

How Classical is Nucleation?

With:

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Outline

We use simulations to test

1. Classical Nucleation Theory
2. Ostwald's rule
3. Alexander-McTague rule
4. Turnbull's rule
5. The "nucleation theorem"

...and find a few surprises.



Wilhelm Ostwald

Intermezzo:

Why Ostwald would have disagreed with this talk.

I am a molecular simulator.. But Ostwald did not believe * in the reality of atoms:

"Only energy is real – believing in atoms is like worshipping idols..."

W. Ostwald

* But after Perrin's experiments on colloids, Ostwald was "converted"

Why do I defy Ostwald?

Simulations tell us things
about nucleation that
experiments cannot tell us...

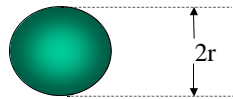
...yet

Homogeneous nucleation

(e.g. Liquid \Rightarrow Solid)

Crystallization requires supercooling

($\mu_{\text{solid}} < \mu_{\text{liquid}}$)



Crystal nucleus

Free - energy gain

$$\Delta G_{\text{Bulk}} = \frac{4\pi}{3} \rho r^3 \Delta\mu_{s,l} < 0$$

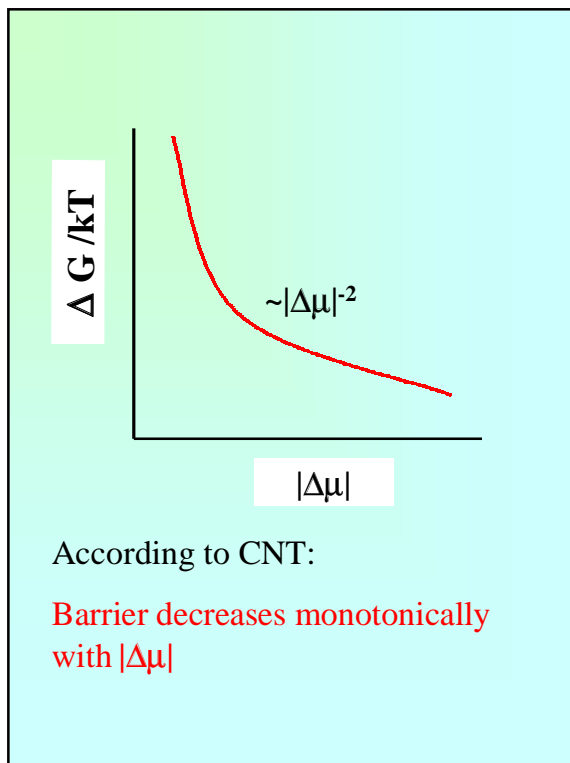
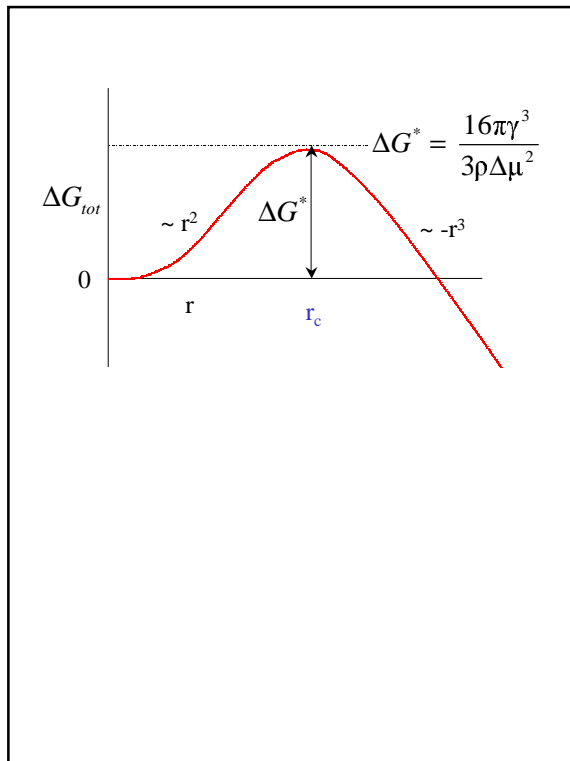
Free - energy loss :

$$\Delta G_{\text{Surface}} = 4\pi r^2 \gamma_{s,l} > 0$$

r^3

r^2

How "Classical" is Nucleation? A Computer-Simulation Study



What does this imply for simulation?

Consider "realistic" supercooling
(10%-20%)

Experimental nucleation rates:
 $O(1) \text{ cm}^{-3} \text{ s}^{-1}$

Simulation:
Volume is much smaller (e.g. for one million particles): $V = O(10^{-15}) \text{ cm}^3$

\Rightarrow Nucleation rate is $O(10^{-15}) \text{ s}^{-1} !!$

\Rightarrow One event per 10^{15} s

\Rightarrow One event per $10^{30} \text{ MD time steps}$

BRUTE FORCE WON'T WORK...

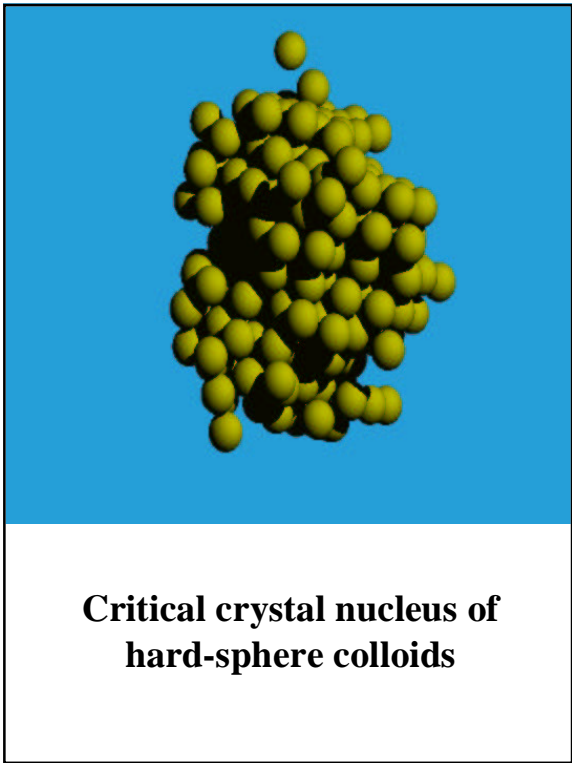
Solution:

1. Compute height of the free-energy barrier ΔG^* (MC/MD)
2. Compute transmission coefficient Γ (MD)

Rate = $\Gamma \exp(-\beta\Delta G^*)$


Probability of "critical" fluctuation
(strong function of T)

Kinetic Prefactor
(usually weak function of T)



Ostwald's Rule(1897)

“The phase that nucleates need not be the **stable phase**, but the one that is **closest in free energy** to the parent phase...”



Stranski & Totomanov (1930's)

“The phase that nucleates is the one with the **lowest nucleation barrier..**”

Alexander & McTague (80)

“On basis of Landau theory, one would expect the following crystal phases to form easily from the melt:

1. **Hexagonal** (2D crystal)
2. **Icosahedral** (...)
3. **BCC crystal**

Examples:

- 1. Condensation of polar liquids**
- 2. "Protein" crystallization**
- 3. Crystallization of Charged colloids**

Experimental observation:

Classical Nucleation Theory works well for non-polar molecules (e.g. CH₄)...

... but not for polar molecules

(e.g. CH₃CN)


Why???

Experiments ...

...cannot probe the critical nucleus

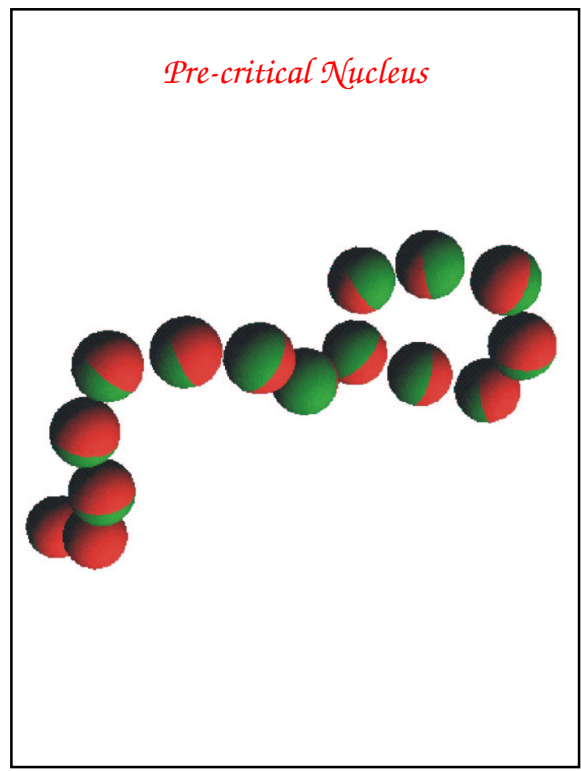
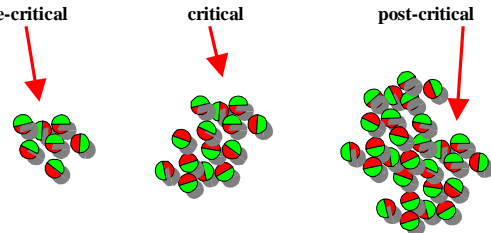
SIMULATION:

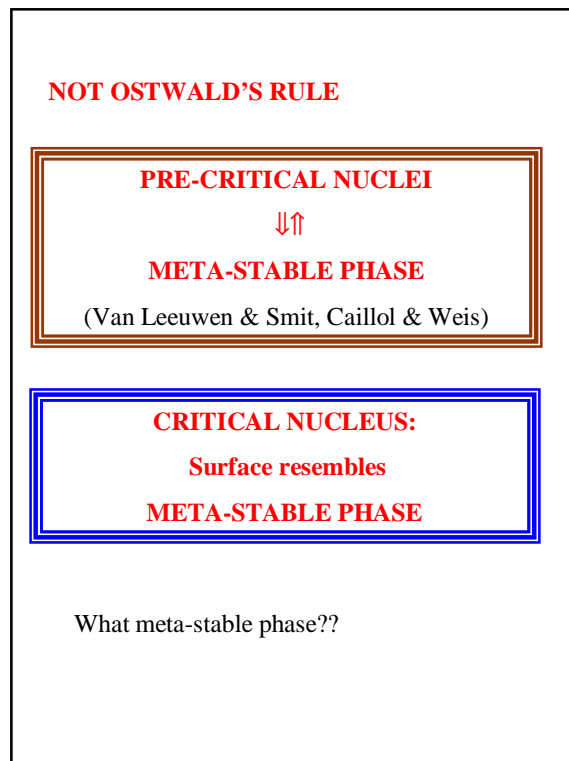
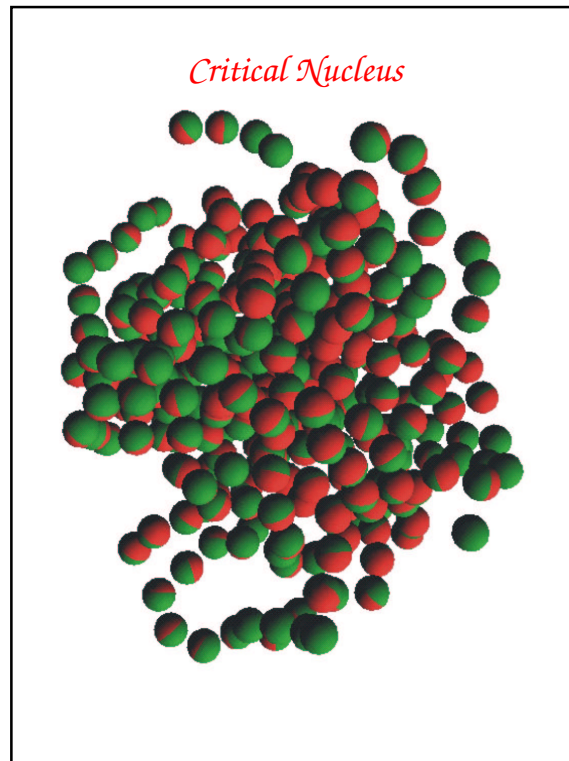
Stockmayer fluid
(= Lennard-Jones + embedded dipole)

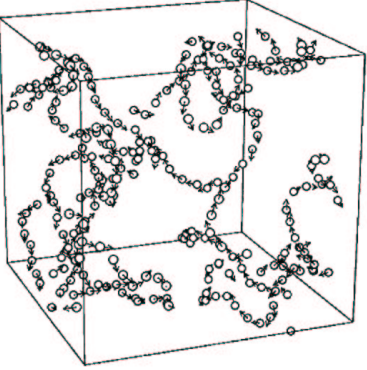


WHAT DO WE EXPECT TO SEE?

Pre-critical critical post-critical







“Gel” phase of dipolar hard spheres
Camp, Shelley, Patey, PRL 88,115(1999)


This is NOT a stable phase of the Stockmayer fluid.

Crystallization of globular proteins

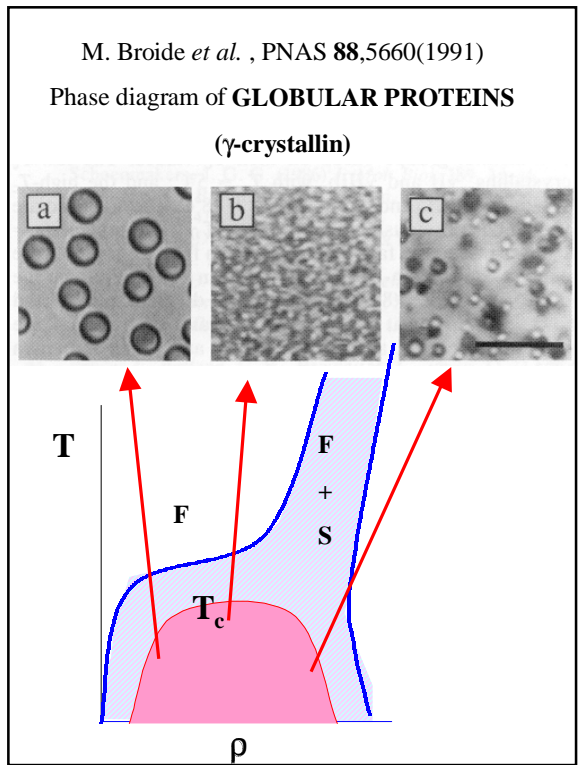
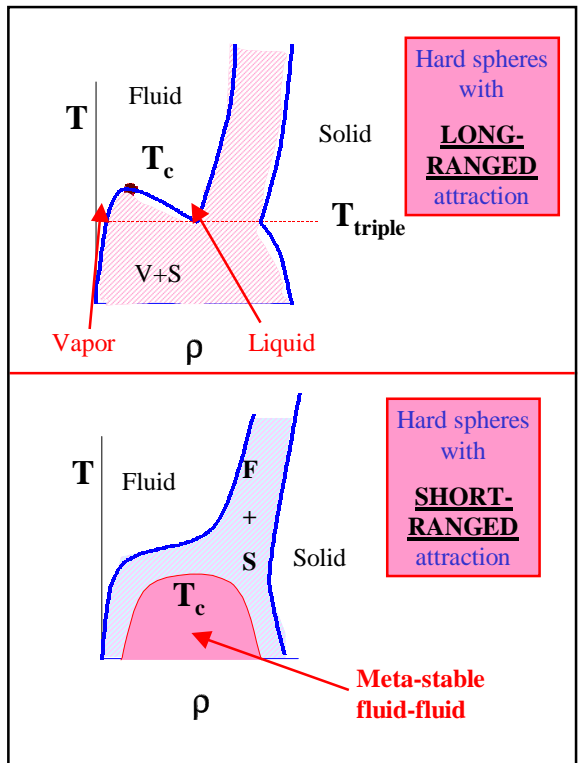
QUOTE:

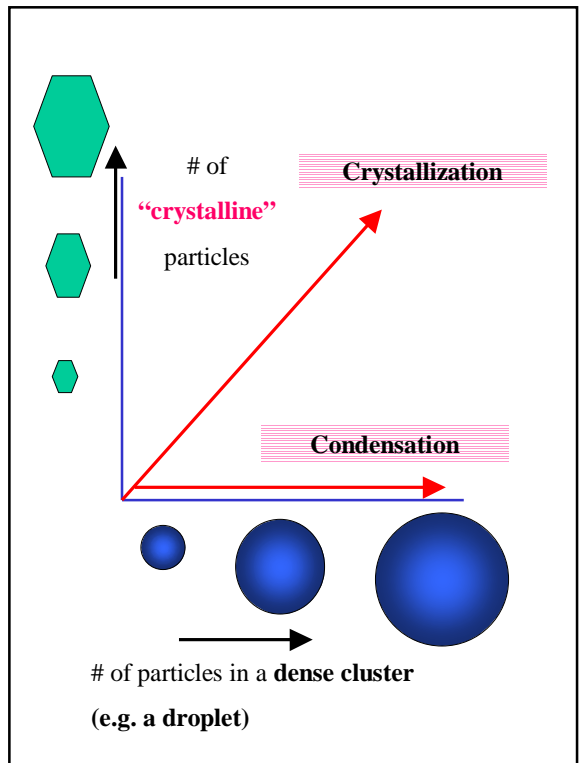
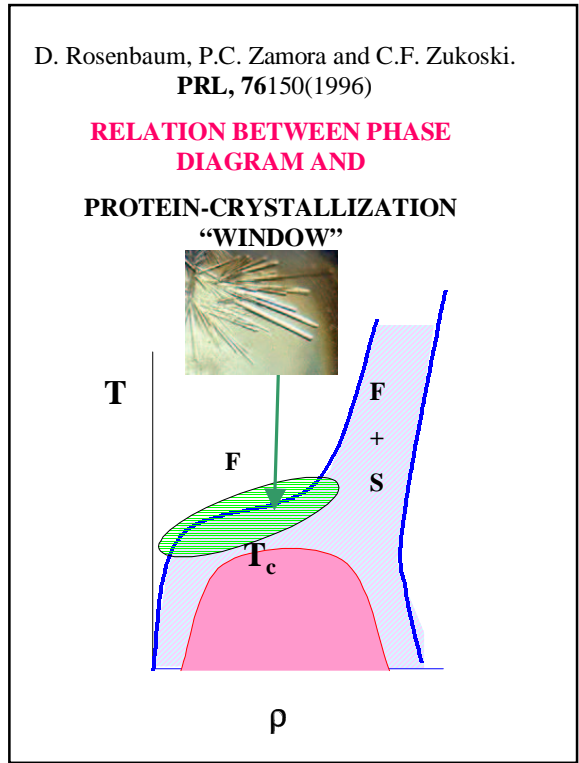
“... mainly trial and error... much like prospecting for gold...”

(McPherson “Preparation and Analysis of Protein Crystals”)

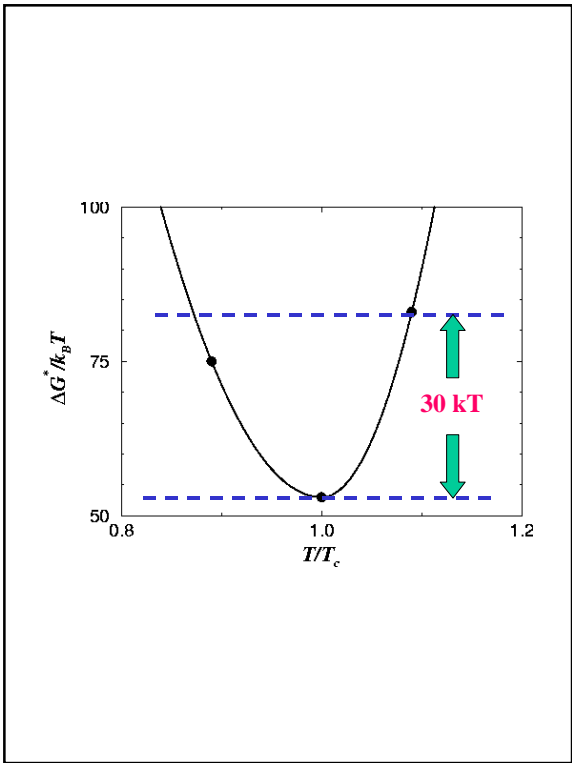
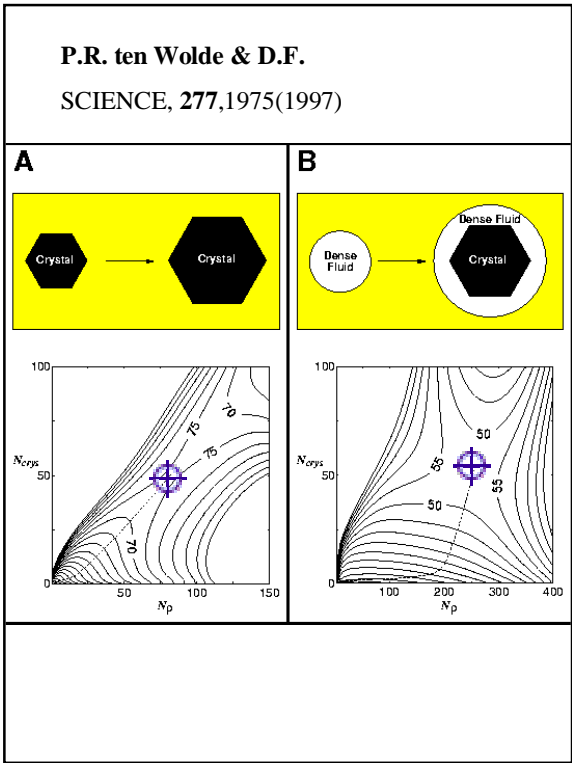


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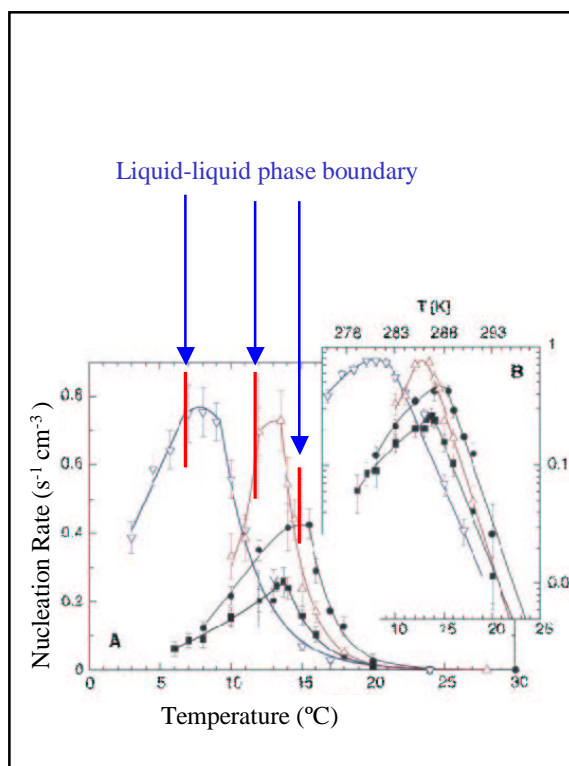


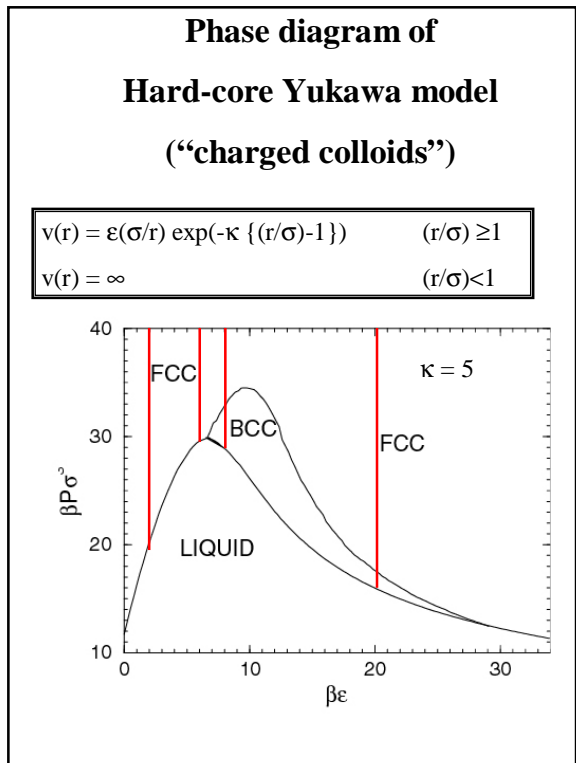
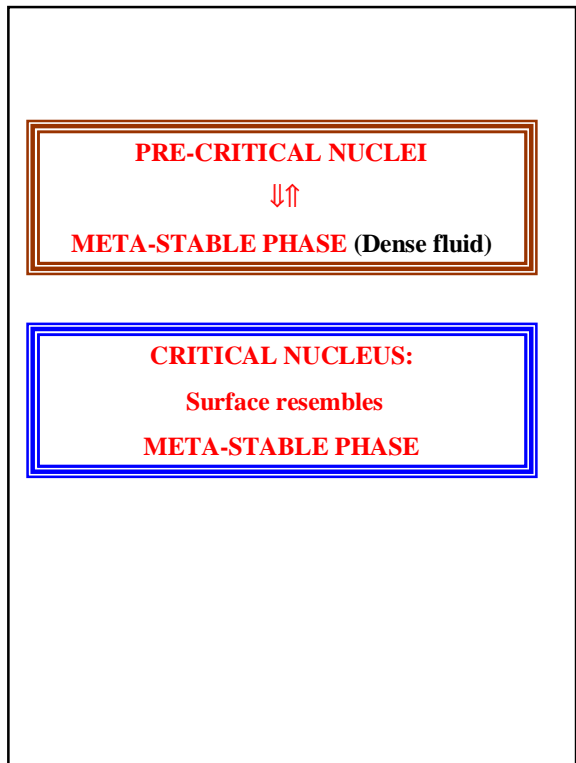
Experimental Evidence??

Recent experiments by Galkin and Vekilov

PNAS 97, 6277–6281(2000)

Lyzosyme crystallization





Ostwald

vs

Alexander-McTague

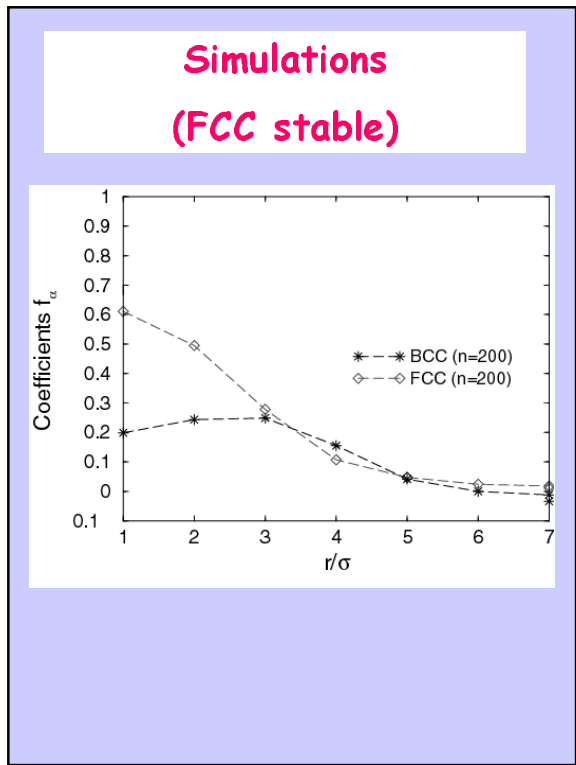
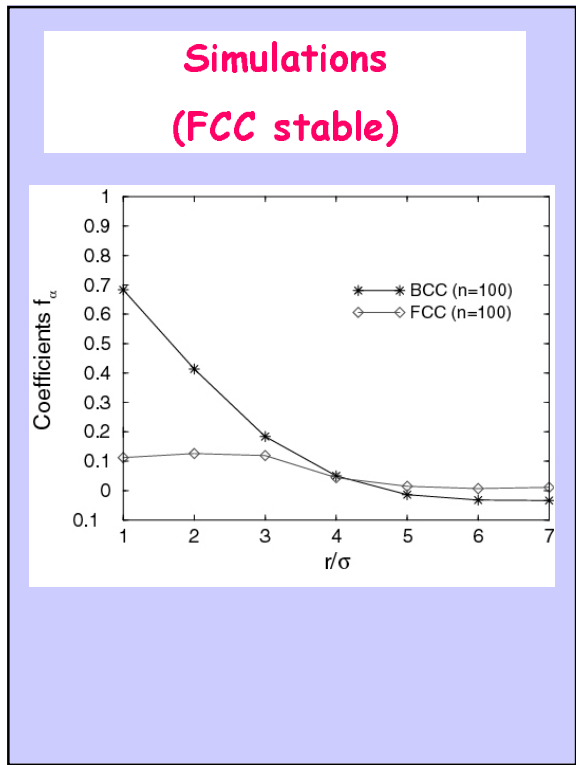
Ostwald:
When FCC is stable, nucleus should be BCC
When BCC is stable, nucleus should be FCC

Alexander-McTague
When FCC is stable, nucleus should be BCC
When BCC is stable,
...nucleus should also be BCC

Simulations

Small nuclei are **ALWAYS** predominantly **BCC**

Large nuclei may be **FCC or BCC**,
depending on which phase is stable

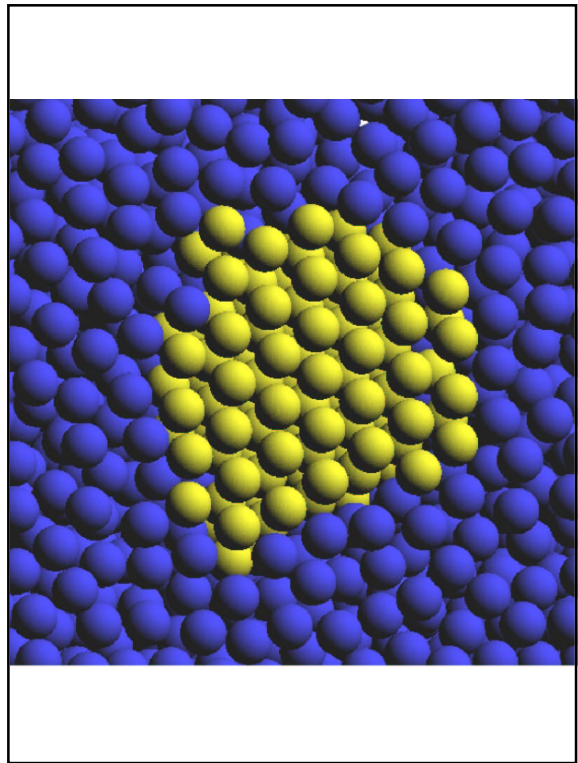


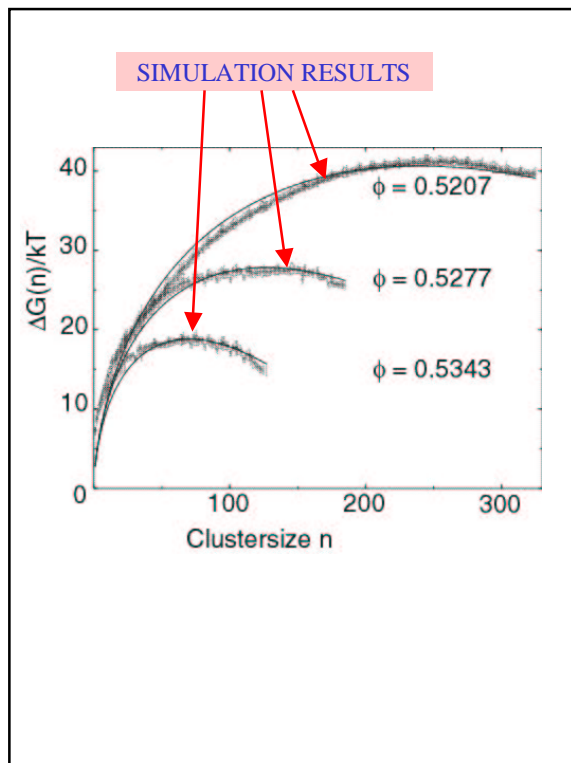
**COMPARISON WITH
EXPERIMENTS
AND
TESTING CLASSICAL
NUCLEATION THEORY**

TEST CASE
**CRYSTAL NUCLEATION of
COLLOIDAL HARD SPHERES**

WHY THIS SYSTEM?

- 1. We know “everything” about the equilibrium properties of hard spheres.**
- 2. Suspensions of uncharged *silica* or *PMMA* colloids really behave like hard-sphere systems**
- 3. There is experimental information on hard-sphere nucleation.**(Ackerson & Schaetzel, Harland & van Megen: on earth. Cheng, Zhu, Chaikin et al.: in μ -gravity)





BEST FIT to Classical Nucleation Theory

SIMULATIONS:

Supersaturated: $\gamma_{\text{eff}} \approx 0.72 \text{ kT}/\sigma^2$

At coexistence: $\gamma \approx 0.62 \text{ kT}/\sigma^2$

Experiments: $\gamma \approx 0.5 \text{ kT}/\sigma^2$

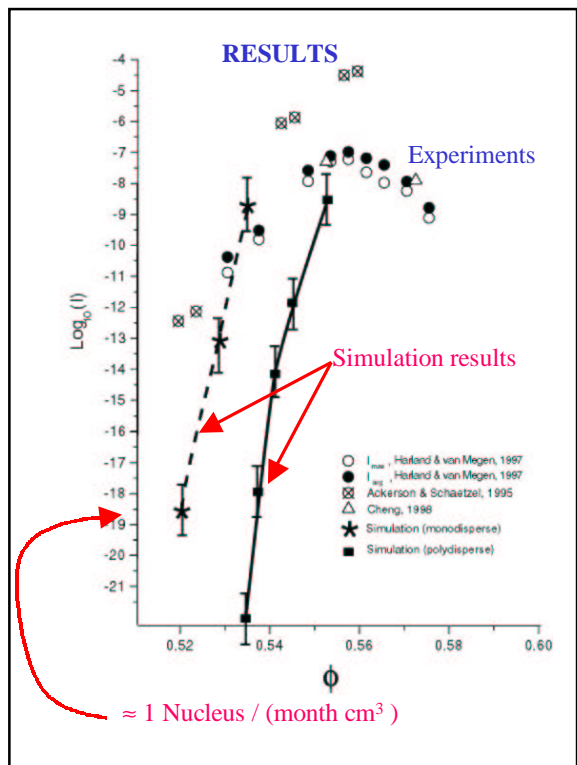
Some disagreement with CNT and with experimental estimates.

$\Delta G^* \sim \gamma^3$

Experiments underestimate barrier height by a factor 3 !

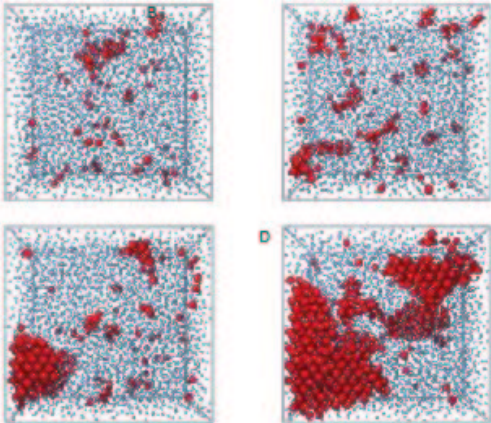
But remember, the nucleation rate is proportional to

$\exp[-16\pi\gamma^3/(3\rho^2\Delta\mu^2kT)]$



Such disagreement is disastrous.

But wait, things get even worse...



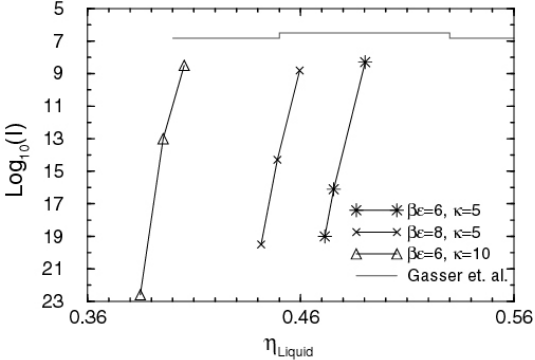
Confocal microscopy – Gasser et al. – Science, 292:258-262(2001)

Slightly charged colloidal spheres

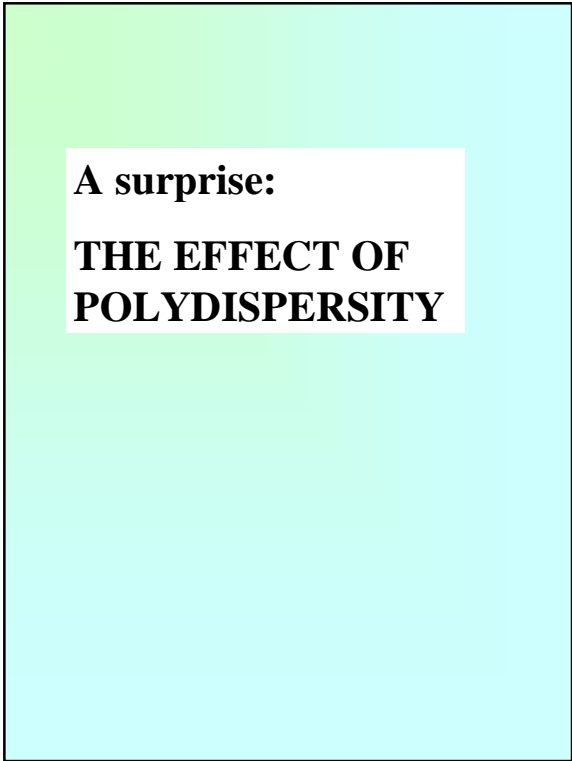
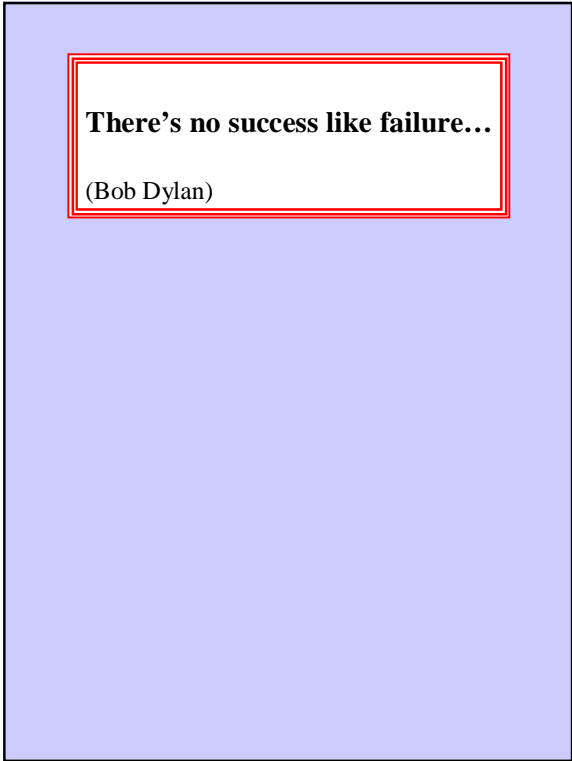
Such experiments on provide DIRECT, MICROSCOPIC INFORMATION ON CRYSTAL NUCLEATION

Simulations and experiments disagree about:

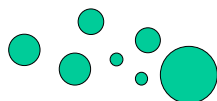
- 1. The structure of the nucleus**
- 2. The interfacial free energy**
- 3. The nucleation rate**



η_{Liquid}	$\text{Log}_{10}(I)$ (βε=6, κ=10)	$\text{Log}_{10}(I)$ (βε=8, κ=5)	$\text{Log}_{10}(I)$ (βε=6, κ=5)
0.38	23		
0.39	13		
0.40	9		
0.44		19	
0.45		15	
0.46		9	
0.48			19
0.49			17
0.50			9



POLYDISPERSITY IN HARD-SPHERE COLLOIDS

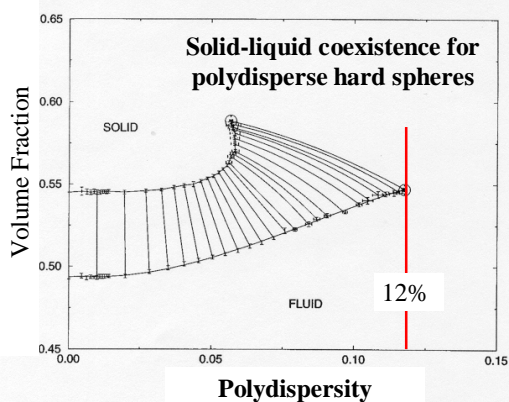


Polydispersity postpones, and eventually suppresses, hard-sphere freezing)

Polydispersity:

$$s \equiv (\langle r^2 \rangle - \langle r \rangle^2)^{1/2} / \langle r \rangle$$

Phase diagram of polydisperse hard spheres



(Bolhuis & Kofke, PRE, 54:634(1996))

What does this imply for the behavior of the nucleation barrier?

$$\Delta G^* = (16 \pi / 3) \gamma^3 / (\rho \Delta \mu)^2$$

Two “predictions”:

- 1. The barrier decreases**
- 2. The barrier increases...**

Prediction 1

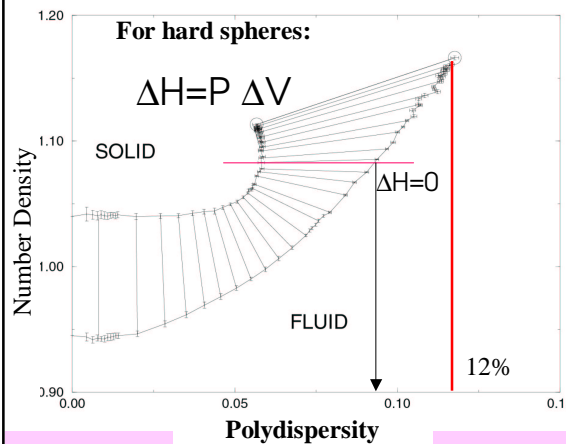
Use Turnbull's rule to estimate surface free energies (γ):

$$\gamma \approx 0.3 \Delta h / v^{2/3}$$

Where Δh is the enthalpy of fusion per particle, and v is the volume per particle (in the crystal)

"Turnbull's Rule"

$$\gamma = c \Delta H$$



Hence...

According to Turnbull, γ should go through zero, and hence the nucleation barrier

$$\Delta G^* = (16 \pi / 3) \gamma^3 / (\rho \Delta \mu)^2$$

should vanish for polydispersities around 9%

Prediction 2

The “nucleation theorem” (Kashiev, Oxtoby, Viisanen, Strey, Reiss...) establishes a relation between barrier height and nucleus size:

$$\frac{\partial \Delta G}{\partial \mu} = -n^*$$

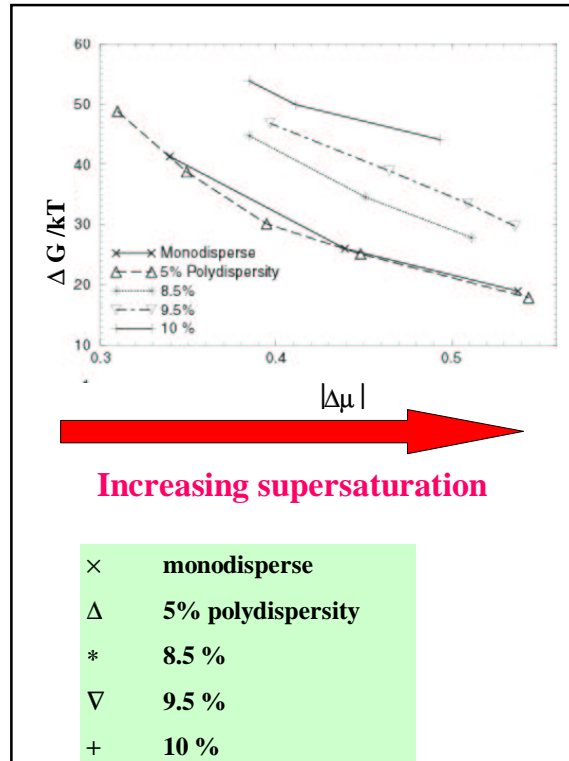
Where n^* is the excess number of particles in the critical nucleus

Hence:

When the number density of the critical nucleus is equal to that of the supersaturated liquid, then:

$$\frac{\partial \Delta G}{\partial \mu} = 0$$

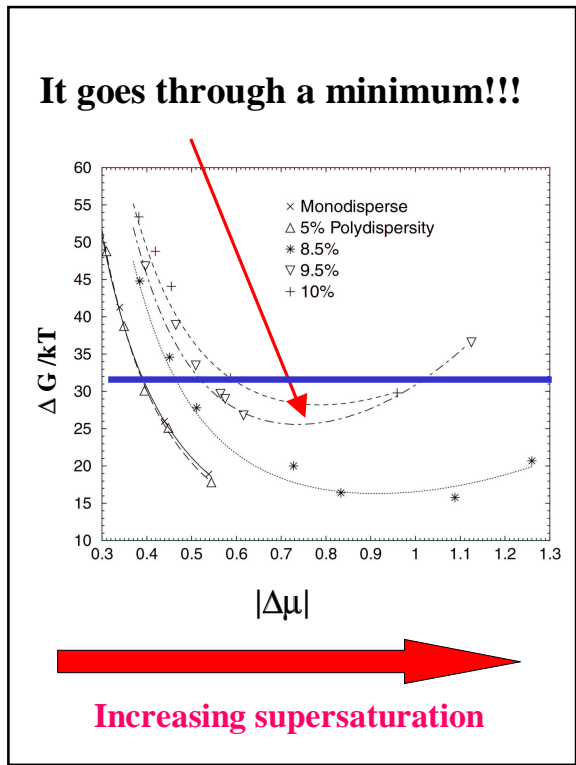
What do the simulations say???



Barrier INCREASES with polydispersity...

...also in the range where Turnbull's rule would predict that it should be zero!

Moreover...



Prediction 2 wins!

$$\frac{\partial \Delta G}{\partial \mu} = n^*$$

The simulations suggest that γ increases with $|\Delta\mu|$, e.g.:

$$\Delta G^* = \frac{16\pi\gamma_0^3 (1 + a\Delta\mu)^3}{3\rho\Delta\mu^2}$$

SUMMARY:

- To understand Nucleation, we need to study the **Critical Nucleus**.
- The **structure** of the critical nucleus is often **NOT** as predicted by CNT
- We find that the **barrier height** also differs from the CNT predictions...
- ...and the **rates** disagree with the analysis of the available experiments

**In short: we need better
experiments and better
theories...**

...or better simulations!