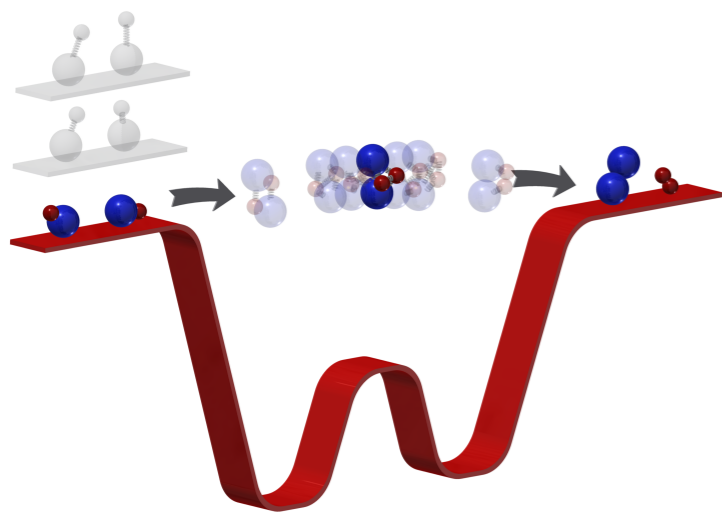


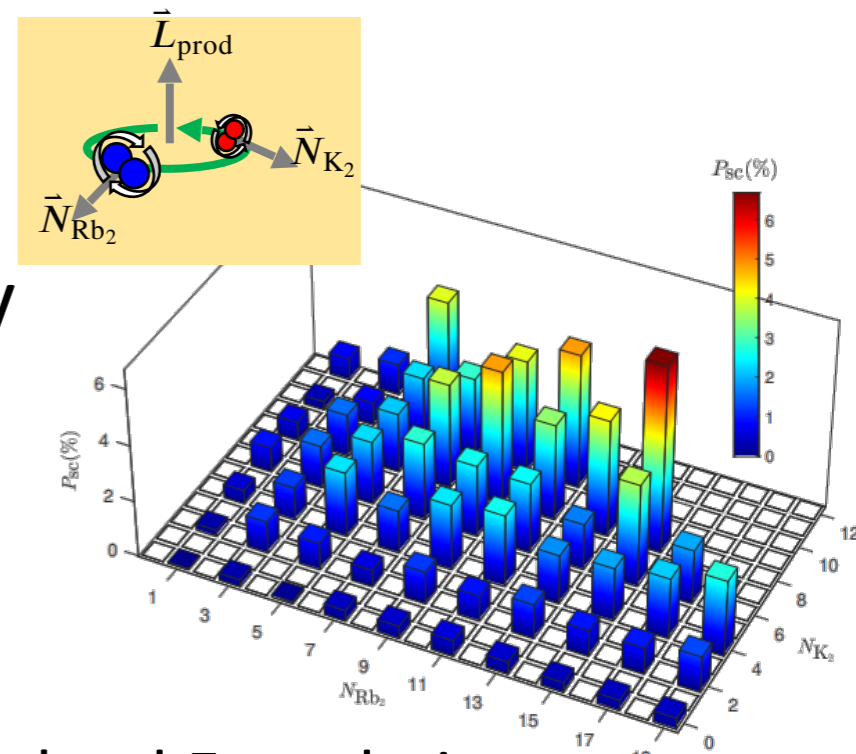


Precision test of statistical dynamics in molecular collisions and reactions



Kang-Kuen Ni

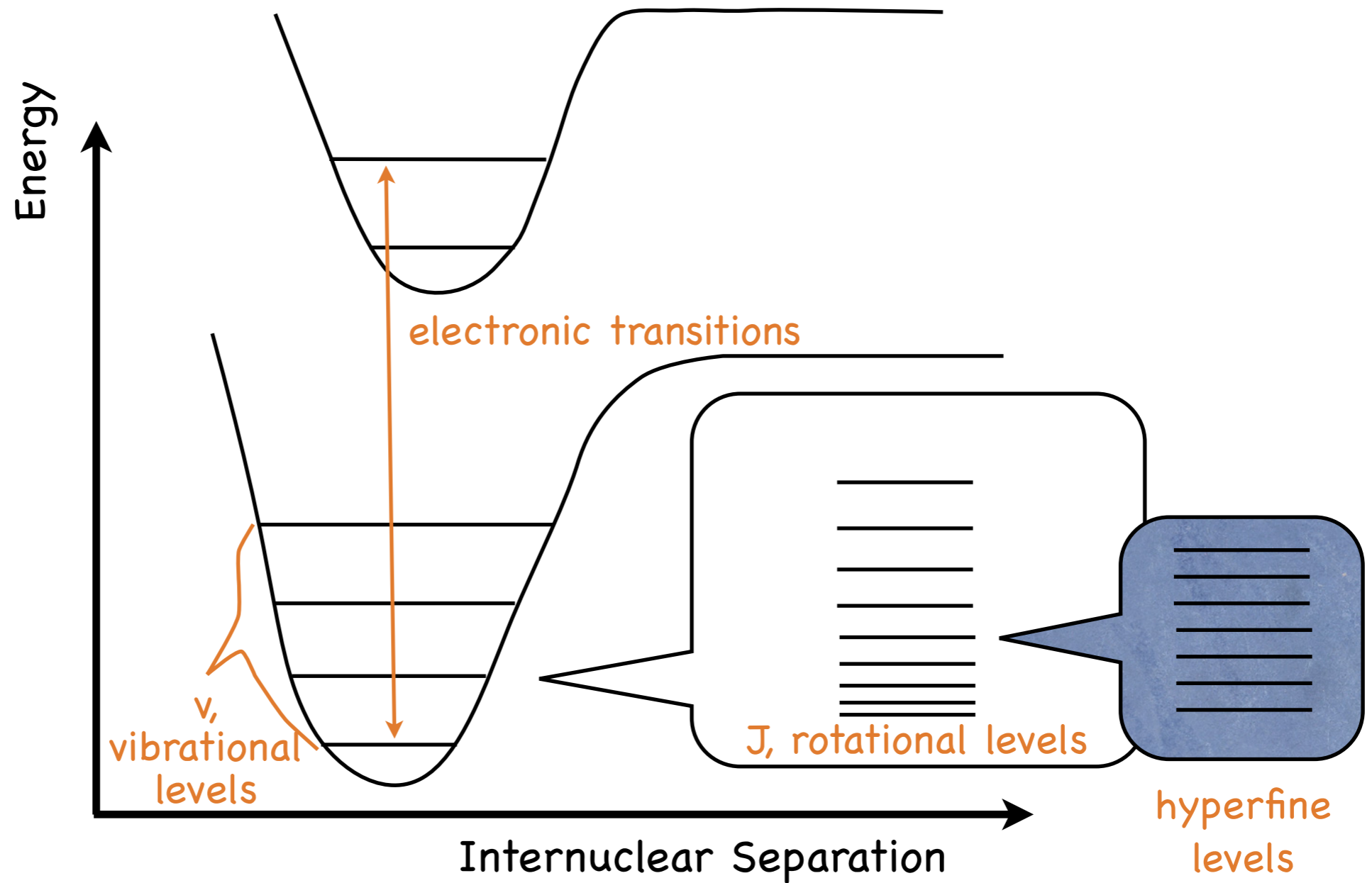
Department of Chemistry
and Chemical Biology
Harvard University



Funding: DOE-YIP, NSF-CUA, David and Lucile Packard Foundation

KITP Few-body Physics Conference, May 23 -26, 2022

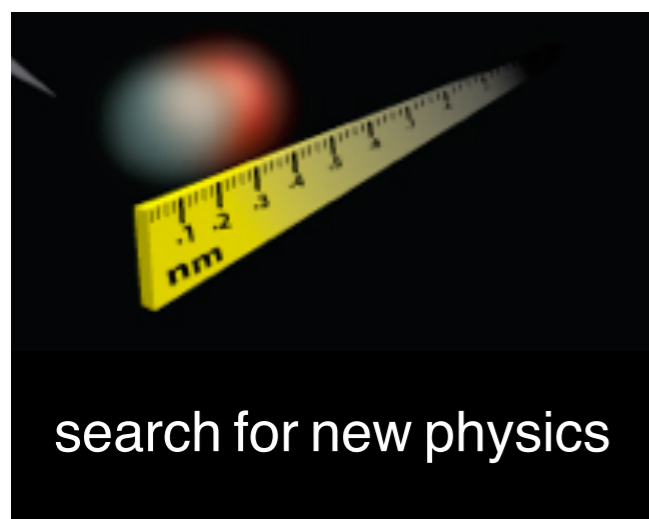
Molecular Quantum Degrees of Freedom



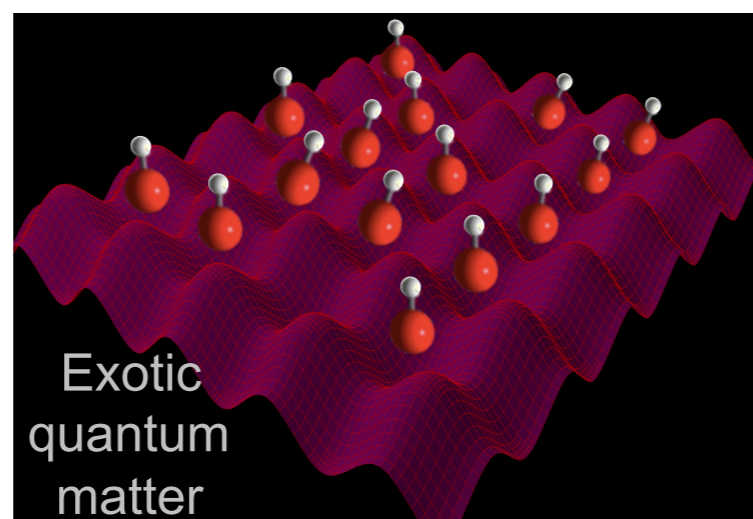
Quantum Control of Molecules

- * cooling and trapping
- * long interaction and probe time
- * building complex system from the bottom up

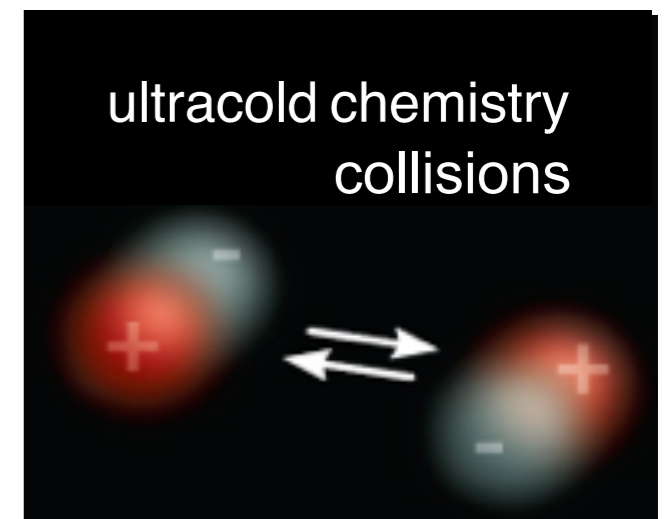
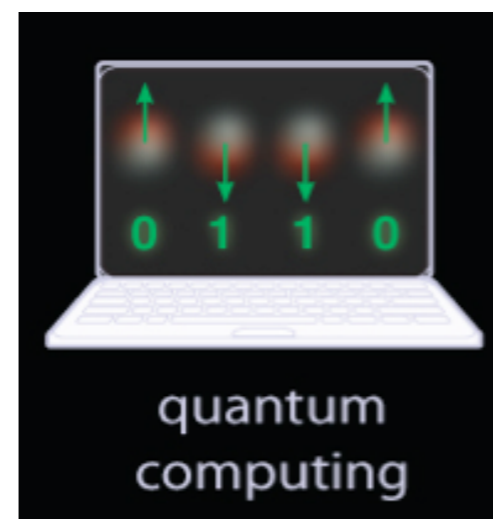
Why Ultracold Molecules?



single-body physics



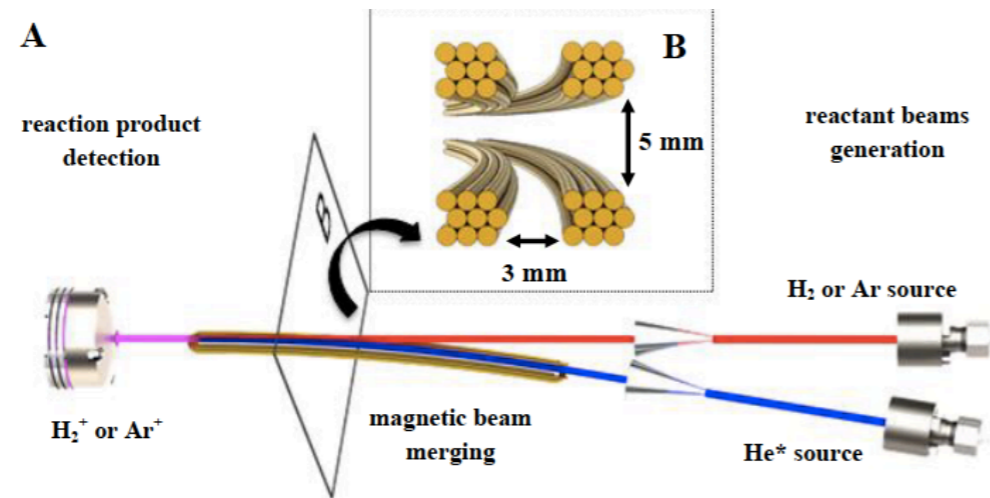
two- to many-body physics



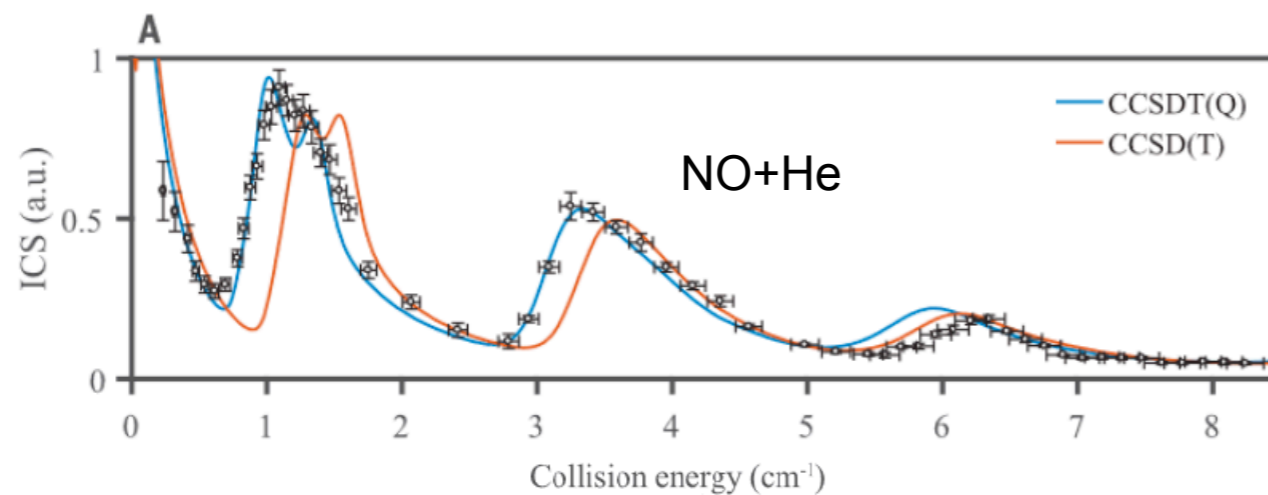
two- to few-body physics

Chemistry and Collisions with Cold Molecules

probing potential energy surfaces beyond
“gold-standard” quantum chemistry calculation

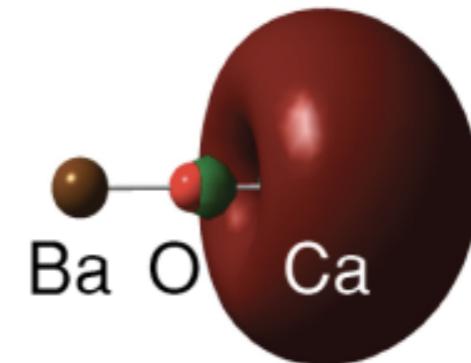


Narevicius (Wiezmann)
Nature Phys. 13, 35 (2017)



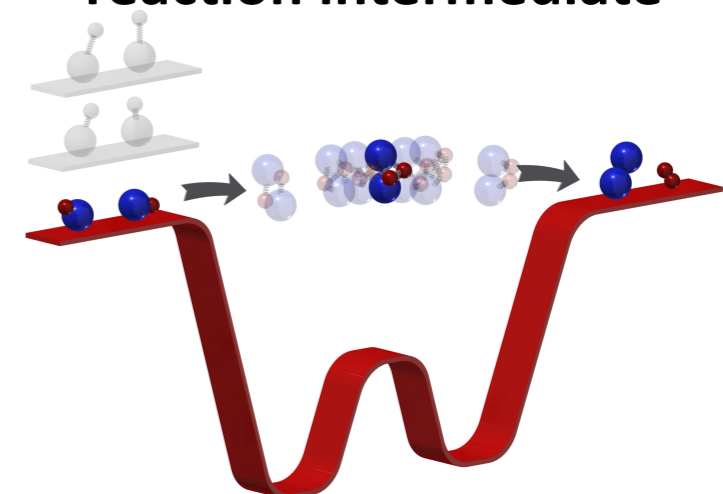
van de Meerakker (Radboud)
Science 368, 626 (2020)

synthesizing new
chemical species



Hudson (UCLA)
Science 357, 1370 (2017)

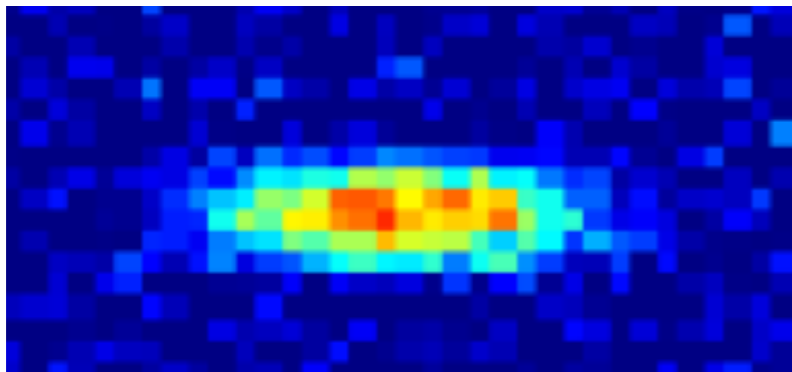
Direct detection of
reaction intermediate



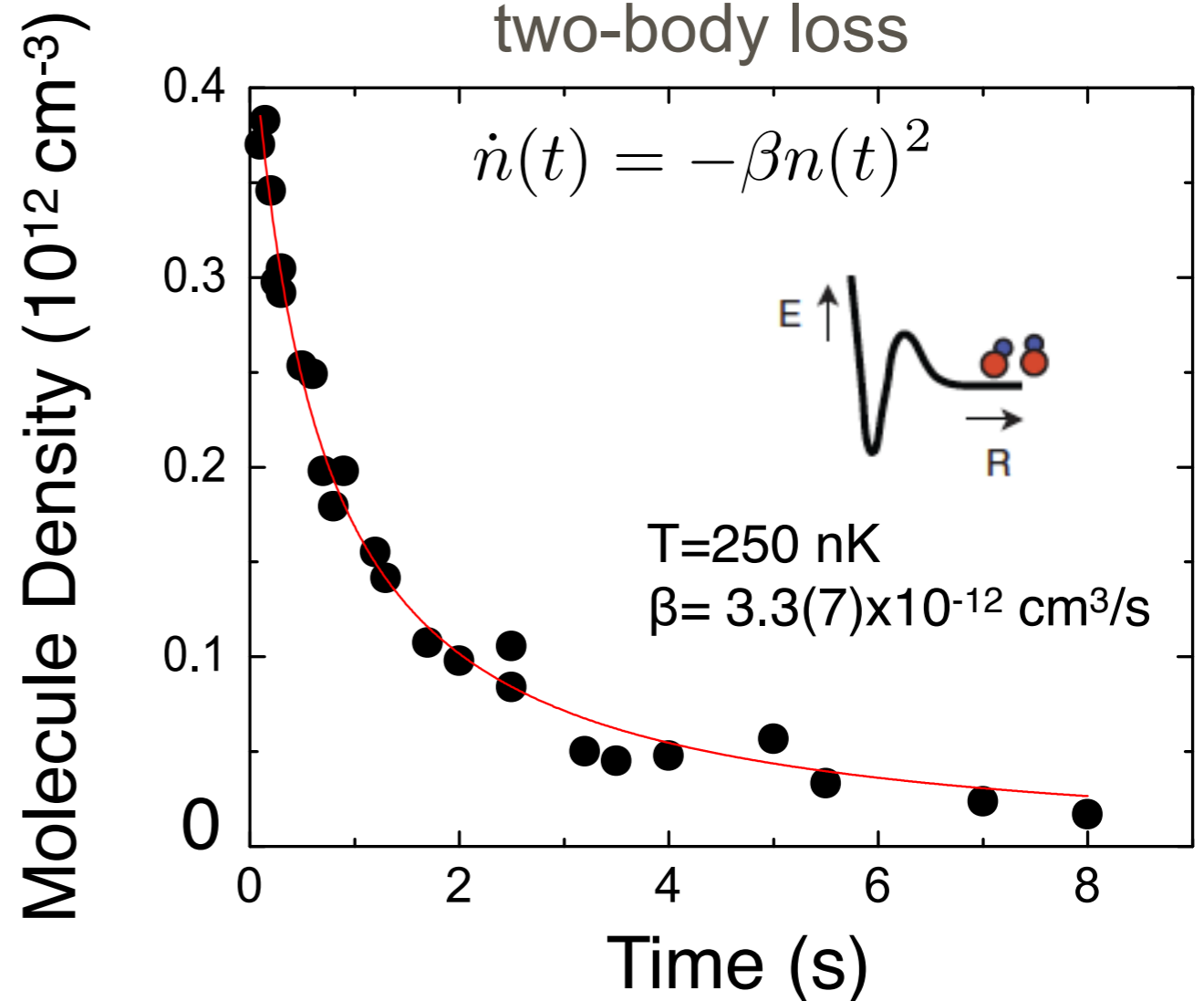
Hu, Liu, et al. (Harvard),
Science 366, 1111 (2019)

JILA KRb Experiment (2008-2010)

A trapped *gas* of molecules in a single and lowest hyperfine, rotational, vibrational, electronic ground state!



peak density = $10^{12}/\text{cm}^3$
temperature = 200 nK



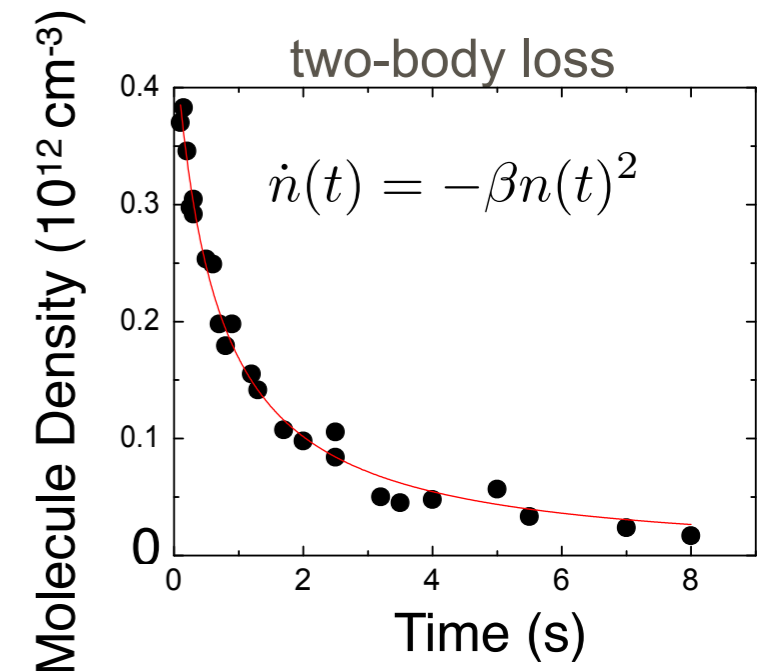
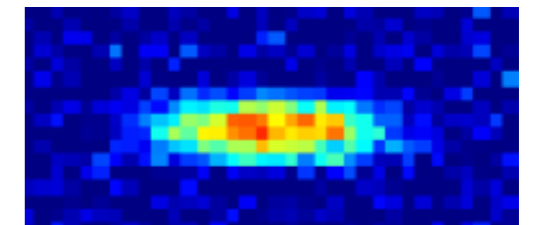
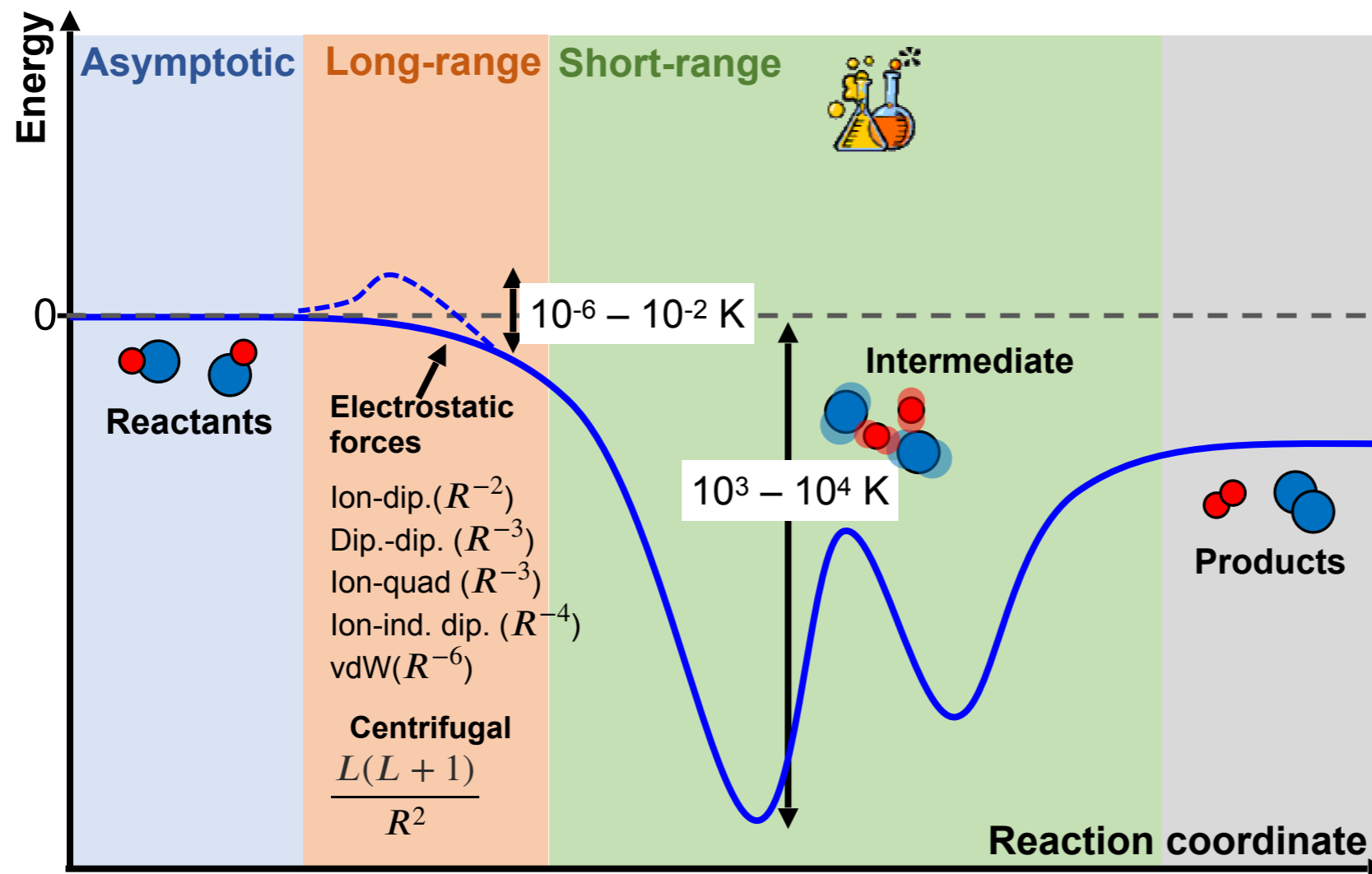
Ni, Ospelkaus et al., Science **322**, 231 (2008)

Wang *et al.*, PRA **81**, 061404 (2010)

Ospelkaus, Ni et al., Science **327**, 853 (2010)

Innsbruck, Durham, MIT, Hong Kong, USTC, MPQ, Hannover, ...

Ultracold Chemical Reaction?



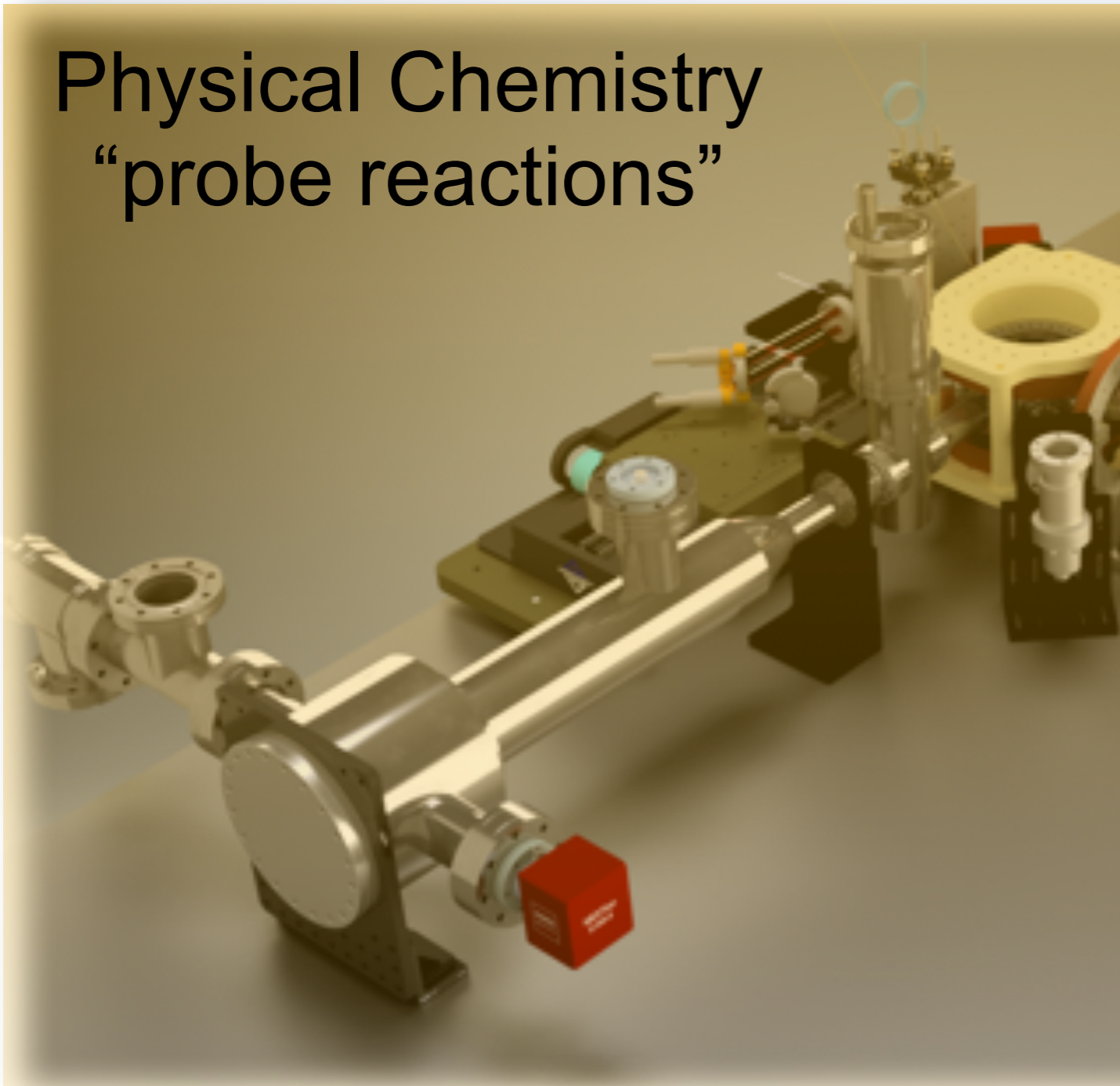
Direct detection needed

Idziaszek and Julienne, PRL 104, 113202 (2010)
 Quéméner and Bohn, PRA 81 022702 (2010)
 Ospelkaus, Ni, ... Jin, Ye, Science, 327, 853 (2010)
 Ni, Ospelkaus, ... Ye, Jin, Nature 464, 1324 (2010)

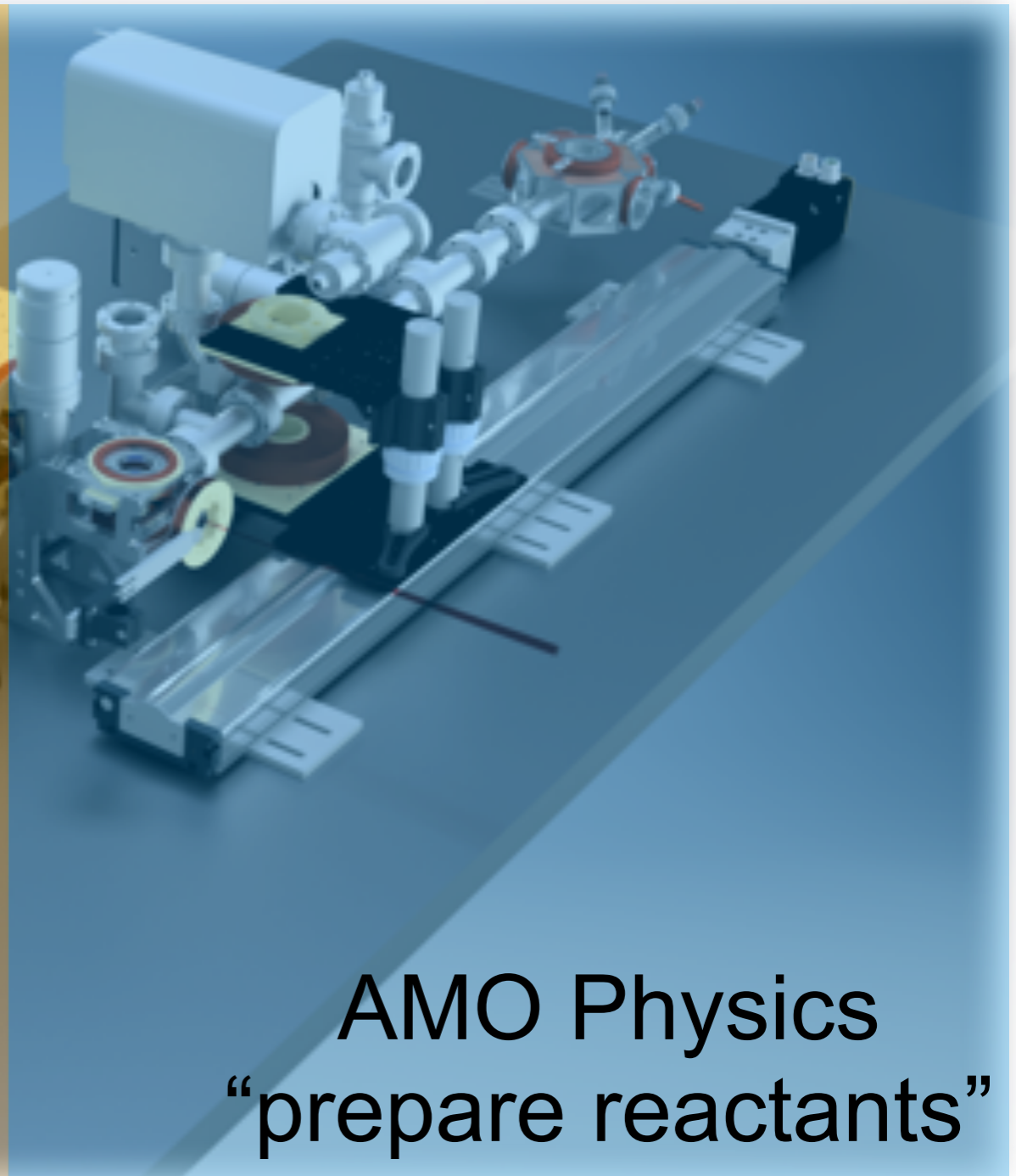
Guo et al., PRX 8, 041044 (2018)
 Ye et al., Sci. Adv. (2018)
 Gregory et al., Nat. Commun. (2019)
 Nesbitt, Chem. Rev. 122, 9, 5062 (2012)

Combining Chemistry and Physics tools

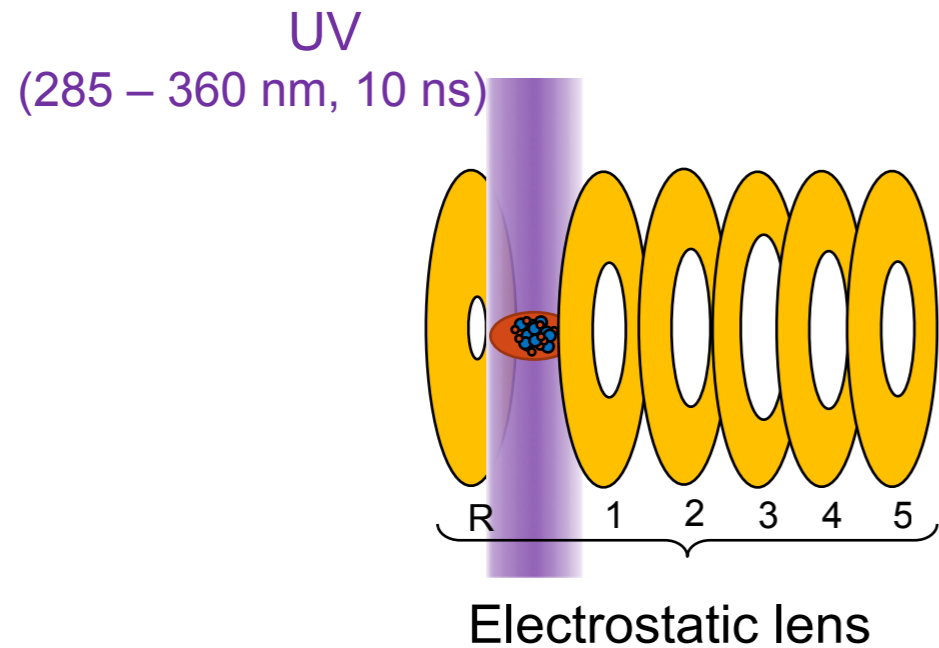
Physical Chemistry
“probe reactions”



AMO Physics
“prepare reactants”

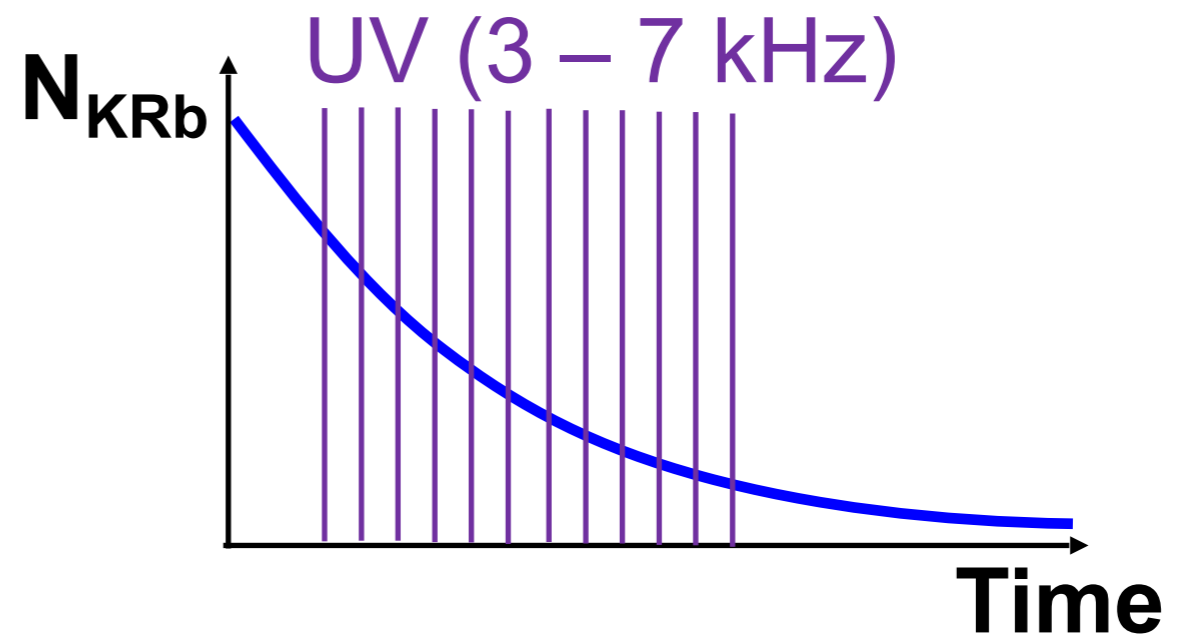
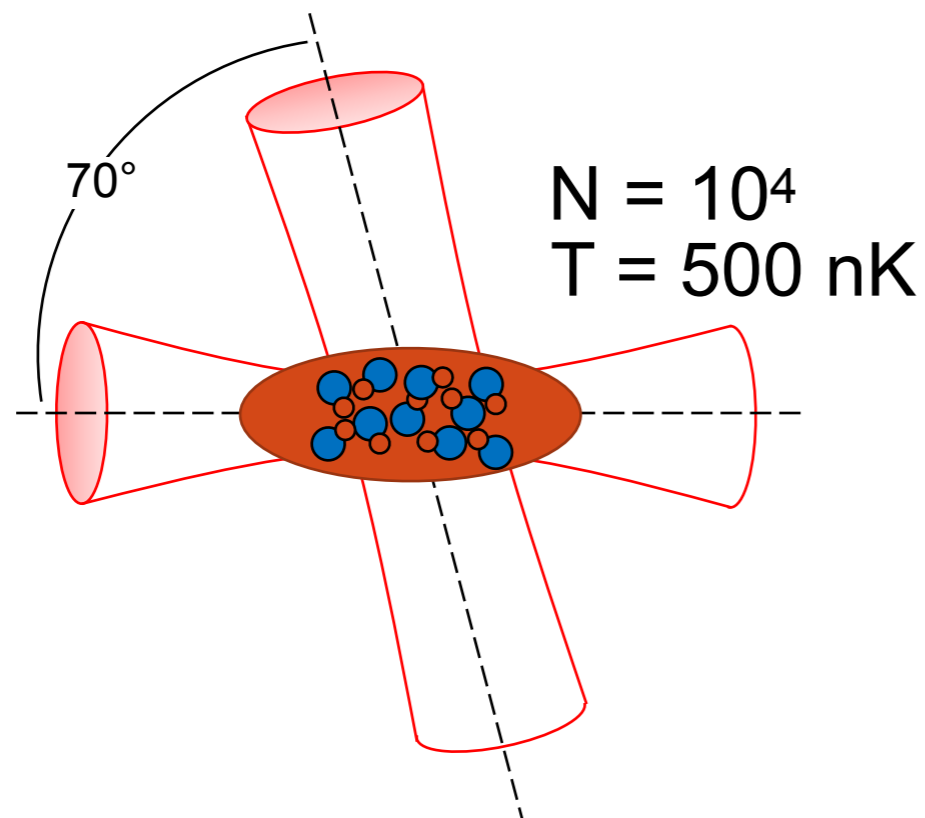
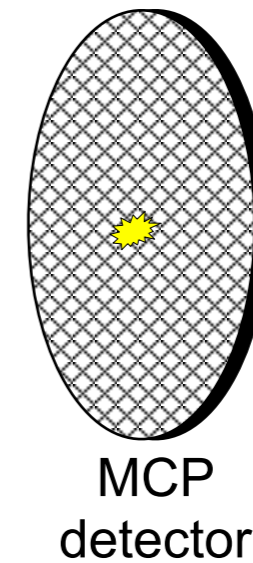


Probing the reaction via ion spectrometry

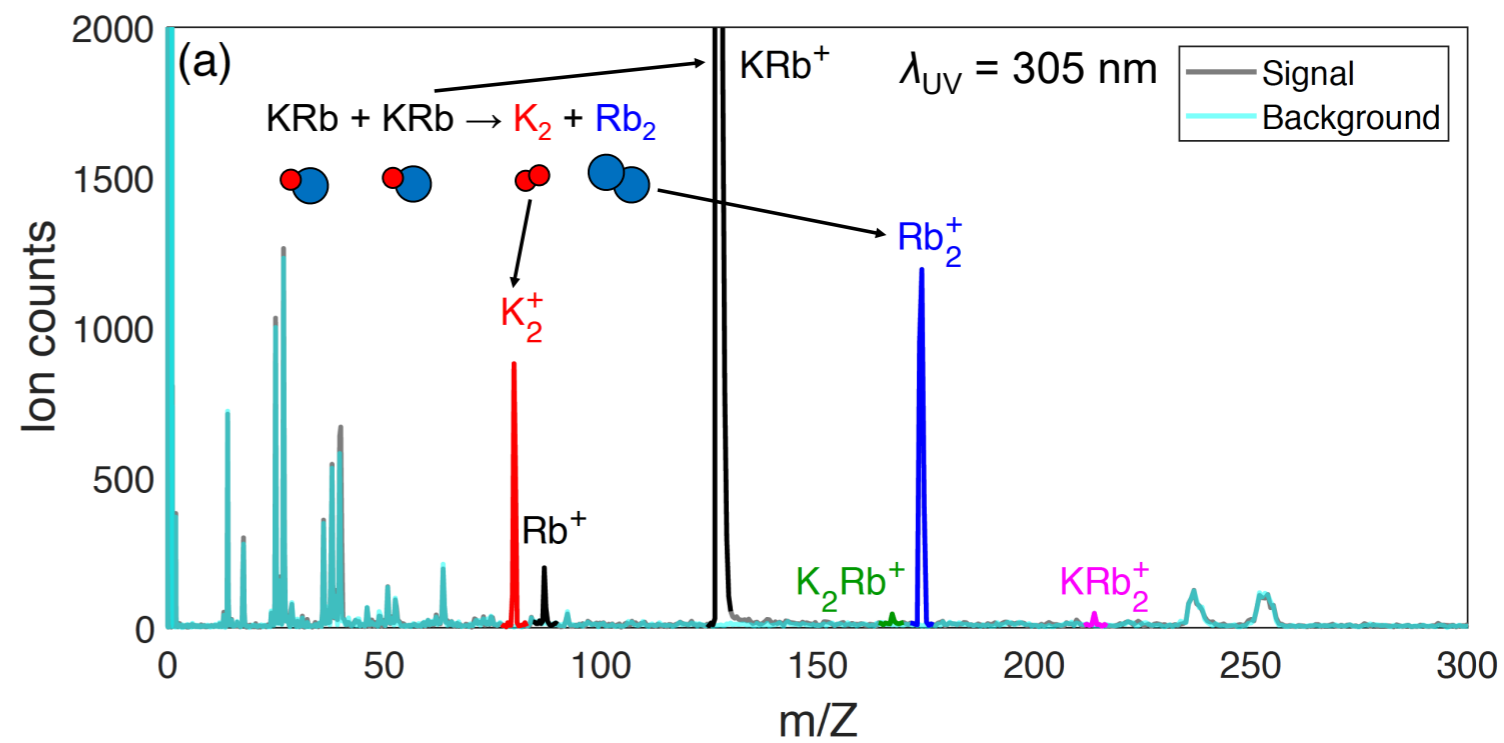
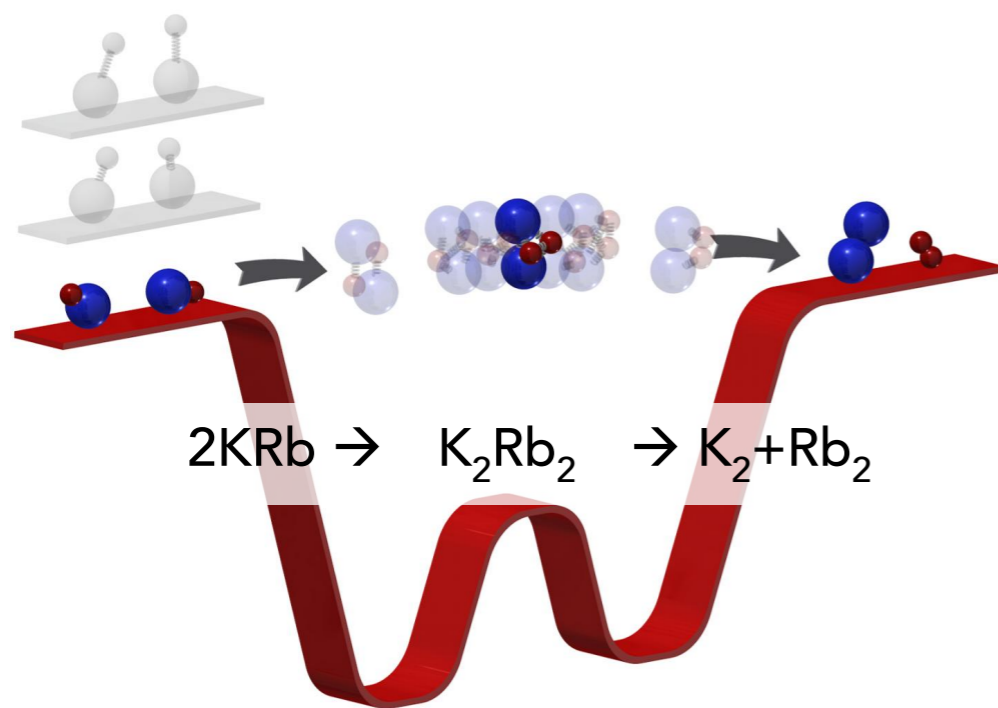
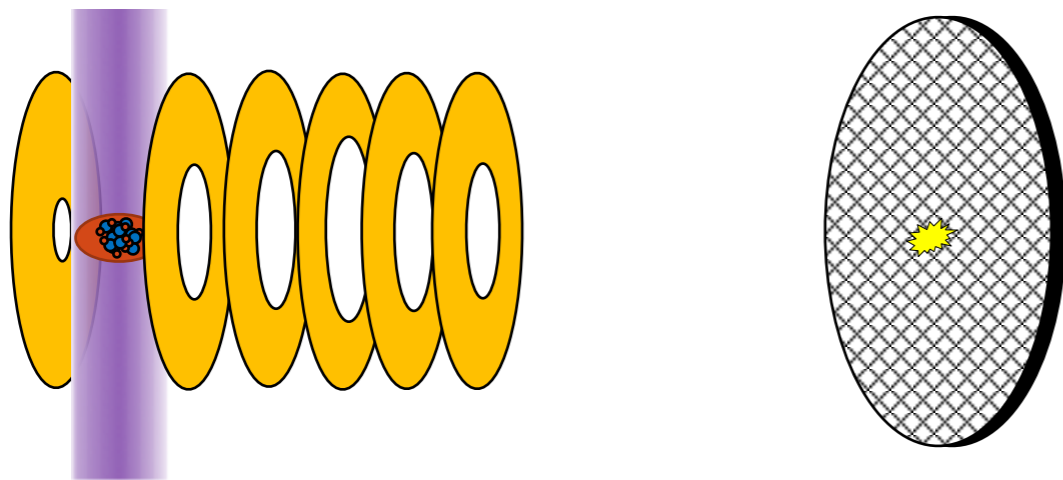


Time-of-flight
Mass/species

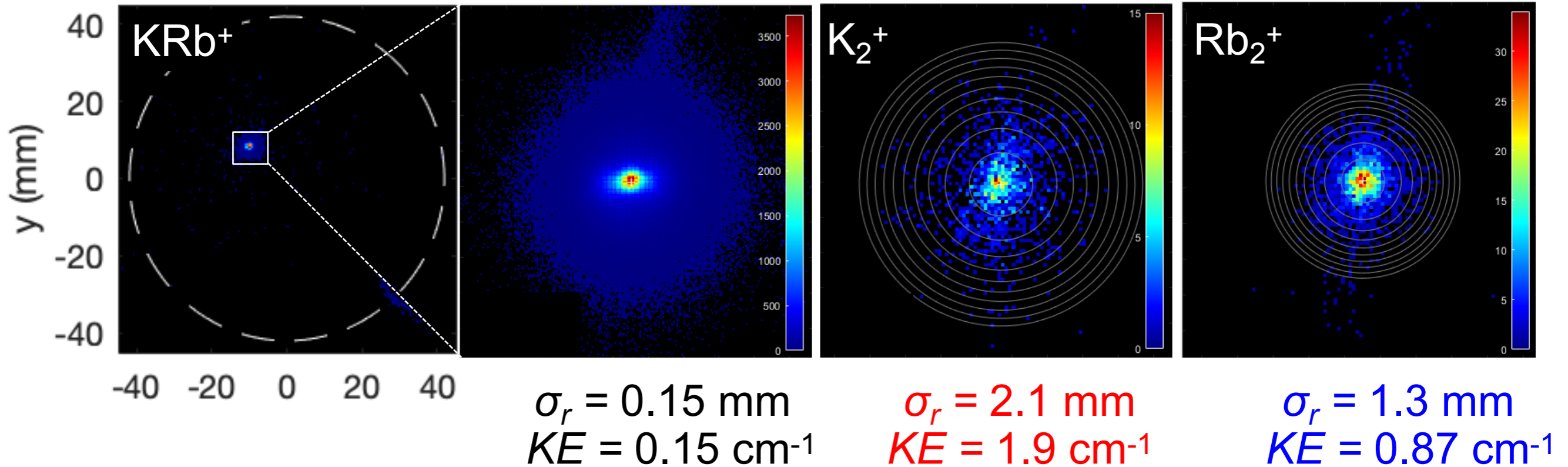
Spatial distribution
Kinetic energy



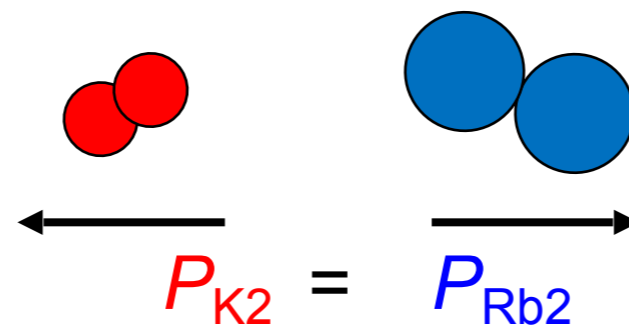
Product Detection via mass spectrometry



Verifying the product signal: KE distribution



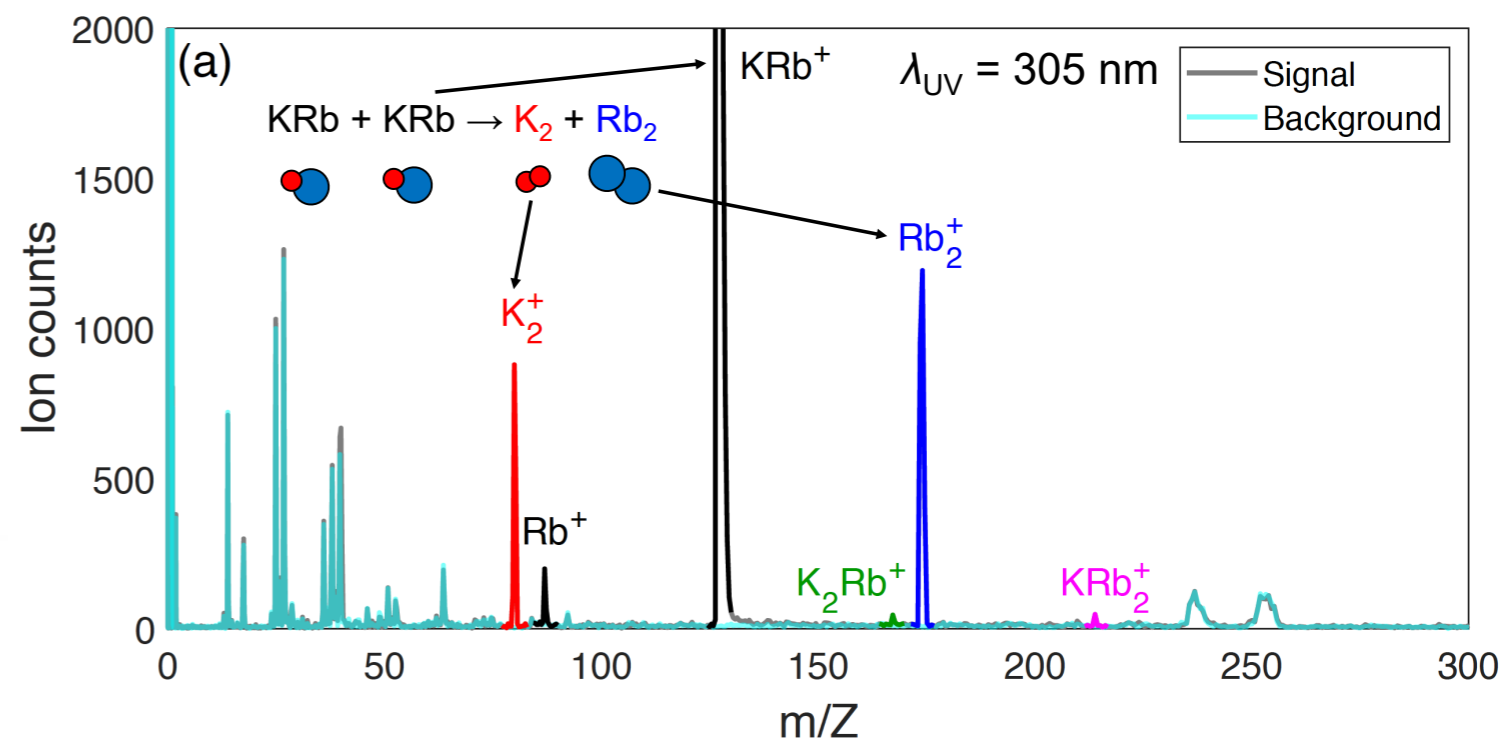
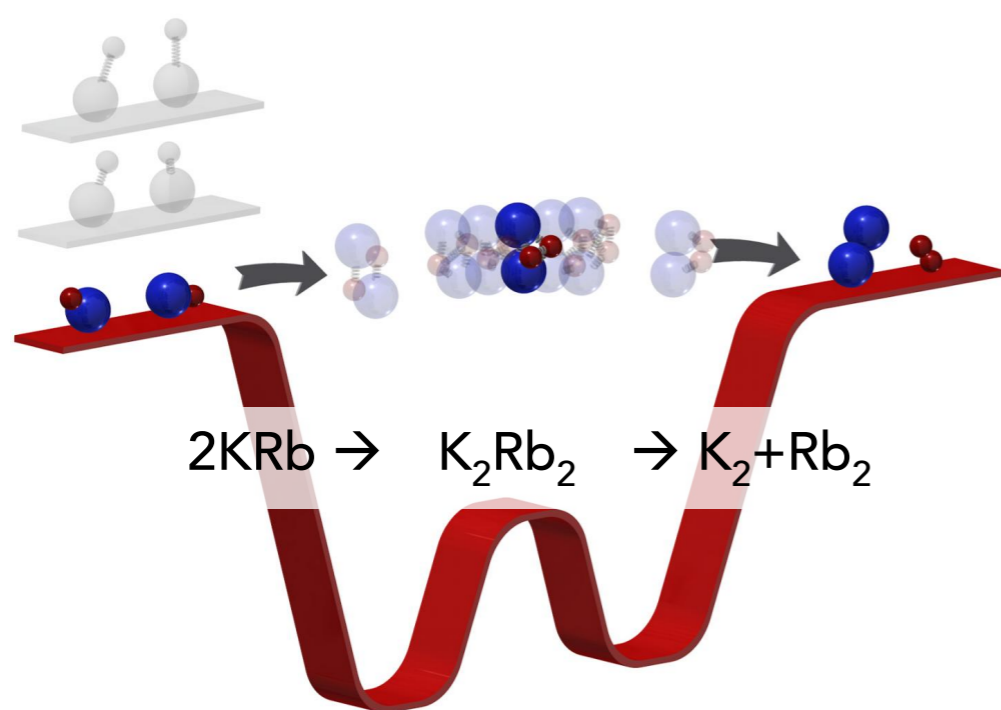
$$\frac{0.87 \text{ cm}^{-1}}{1.9 \text{ cm}^{-1}} = 0.46$$



$$\frac{KE_{Rb_2}}{KE_{K_2}} = \frac{m_{K_2}}{m_{Rb_2}} = 0.46$$

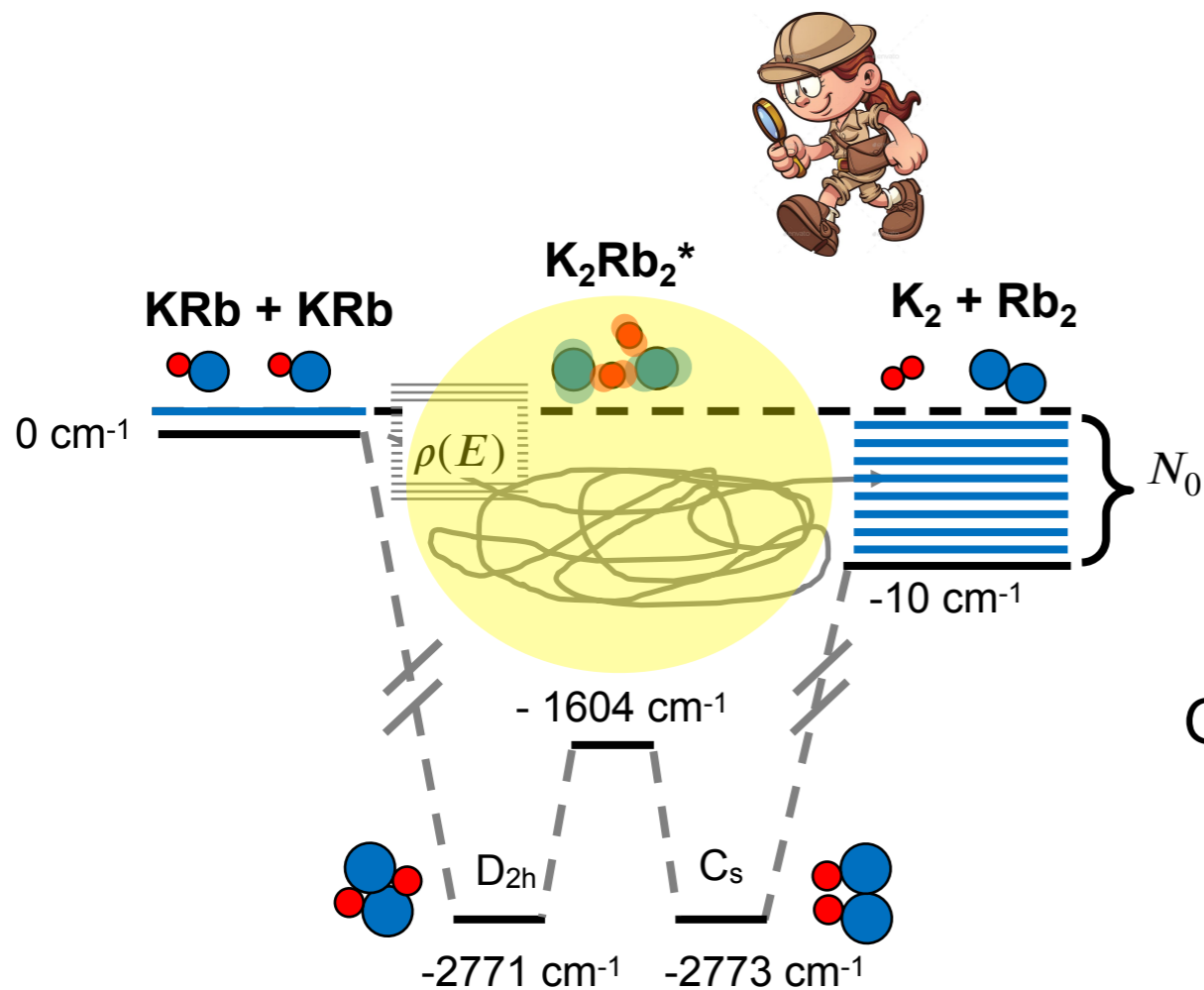
Surprise #1: ultracold chemical reaction!

Surprise #2: Reaction Complex Detection



The long-lived intermediate complex

>> ps



Rice-Ramsperger-Kassel-Marcus
RRKM theory

$$\tau_c = \frac{h \rho(E)}{N_0}$$

theory estimated lifetimes

Bohn et al.(2013): $\tau_c = 3 \mu\text{s}$

Groenenboom et al. (2019): $\tau_c \sim 0.1 \mu\text{s}$

experiment estimated lifetime

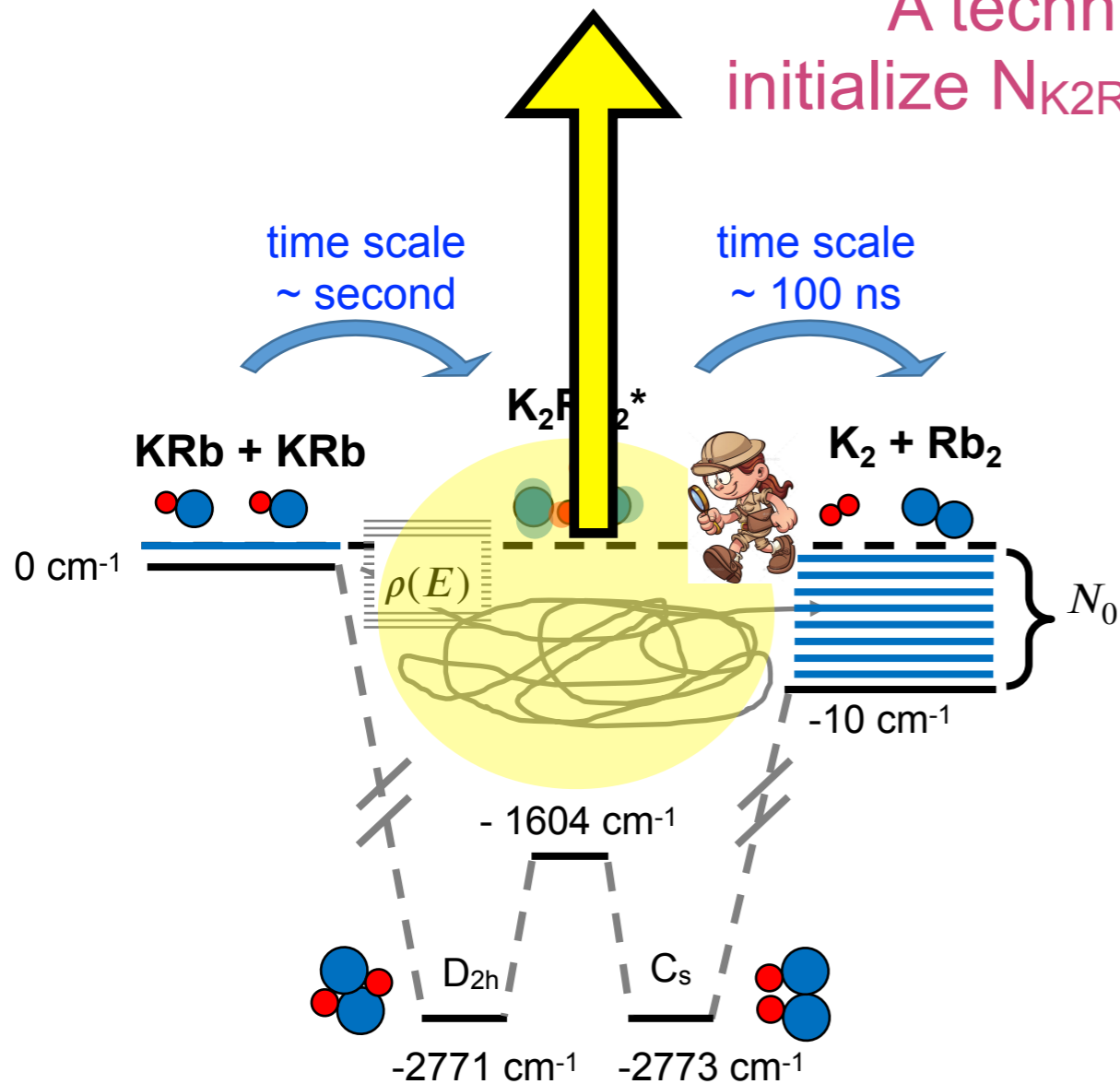
$$\tau_c \sim 0.35 - 3 \mu\text{s}$$

$$\sigma(\text{K}_2\text{Rb}_2^* \rightarrow \text{K}_2\text{Rb}_2^+) \sim 10^{-17} - 10^{-18} \text{ cm}^2$$

The long-lived intermediate complex

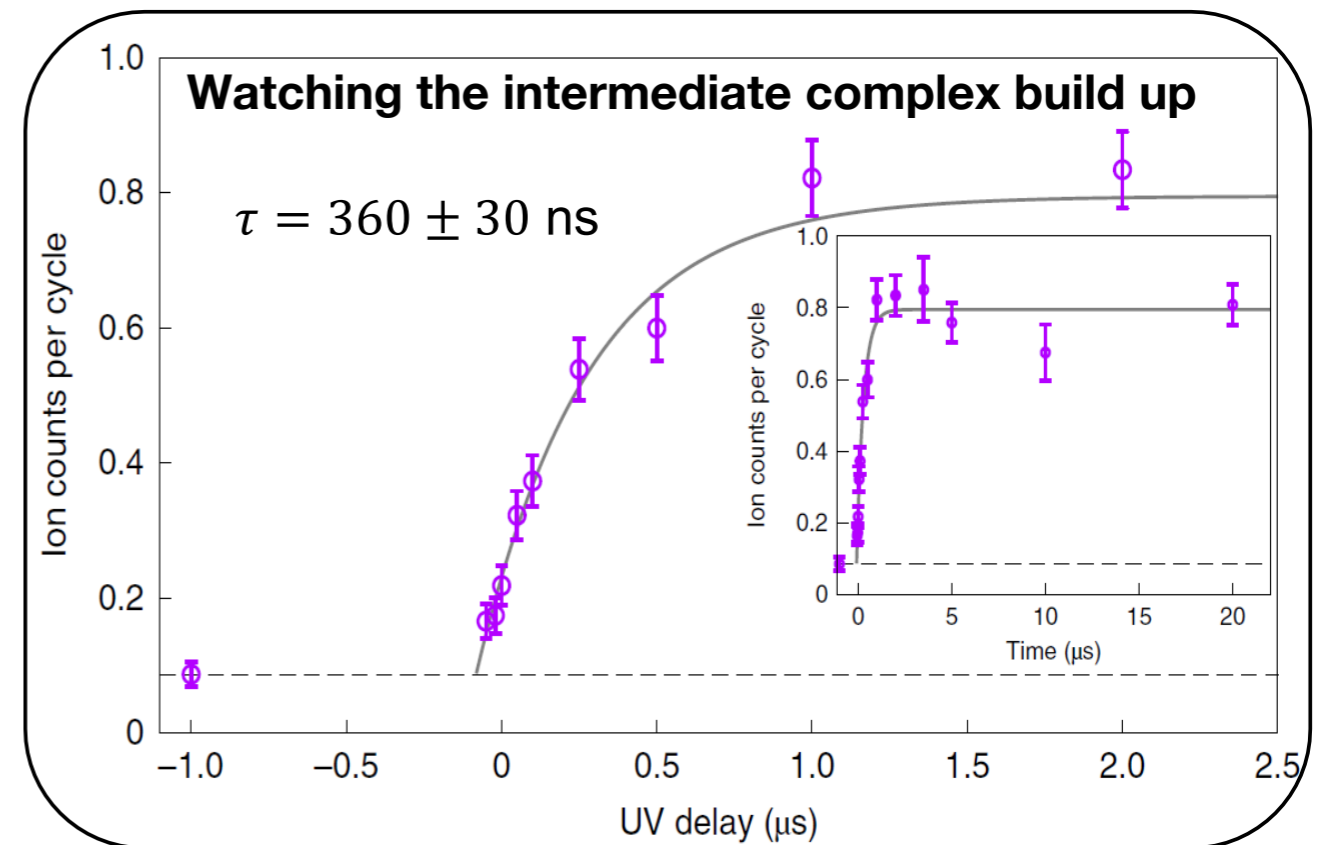
A technical challenge:
initialize $N_{K_2Rb_2}$ at $t=0$ to <100 ns

\gg ps



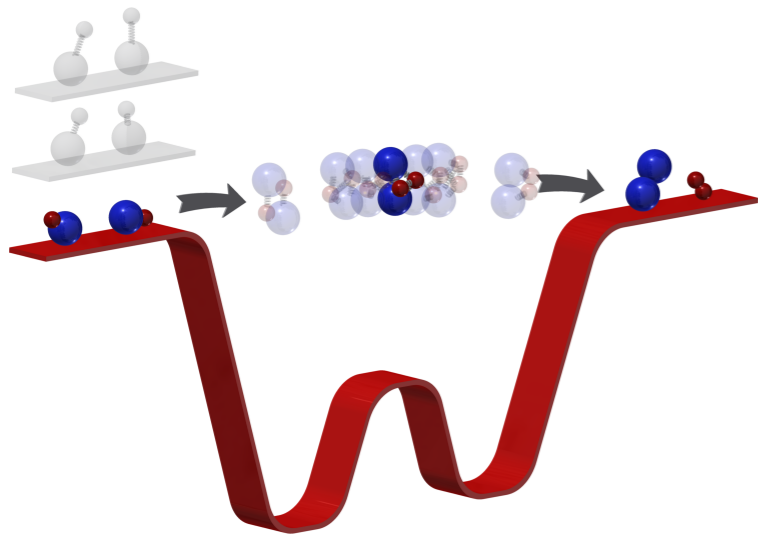
RRKM (from Tijs Karman)

$$\tau_c^{RRKM} = 170(60) \text{ ns}$$



Liu*, Hu* et al., Nature Physics **16**, 1132 (2020)

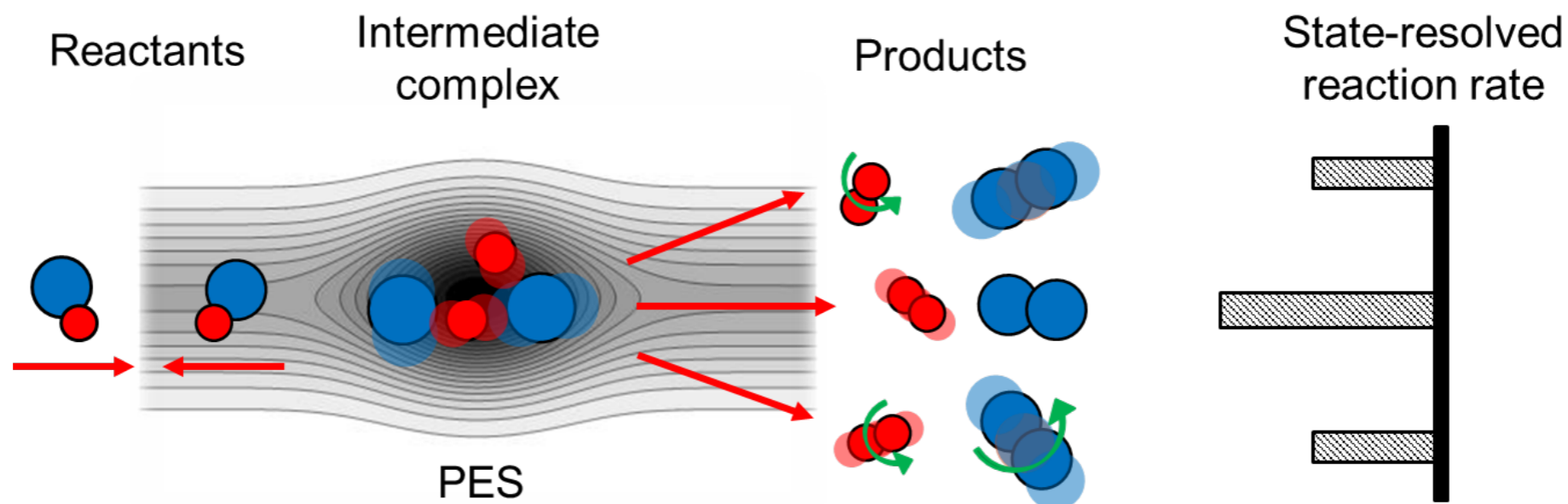
Related work for RbCs + RbCs (Durham), PRL 124, 163402 (2020)



Outline

- * Total quantum control of molecules from association of atoms
- * Surprise 1 - chemical reactions at ultralow temperatures
- * Surprise 2 - reactions play out in “slow motion”
- * Surprise 3 - steering reactions with light
- * Complete characterization of reactive process - mapping quantum states of correlated reaction product pairs
- * Surprise 4 - control reaction product parity state
- * More surprises - long-lived atom-molecule complexes

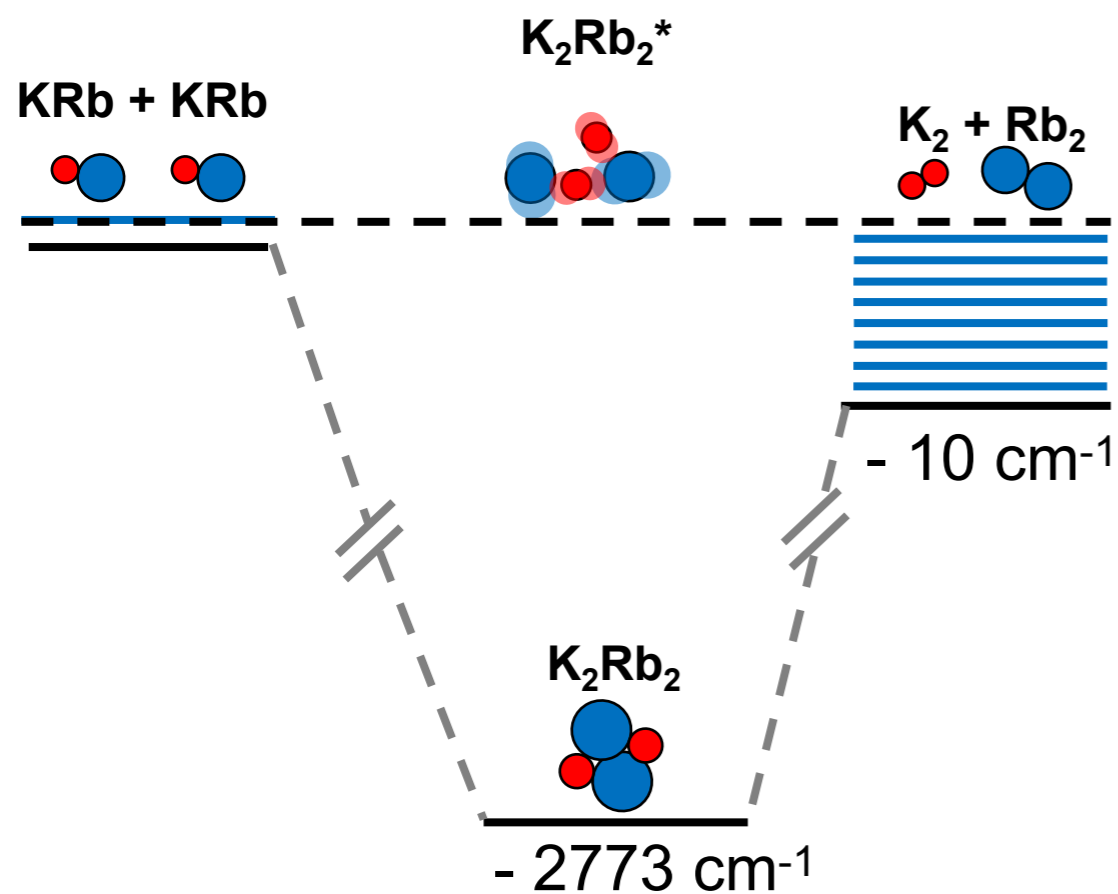
Chemical Reaction: a scattering problem



State-of-the-art theory cannot exactly predict the quantum state distribution for reactions involving 4 heavy atoms

but the observation of a long-lived complex suggests a statistical product state distribution

Possible quantum states of the products



No vibrational excitation

$$(\nu_{\text{K}_2} = \nu_{\text{Rb}_2} = 0)$$

Rotational excitation

$$N_{\text{K}_2}^{\text{max}} = 12 \quad \text{---} \quad N_{\text{Rb}_2} = 7$$

$$N_{\text{K}_2} = 2 \quad \text{---} \quad N_{\text{Rb}_2}^{\text{max}} = 19$$

⋮

$$N_{\text{K}_2} = 1 \quad \text{---} \quad N_{\text{Rb}_2} = 0$$

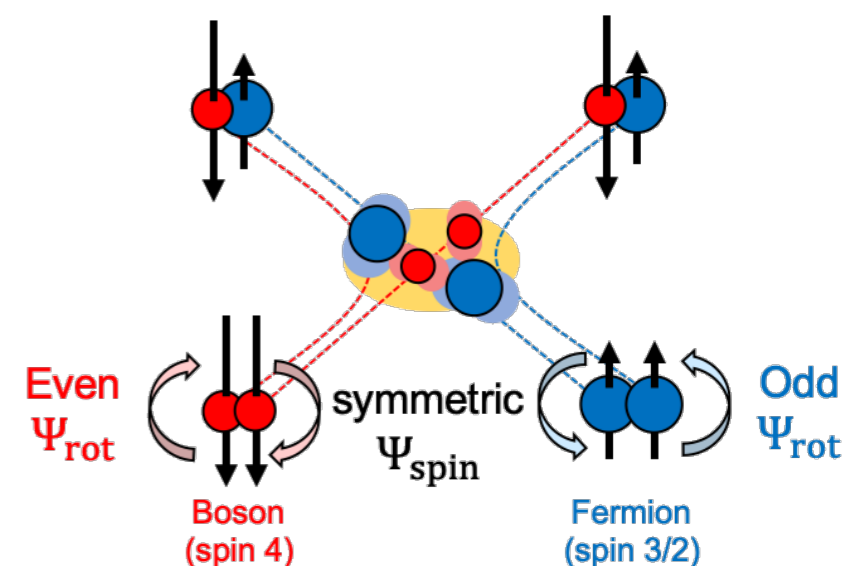
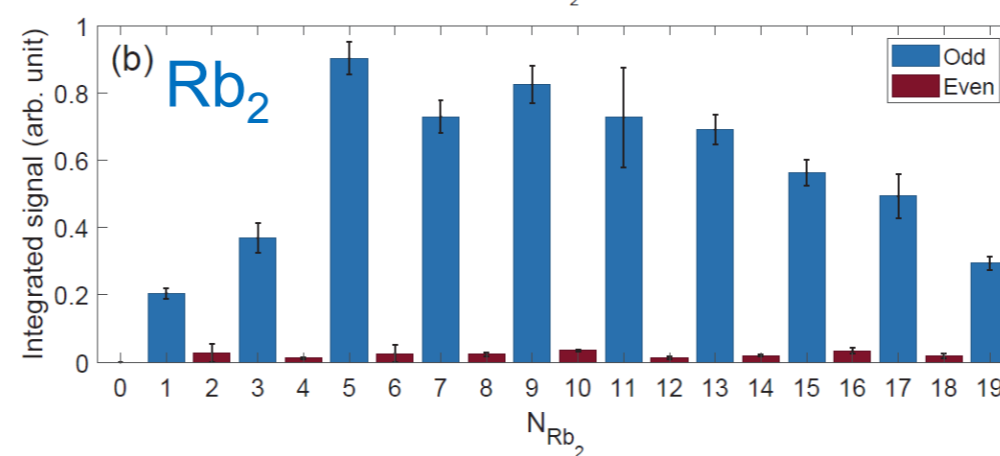
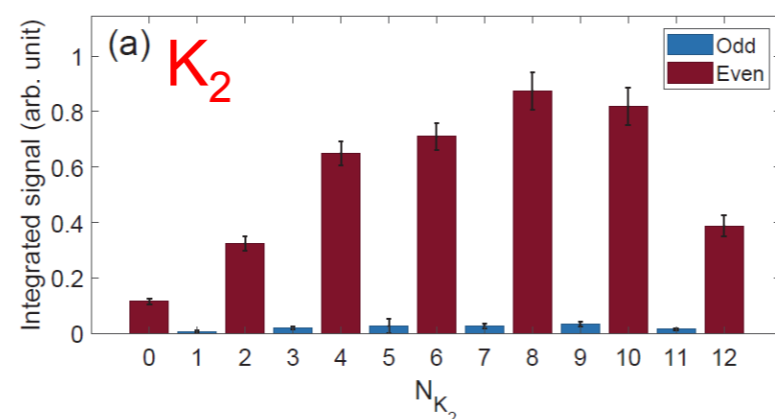
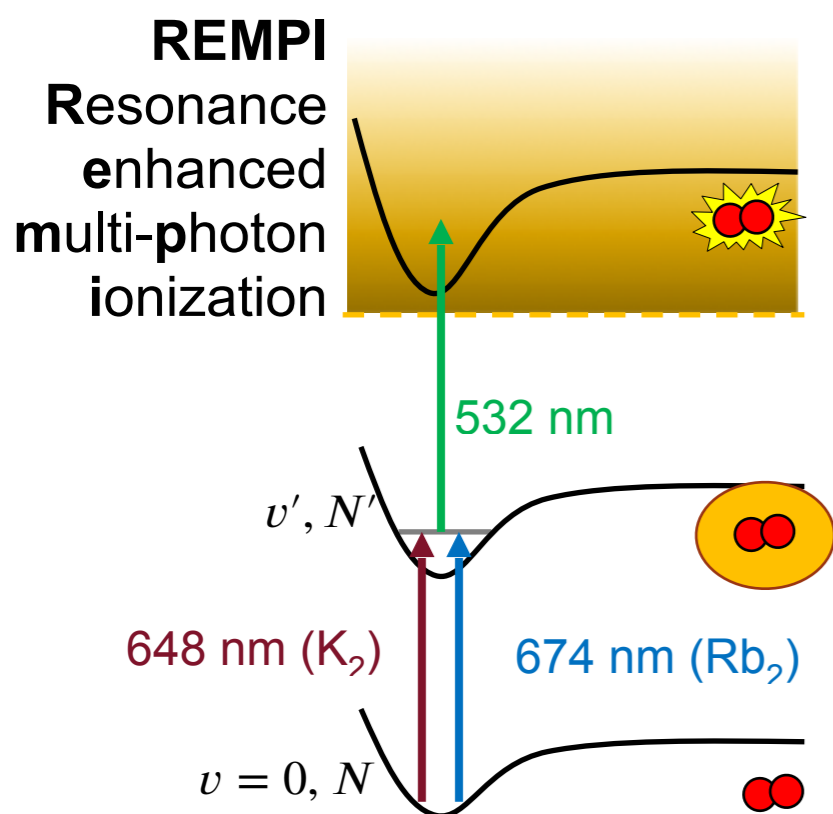
$$N_{\text{K}_2} = 0 \quad \text{---} \quad N_{\text{Rb}_2} = 1$$

$$N_{\text{K}_2} = 0 \quad \text{---} \quad N_{\text{Rb}_2} = 0$$

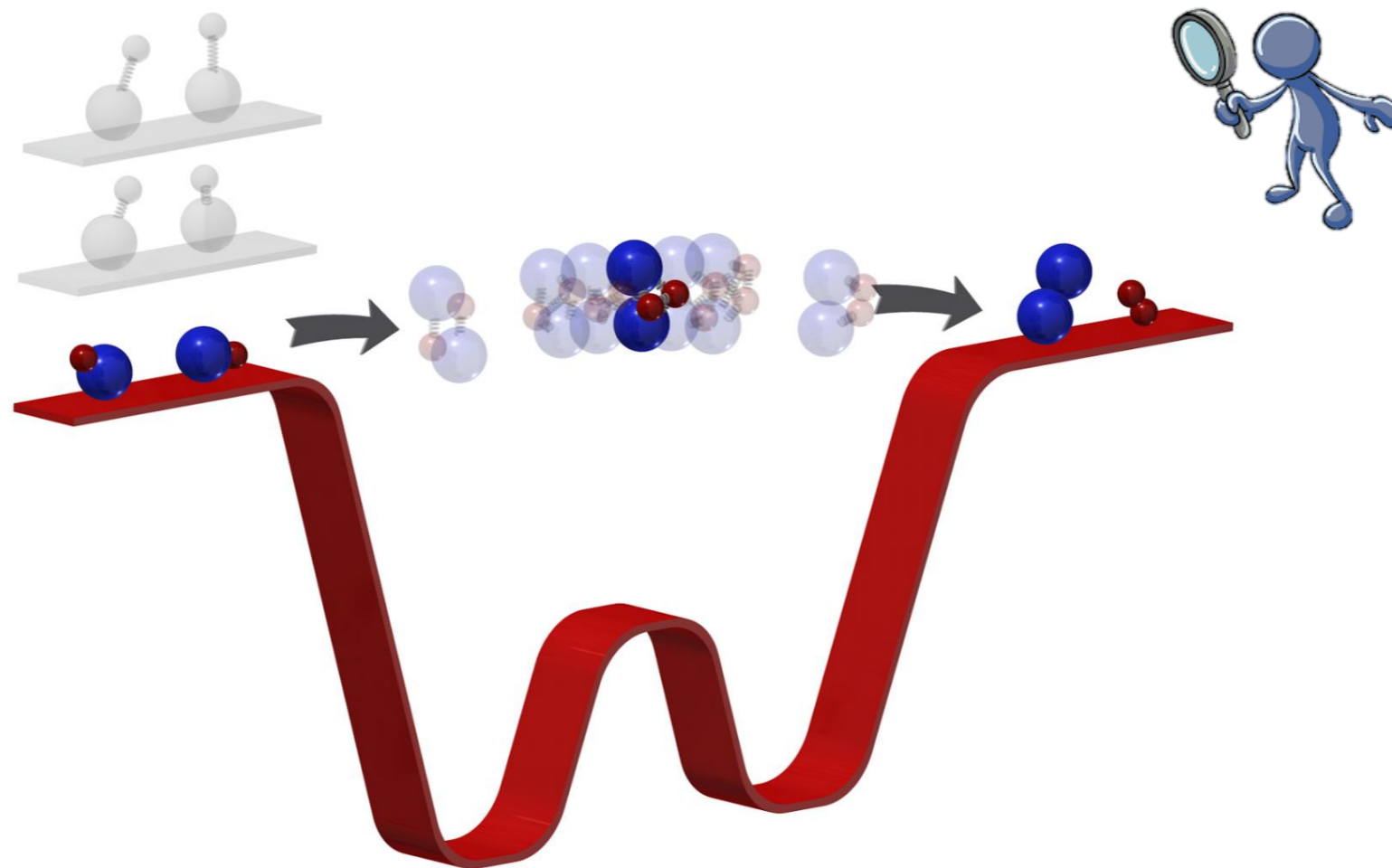
$$1\text{K} = 0.002 \text{ kcal/mol} \sim 1 \text{ cm}^{-1}$$

Rotational state distributions of K_2 and Rb_2

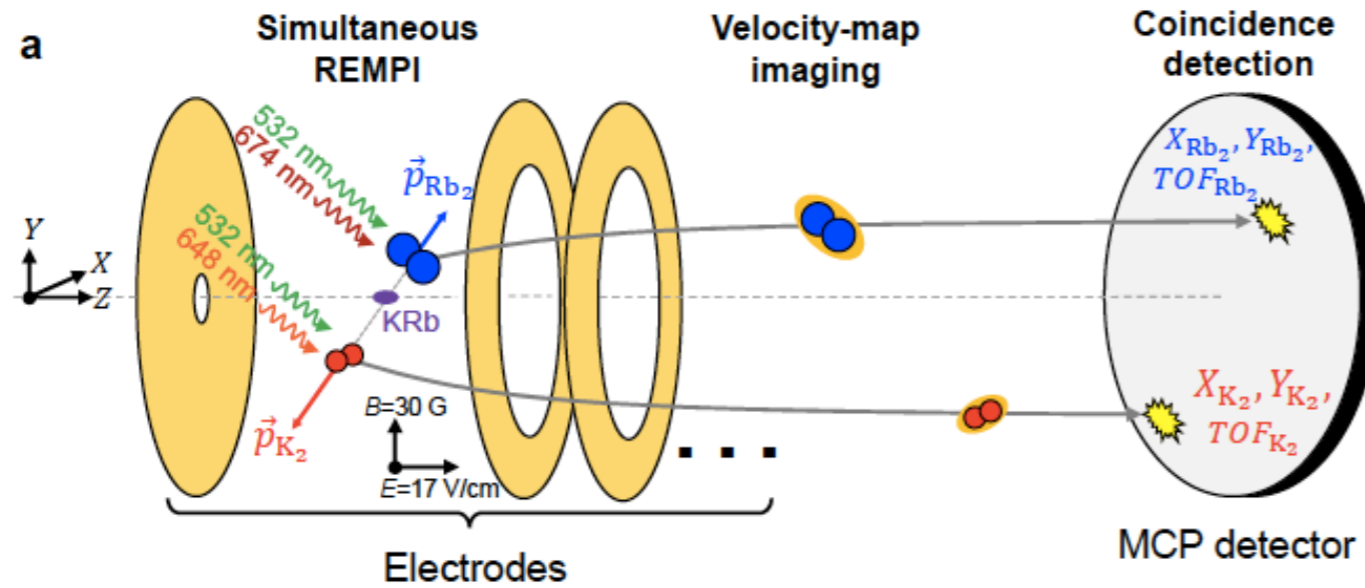
Strong parity preference \rightarrow Nuclear spin conservation!



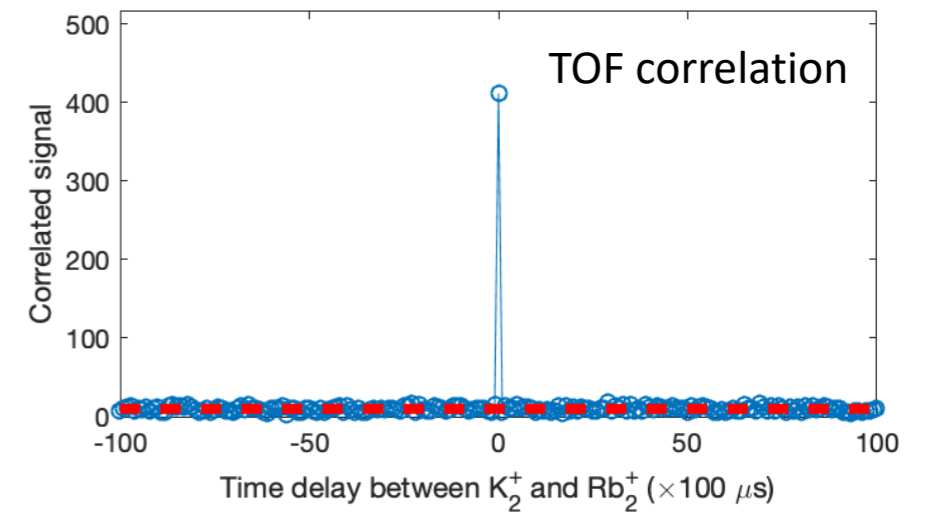
Resolve products from individual reaction events?



Coincidence detection



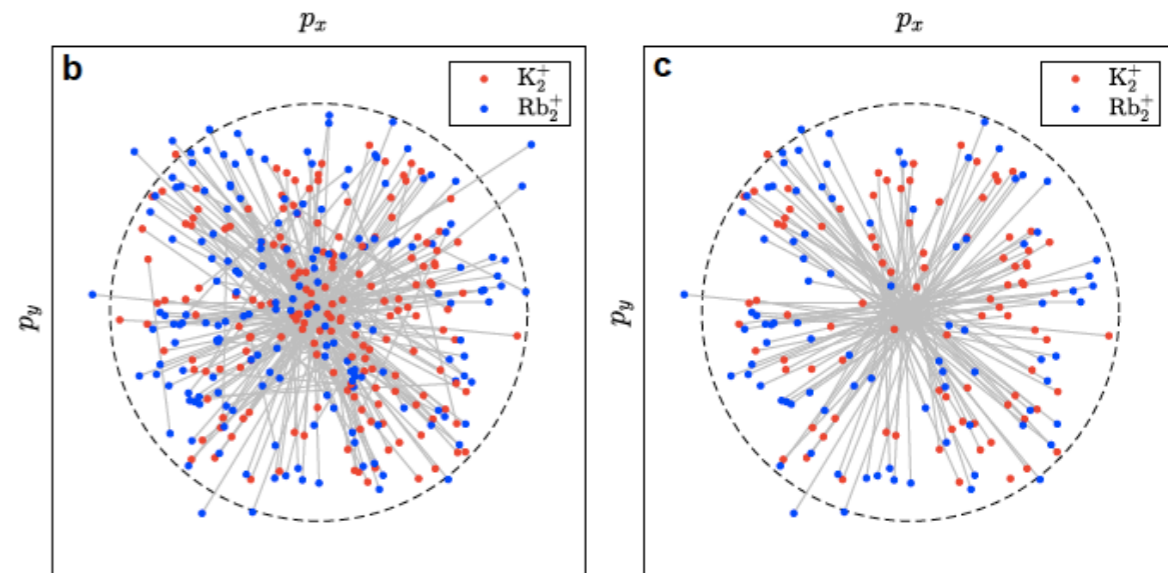
$$(N_{\text{K}_2} = 12, N_{\text{Rb}_2} = 5)$$



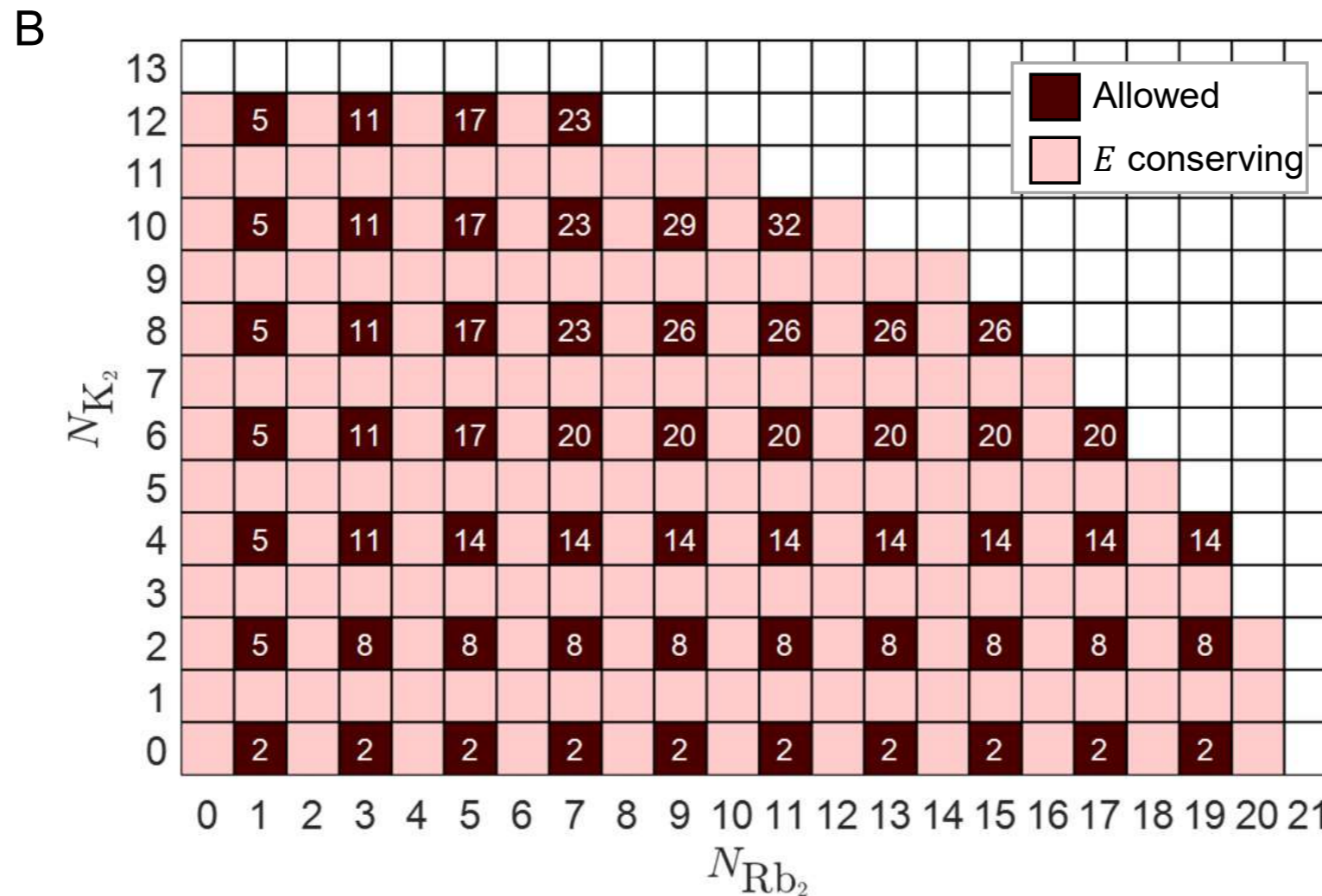
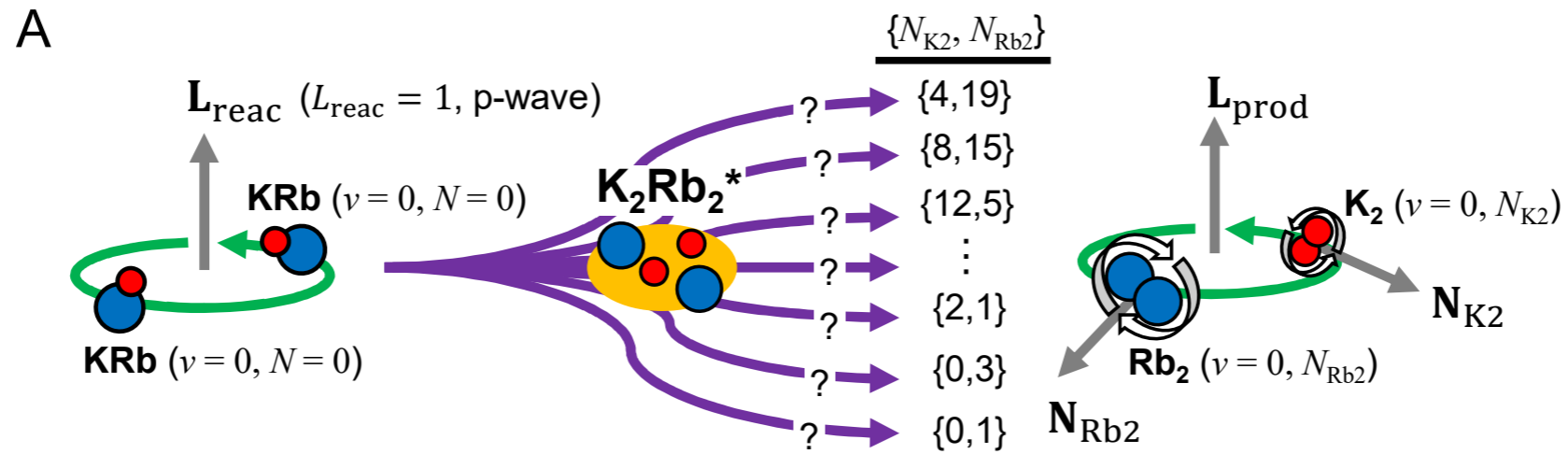
Momentum conservation constrains:

$$\vec{P}_{\text{K}_2} + \vec{P}_{\text{Rb}_2} = \vec{P}_{\text{KRb}} + \vec{P}_{\text{KRb}} \rightarrow 0$$

- Z momentum – TOF information filter
- X, Y momentum – X, Y spatial information filter

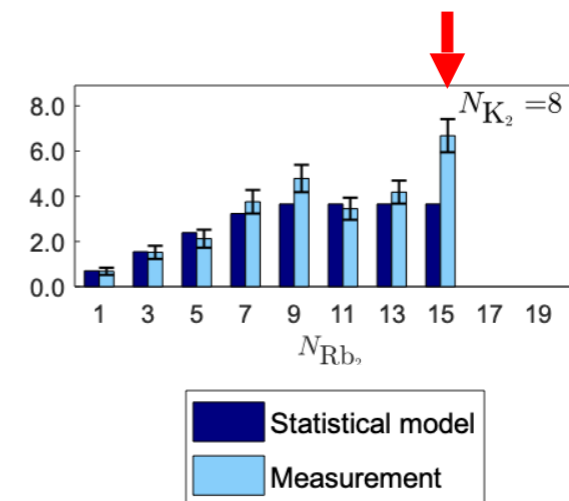
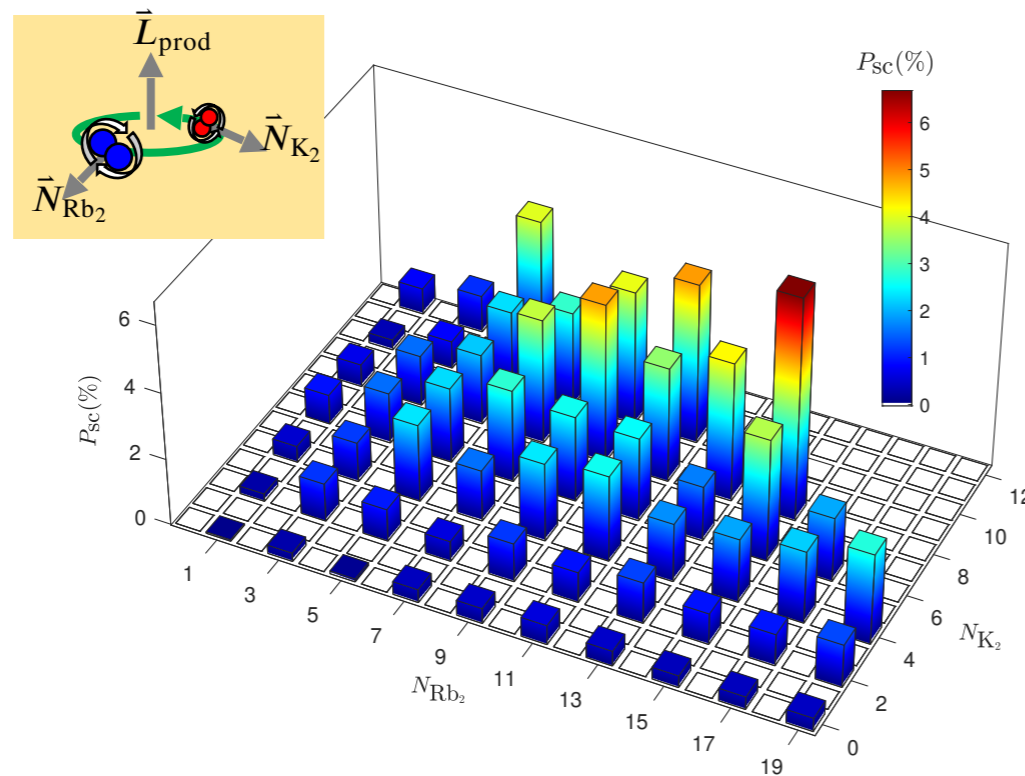


Non-trivial statistical distribution



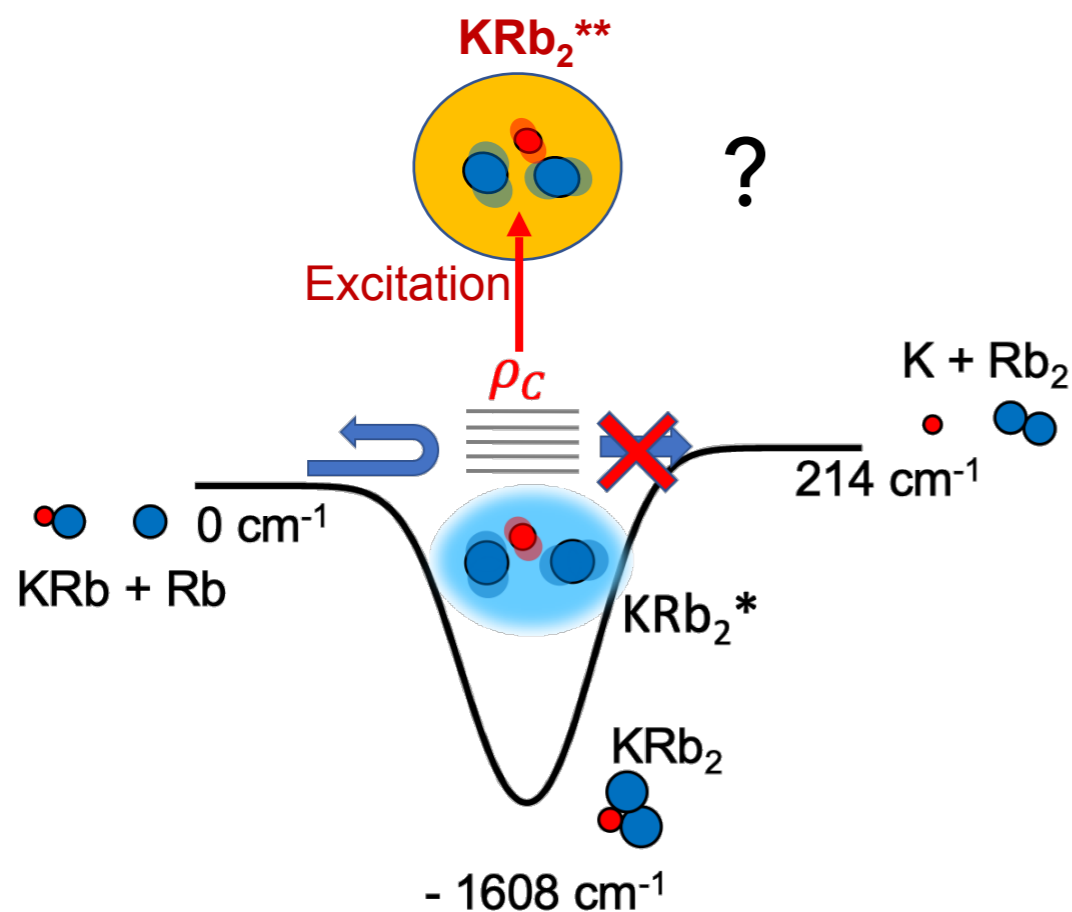
Complete Characterization of a Chemical Reaction

- Full quantum state outcome measured for the molecule+molecule reaction. Our analysis (likelihood ratio test) shows that after removing 7 deviating channels, the data matches the prediction based on statistical state counting (exact quantum calculation is beyond the state-of-the-art theory)



What's Next?

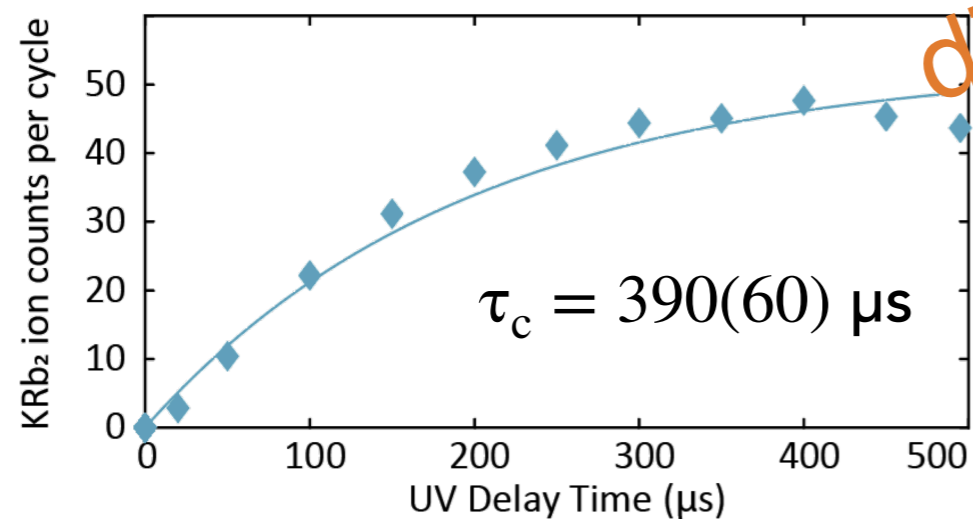
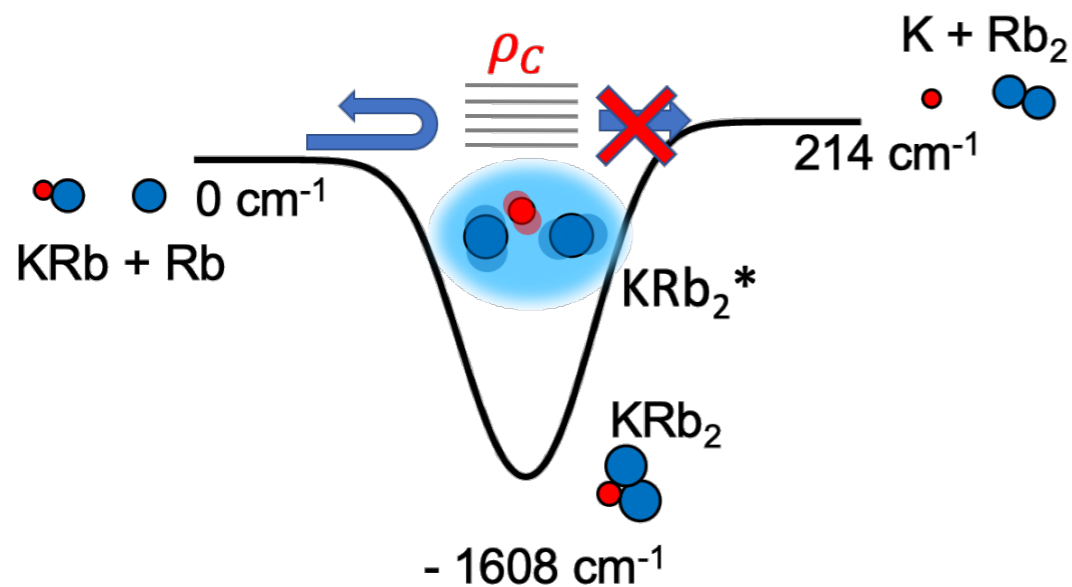
“more tractable” atom-molecule reactions/collisions



- * $\text{KRb} + \text{Rb}$ reaction is energetically forbidden
- * Long-lived complex?
- * Photo-excitation of complexes?
- * $\text{KRb} + \text{K}$ reaction is energetically allowed

Exp vs RRKM predicted complex lifetime

$$\tau_{\text{RRKM}} = \frac{h \rho_C}{N_0}$$



	Measurement	RRKM prediction
KRb + KRb [1]	360(30) ns	170(60) ns
RbCs + RbCs [2]	0.53 ms	0.25 ms

- [1] Liu*, Hu*, et al., Nature Physics (2020)
 [2] Gregory et al, PRL 124, 163402 (2020)

KRb + Rb

RRKM prediction
1 ns [3]
270 ns [4]

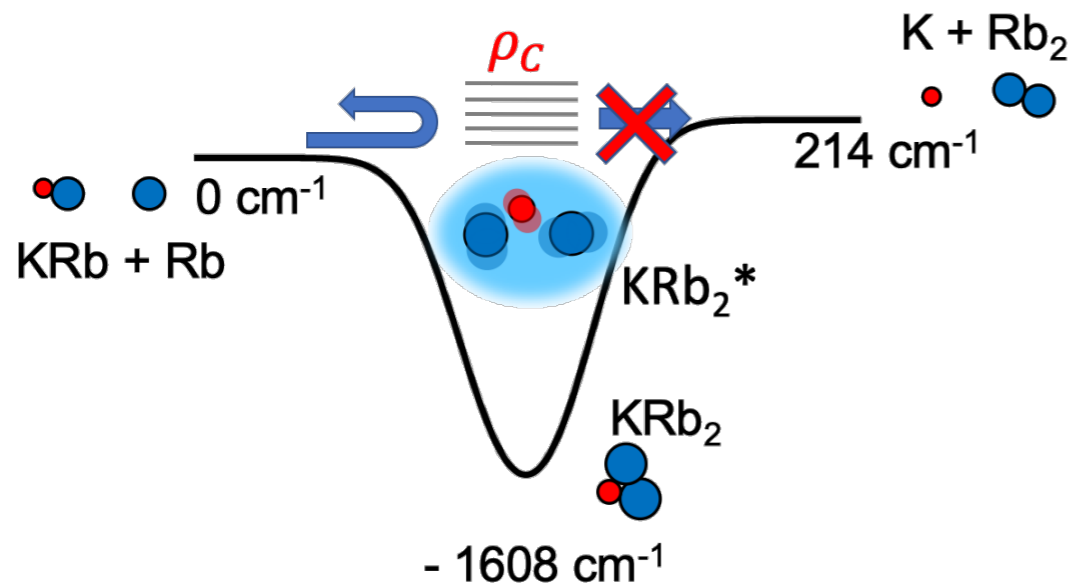
differs by 10⁵!!?

- [3] Private Communication with Tijs Karman & Christianen, et al, Phys. Rev. A 100, 032708 (2019)
 [4] Mayle, et al, Phys. Rev. A 85, 062712 (2012)

Other systems	Measurement*	RRKM prediction
NaK + NaK (boson, fermion)	>0.35 ms, >2.6 ms	6 μs , 18 μs
NaRb + NaRb	>1.2 ms	13 μs

What could explain the $\times 10^5$ difference?

$$\tau_{\text{RRKM}} = \frac{h \rho_C}{N_0}$$



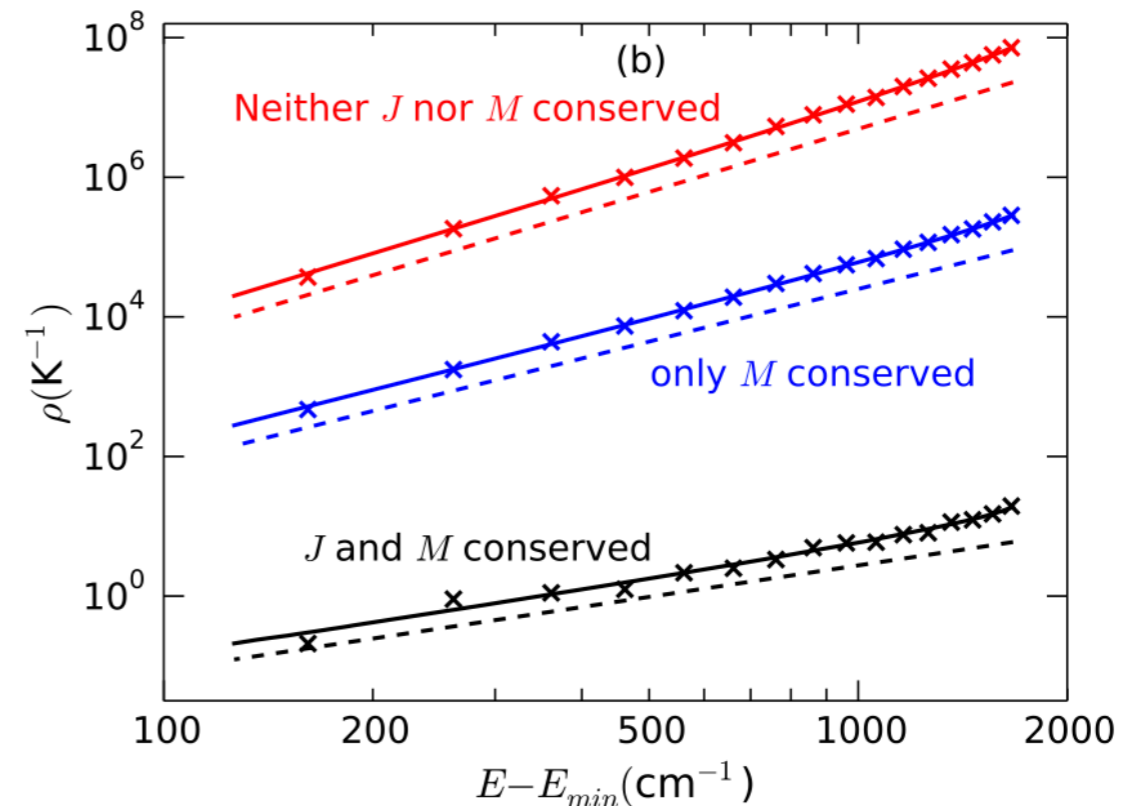
RRKM assumption

- DOS large
- breaks down?

Croft et al. arXiv:2111.09956

RRKM holds but

- Angular momentum is not conserved?

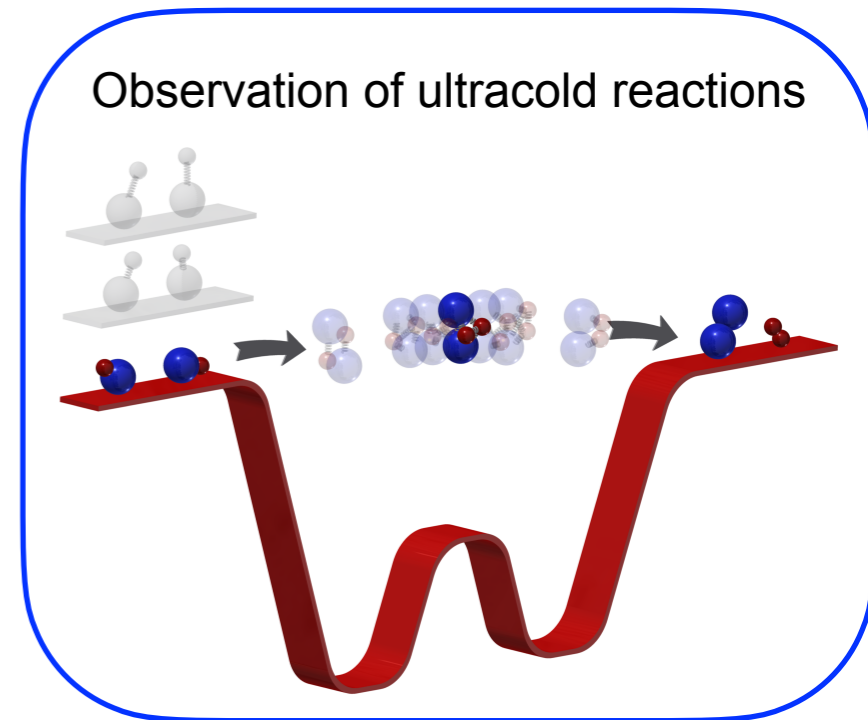


Christianen, et al, Phys. Rev. A 100, 032708 (2019)

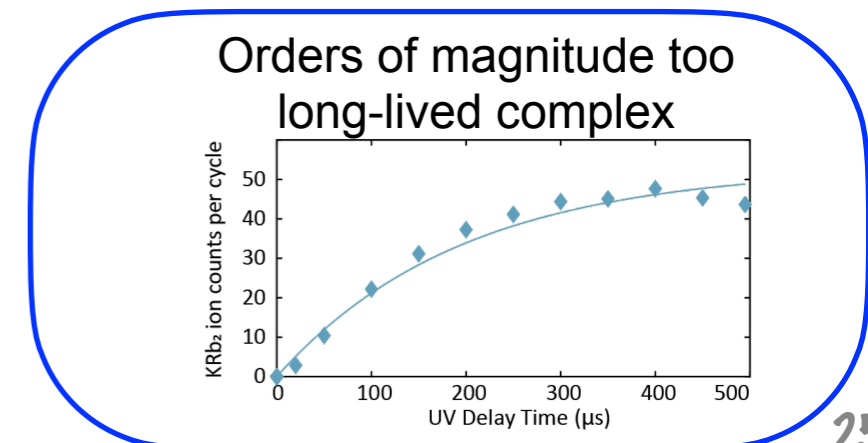
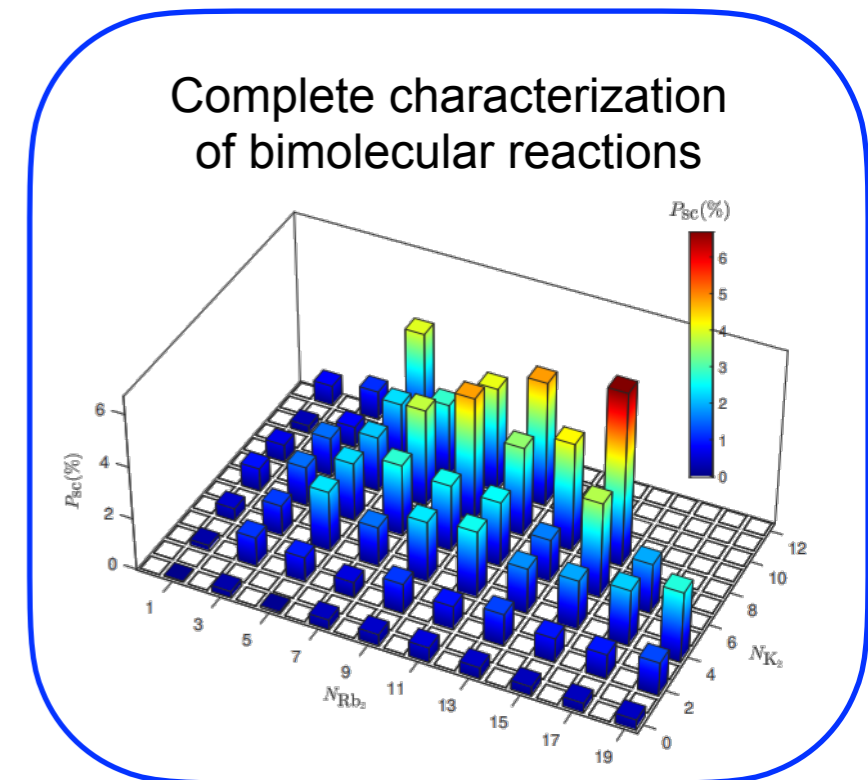
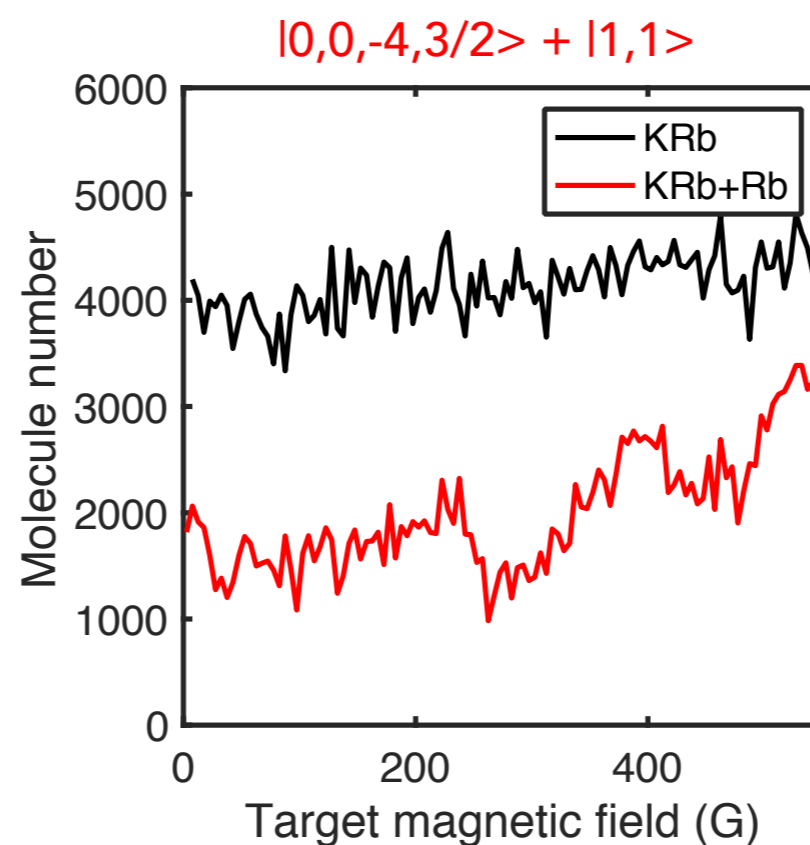
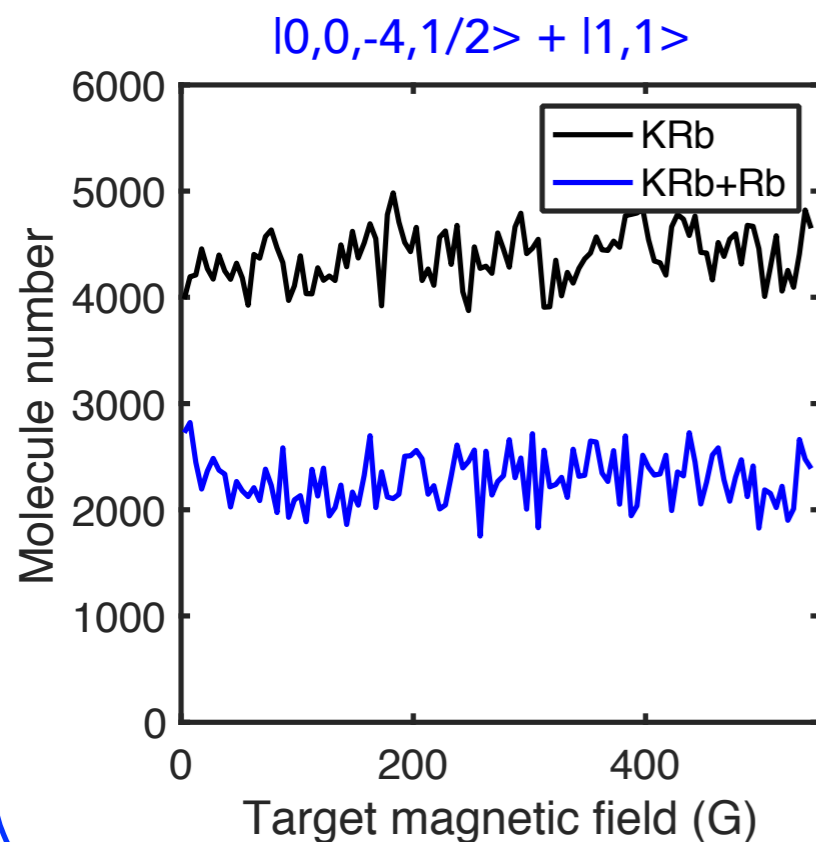
Also: Frye, et al, NJP 23 125008 (2021)

Summary and Outlook

- * We brought chemistry and physics tools together to completely characterize a molecule-molecule reactive process with full state-to-state details
- * We now turn our attentions to atom-molecule reactions/collisions where exact quantum calculations are more tractable, but new questions arise



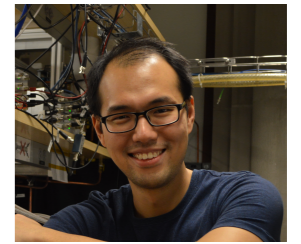
Magnetic field dependent atom-molecule loss (*preliminary*)



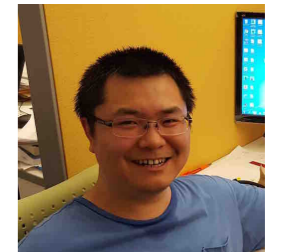
Acknowledgments

Thank you!

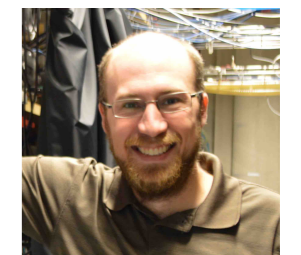
KRb Alumni



Yu Liu, PhD
=> NIST Boulder



Dr. Ming-Guang Hu
=> QuEra Computing



Dr. David Grimes
=> MIT



Dr. Matt Nichols
=> Apple



Funding: DOE, Packard foundation, NSF, AFOSR, Dreyfus foundation, MURI-AFOSR, NSF-CUA, DOE-LBNL
Past: Beckman foundation, ARO, Sloan fellowship

