

## Flow and phase transitions in micellar systems

Shear thickening & time-dependent rheology

### Outline

Basic shear thickening phenomena  
Wormlike micelles  
Experimental approach – light scattering microscopy  
Rheological & structural behavior

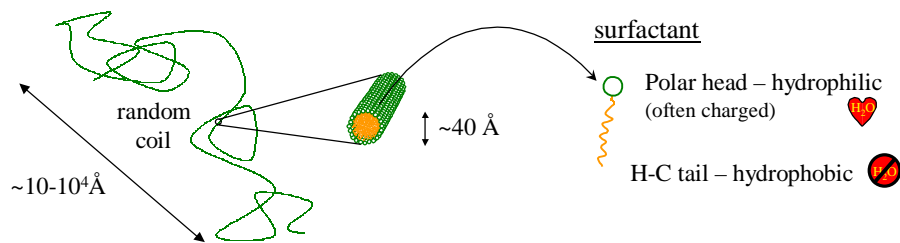
### Collaborators

Chu-heng Liu (Xerox) – light scattering microscopy  
Philippe Boltenhagen (Strasbourg) – rheo-scattering  
Yuntao Hu (Unilever) – rheo-scattering  
Jacqueline Goveas (Rice Univ.) – theory

1

## Worm-like micellar solutions

(in water)



### Concentration regimes

- critical micelle concentration ( $cmc$ )

$c < cmc \rightarrow$  only free surfactant molecules

$c > cmc \rightarrow$  micelles + free surfactant molecules

- overlap concentration ( $c^*$ )


micelles overlap each other above this concentration

experimentally often defined by the concentration where  
solution viscosity = twice solvent viscosity

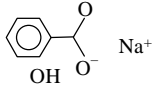
2

### Shear-thickening worm-like micellar solutions

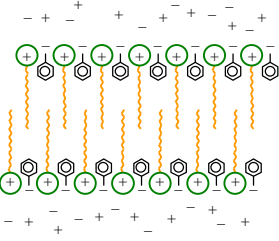
CTAB + NaSal + H<sub>2</sub>O (cetyltrimethylammonium bromide + sodium salicylate + water)



Br<sup>-</sup>



Na<sup>+</sup>



- Sal<sup>-</sup> ions are incorporated into micelles ~1:1 (NMR)  
(this is unusual – most charged micelles are not neutral!)
- c\* ~ a few millimolar

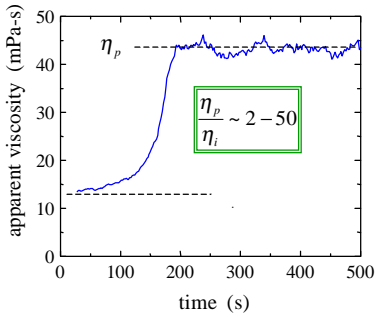
Most micellar systems which exhibit shear thickening are like this

3

### Shear thickening in dilute worm-like micellar solutions

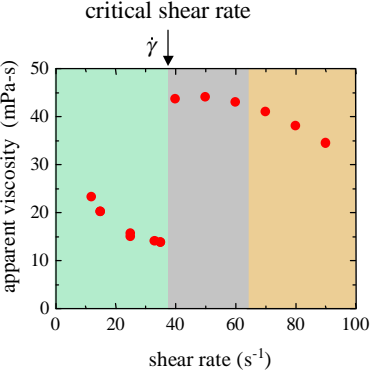
(basic rheology)

Response after turning on steady shear rate greater than some threshold



Note long latency time prior to shear thickening

critical shear rate



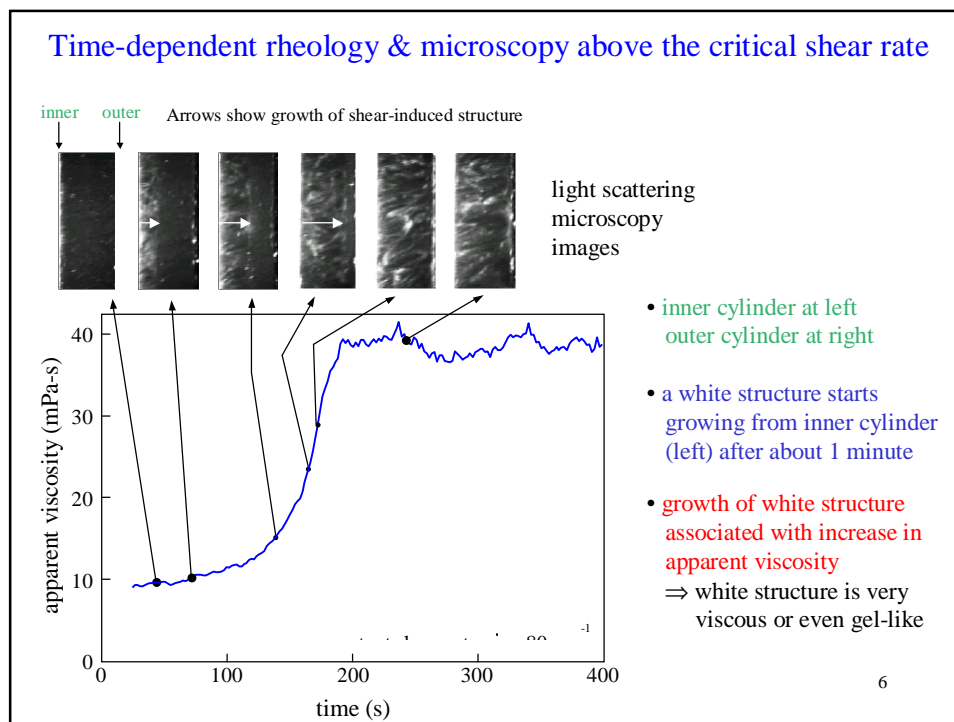
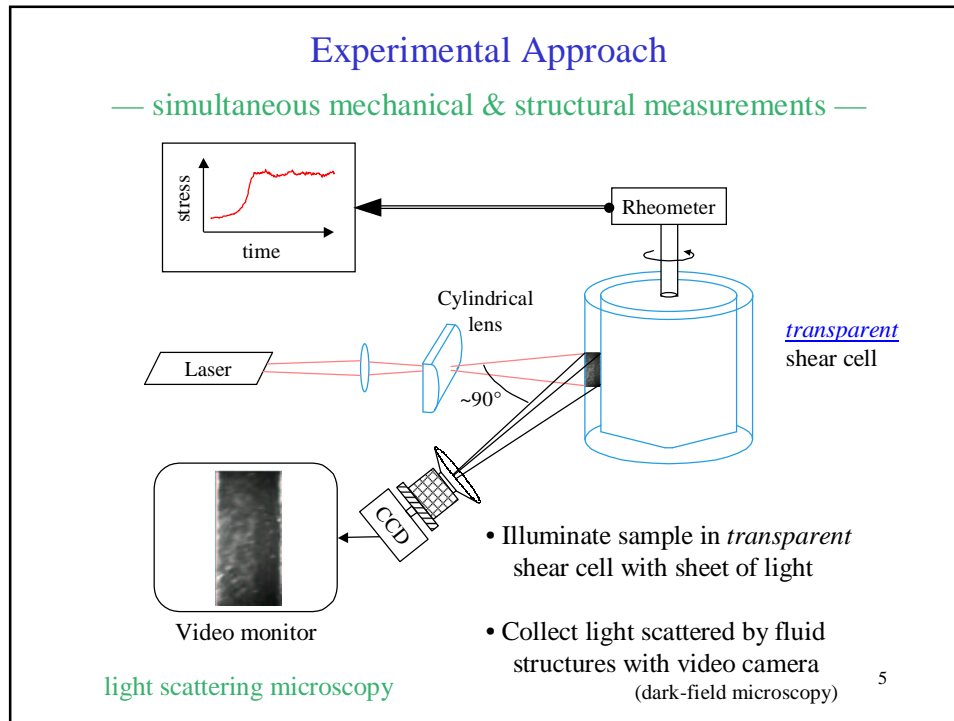
█ Shear thinning

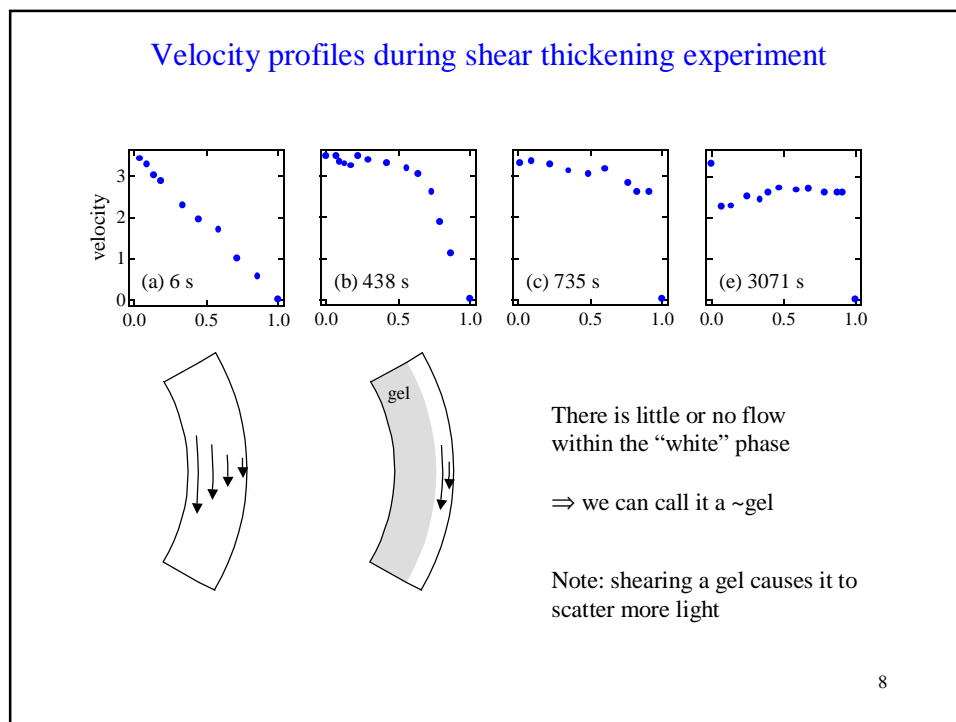
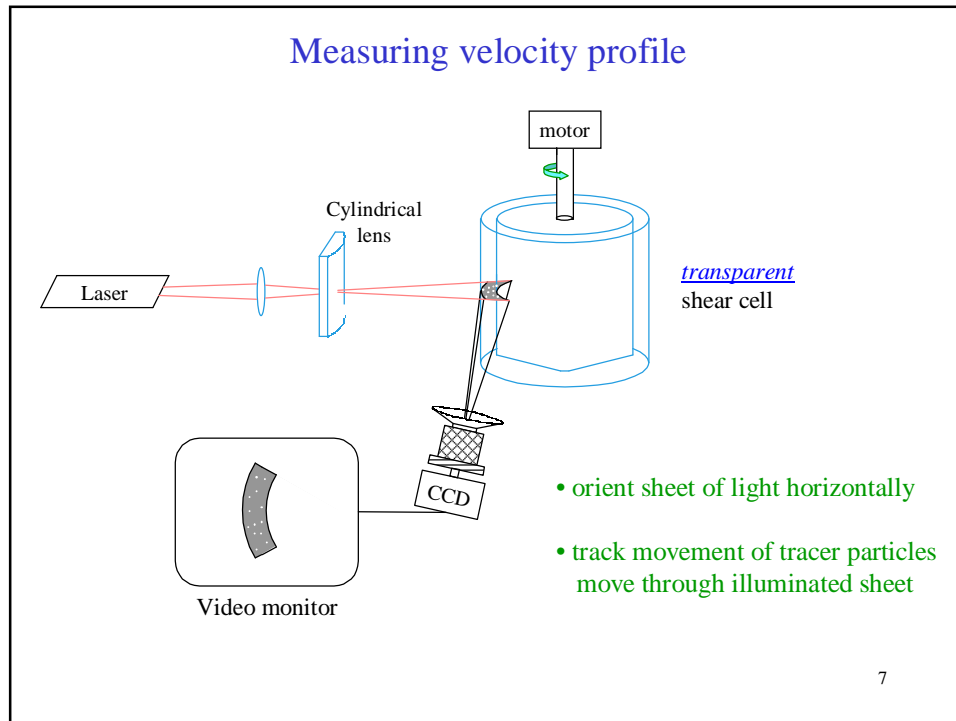
█ Shear thickened state

█ Shear thinning (after shear thickening)

Concentrations for shear thickening: ~cmc < c < ~3-5 c\*

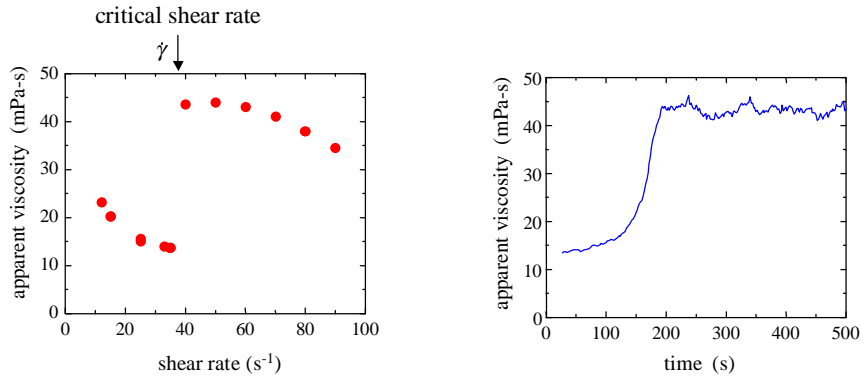
4





# Rheology of Shear-Thickening Micellar Solutions

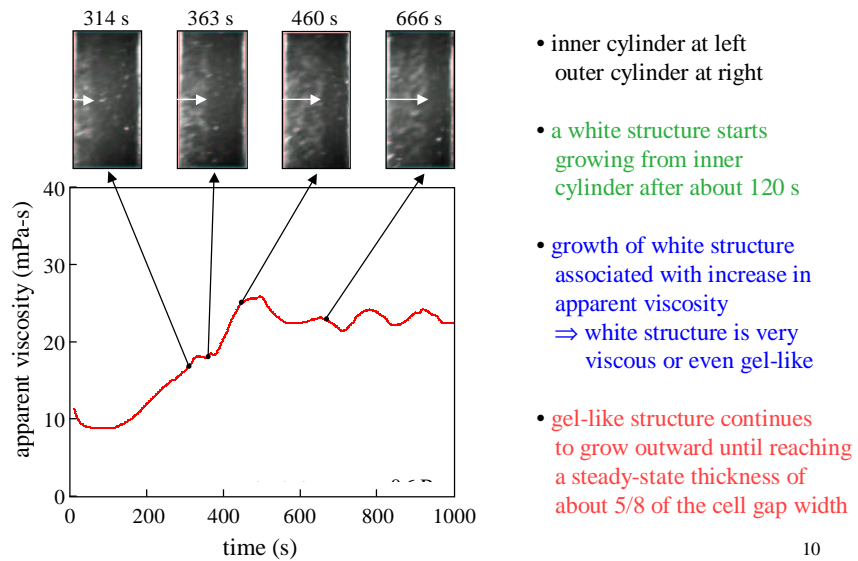
Why the steady state gel thickness & apparent viscosity increase discontinuously ...



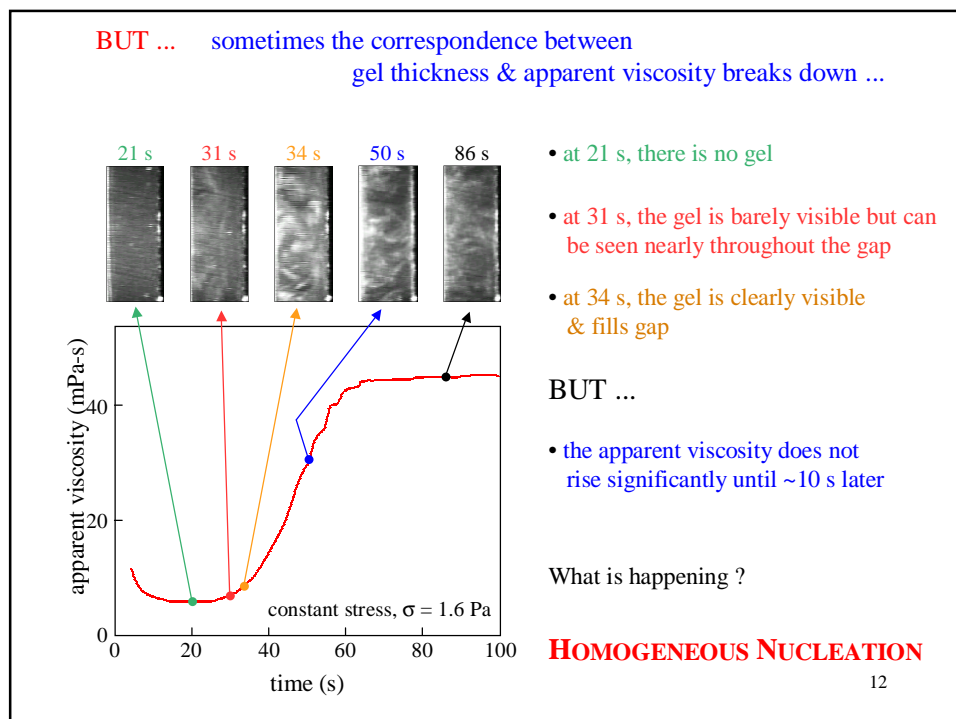
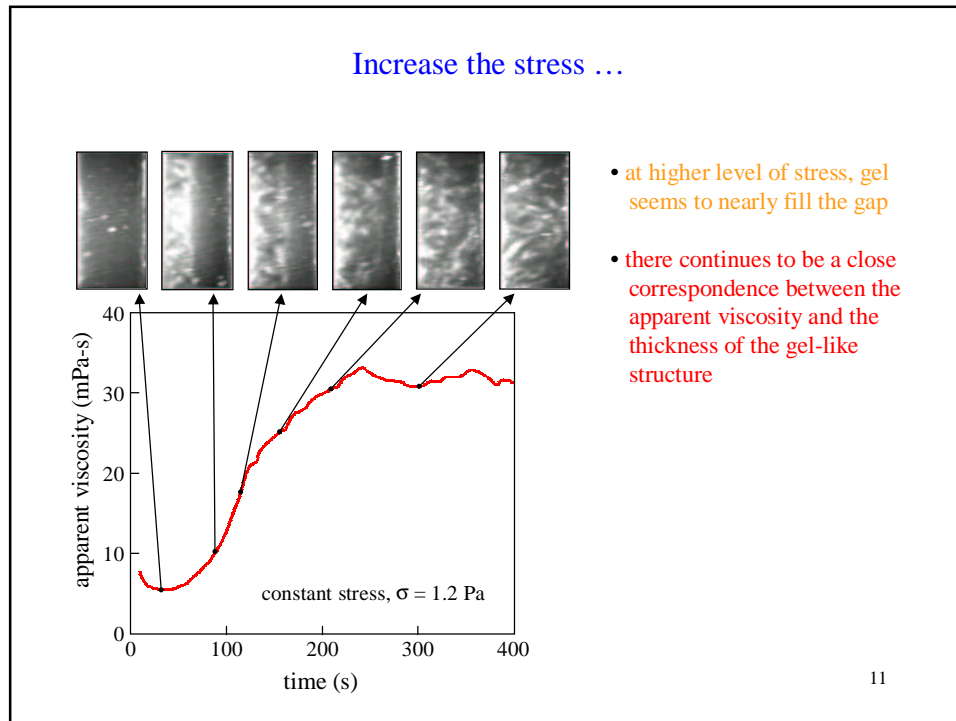
- (1) When  $\dot{\gamma} > \dot{\gamma}_c$ , "gel" begins to grow from inner cylinder
- (2) Since the gel doesn't flow, the shear rate and stress in the remaining fluid increase
- (3) This causes more gel to form and further increases the shear rate & stress
- (4) This process continues until there are no more micelles available to attach to gel

9

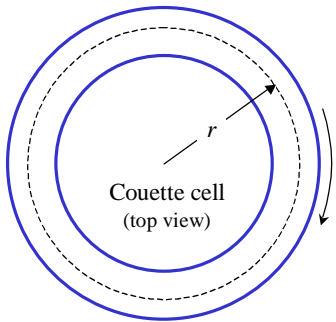
Time-dependent rheology & microscopy for controlled stress



10



### Nucleation and stability of gel phase



Mechanical equilibrium  $\Rightarrow \sigma \propto \frac{1}{r^2}$   
(balance torque at  $r$ )

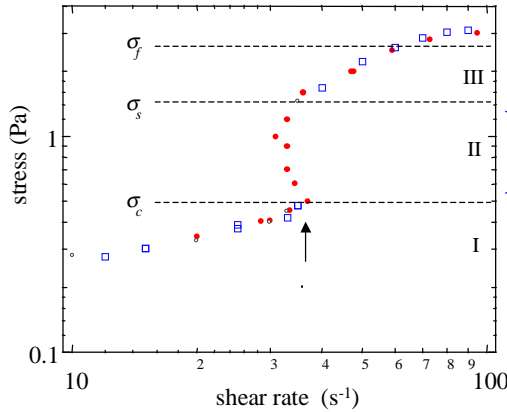
$\Rightarrow$  Stress is greatest near inner cylinder

- when stress or shear rate is increased incrementally, gel nucleates only at the inner cylinder
- when stress or shear rate is increased such that  $\sigma \gg \sigma_c$ , gel nucleates everywhere in the cell where  $\sigma \gg \sigma_c$

**HOMOGENEOUS NUCLEATION**

13

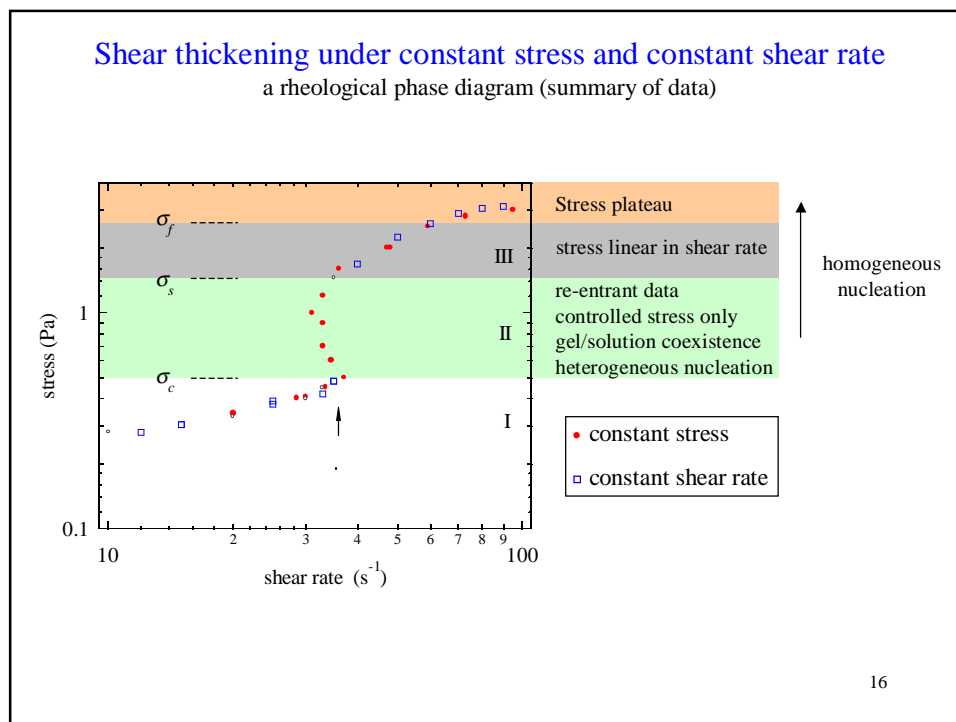
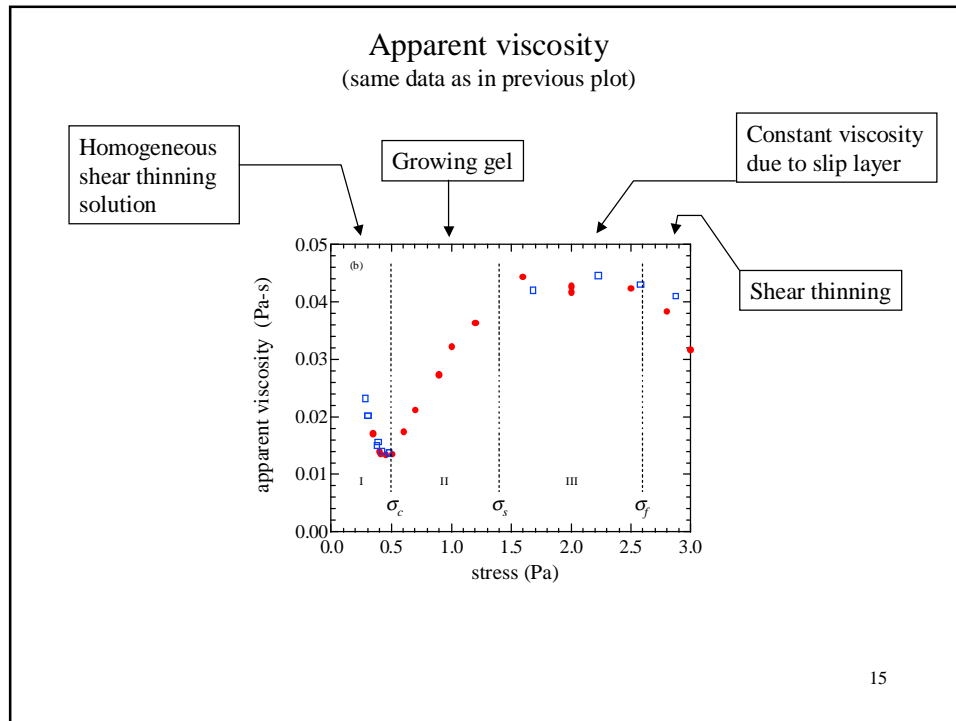
### Role of stress and shear rate



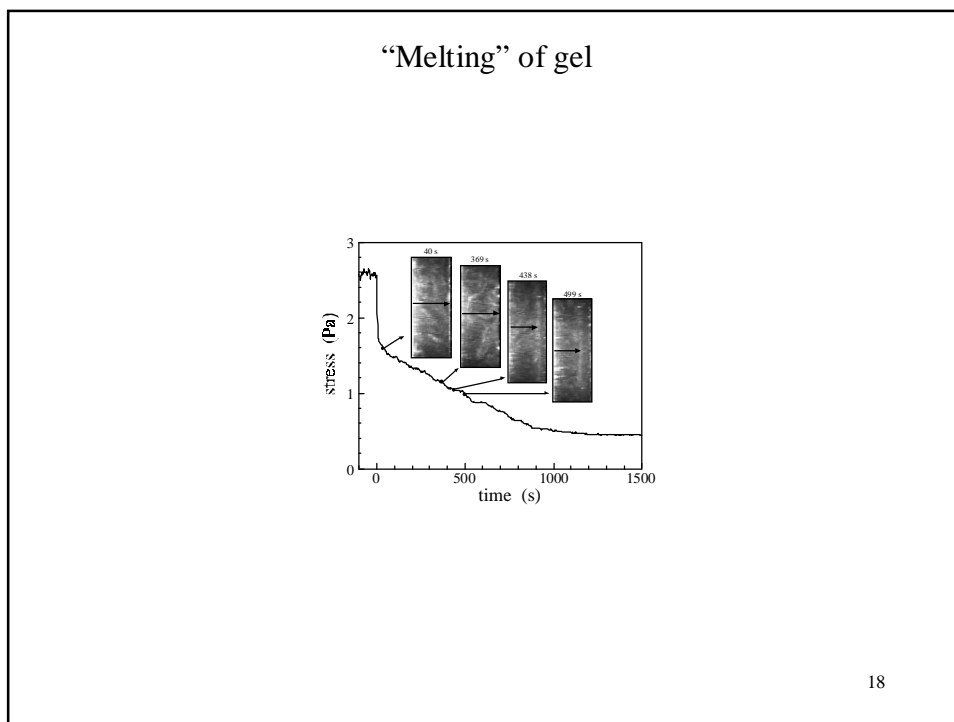
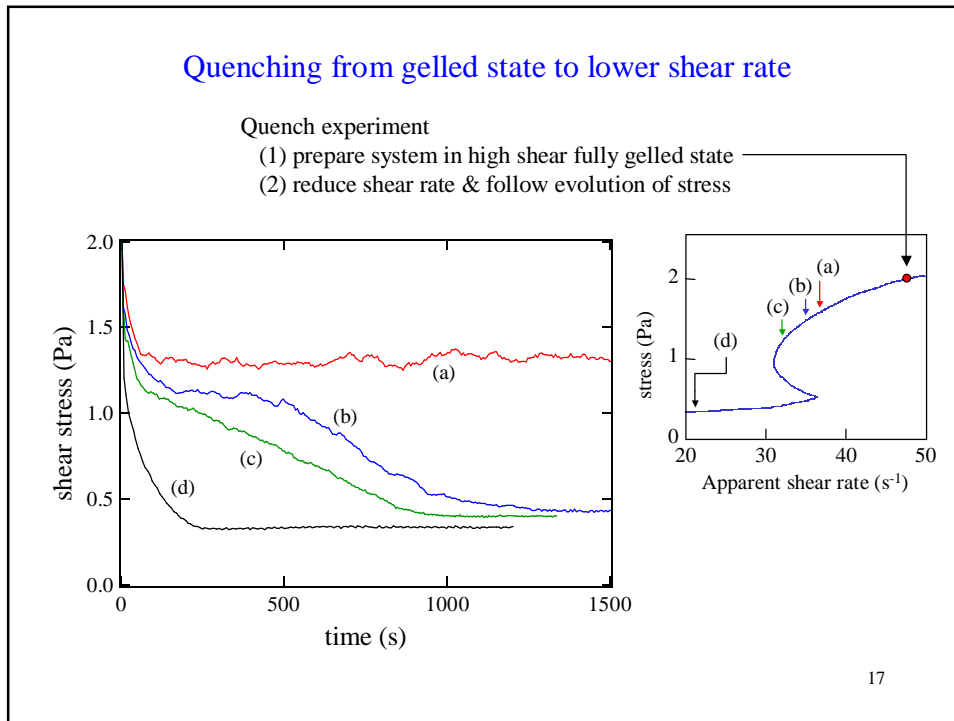
- In region II,  $\dot{\gamma} < \dot{\gamma}_c$ , but gel remains under controlled stress
- Shear rate in gel phase is nearly zero but gel phase persists for  $\sigma > \sigma_c$
- Gel phase is observed iff  $\sigma > \sigma_c$

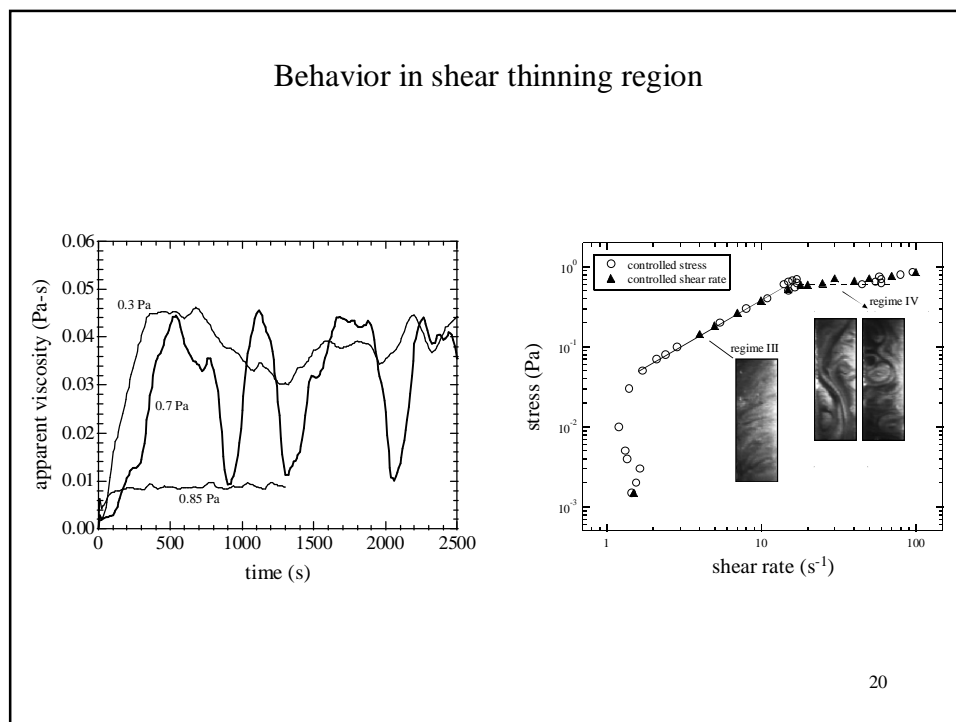
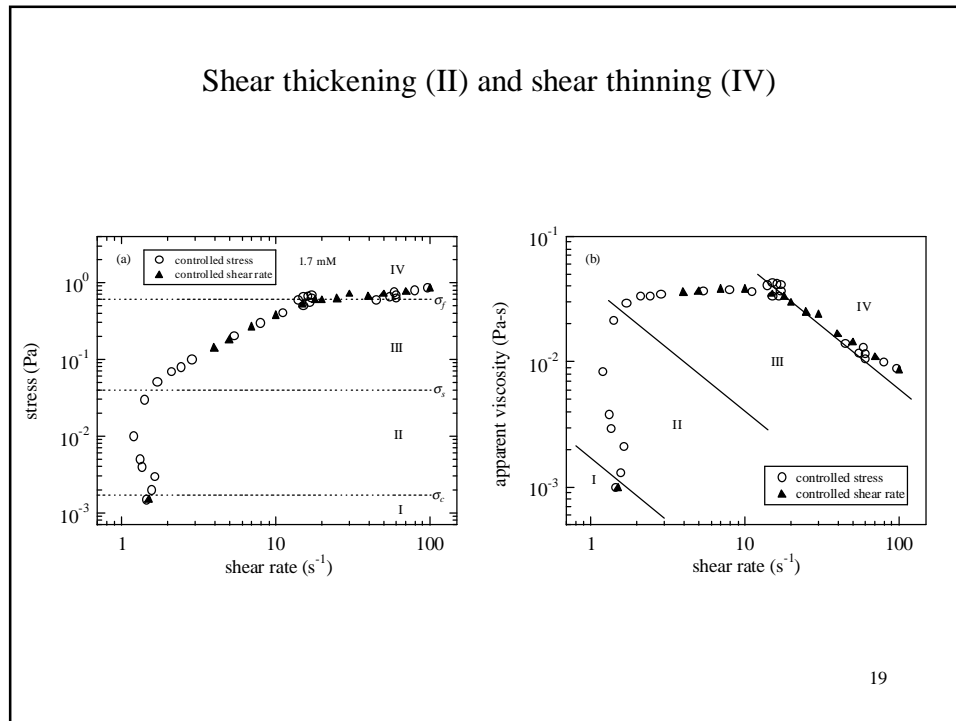
$\Rightarrow$  stress (and not shear rate) controls the formation of gel

14

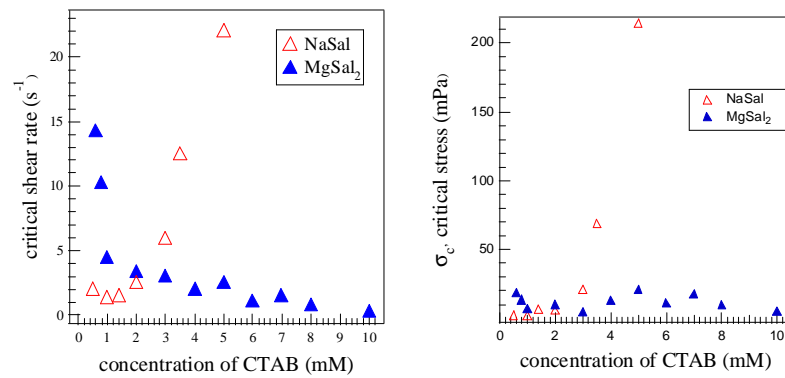






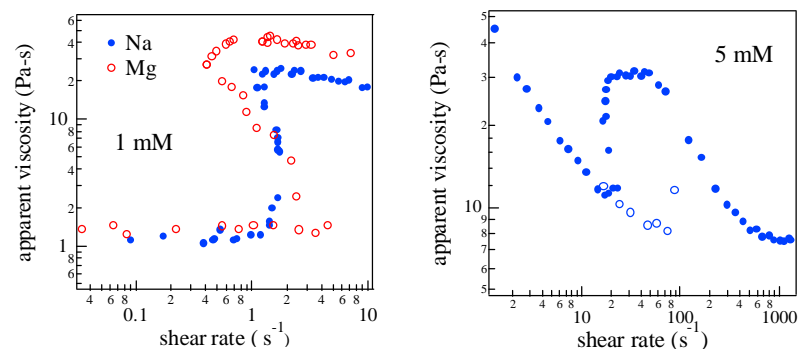


Effect of counterion valence on critical shear rate & stress



21

More rheology ...



22