

# The Atmospheres of the Terrestrial Worlds

## Earth-Venus-Mars-Titan



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# Map

- The basics
  - Atmospheres make-up
  - Temperatures
  - Greenhouse now, greenhouse then; aerosols too
- Nitrogen, nitrogen everywhere, but where from?
- Methane only somewhere: does it mean life?
- What next?

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# Earth



1 AU  
288 K  
1 bar

N<sub>2</sub> 78% by vol  
O<sub>2</sub> 21%  
O<sub>3</sub> 0.3 ppm  
CH<sub>4</sub> 1.78 ppm  
H<sub>2</sub>O ≤1%  
CO<sub>2</sub> 400 ppm  
SO<sub>2</sub>, HCl

# Mars



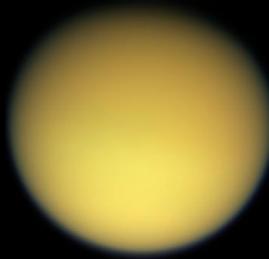
1.5 AU  
220 K  
6 mb

CO <sub>2</sub>	96%
N <sub>2</sub>	2%
O <sub>2</sub>	0.1%
O <sub>3</sub>	0.01-1 ppm
CH <sub>4</sub>	0.7-7 ppb
H <sub>2</sub> O	150 ppm
H <sub>2</sub> O <sub>2</sub>	0-20 ppb

9.5 AU  
94 K  
1.5 bar

N<sub>2</sub> 94%  
CH<sub>4</sub> 6%  
C<sub>x</sub>H<sub>y</sub>-N<sub>z</sub>

# Titan



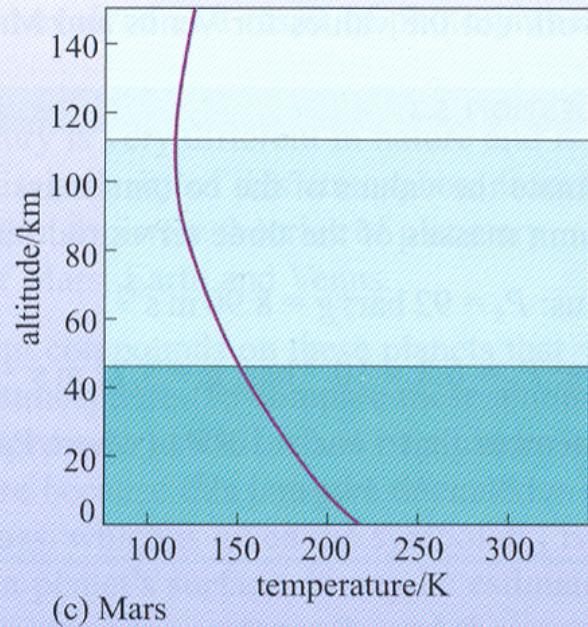
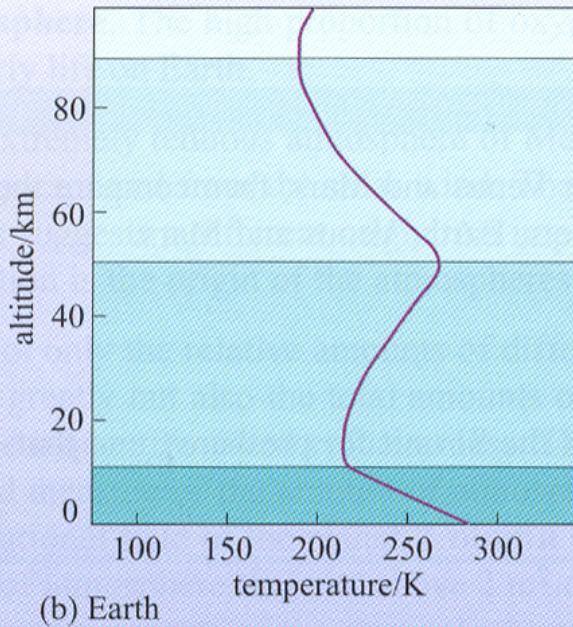
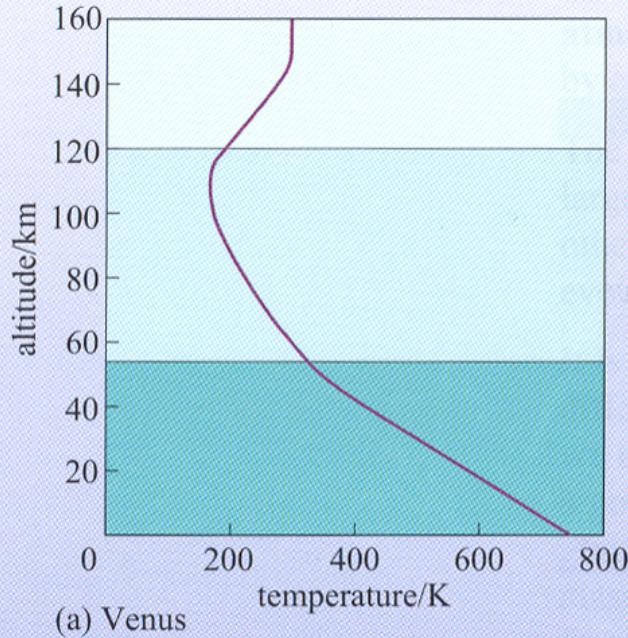
# Venus (climate gone wild)



0.72 AU  
735 K  
92 bar

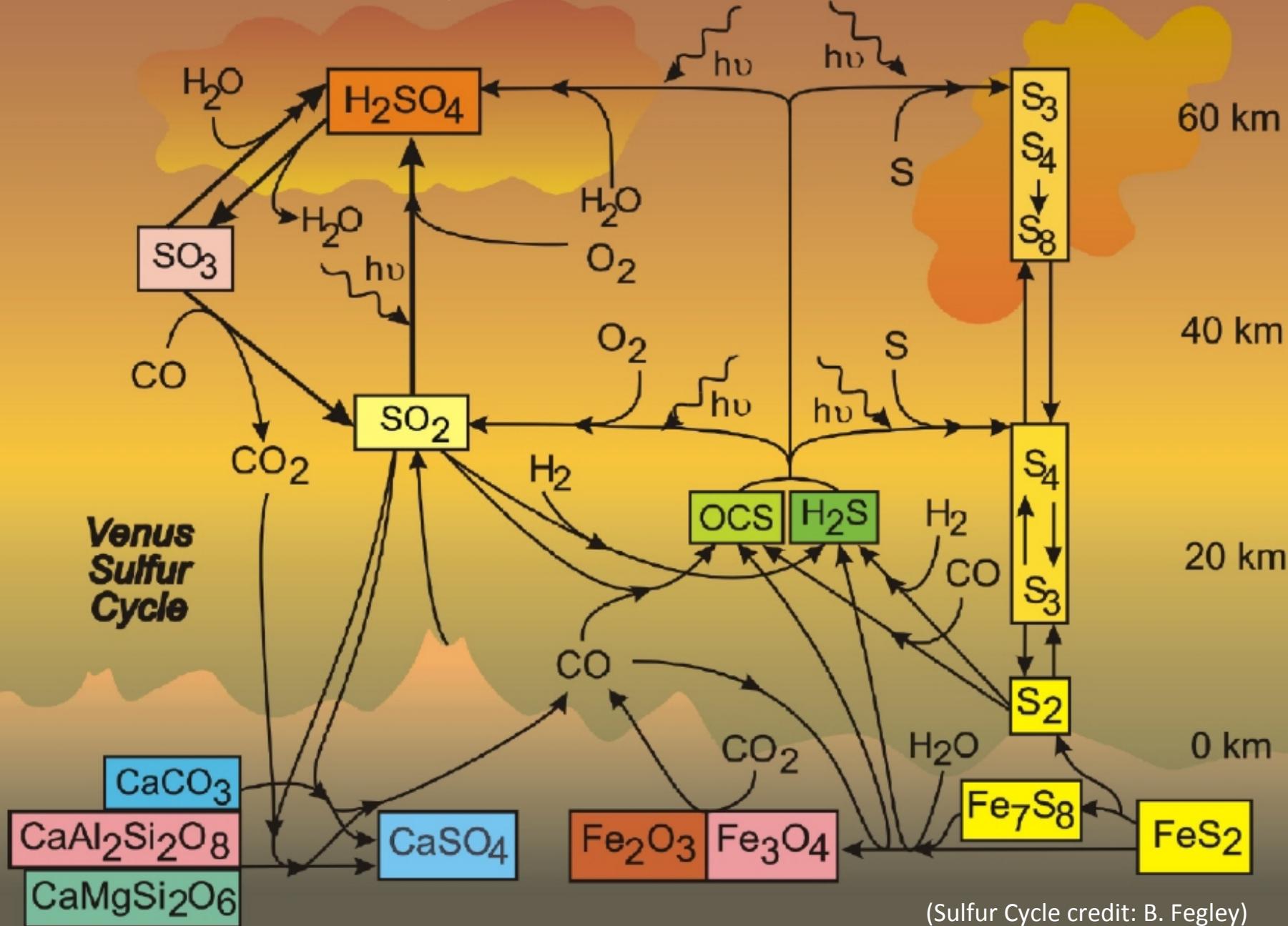
CO <sub>2</sub>	96%
N <sub>2</sub>	3%
O <sub>2</sub>	ppm
H <sub>2</sub> O	10 ppm
SO <sub>2</sub> , H <sub>2</sub> SO <sub>4</sub>	
HCl, HF	

# Venus-Earth-Mars Thermal Structure



$T_{eq}$	231 K	253 K	212 K
$T_s$	735 K	288 K	220 K
Greenhouse:	$\text{CO}_2 \sim 500 \text{ K}$	$\text{H}_2\text{O } 30 \text{ K}, \text{CO}_2 5 \text{ K}$	aerosols $\sim 8 \text{ K}$

# $H_2SO_4$ aerosols cool Venus

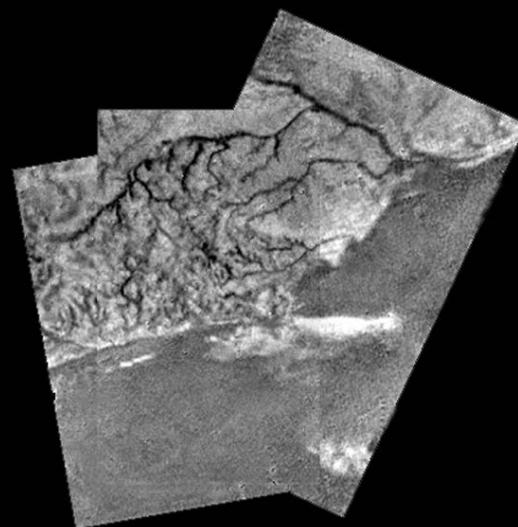


# Titan

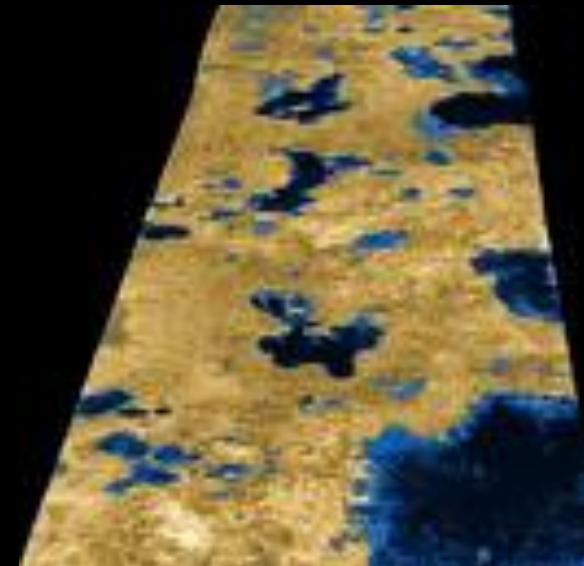
## Aerosols *heat* the atmosphere, and control its fate



**haze/Voyager  
1980**

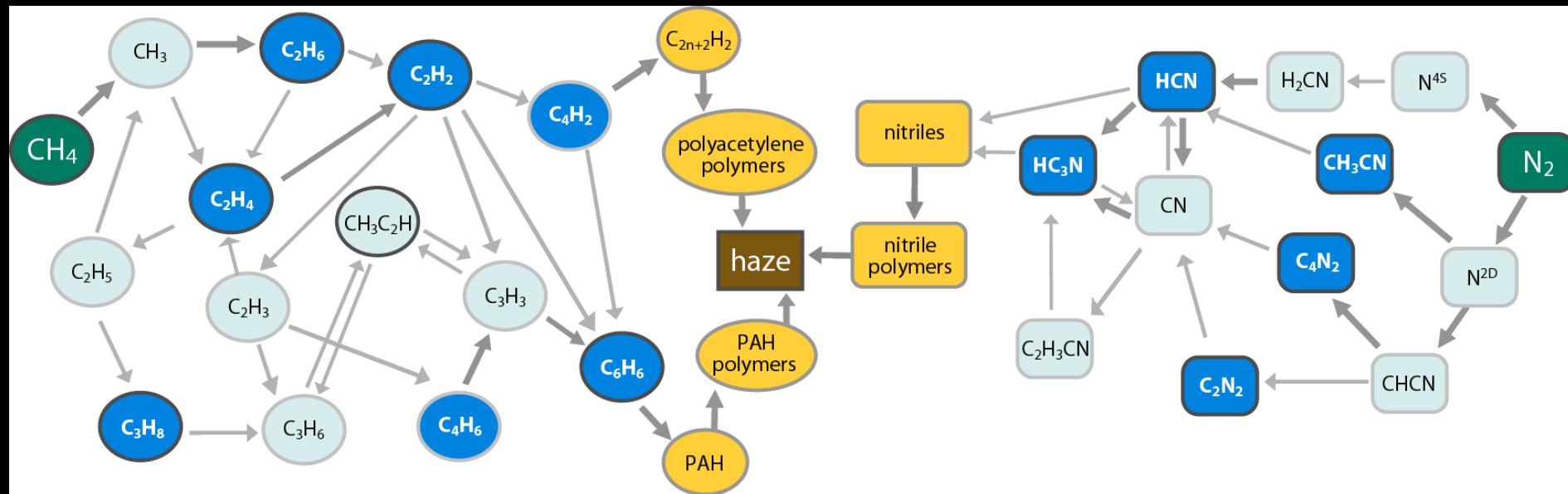


**rivers/Huygens  
2005**



**lakes/Cassini  
2004-2017**

# Methane converts *irreversibly* to heavier hydrocarbons, nitriles, hazes in ~30 My by neutral and ion chemistry

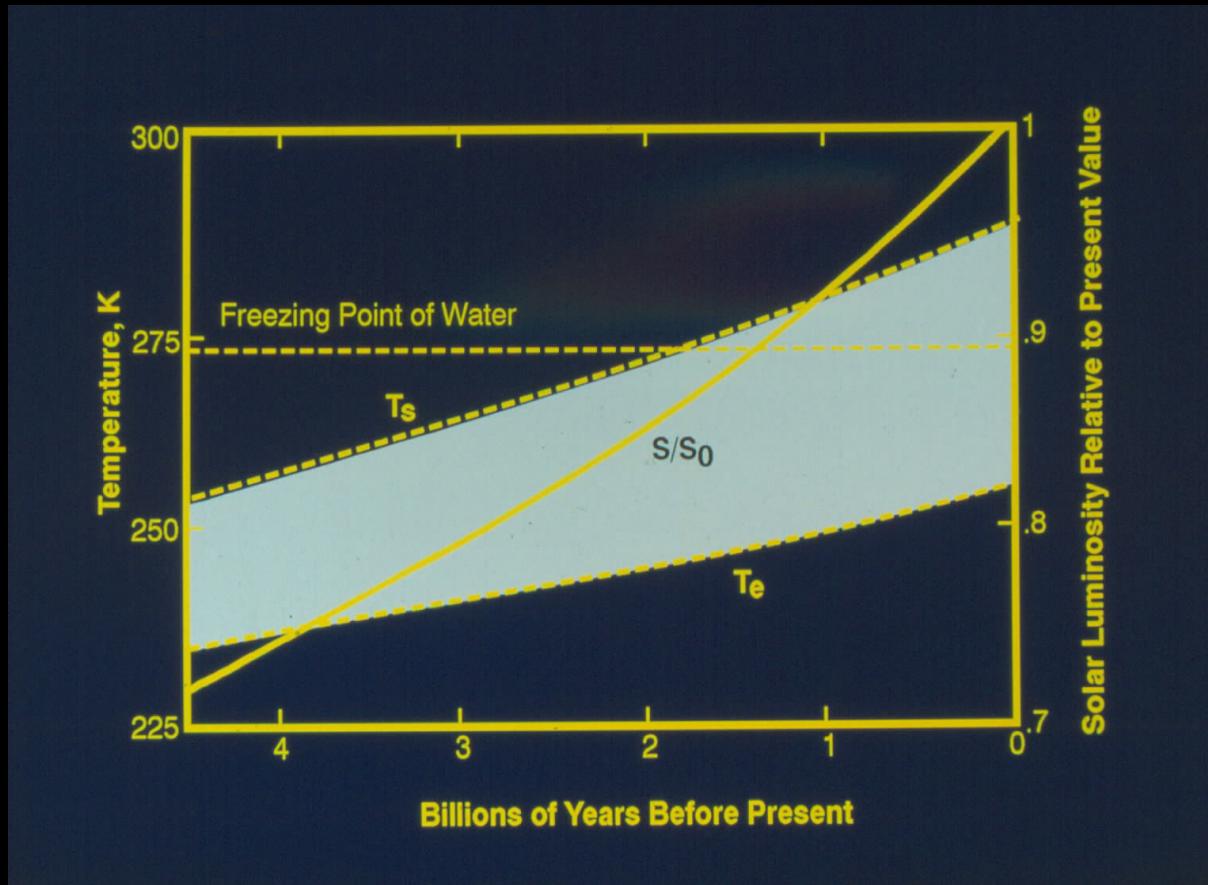


No CH<sub>4</sub> → N<sub>2</sub> condenses → Little atmosphere!

Hazes from CH<sub>4</sub>-N<sub>2</sub> → 120 K warming in the stratosphere

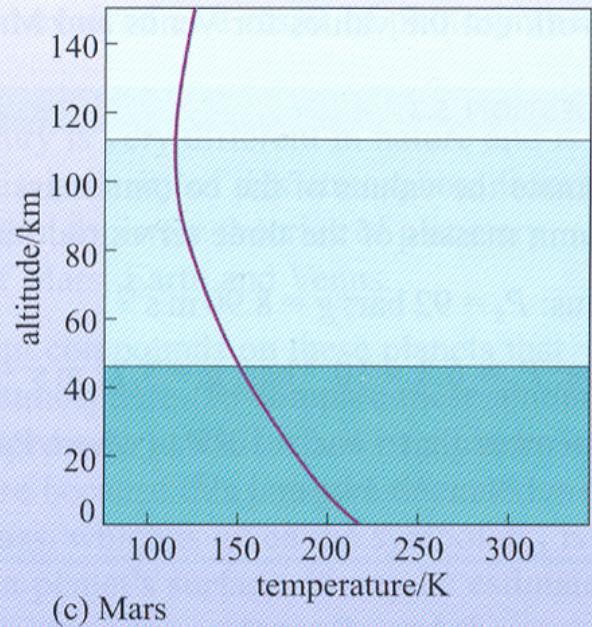
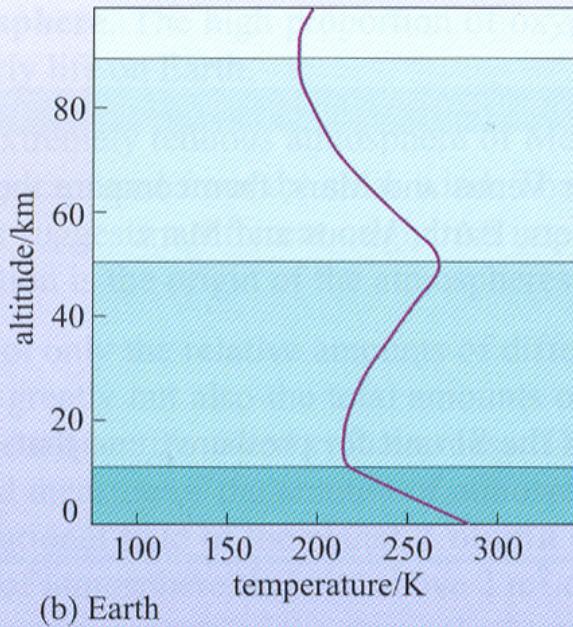
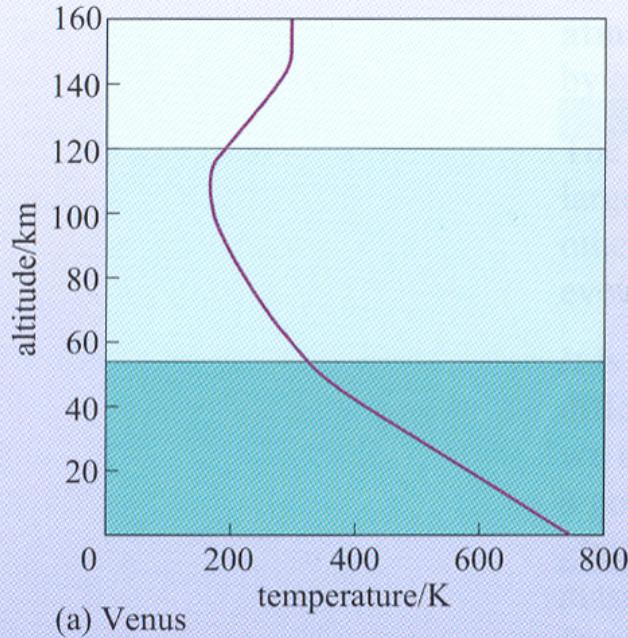
CIA (CH<sub>4</sub>-N<sub>2</sub>, CH<sub>4</sub>-H<sub>2</sub>...) 20 K warming in the troposphere

# The Faint Young Sun Problem



Earth's surface temperature would have been below freezing prior to 2 Gy, but liquid water has been present from the beginning  
(Kasting 1988)

# Venus-Earth-Mars Thermal Structure



thermosphere

mesosphere

stratosphere

troposphere

D/H    **150× SMOW**

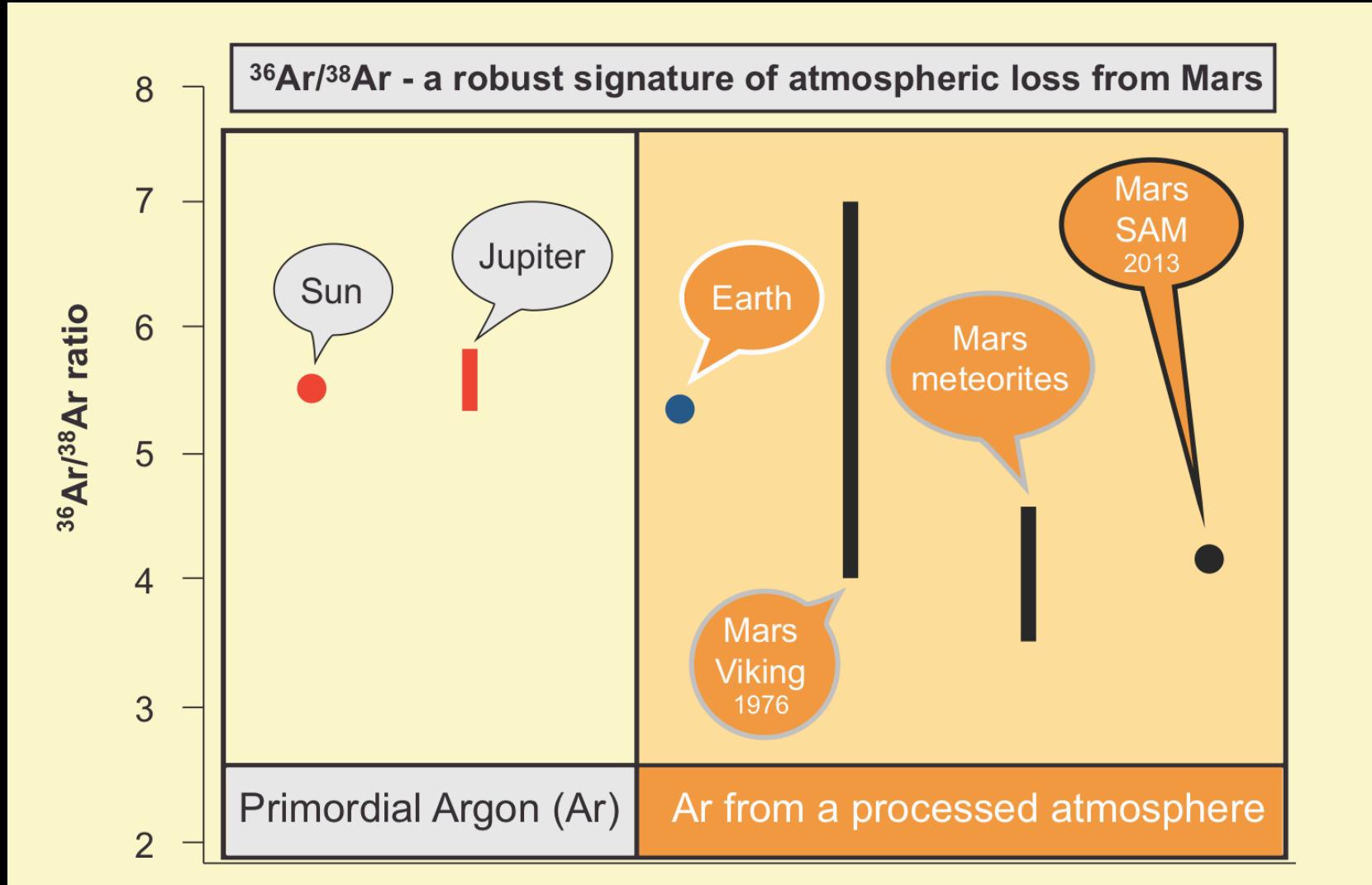
**$1.6 \times 10^{-4}$  (SMOW)**

**5-8× SMOW**

Greenhouse gases in the past  
H<sub>2</sub>O initially, then CO<sub>2</sub>

CO<sub>2</sub> (~~NH<sub>3</sub>, CH<sub>4</sub>~~)

CO<sub>2</sub>



Argon isotope ratio gives the most compelling evidence that the so-called “martian” meteorites are indeed rocks from Mars (Atreya et al. 2013)

# Isotopes point to atmospheric loss from Mars

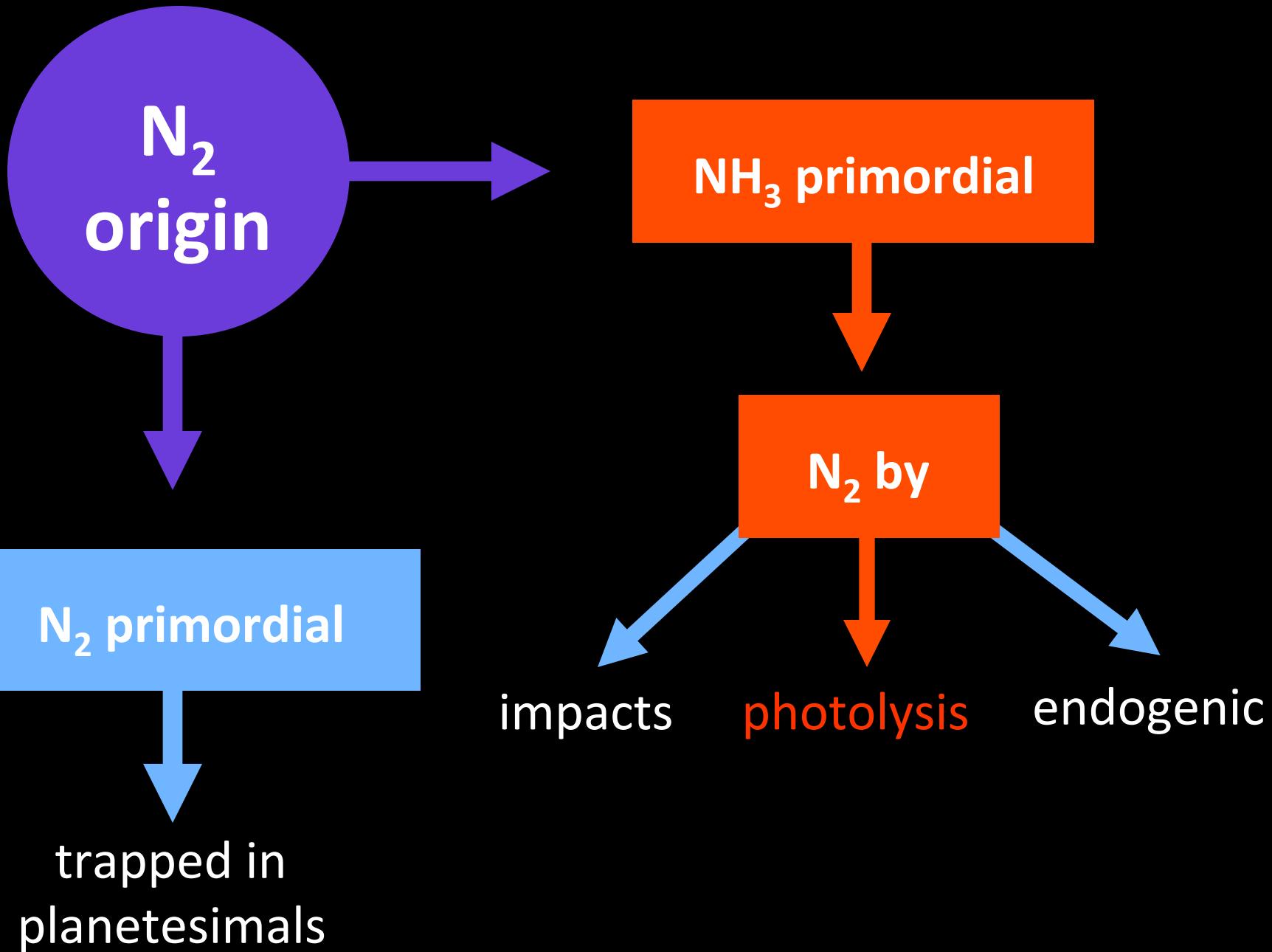
Isotopes	Mars value ‰	SAM suite instrument	Reference
$\delta^{38}\text{Ar}_{\text{Sun}}$	$310 \pm 31$	QMS	Atreya et al. (2013, <i>GRL</i> )
$\delta^{40}\text{Ar}_{\text{Earth}}$	$5419 \pm 1013$	QMS	Mahaffy et al. (2013, <i>Science</i> )
$\delta^{15}\text{N}_{\text{Earth}}$	$572 \pm 82$	QMS	Wong et al. (2013, <i>GRL</i> )
$\delta^{13}\text{C}_{\text{VPDB}}$	$45 \pm 12$	QMS	Mahaffy et al. (2013, <i>Science</i> )
$\delta^{13}\text{C}_{\text{VPDB}}$	$46 \pm 4$	TLS	Webster et al. (2013, <i>Science</i> )
$\delta^{18}\text{O}_{\text{SMOW}}$	$48 \pm 5$	TLS	Webster et al. (2013, <i>Science</i> )
$\delta\text{D}_{\text{SMOW}}$	$4950 \pm 1080$	TLS	Webster et al. (2013, <i>Science</i> )

Curiosity's SAM measures atmospheric isotope ratios

[ $\delta$  vs R: e.g.  $\delta^{13}\text{C}(\text{\%}) = 1000 (\text{R}/\text{R}_{\text{Std}} - 1)$ ;  $\text{R} = ^{13}\text{C}/^{12}\text{C}$ ]

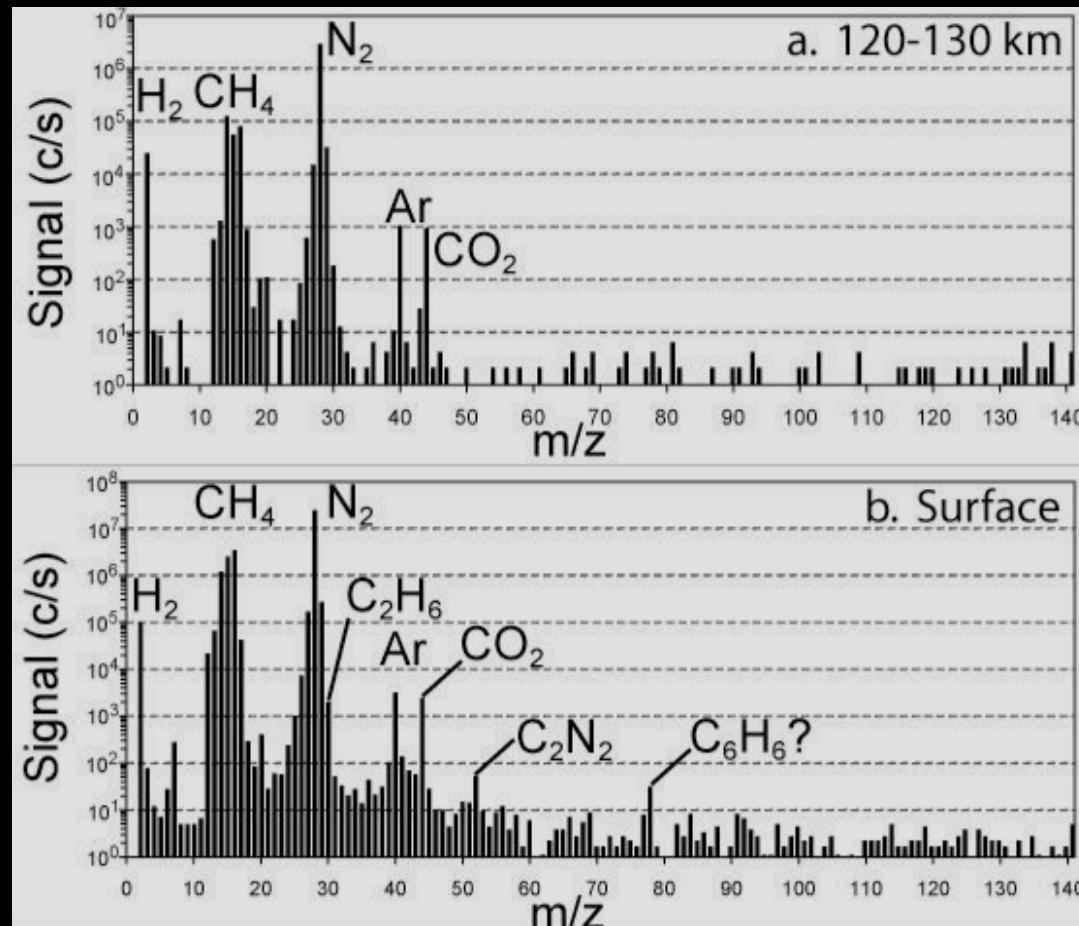
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# Nitrogen did *not* arrive as N<sub>2</sub> on Titan

<sup>36</sup>Ar/N<sub>2</sub> Titan:  $2.1 \times 10^{-7}$  solar: 0.11 Titan/Sun =  $2 \times 10^{-6}$

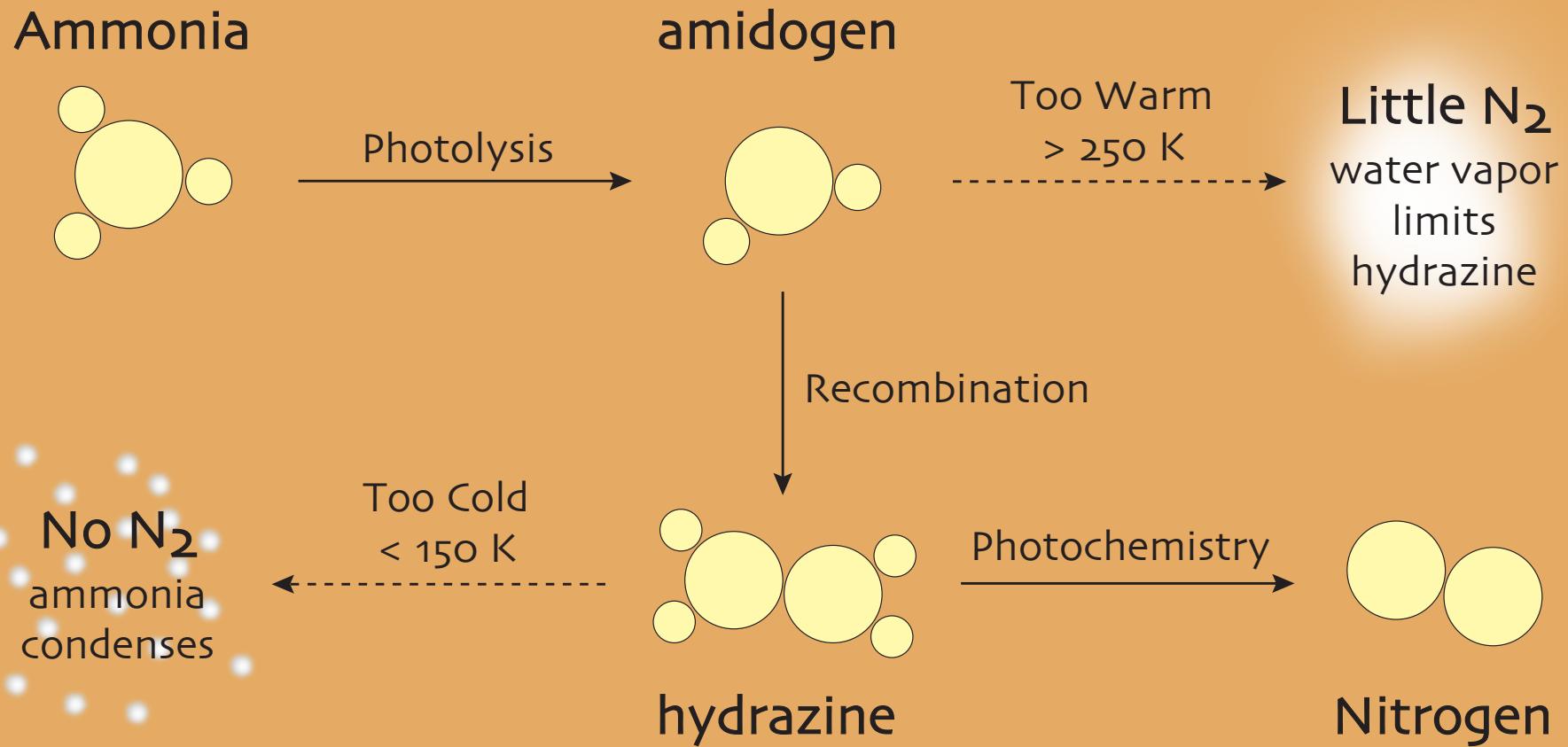


(Niemann et al. 2005, 2010 Huygens GCMS)

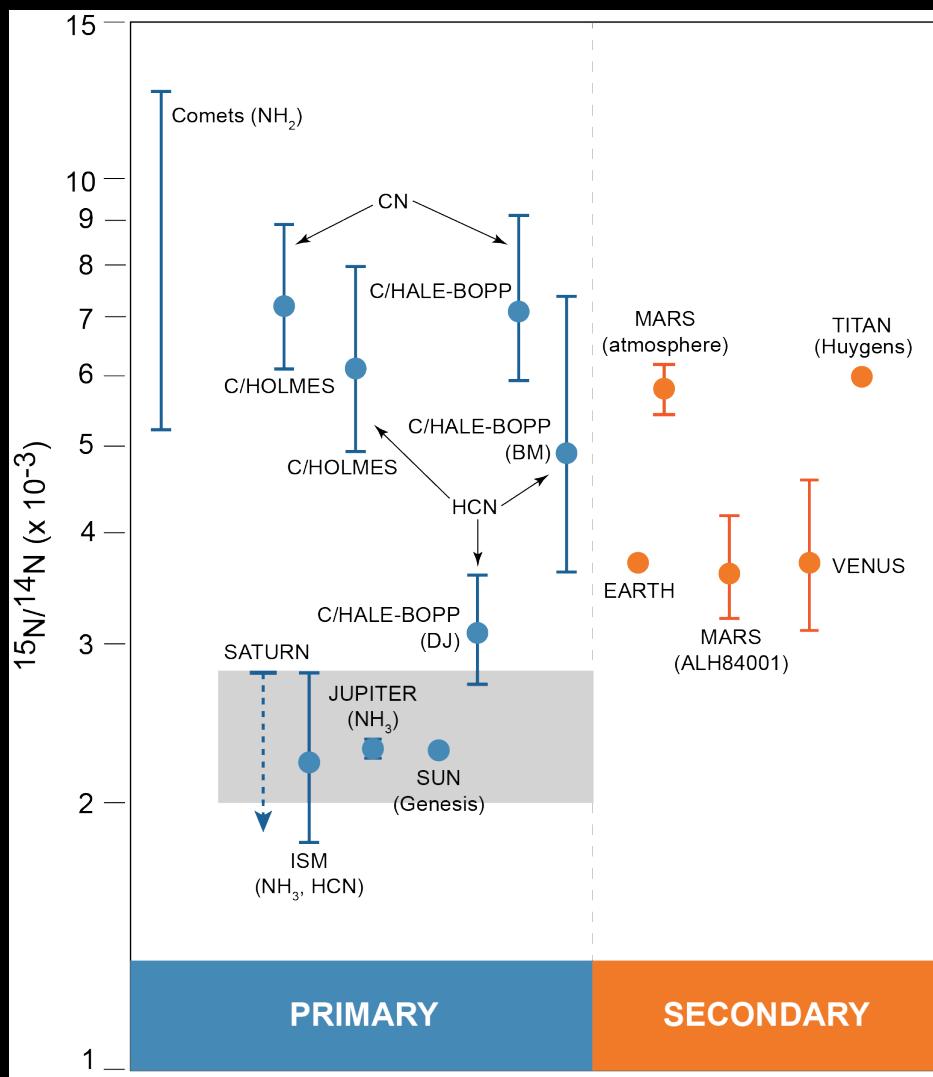
# $\text{N}_2$ from $\text{NH}_3$ on *primordial* Titan

## 2 bars in <5 Myr; 10 bars in <20 Myr

(Atreya et al. 1978, 2006; Adams and Atreya, 2005)



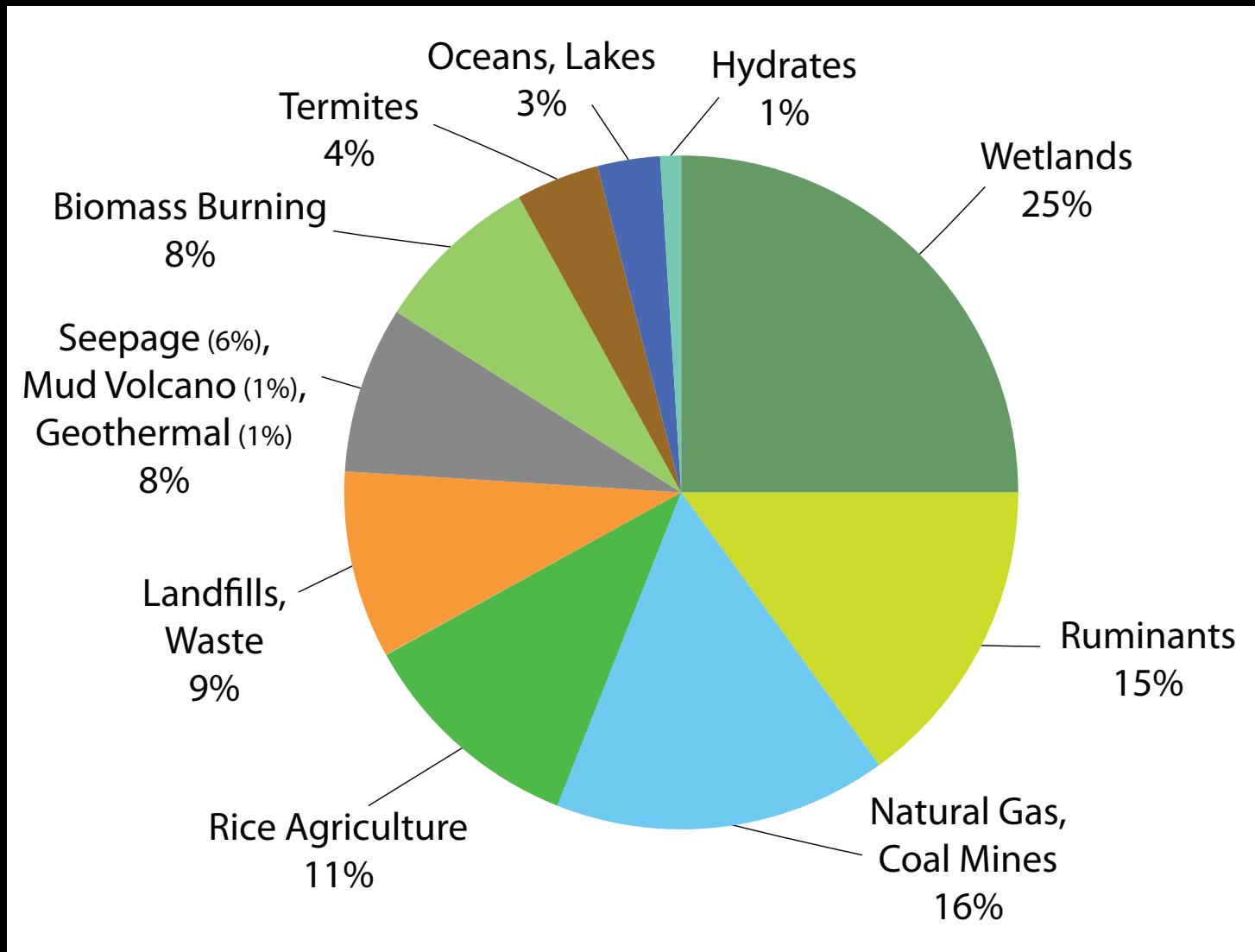
# Nitrogen isotopes: Titan and the rest



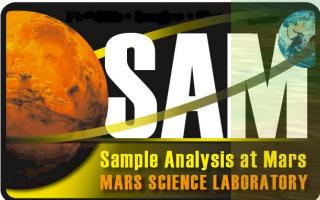
(Atreya et al. 2015)

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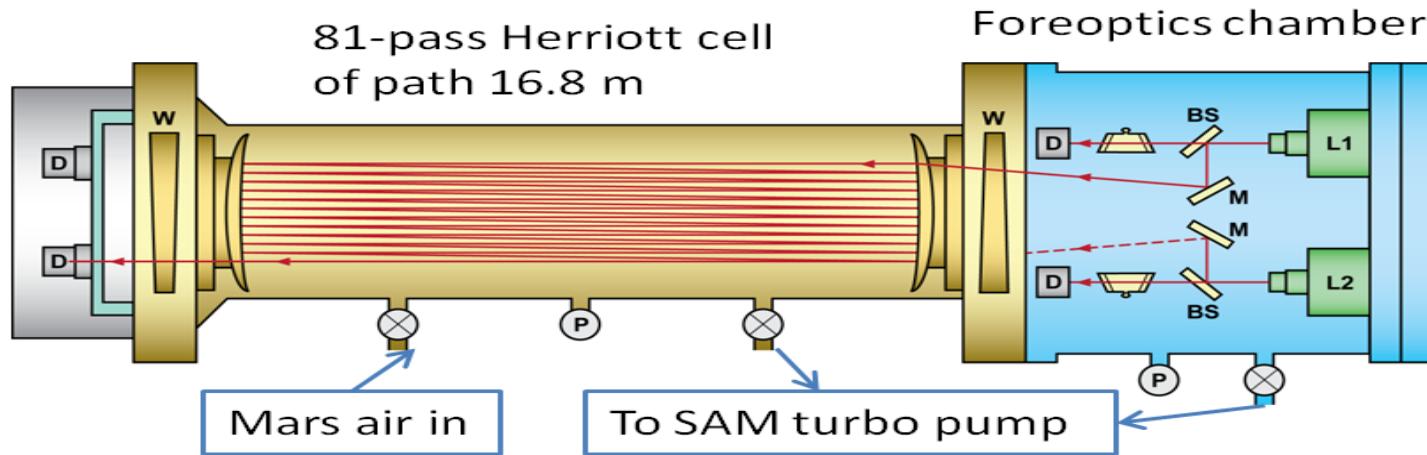
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**Life as we know it produces methane,  
90-95% of the 1775 ppbv on Earth**

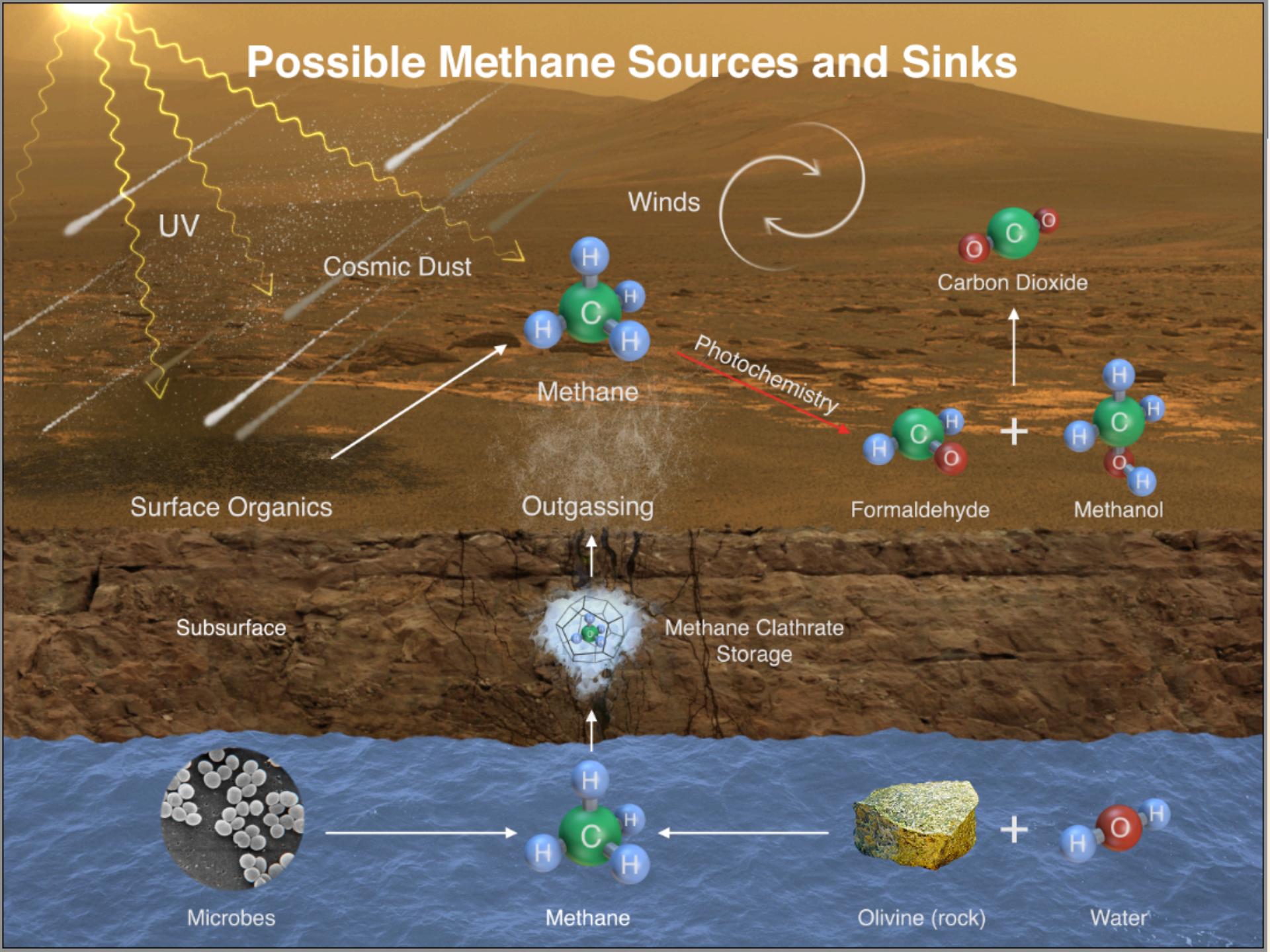


# Tunable Laser Spectrometer measured a constant methane background level of 0.7 ppbv, and a spike of 7 ppbv over two months



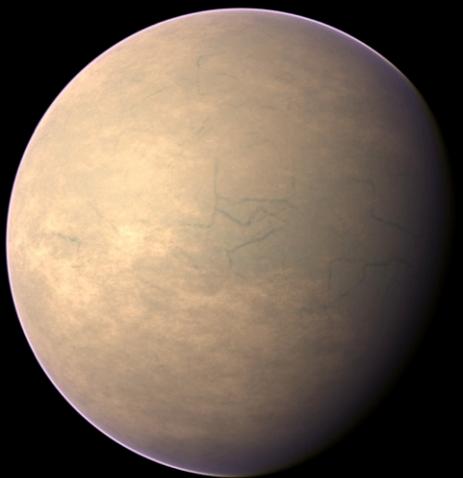
(Webster et al. Science 2015)

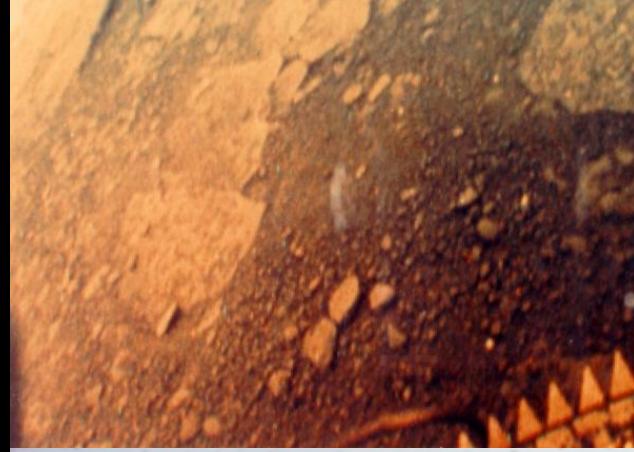
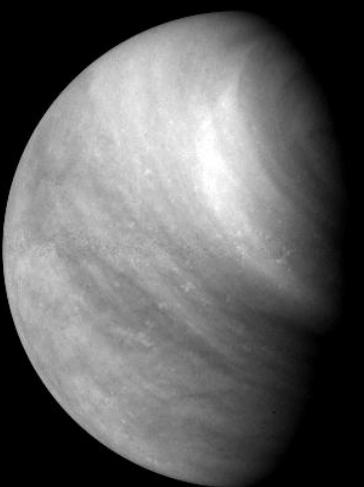
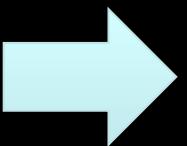
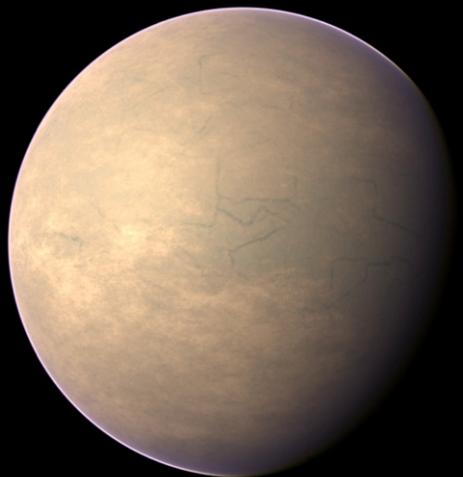
# Possible Methane Sources and Sinks



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(Credit: D. Grinspoon; Venera; Viking)

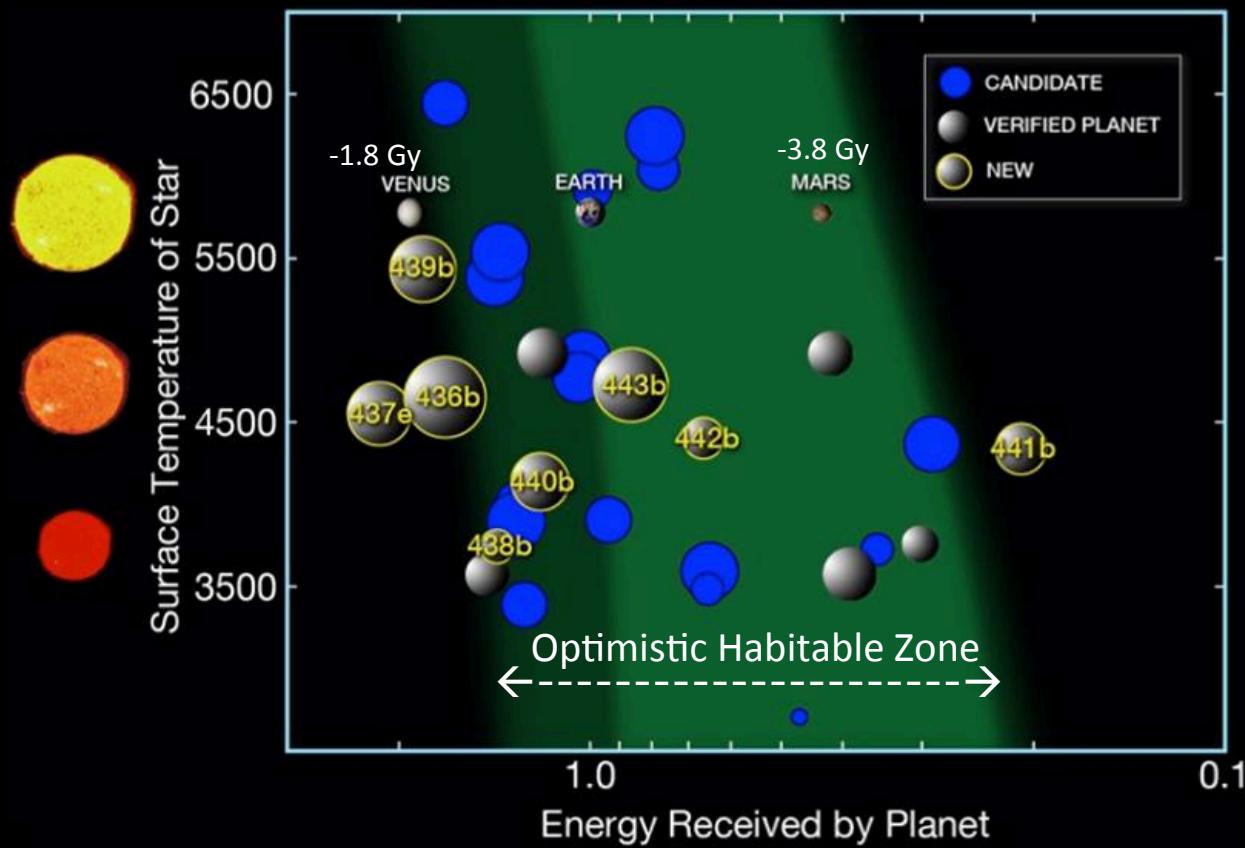
# Habitable Zone Planets



# Kepler's New Planets



*As of January 2015*



# Looking to the future: from Earth-size to Earth-like, i.e. Habitable Planets

- Life as we know it
  - Is carbon based (RNA, DNA, proteins), and requires
  - Liquid water, for biochem rxns and nutrient transport
  - Nutrients (CHNOPS)
  - Energy (stellar/chemical)
- SLWHZ (surface liquid water habitable zone) is excellent start, and identification of
- $\text{H}_2\text{O}$ ,  $\text{CH}_4$ ,  $\text{O}_2$ ,  $\text{O}_3$ ,  $\text{CO}_2$ ,  $\text{SO}_2$ ,  $\text{C}_n\text{H}_m$ ,  $\text{HCl}$ ,  $^{13}\text{C}/^{12}\text{C}$  and the aerosols etc. in the atmospheres of earth-size exoplanets could give clues to their habitability