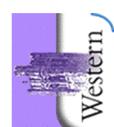
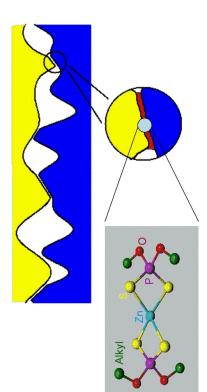
## Energy Dissipation Mechanisms at the Nanometer Scale: Computer Simulations



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Nanotips:

Ludgar Wenning University of Mainz

Nanotubes:

Steve Stuart Clemson University

Rough surfaces:

Carlos Campana University of Western Ontario

Anti-wear films:

Tom Woo University of Ottawa

Nick Mosey University of Western Ontario

#### Outline

Small-Scale Mechanisms of <u>interlocking</u> and <u>dissipation</u>

- geometric interlocking (why egg cartons don't explain much)
- other interlocking mechanisms
- role of instabilities / molecular hysteresis
- boundary lubricants (curved surfaces) unlubricated, rough, elastic contacts boundary lubricants (flat surfaces) - interlocking and instabilities in:
- chemistry-induced interlocking anti-wear films nanotubes

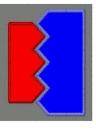
Energy Dissipation Mechanisms ...

## Geometríc ínterfockíng

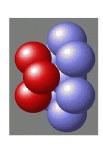
Amontons, Euler, Coulomb: geometric interlocking at surfaces are rough 🕁 small length scales

spherical caps on surfaces  $\mu_{\rm s} \approx 0.34$ Bélidor (≈1750)

But: Surfaces usually don't match and what about kinetic friction?

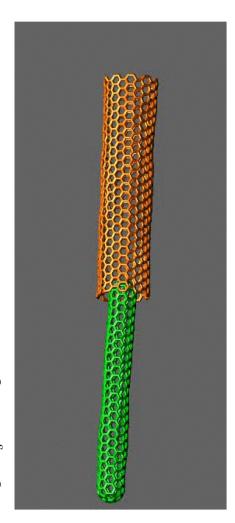






## Geometric interlocking

Nanotube embedded in another commensurate nanotubes ⇒ large, instantaneous, lateral forces  $\Rightarrow$  large  $F_s$ 

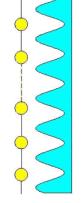


Tangney, Louie, Cohen, PRL 93, 065503 (2004) Thanks to Paul Tangney

Energy Dissipation Mechanisms

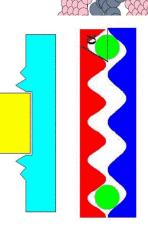
# Other interlocking mechanisms

for interlocking (Coulomb) elastic deformation allows



(Bowden and Tabor) - plastic deformation

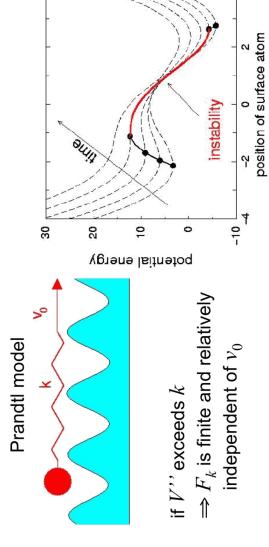
boundary lubricants



- cold welding, tribo-chemical reactions

## Role of instabilities

· interlocking does not explain kinetic friction or dissipation



Energy Dissipation Mechanisms ...

### Instabílíties in boundary lubricants



qualitative argument:
surface has to climb
up a slope  $\alpha \rightarrow \mu = ta$ 

accounts for static friction

G. He, MHM, M.O. Robbins, Science 284, 1650 (1999)

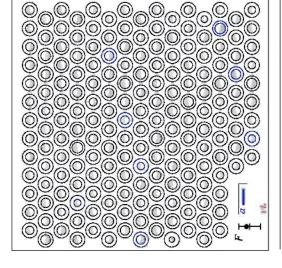
Do boundary lubricant **automatically** lead to  $F_{\mathbf{k}}$ ? Do boundary lubricants **typically** lead to  $F_{f k}$ ?

Yes.

Š.

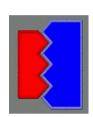
### Instabílíties in boundary lubricants

Do boundary lubricants **automatically** lead to  $F_{\mathbf{k}}$ ?



Energy Dissipation Mechanisms ...

Two **commensurate** walls separated by a **single-layer** lubricant:

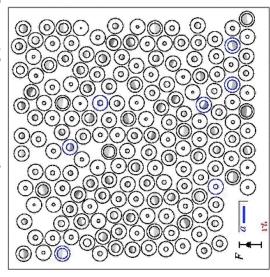


Theory: MHM, PRL 89, 224301 (2002)
Snapshots: MHM, MWWT 35, 603 (04)
Movies: publish.uwo.ca/~mmuser

show movie in external player

### Instabílities in boundary lubricants

Do boundary lubricants **typically** lead to  $F_{
m k}$ ?



Energy Dissipation Mechanisms ...

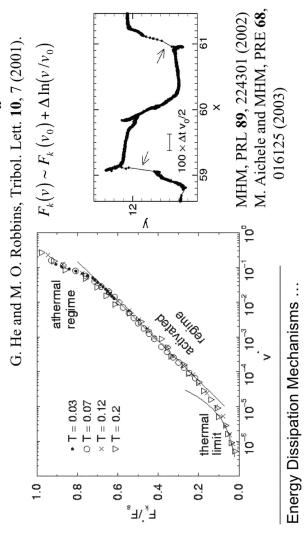
Two incommensurate walls separated by a single-layer lubricant:

Theory: MHM, PRL 89, 224301 (2002)
Snapshots: MHM, MWWT 35, 603 (04)
Movies: publish.uwo.ca/~mmuser

show movie in external player

### Instabílities in boundary lubricants

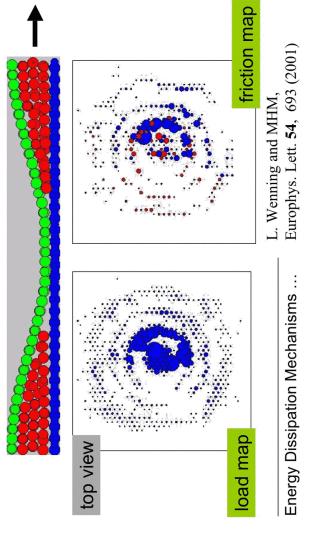
Do boundary lubricants  ${f typically}$  lead to  ${f typical}$   $F_k$ ?



## Boundary fubricants

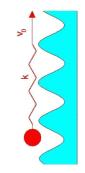


curved interfaces: may be different in detail (squeeze out)



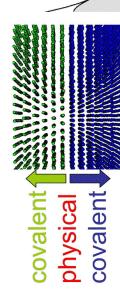
## Elastic instabilities

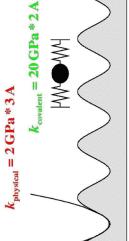
Can the competition of elasticity and roughness (on long length scales) lead to friction without local friction mechanisms?



if  $V^{"}>k\Rightarrow F_k$  is finite

on local scale: atypical





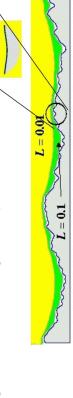
Do larger length scales change the picture?

Energy Dissipation Mechanisms ...

## Elastic instabilities

- fractal on fractal -

Do larger length scales change the picture?



- ullet potential such that "Amontons's law" holds locally:  $F_{ ext{lat}} = (
  abla h) L$
- Hurst roughness exponent  $\frac{2\pi}{\lambda_{\text{long}}} \le q \le \frac{2\pi}{\lambda_{\text{short}}}$  $\tilde{S}(\vec{q}) \equiv \left\langle \tilde{h}(\vec{q})\tilde{h}^*(\vec{q}) \right\rangle \propto q^{-2H-d}\delta(\vec{q}-\vec{q}')$  for  $C(\Delta r) \equiv \left\langle \left\{ h(\vec{r}) - h(\vec{r} + \Delta \vec{r}) \right\}^2 \right\rangle \propto \Delta r^{2H}$ 
  - elastic slider: mass points m = 1, springs k = 1, length a = 1
- 1D: elastic chain (maps onto Frenkel-Kontorova model) 2D: trigonal lattice (Poisson ratio: v=0; v=1/3)
  - 3D: s.c. lattice with 2nd neighbor coupling

Energy Dissipation Mechanisms ...

C. Campana and MHM, to be submitted

**Energy Dissipation Mechanisms** 

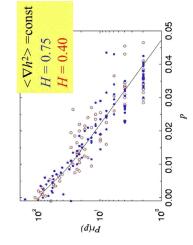
## Elastic instabilities

fractal on fractal

reproduces almost exactly results for the 1D FK model

models of elastic, non-adhesive contacts suggest:

 $Pr(p)=f(<\nabla h^2>)\sim \exp(-p/p$  ') for  $p\to\infty$  Hyun et al., PRE 70, 026117 (2004)



Measurable friction coefficients: Energy Dissipation Mechanisms ... Presence of (local) buckles

1+1 dimensional surface Friction:

$$H = 0.75 : \sqrt{\langle (\nabla h)^2 \rangle} = 0.35$$
$$L = 0.1B$$

$$\lambda_{c}^{\text{short}} = 4a \; ; \; \lambda_{c}^{\text{long}} = 1024 \, a$$

$$\Rightarrow \mu_{s} = 2 \times 10^{-4}$$

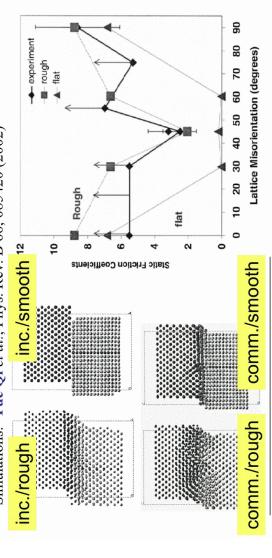
C. Campana and MHM, to be submitted

## Elastic instabilities

Ní(100) on Ní (100)

 local roughness seems key factor for friction coefficients see also Mark's recent Nature paper

Simulations: Yue Qi et al., Phys. Rev. B 66, 085420 (2002)



# Chemistry-induced interlocking

carbon nanotubes

 friction between nanotubes (chemically realistic potentials) Stuart, Piotrowski, and MHM, J. Phys. Chem. B (to be submitted)

H-terminated infinite (PBC) inner: outer:

superlubric

unterminated outer:

point defects superlubric superlubric outer:

F < 0.1 nN / vacancy

Energy Dissipation Mechanisms ...

# Chemistry-induced interlocking

carbon nanotubes -

 friction between nanotubes (chemically realistic potentials) Stuart, Piotrowski, and MHM, J. Phys. Chem. B (to be submitted)

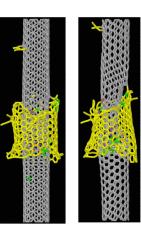
;  $\sigma < 0.02 \text{ MPa}$ F < 0.1 nNsimulations:

experiments: F = 100 nN

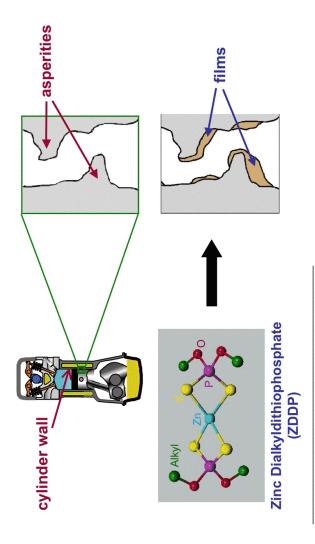
simulations do not include fracture process of outer tube

snapshot following fracture

snapshot following sliding 50 nN H simulations:



### -formation and functionality of anti-wear additives Chemistry-induced interlocking



Energy Dissipation Mechanisms

### Antí-wear fílms



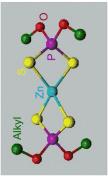
- invented in the 1930's
- optimized for steel and cast iron surfaces

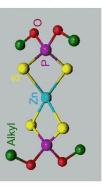
new additives sought-after

- ZDDPs ineffective on lightweight materials, such as aluminum
- Zn, S and P atoms damage catalytic converters

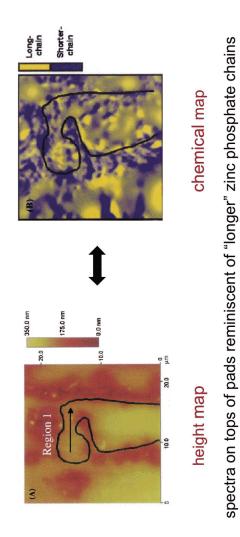
phenomenology of ZDDP anti-wear pads well known

- theories for anti-wear pad (AWP) formation unconvincing - no molecular theory of AWP functionality
  - rational design of new AW additives not possible





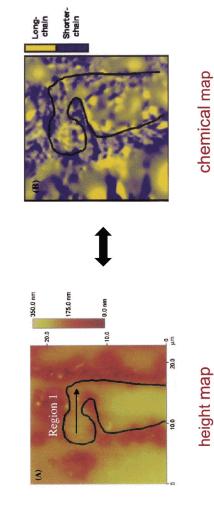
### Antí-wear fílms - ZDDP film phenomenology -



Energy Dissipation Mechanisms ...

Nicholls et al., Tribol. Lett. 17, 205 (2004)

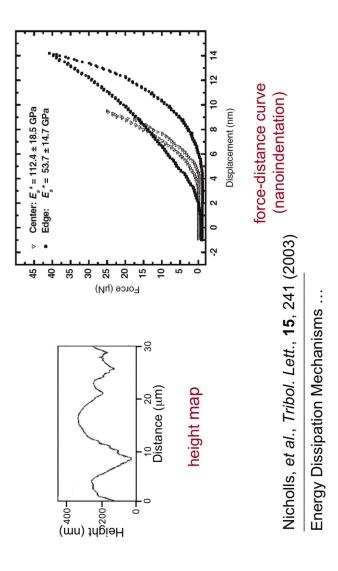
#### Anti-wear films - ZDDP film phenomenology



spectra on tops of pads reminiscent of "longer" zinc phosphate chains

Nicholls et al., Tribol. Lett. 17, 205 (2004)

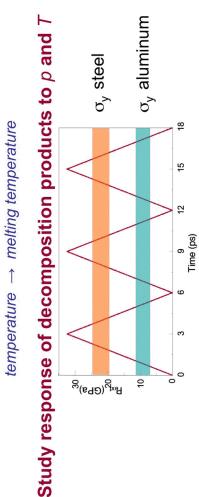
### Antí-wear fílms - ZDDP fílm phenomenology -



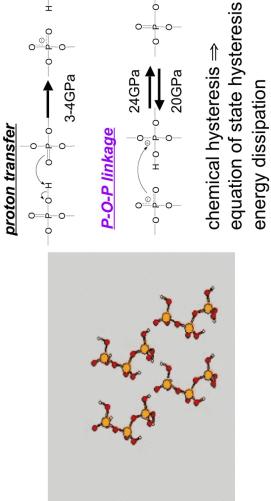
### Antí-wear fílms - ZPs ín mícrocontacts -

Pads form and function on asperities

Theory must consider extreme conditions in contacts theoretical yield strength pressure



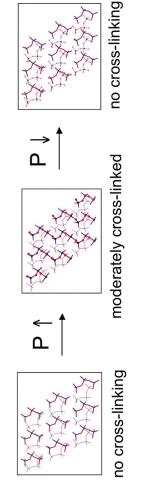
#### ZPs in microcontacts Anti-wear films



24GPa

Mosey, Woo, MHM, PRB 72, 054124 (2005) Energy Dissipation Mechanisms ...

### structural changes in phosphates Anti-wear films



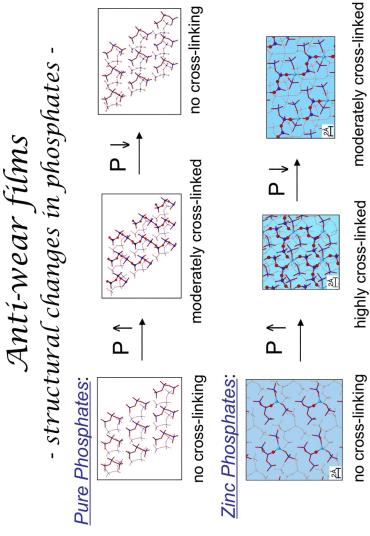
Pure Phosphates.

cannot serve as AWPs, because do not form pads

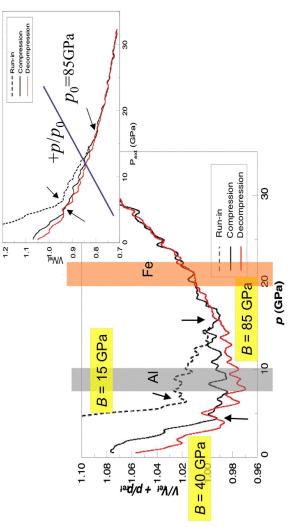
Mosey, MHM, Woo, Science 307, 1612 (2005)

Energy Dissipation Mechanisms ...

# Antí-wear fílms



### equation of state of ZPs Anti-wear films



Mosey, MHM, Woo, Science 307, 1612 (2005) Energy Dissipation Mechanisms ...

## Anti-wear films - comparison to experiment

chemically-connected, hard pads form where p is high

Nicholls, et al., Tribol. Lett., 2004, 17, 205 Nicholls, et al., Tribol. Lett., 2003, 15, 241 explanation of why ZDDPs do not work on Al

effects of Ca-containing detergents can be rationalized

reduces wear on Al (Norton group)

increases wear on steel (Varlot, et al., Wear, 2001, 249, 1029)

... and many other observation (growth rate, tempering,

Energy Dissipation Mechanisms ...

### Conclusions

- kinetic friction requires the presence of instabilities
- interlocking alone may not be enough
- instabilities can often be categorized
- elastic instabilities
- plastic instabilities
- erratic motion in boundary lubricants
- pressure or sliding-induced chemical reactions
- the last layer plays a tremendously important role