \varpi_j \text{ are reaction rates of individual process "j"} \end{array}$$

Example:



ϖ

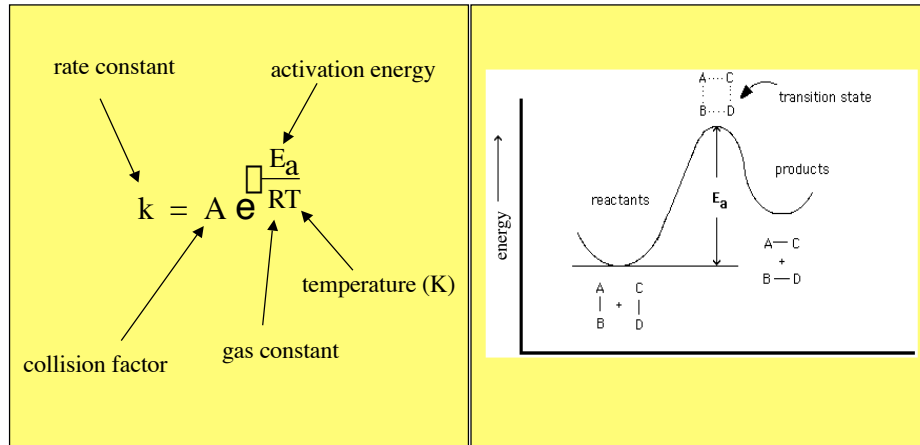
$$\frac{dC}{dt} = 2\varpi_1 - \varpi_2$$

$$\varpi_1 = k_1 AB$$

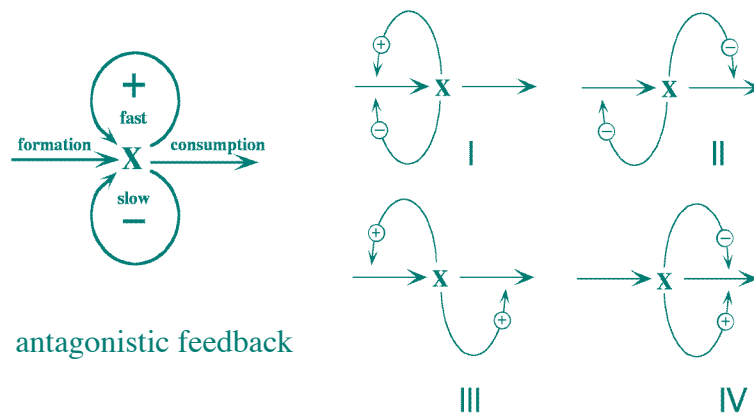
$$\varpi_2 = k_2 CD$$

k_i : rate constant

Influence of temperature on reaction rates:
the Arrhenius equation

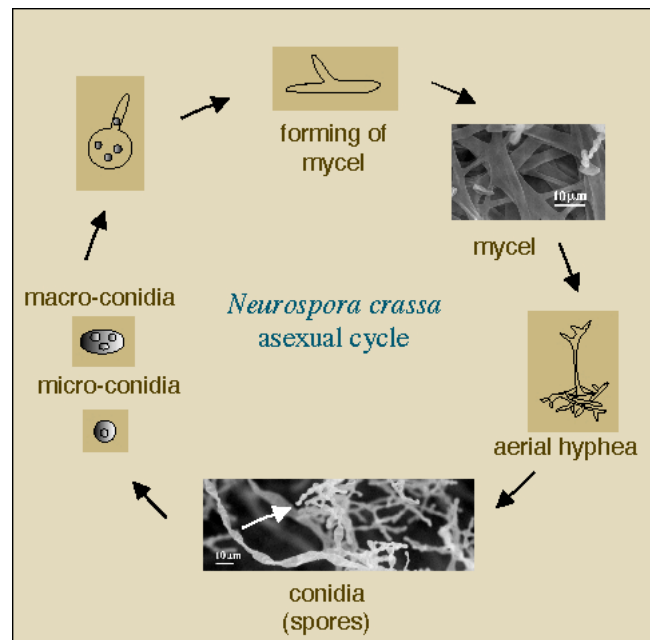


Franck's antagonistic feedback concept
in physico-chemical oscillators

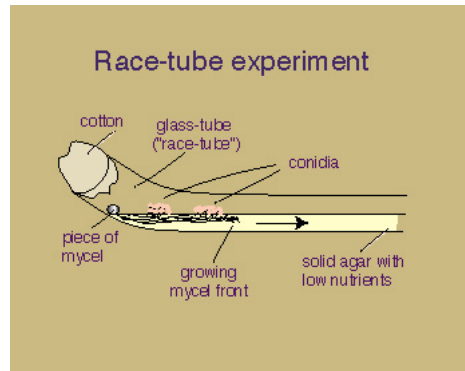


Franck, Angew. Chem. Int. Ed. Engl. 17 (1978) 1-15

Neurospora crassa: A Model Organism for the Study of Circadian Rhythms



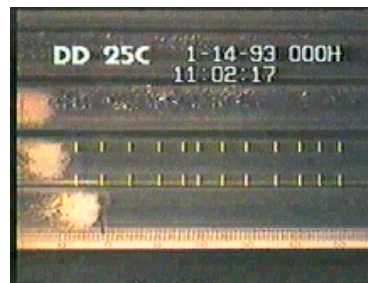
Racetube assay of clock rhythm in *Neurospora crassa*



Neurospora crassa: A Model Organism for the Study of Circadian Rhythms

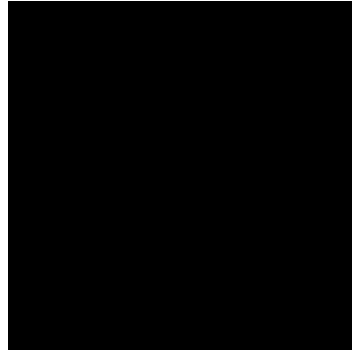


Influence of extracellular pH on *Neurospora*'s circadian rhythm

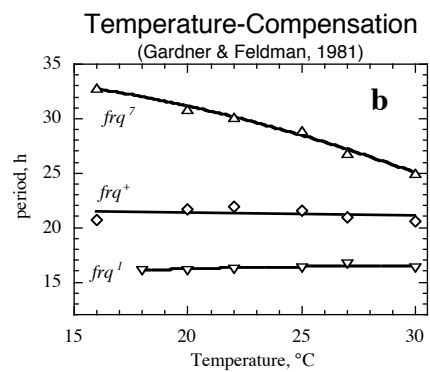
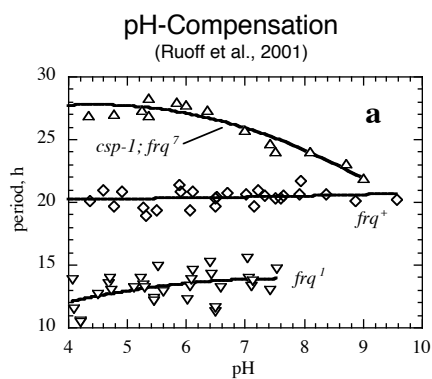


Jason C. Thoen and Van Gooch:
"Time Lapse Video Showing an Internal Circadian Clock in Mold (*Neurospora*) Growth"

Neurospora crassa growth and banding
in a Petri dish



pH- and temperature-compensation in
mutants of *Neurospora crassa*



Temperature-compensation in pH-oscillators

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PCCP

general scheme of a pH oscillator:

$$A^- + H^+ \rightleftharpoons AH \quad (1)$$

$$AH + H^+ + (B) \rightarrow 2H^+ + P^- \quad (2)$$

$$H^+ + (C^-) \rightarrow CH \quad (3)$$

Fig. 2 Temperature compensation of the frequency in the calculated pH time series. Curve (a) was simulated with input concentrations, flow rate, and the basic set of the rate constants shown in Fig. 1(a). All the rate constants were 5 times higher than their basic values for curve (b) resulting in high frequencies. The same frequency as in curve (a) was obtained with the following constants: $k_1 = 3.0 \times 10^{-6} \text{ M}^{-1} \text{ s}^{-1}$; $k_2 = 1.1 \times 10^4 \text{ s}^{-1}$; $k_3 = 3.0 \times 10^4 \text{ M}^{-1} \text{ s}^{-1}$; $k_4 = 0.40 \text{ s}^{-1}$ (curve c).

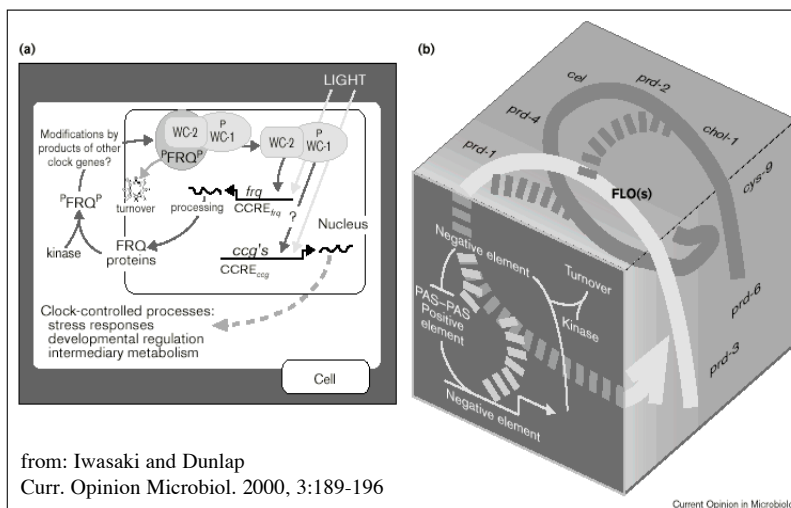
opposing effects on period from process (2) and (3)

Fig. 4 Period length as a function of temperature at different input sulfite concentrations $[H_2O_2] = 0.0135$; $[SO_3^{2-}] = 0.0050$; $[H^+] = 5.0 \times 10^{-8} \text{ M}$; $k_1 = 3.0 \times 10^{-6} \text{ s}^{-1}$; $[SO_3^{2-}] = 0.0018$ (triangles), 0.0017 (full circles), 0.0015 M (open circles).

Phys. Chem. Chem. Phys., 2002, 4, 5265–5269 5267

experimental finding of TempComp in the H₂O₂-sulfite-thiosulfate system

Negative feedback loop of *frequency* gene: Putative circadian pacemaker of the *Neurospora* clock

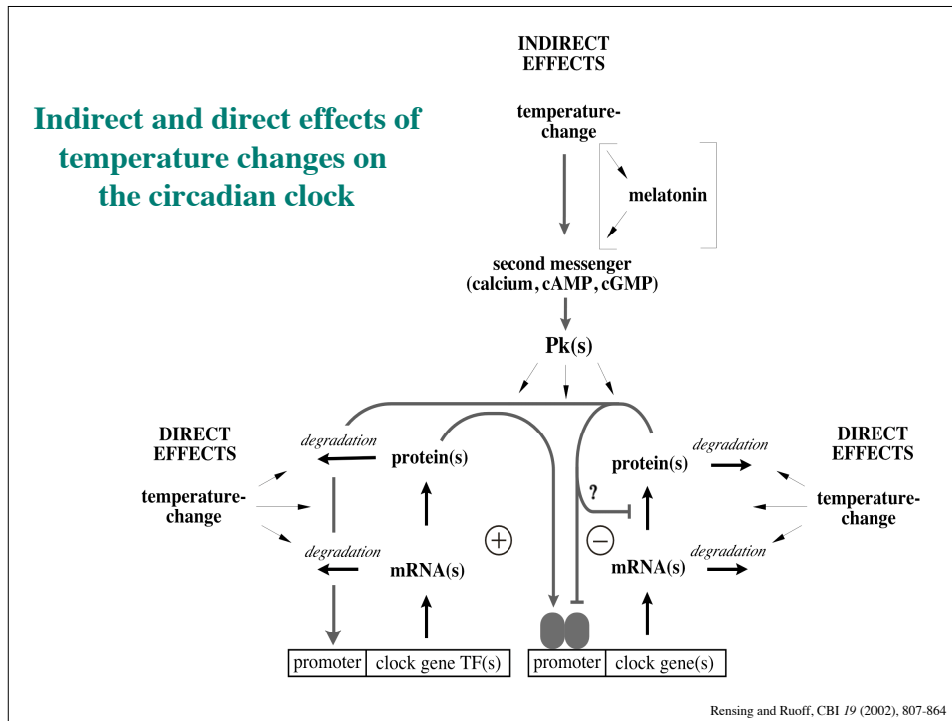


FRQ-protein

```

>sp|P19970|FRQ_NEUCR FREQUENCY CLOCK PROTEIN - Neurospora crassa.
MADSGDKSQGMRPPPFDSRGHPLRRASPDKSTITLENHRLARDTSSRVTS
SSALGVTESQPQLKSSPTRRNSSGESEPTNWFNQSNRNPAAAFHDESHIM
EVDPPFFYQKETDSSNEESRYPPGRNPVHPPGGVQLPGFRPVAAHSTAADD      coiled-coil
YRSVIDDLTVENKRLKEELKRYKQFGSDVMRKEKLFKIVHGLPRKKRE
LEATLRHFASLGDSEESTSQRRKTGRHGTA VYSSGVLSKHDSSSSRS
RPVDSAYNSMSTGRSSHAPHSSGSLGRPSLTRAHSVGTQKVENYL RDTP
DGLLPHHIVMTDKEKKLVVRRLEQLFTGKISGRNQSNRNSMPMSMDAPLA
PEGTNMAPP RPPPEGLREACIQLDGDNPRKNRSSKDNGSASNSGGDQTE
LGGTGTGSGDGS GSGGRGTGNNTSPPGAIAPDQRPTRPRDLDPDRVQIPSE
NMDYIRHLGLVSPPEFLQGSRTSYQDVAPDAEGWVYLNLLCNLAQLHMVNV
TPSFIRQAVSEKTKFQLSADGRKIRWRGGTDGTFSSDSEDKSQQSPM
TEDTEGSDKNGRRKKRKTQASSEIGRFGPSRSPSDFHYKPMFVHRNS
SSIETSLEESMSQGSSEDAVDES NMGNSKWDFSGSGTTQRRKRRYDGAIV
YYTGAPFCTDLSGDPGDMSPTAQMTAGREVEGSGGDEVEHVLQRTLSGS
SLPIRPLSDDRARVAEVLDFDPGNPELVADDGSSPNDEDFVFPWCEDPA
KVRIQPIAKEVMEPSGLGGVLPDDHFVMLVTRRVVRPILQRQLSRSTTS
EDTAEFIAERLAAIRTS SLP RSHRLTVAPLQVEYVSGQFRRLNPAPLP      PEST-2
PPAIFYPPFSTDSSWDDGDLASDDEVEVEEEDSYSEGQISRRANPHFS      PEST-2
DNNTYMRKDDLAFDTE TDV RMDSDNRLSDSGHNMRRAMPRAEAVDGD DS
PLAAVTGKEVDIVHTGSSVATAGGAESGYSSSMEDVSSS
    
```

modelling temperature-compensation



A theory of temperature-compensation

Assume that we know all (elementary) reactions $\{R_i\}$ defining a cell's physiology and its circadian rhythm.

$$\text{process } R_i(T) \quad k_i = A_i \exp(-E_i/RT)$$

$$\text{period } P = f(k_1, k_2, k_3, \dots, k_i, \dots)$$

Antagonistic balance in temperature

$$P = f(k_1, k_2, k_3, \dots, k_i, \dots)$$

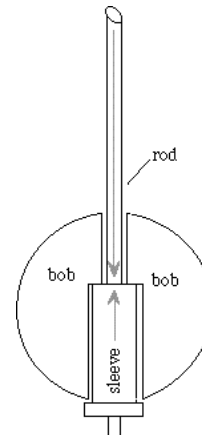
$$\frac{dP}{dT} = \sum_i \left(\frac{\partial f}{\partial k_i} \right) \left(\frac{\partial k_i}{\partial T} \right) = 0$$

$$\sum_i \left(\frac{\partial \ln(f)}{\partial \ln(k_i)} \right) E_i = 0$$

Concept of opposing reactions
(antagonistic balance in temperature)

$$\sum_j \left(\frac{\partial \ln(f)}{\partial \ln(k_j)} \right) E_j = \sum_i \left(\frac{\partial \ln(f)}{\partial \ln(k_i)} \right) E_i$$

(P-increasing) (P-decreasing)



Mechanical analogy:
Temp-Comp Invar pendulum

Infinite number of possibilities to realize temperature-compensation!

- For a given reaction kinetic oscillator model there is an infinite number of activation energy combinations that will lead to temperature compensation. Evolution has "realized" some of them.

$$\sum_j \left(\frac{\partial \ln(f)}{\partial \ln(k_j)} \right) E_j = \sum_i \left(\frac{\partial \ln(f)}{\partial \ln(k_i)} \right) E_i$$

(P-increasing) (P-decreasing)

- Analogous homeostasis conditions may be formulated for other physico-chemical properties, as for example, pH or salinity:

salinity: $k_i = k_{oi} \exp(\pm \sqrt{I})$, I = ionic strength
 pH: $k_i = k_{oi} \exp(-\sqrt{I} (\text{pH} - \text{pH}_{i,\text{opt}})^2)$

Euler's summation theorem

$$P = f(k_1, k_2, k_3, \dots, k_i, \dots)$$

$\left(\frac{\partial \ln P}{\partial \ln k_i}\right)$ are called *control coefficients* C_i , *period elasticities*, or *sensitivity coefficients*.

The period function $P = f$ is called homogenous to degree -1 if:

$$f(tk_1, tk_2, tk_3, \dots, tk_i, \dots) = t^{-1}f(k_1, k_2, k_3, \dots, k_i, \dots) = t^{-1}P$$

□

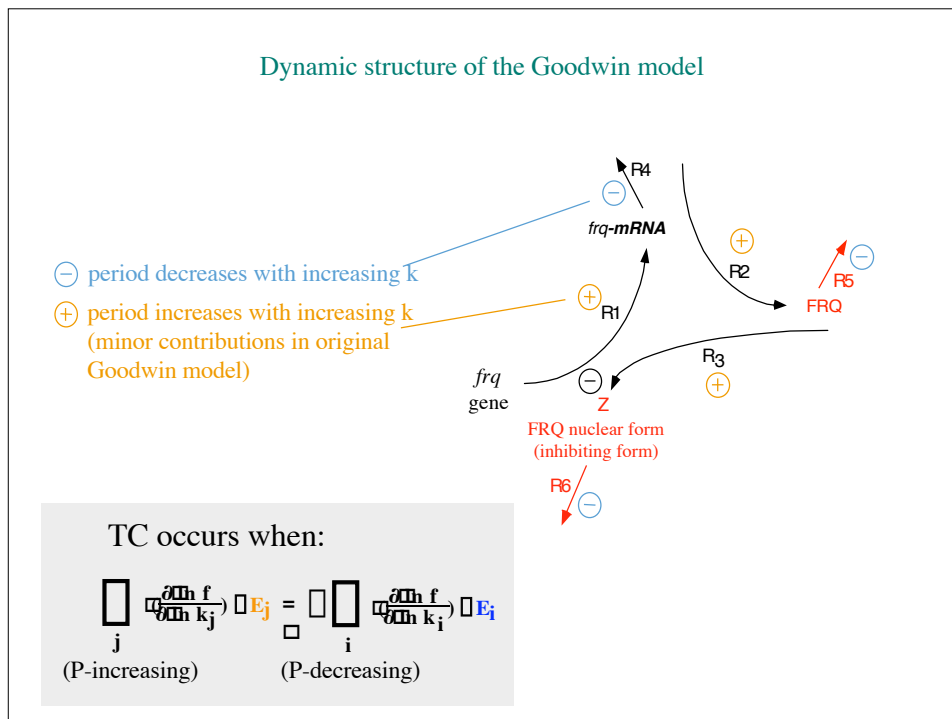
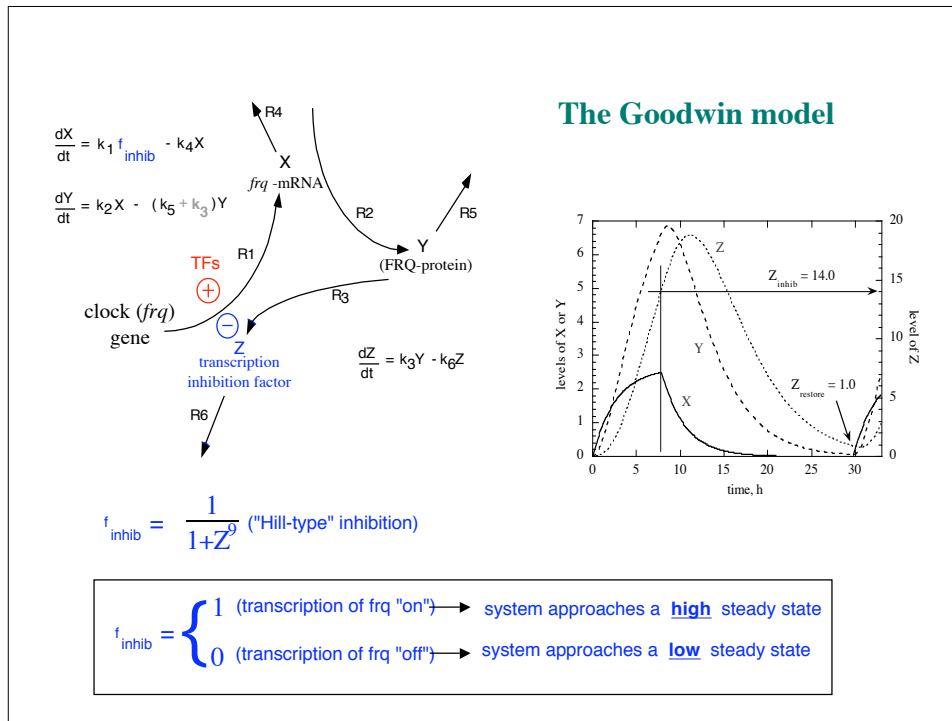
$$\sum_i \left(\frac{\partial \ln P}{\partial \ln k_i}\right) = \square 1$$

A reaction kinetic modelling strategy: working with minimal models

Ockham's razor:

“Plurality is not to be assumed without necessity”

William of Ockham, 1285-1349



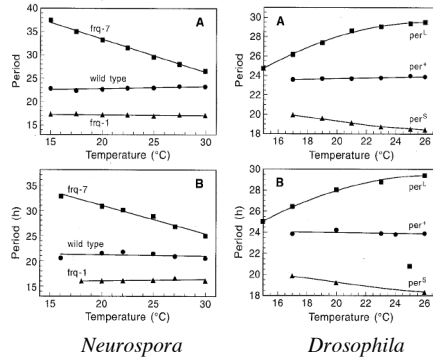
Prediction that clock protein degradation plays an important role for circadian period length and temperature-compensation

Table 1. Effect of loop rate constant values k_1, k_2, k_3 and turnover rate constant values k_4, k_5, k_6 on period length

k_1	Period ^a	k_2	Period ^b	k_3	Period ^c
0.1	23.3	0.1	23.1	0.1	23.1
1.0	23.4	1.0	23.4	1.0	23.4
10.0	23.6	10.0	23.6	10.0	23.6
100.0	23.6	100.0	24.0	100.0	24.0
1000.0	23.9	1000.0	24.6	1000.0	24.6

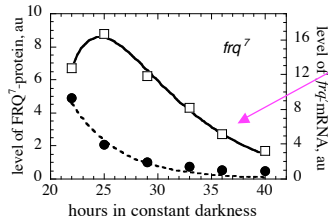
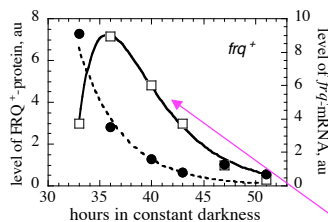
k_4	Period ^d	k_5	Period ^e	k_6	Period ^f
0.05	39.3	0.05	39.0	0.05	29.1
0.15	26.2	0.15	26.2	0.15	21.0
0.3	20.0	0.3	20.0	0.3	16.3
0.45	17.7	0.45	17.7	0.45	13.8
0.6	16.6	0.6	16.4	0.6	12.7

(Ruoff et al., Naturwissenschaften (1996) 83: 514-517)

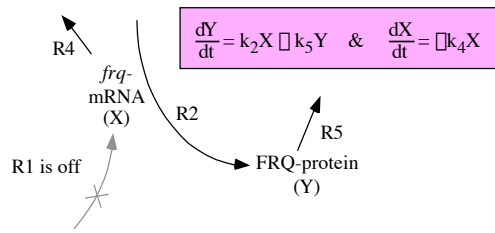


(A: Goodwin model; B: experiment)

Estimating *frq*-mRNA and FRQ-protein half-lives in *frq*[+] and *frq*[7]



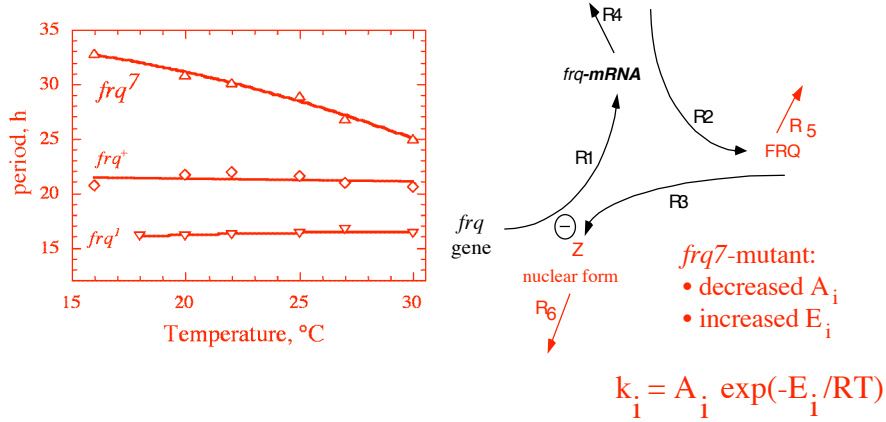
Garceau et al., Cell 89 (1997) 469-476 (exp data)
Ruoff et al, JBR 14 (1999) 469-479 (model)



$$Y(t) = Y_{start} \exp(-k_5 t) + \frac{k_2 X_{start}}{k_5 - k_4} (\exp(-k_4 t) - \exp(-k_5 t))$$

Species	Half-Life, h
<i>frq</i> [+]-mRNA	3.39 ± 0.42
<i>frq</i> [7]-mRNA	2.86 ± 0.31
FRQ[+]	2.9
FRQ[7]	6.2

FRQ-Stability and Temperature Compensation



FRQ^{S513I}

>spI19970|FRQ_NEUCR FREQUENCY CLOCK PROTEIN - *Neurospora crassa*.

```

MADSGDKSQGMRPPPFDSRGHPLPRRASPKSITLENHRLARDTSSRVTS
SSALGVTESQPQLKSSPTRRNSGSESEPTNWFNQSNRNPAAAFHDESHIM
EVDPPFYQKETDSSNEESRYPPGRNPVHPPGGVQLPGFRPVAHSTAADD
YRSVIDDLTVENKRLKEELKRYKQFGSDVMRKEKLFKIVHGLPRRKKRE
LEATLRHFAASLGDSSESTSQRRKTGRHGTAVYSSGVLSKHDSSSSRS
RPVDSAYNSMSTGRSSHAPHSSGSPSLGRPSLTRAHSVGTQKVENYLDRTP
DGLLPHHIVMTDKEKKLVVRRLEQLFTGKISGRNMQRNQSMPSMDAPLA
PEGTNMAPP RPPPEGLREACIQLDGDNPRKNRSSKDNGSASNSGGDQTE
LGGTGTGSGDGS GSGGRTGNNTSPPGAIAPDORPTRPRDLDPDRVQIPSE
NMDYIRHLGIVGTFFLQGSRTSYQDVAPDAEGWVYLNLLCNLAQLHMNV
TPSFIRQAVS PK Q T F Q L S A D G R K I R W R G T D G T K F S S S S S D K S Q S P M
T E D T E D G S D K N S P K K R K T Q Q A S E I G R F G P S R S P S D T F H Y K M F V H R N
SSIETSLEESMSQGSSEDAVDES NMGNSKWDFSGSGTTQRRKRKYDGAIV
YYTGAPFCTDLSGDPGDMSPTAQMTAGREVEGSGSDEVEHVLQRTLSGS
SLPIRPLSDDRARVAEVLDFDPGNPELVADDGSSPNDEDFVFPWCEDPA
KVRIQPIAEKVMPSGLGGVLPDDHFVMLVTRRVVRPILQRLSRSTTS
EDTAEFIAERLAAIRTSSPLPPRSHRLTVAPLQVEYVSGQFR L N P A P L P
P P A I F Y P P F T D S S W D D L A S D D E V E V E E D S Y S E G I S R R A N P H F S
DNNTYMRKDDLAFDTE TDV R M S D D N R L S D S G H N M R A M M P R A E A V D G D D S
PLAAVTGKEVDIVHTGSSVATAGGAESGYSSMEDVSSS
    
```

frq[7]:G[] D frq[1]:G[] S

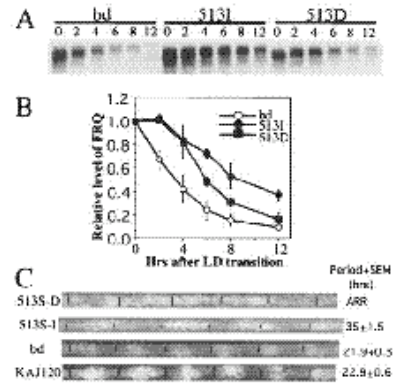
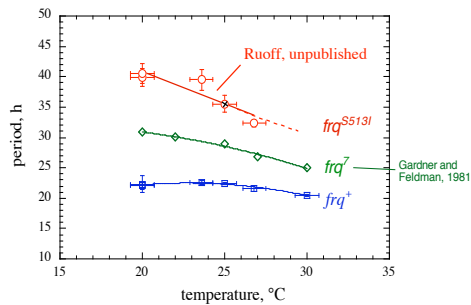
PEST-1 PEST-1

PEST-2 PEST-2

PEST-2

Importance of FRQ-degradation on circadian period length

Replacement of serine-513 by isoleucine (I) blocks phosphorylation at this site and results in a significantly slower FRQ-degradation and in a long circadian period length ≈ 35 h as well as loss in temperature-compensation.



(Liu *et al.*, PNAS (2000) 97:234-239)

FRQ PEST-1

>spI19970FRQ_NEUCR FREQUENCY CLOCK PROTEIN - *Neurospora crassa*.

```

MADSGDKSQGMRPPPFDSRGHPLRRASPDKSIITLENHRLARDTSSRVTS
SSALGVTESQPLKSSPTRRNSGSEPTNWFNQSNRNPAAAFHDESHIM
EVDPPFYQKETDSSNEESRYPPGRNPVHPPGGVQLPGFRPVAAHSTAADD
YRSVIDDLTVENKRLKEELKRYKQFGSDVMRKEKLFKIKVHGLPRRKKRE
LEATLRHFAASLGDSSSESTSQRRTGRHGTAIVYSSGVLSKHDSSSSRS
RPVDSAYNSMSTGRSSHAPHSSGSLGRPSLTRAHSVGTQKVENYLRDTP
DGLLPHHIVMTDKKKLKVRRLEQLFTGKISGRNMQRNQSMPSMDAPLA
PEGTNMAPPRPPEGLREACIQLDGDNPRKNRSSKDNGSASNSGGDQTE
LGGTGTGSGDGS GGRTGNNTSPPGAIAPDQRPTRPRDLDPDRVQIPSE
NMDYIRHLGLVSPFLQGSRTSYQDVAPDAEHWVYLNLLCNLAQLHMVNV
TDSIIRQAVSEKTKFQLSADGRKIRWRGGTDGTFSSDSSDKSQOSPM
TEDTEDGSDKNRRKKRKTQOASSEIGRFGPSRSPSDTFHYKPMVHRNS
SSTETSLSDMSQGSSEDAVDESNMGNSKWDFSGSGTTQORRKRKYDGAIV
YYTGAPFCTDLSDPGDMSPTAQM TAGREVEGSGGDEVEHVLQRTLSPG
SLP IRLSDDRARVAEVLDFDPGNPELVADDGSSPNDEFVFPWCEDPA
KVR IQP IAKEVMEPSGLGGVLPDDHFVMLVTRRVVRLQRLSRSTTS
EDTAEFIAERLAAIRTSSPLPPSRHRLTVAPLQVEYVSGQFRRLNPAPLP
PPAIFYPPFSTDSSWDDGDDLASDDEEVEEVEEEDSYSEGQISRANPHFS
DNNTYMRKDDLAFDTETDVRMDSDDNRLSDSGHNMRAMPRAEAVDGDD
PLAAVTGKEVDIVHTGSSVATAGGAESGYSSSMEDVSS
    
```

frq[7]:G D frq[1]:G S
 PEST-1 PEST-1
 PEST-1 NLS

PEST-2
 PEST-2

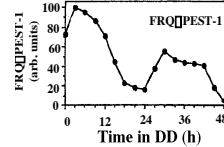
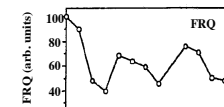
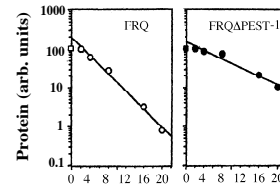
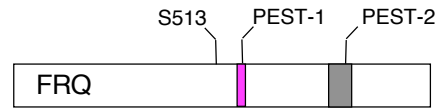
Importance of FRQ-degradation on circadian period length

(Görl *et al.*, EMBO J. (2001) 20:7074-7084)

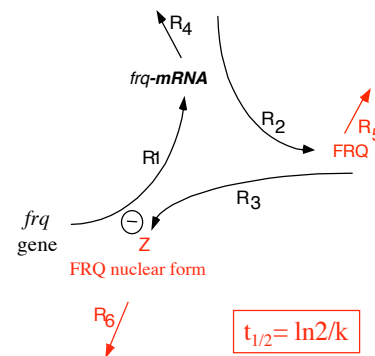
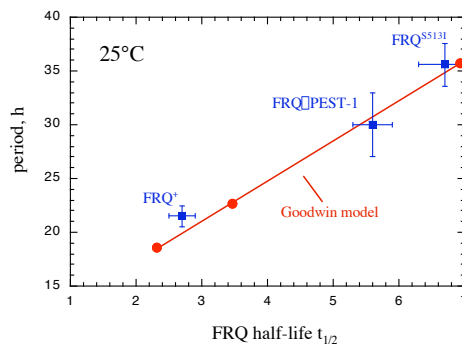
Deletion of PEST-1 leads to an increased FRQ-protein stability.

Both *frq* PEST-1 RNA and FRQ PEST-1 protein are rhythmically expressed with an increased period length of 28-30 h.

However, *frq* PEST-1 has lost the conidiation rhythm.



Relationship between period length and FRQ (first-order) half-life $t_{1/2}$

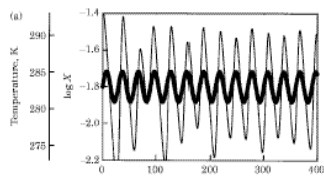


modelling entrainment by temperature
cycles

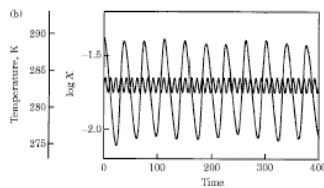
&

temperature-pulse phase response curves

entrainment by temperature cycles in the Goodwin model

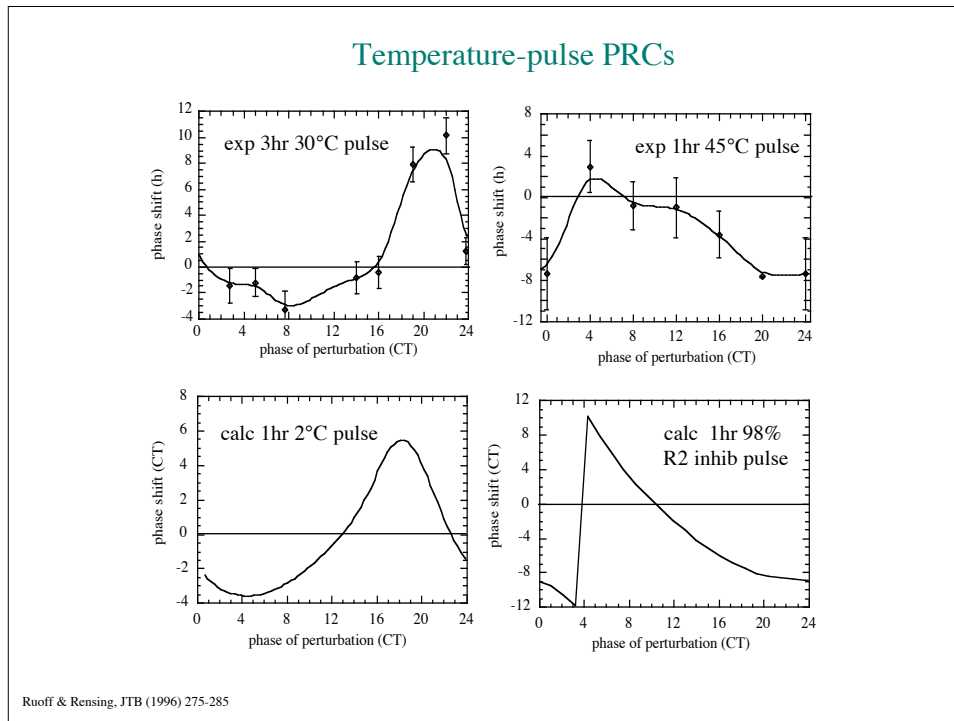


(a) Temperature entrainment using a sinusoidal temperature perturbation with period =30 and a temperature amplitude of 1°C. The inner oscillation is the entraining temperature cycle while the outer oscillation is the responding log X value.

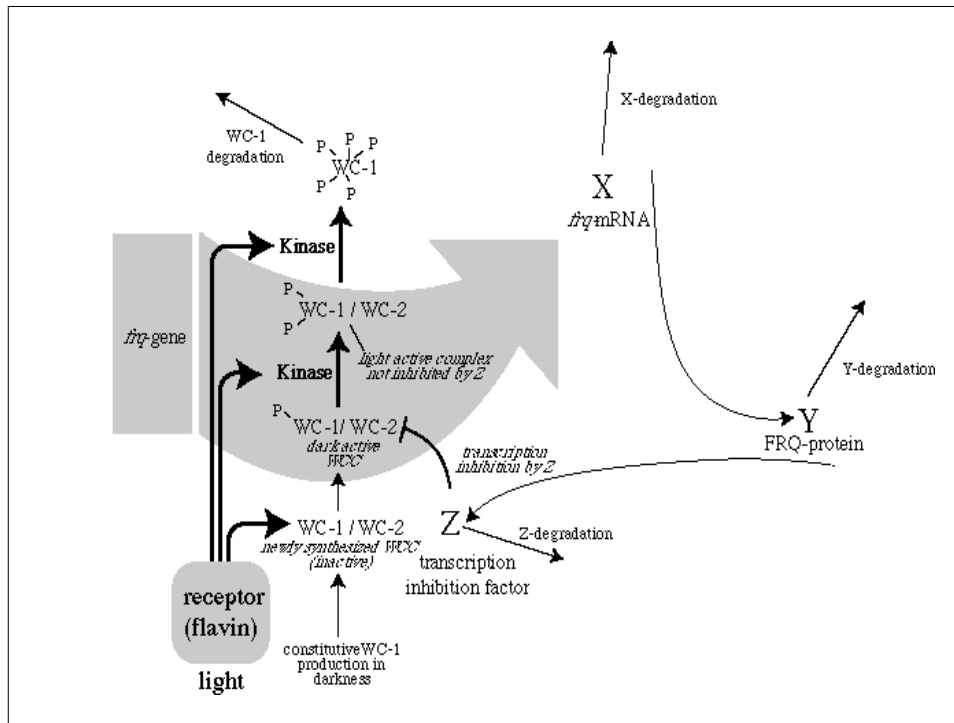


(b) When the period is decreased to 10 the oscillator is no longer able to follow the perturbing rhythm and resets to its original unperturbed period of 38.

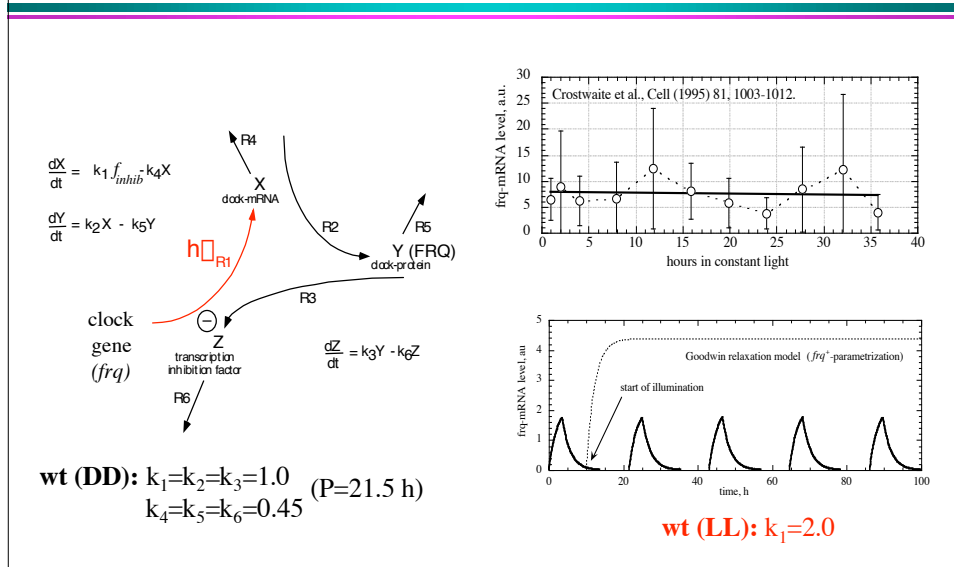
Ruoff & Rensing, JTB (1996) 275-285



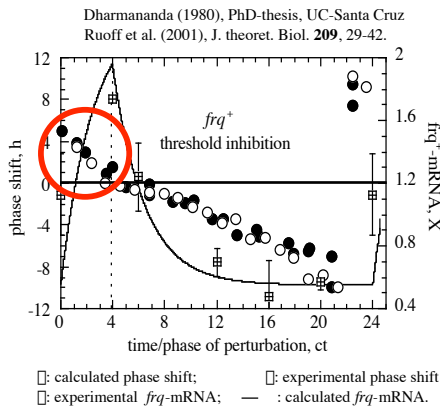
Influence of light on the *Neurospora*
clock:
phase response curves



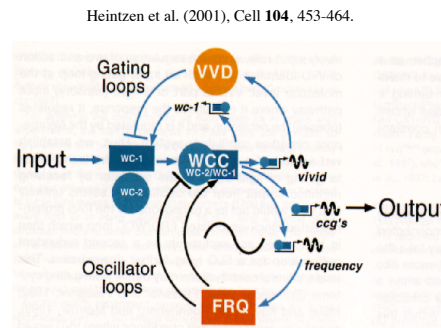
Constant Illumination (Threshold Model)



Model prediction: transcription inhibition directly after light pulse to ensure phase advances



Additional inhibition of *frq*-transcription necessary directly after light pulse to get phase advances for phase of perturbations between ct0-ct4.

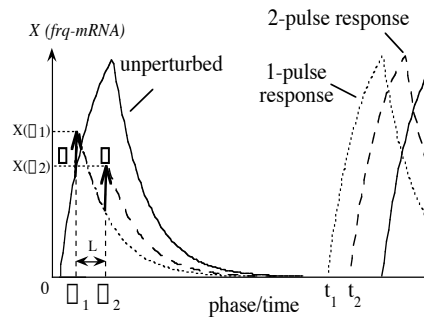


The PAS-protein VIVID (VVD) has been found to repress light input and its function appears to be related to our theoretical findings.

Phase Resetting with Two Light Pulses

Two light pulses with interval L are applied.

$\Delta \phi_{12}$ is the additional phase shift due to the second light pulse

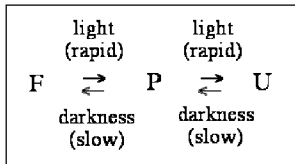


$$\Delta \phi_{12} = \phi L + \frac{1}{k_4} \ln \left(\frac{X(\phi_1)}{X(\phi_2) \exp(\phi k_4 L)} \right)$$

Ruoff et al. (2001), JTB, **209**, 29-42.

2-Pulse Light PRC: Experimental Results and Interpretation

Kinetics of light-signal in 2-pulse light perturbations:



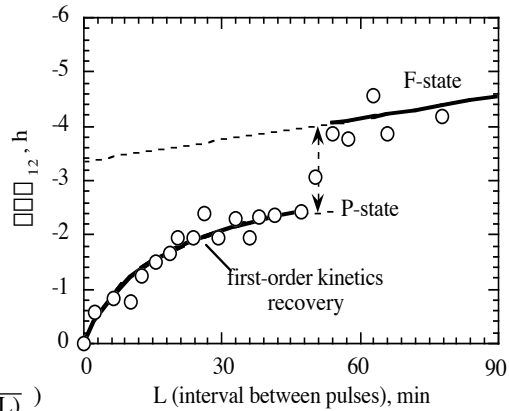
$$P = P_0(1 - \exp(-t/\tau_p))$$

$$X = X_0 + \Delta X(1 - \exp(-t/\tau_p))$$

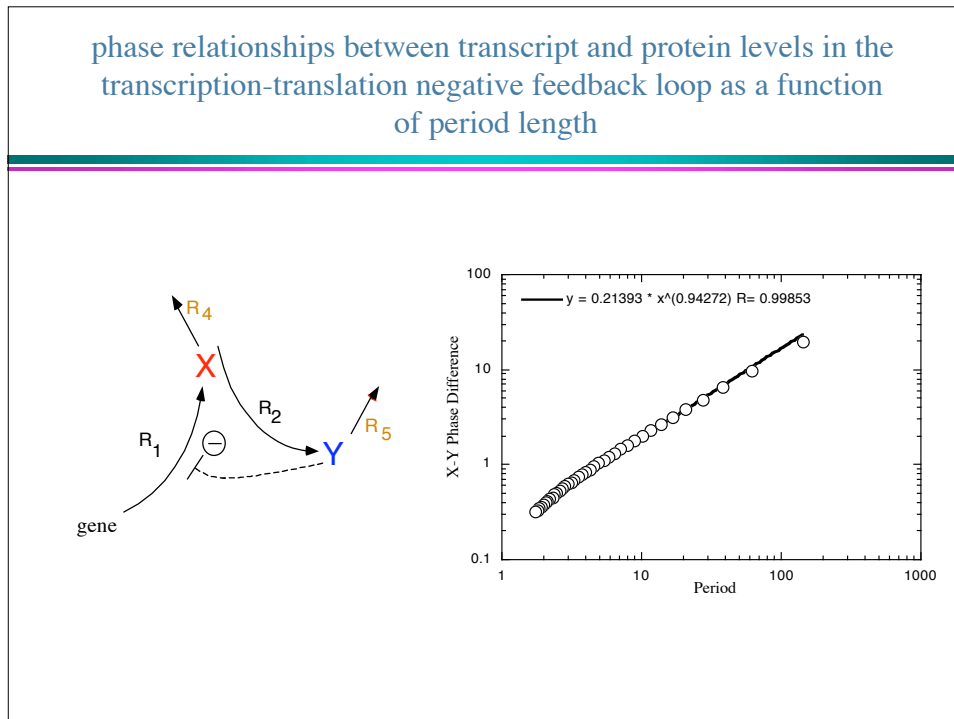
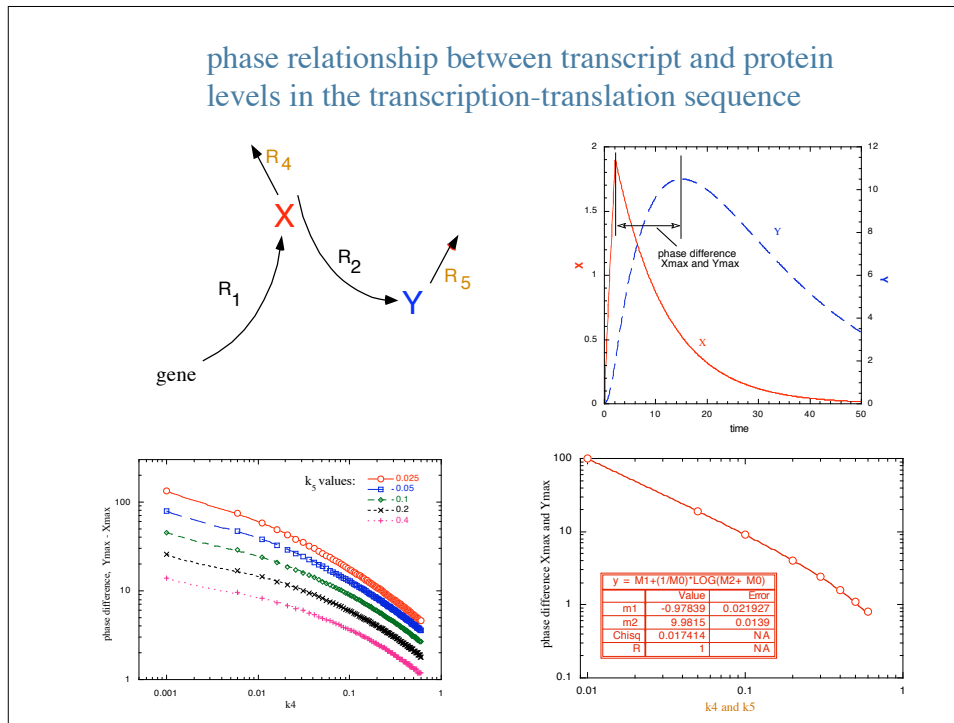
$$= X_0(1 - \exp(-t/\tau_p))$$

$$\Delta X_{12} = \Delta X + \frac{1}{k_4} \ln \left(\frac{X(\Delta t)}{X_0(1 - \exp(-k_4 L))} \right)$$

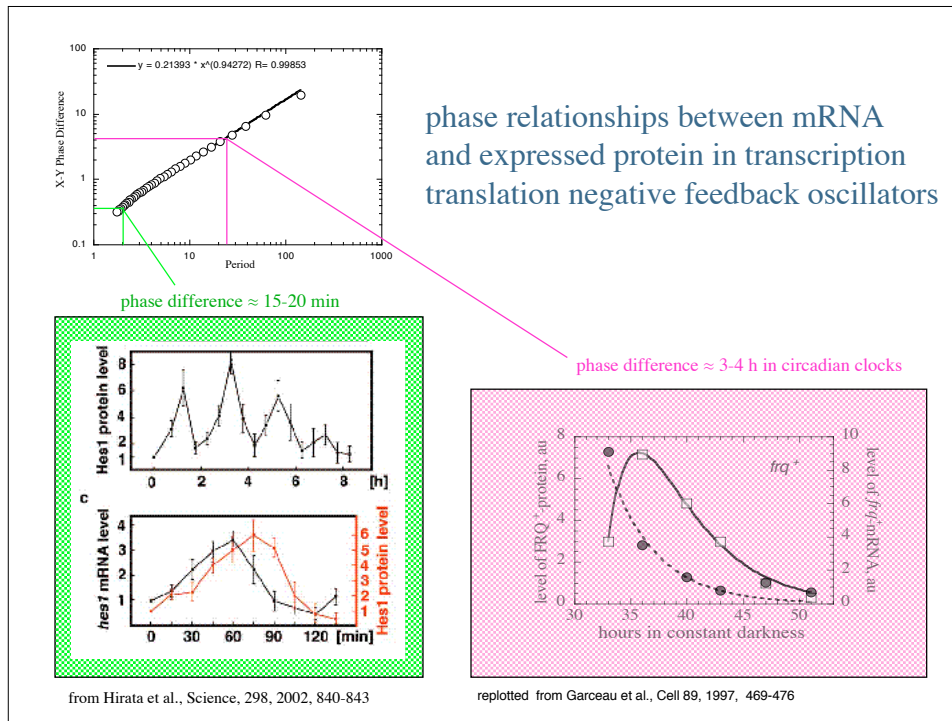
Exp. Result: Dharmananda, S. (1980) PhD thesis.



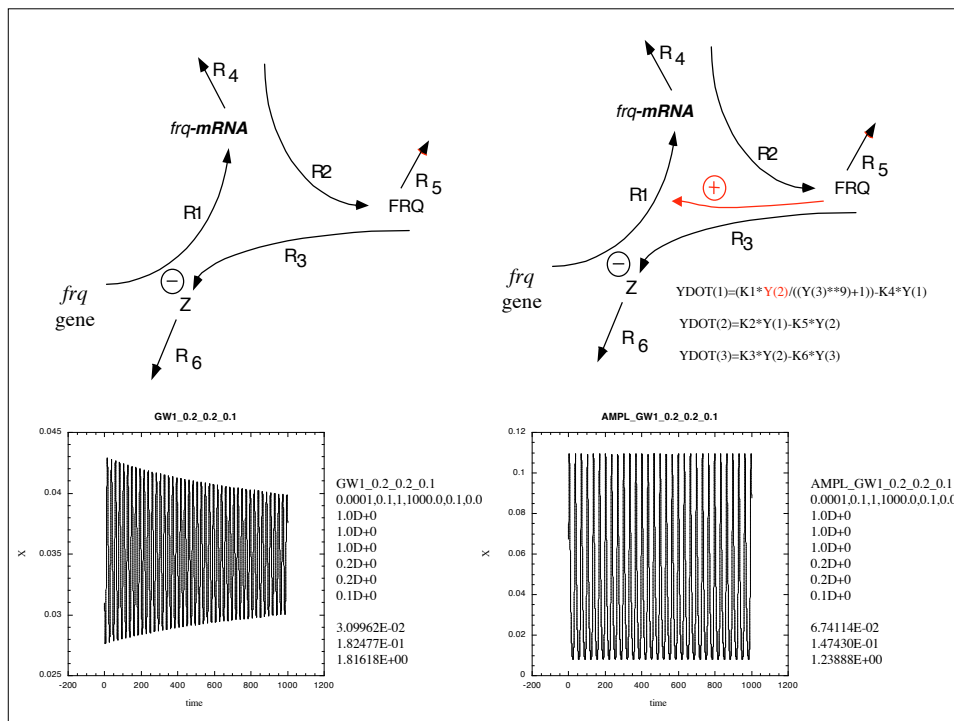
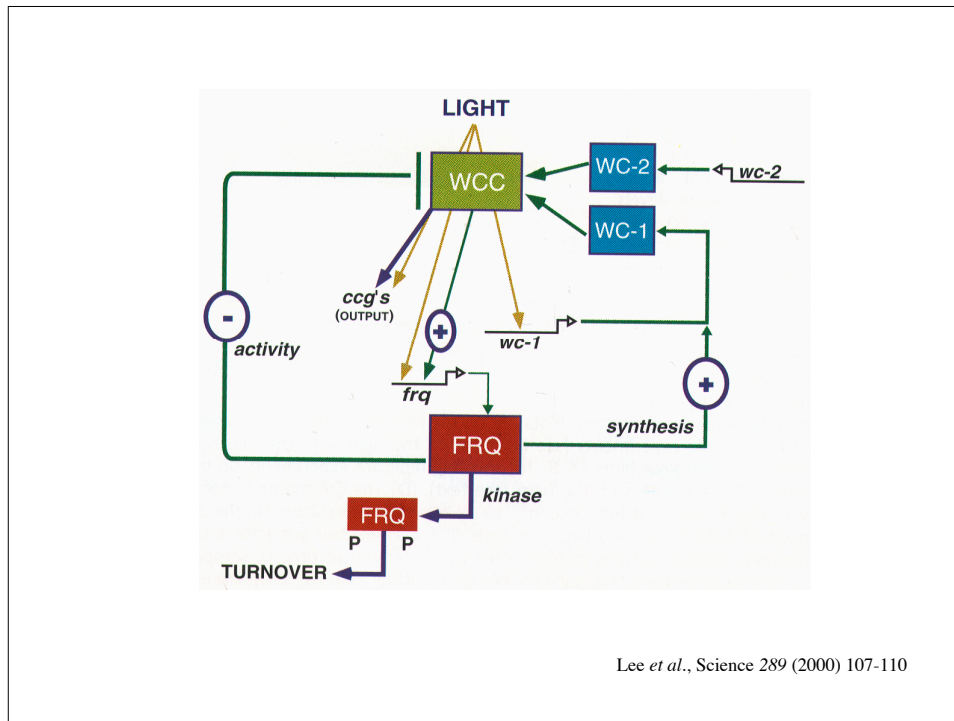
phase relationship between transcript and protein levels

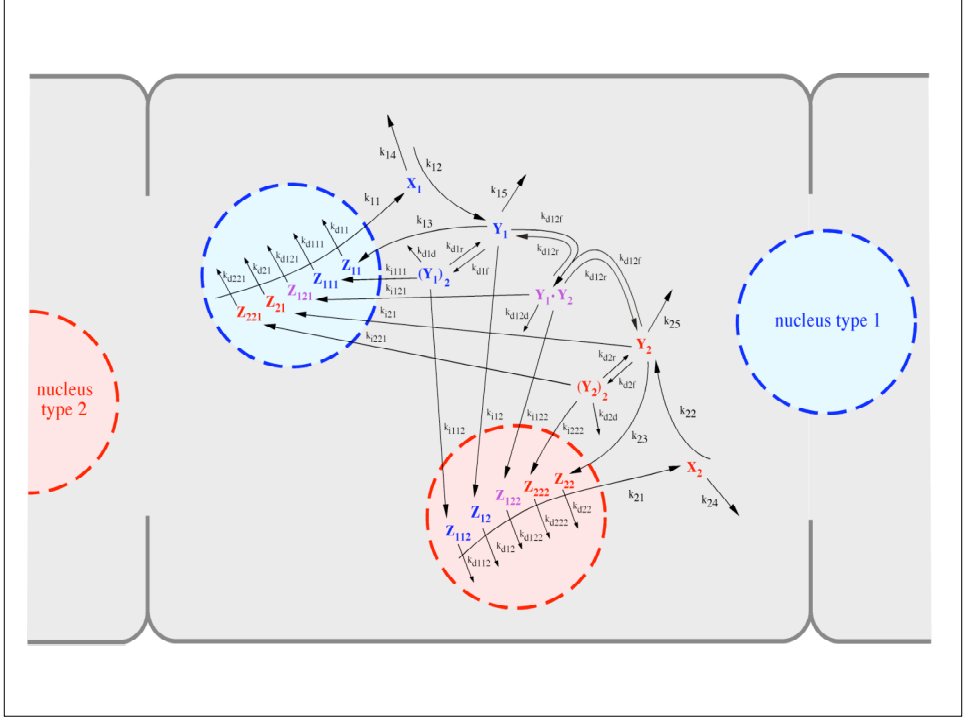
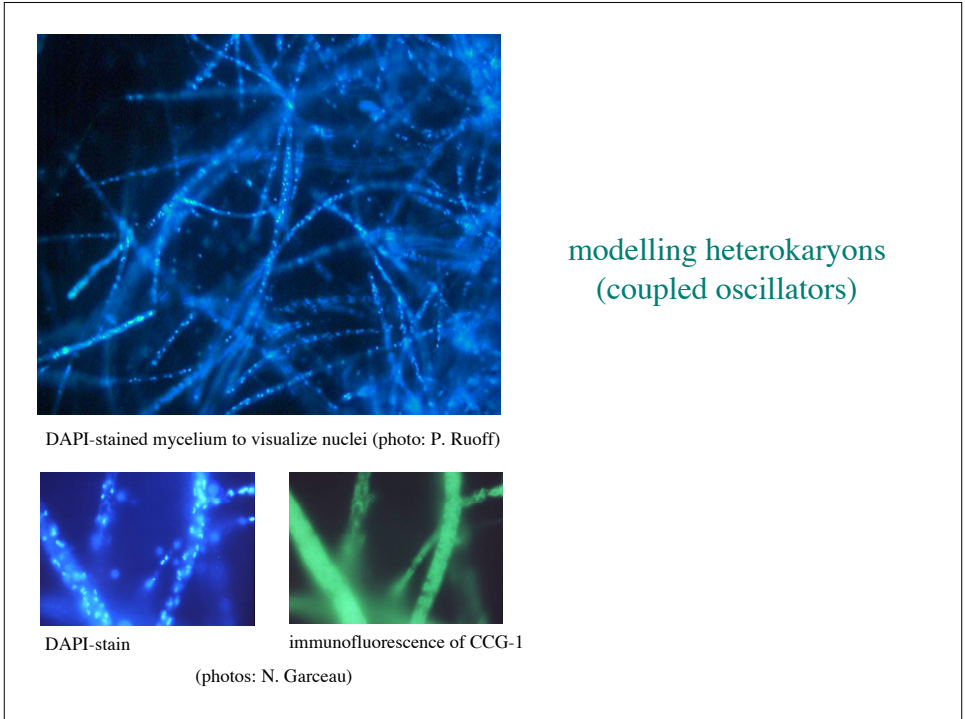


A Theory for Temperature-Compensation and Period Homeostasis in Reaction Kinetic Models of Biological Oscillators



stabilizing role of positive feedback
(autocatalytic loop)





A Theory for Temperature-Compensation and Period Homeostasis in Reaction Kinetic Models of Biological Oscillators

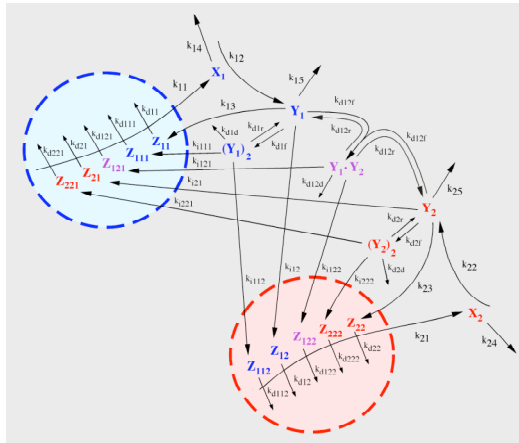
$k_{11}=k_{12}=k_{13}=k_{21}=k_{22}=k_{23}=k_{112}=k_{i21}=1.0$ (transcription, translation, nuclear import)

$k_{14}=k_{24}=0.28$ (mRNA degradation)

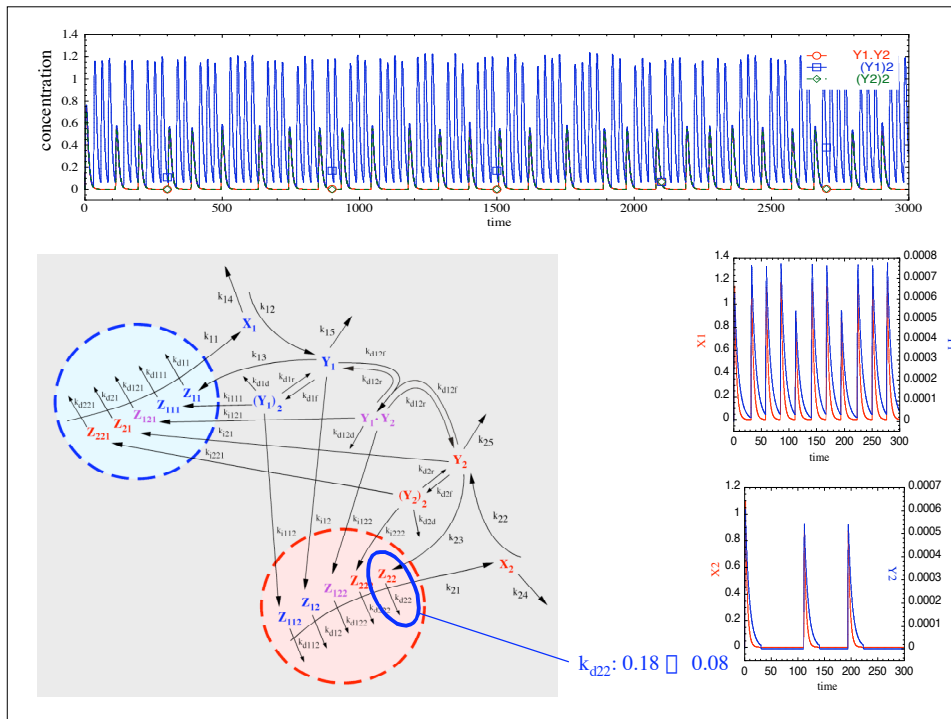
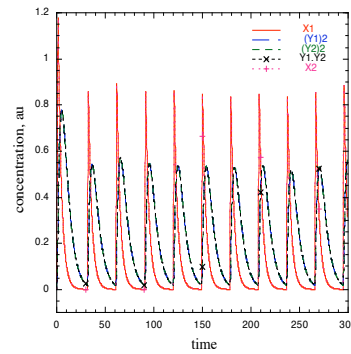
$k_{d11}=k_{d22}=k_{15}=k_{25}=k_{d1d}=k_{d2d}=0.18$ (FRQ, (FRQ)₂ degradation)

$k_{d1f}=k_{d2f}=k_{d12f}=1 \times 10^6$; $k_{d1r}=k_{d2r}=k_{d12r}=0.01$ (reversible (FRQ)₂ formation)

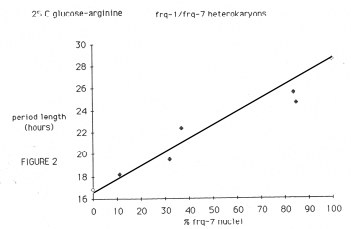
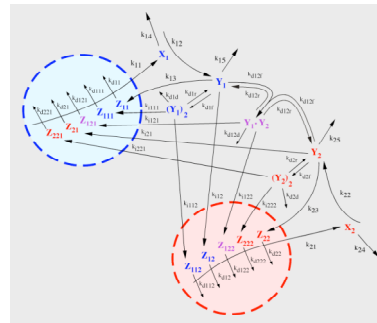
remaining rate constants = 0 (no FRQ-dimer import into nucleus)



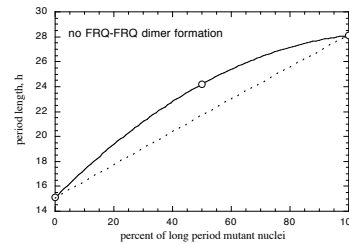
homokaryon example



heterokaryon periods



from J. J. Loros: "Studies on FRQ-9, a Recessive Circadian Clock Mutant of *Neurospora crassa*" University of California Santa Cruz, 1984.



calculated from above model with no FRQ dimer formation

Summary

- Importance of *Ockham's razor* to remove redundancy in models, i.e., create minimal models.
- Minimal models can be further developed in accordance with experimental findings.
- More concerted interactions between experimentalist and theoreticians/modellers are needed in order to do more systematic experimental studies to get data for building models.



Baker Library, Dartmouth College



Lysefjord, near Stavanger

Thanks to

- Prof. Jay Dunlap for support during my stay at the Genetics Department, Dartmouth Medical School.
- Stavanger University College for giving me the possibility for a sabbatical leave.

Appendix

Euler's summation theorem

$$P = f(k_1, k_2, k_3, \dots, k_i, \dots)$$

$\left(\frac{\partial \ln P}{\partial \ln k_i}\right)$ are called *control coefficients* C_i , *period elasticities*, or *sensitivity coefficients*.

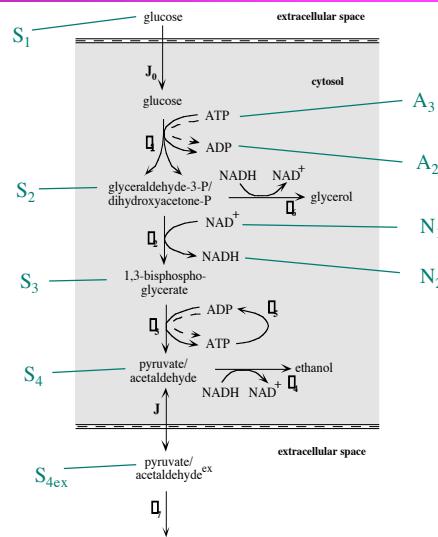
The period function $P = f$ is called homogenous to degree -1 if:

$$f(tk_1, tk_2, tk_3, \dots, tk_i, \dots) = t^{-1}f(k_1, k_2, k_3, \dots, k_i, \dots) = t^{-1}P$$

□

$$\sum_i \left(\frac{\partial \ln P}{\partial \ln k_i}\right) = -1$$

The temperature-compensated glycolytic oscillator: testing Euler's summation theorem and antagonistic balance condition for temperature-compensation



$$\frac{dS_1}{dt} = J_0 - v_1$$

$$v_1 = k_1 S_1 A_3 \left[\frac{1}{K_1} + \left(\frac{A_3}{K_1}\right)^2 \right]^{-1}$$

$$\frac{dS_2}{dt} = 2v_1 - v_2 - v_6$$

$$v_2 = k_2 S_2 N_1$$

$$\frac{dS_3}{dt} = v_2 - v_3$$

$$v_3 = k_3 S_3 A_2$$

$$\frac{dS_4}{dt} = v_3 - v_4 - J$$

$$v_4 = k_4 S_4 N_2$$

$$\frac{dN_2}{dt} = v_2 - v_4 - v_6$$

$$v_5 = k_5 A_3$$

$$\frac{dA_3}{dt} = -2v_1 + 2v_3 - v_5$$

$$v_6 = k_6 S_2 N_2$$

$$\frac{dS_4^{ex}}{dt} = J - v_7$$

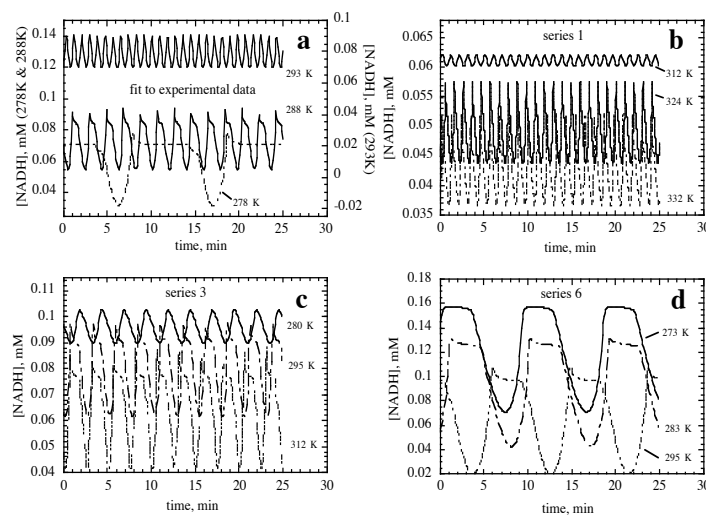
$$v_7 = k_7 S_4^{ex}$$

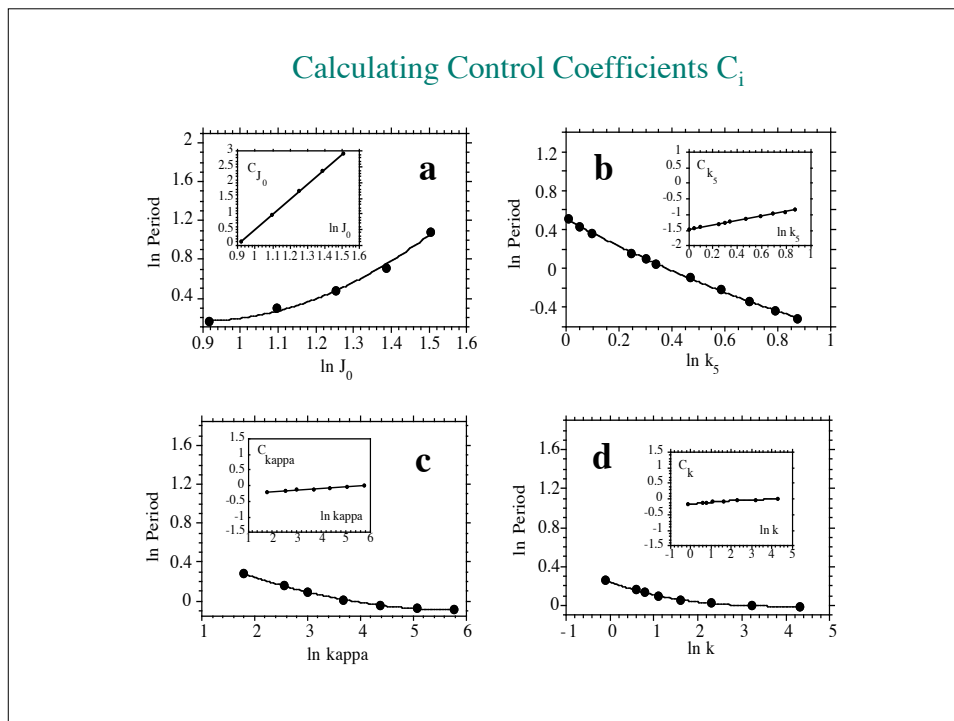
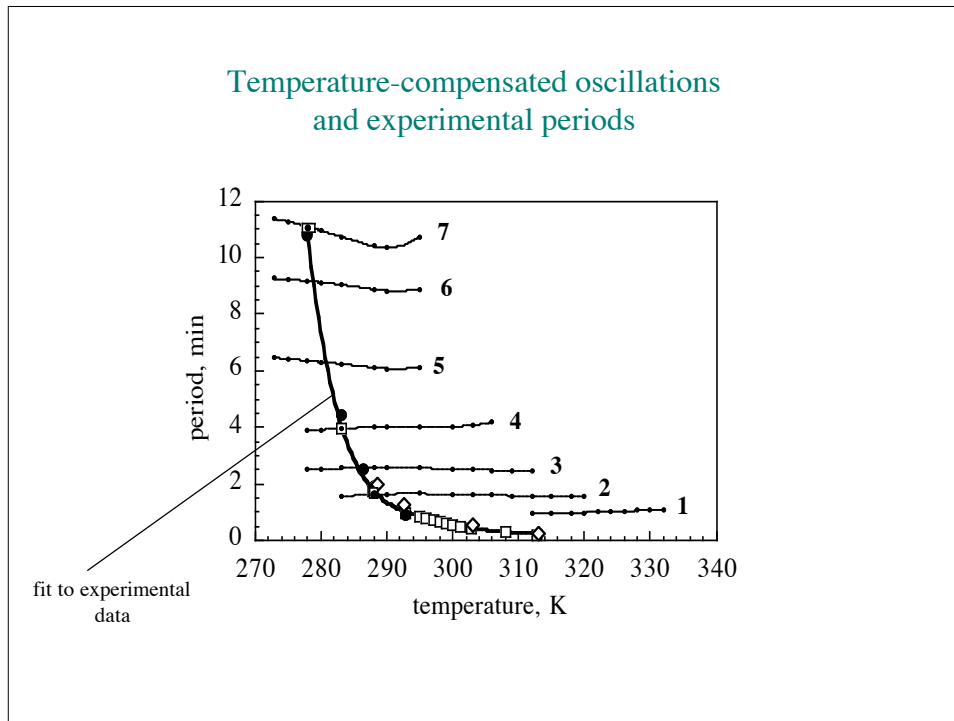
J. Wolf and R. Heinrich, Biochem. J. 345 (2000) 321-334.

Parameter Values of the Reference State ^a		Initial Concentrations for Reference State Oscillations	
Parameter	Value	Compound	Concentration
J_0	2.5 mMmin ⁻¹	glucose	1.187 mM
k_1	100.0 mM ⁻¹ min ⁻¹	glyceraldehyde-3-P/ dihydroxyacetone-P	0.193 mM
k_2	6.0 mM ⁻¹ min ⁻¹	1,3-bisphosphoglycerate	0.050 mM
k_3	16.0 mM ⁻¹ min ⁻¹	pyruvate/acetalddehyde	0.115 mM
k_4	100.0 mM ⁻¹ min ⁻¹	external pyruvate/acetalddehyde	0.077 mM
k_5	1.28 min ⁻¹	ADP	1.525 mM
k_6	12.0 mM ⁻¹ min ⁻¹	ATP	2.475 mM
k	1.8 min ⁻¹	NAD ⁺	0.923 mM
LJ	13.0 min ⁻¹	NADH	0.077 mM
q	4.0		
K_1	0.52 mM		
N	1.0 mM		
A	4.0 mM		
\bar{I}	0.1		

^a resulting in a period length of 1.17 min.

Temperature-compensated and non-temperature-compensated oscillations in the Wolf-Heinrich model





Control coefficients C_i for the oscillation period in the reference state

Testing Euler's summation theorem

$$\frac{dS_1}{dt} = J_0 \square v_1$$

$$\frac{dS_2}{dt} = 2v_1 \square v_2 \square v_6$$

$$\frac{dS_3}{dt} = v_2 \square v_3$$

$$\frac{dS_4}{dt} = v_3 \square v_4 \square J$$

$$\frac{dN_2}{dt} = v_2 \square v_4 \square v_6$$

$$\frac{dA_3}{dt} = \square 2v_1 + 2v_3 \square v_5$$

$$\frac{dS_4^{ex}}{dt} = \square J \square v_7$$

$$v_1 = k_1 S_1 A_3 \square + \left(\frac{A_3}{K_1}\right)^{\square} \square$$

$$v_2 = k_2 S_2 N_1$$

$$v_3 = k_3 S_3 A_2$$

$$v_4 = k_4 S_4 N_2$$

$$v_5 = k_5 A_3$$

$$v_6 = k_6 S_2 N_2$$

$$v_7 = k_7 S_4^{ex}$$

control coefficient	value ($\pm 1\%$ variation)	value ($\pm 10\%$ variation)
C_{J_0}	+ 0.69	+ 0.68
C_{k_1}	\square 0.21	\square 0.23
C_{k_2}	+ 0.04	+ 0.06
C_{k_3}	\square 0.01	\square 0.01
C_{k_4}	+ 0.26	+ 0.27
C_{k_5}	\square 1.24	\square 1.27
C_{k_6}	\square 0.26	\square 0.28
C_k	\square 0.13	\square 0.12
C_J	\square 0.17	\square 0.15
C_{K_1}	\square 0.86	\square 0.84
$\square C_i$ (without K_1)	\square 1.02	\square 1.05
$\square C_i$ (with K_1)	\square 1.88	\square 1.89

Testing the antagonistic balance equation

For calculating the balance sums (last row) values for control coefficients have been used corresponding to a 10% change of parameters (c.f. Table 3)

Parameter	E_i , kJ/mol (experimental) ^a	E_i , kJ/mol (series 1)	E_i , kJ/mol (series 3)	E_i , kJ/mol (series 6)	E_i , kJ/mol (series 7)
J_0	16.2	43.9	46.1	41.4	51.0
k_3	44.9	31.7	48.7	30.4	39.2
k_4	58.7	44.2	38.0	48.1	42.4
k_1	13.8	16.0	19.1	14.0	23.3
k_2	60.7	10.4	11.8	10.8	11.9
k_5	41.2	13.3	17.9	15.1	19.8
k_6	31.4	23.7	16.2	24.0	15.7
\square (kappa)	15.9	22.3	19.0	18.7	19.7
k	24.3	30.0	32.2	34.1	31.5
K_1^b	47.0	12.2	11.4	8.6	10.0
$\square C_i E_i$	-79.0	-2.3	-6.1	-1.8	-3.6

^a fit to Hemker et al. data, Fig. 3 (large solid dots)
^b in case of K_1 , E_i is interpreted as $\square H_{K_1}^0$