



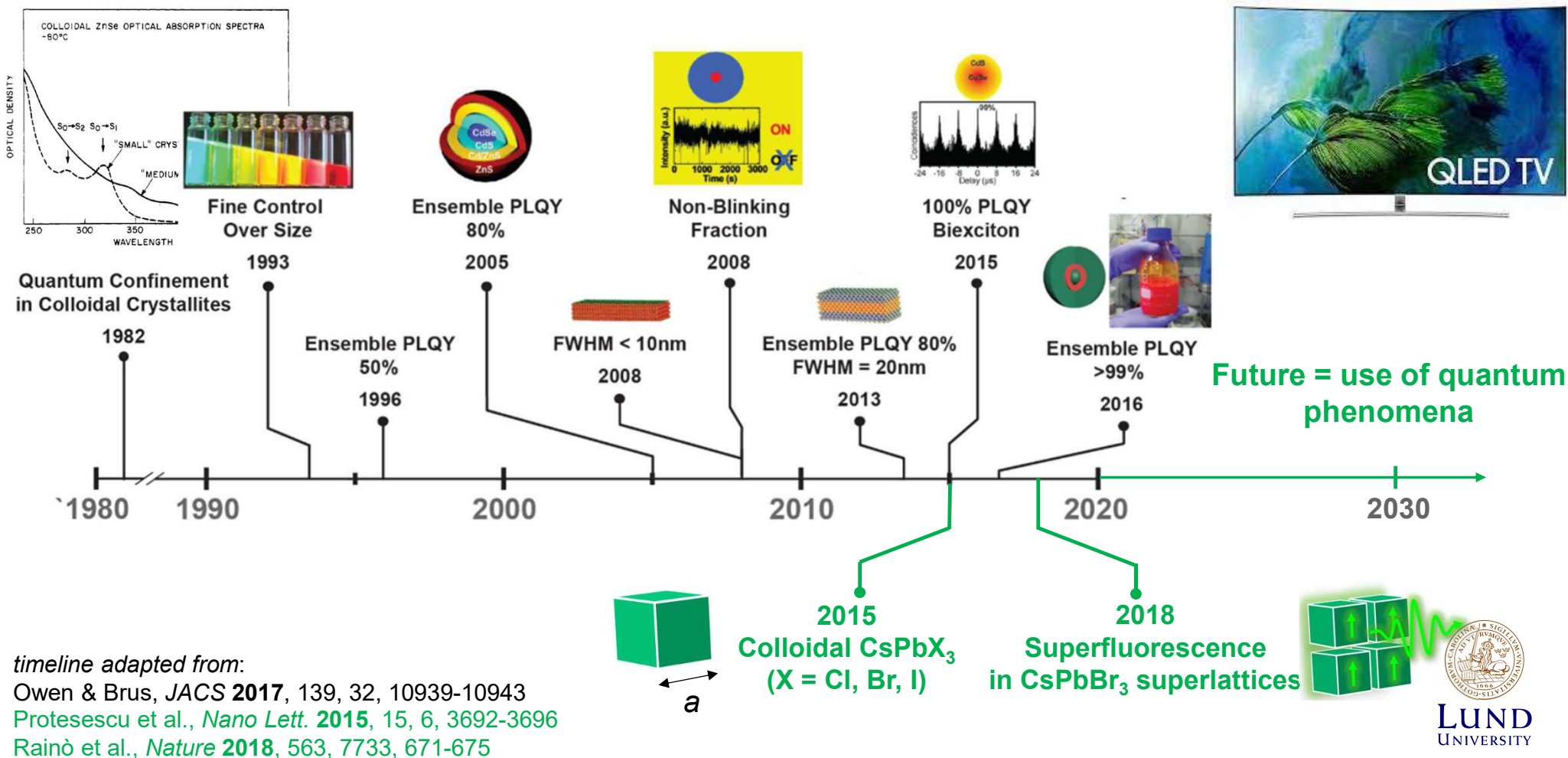
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Collective X-ray Diffraction and Photoluminescence in Perovskite Nanocrystal Superlattices

DMITRY BARANOV, DIVISION OF CHEMICAL PHYSICS, LUND UNIVERSITY
KITP NANOASSEMBLY23 CONFERENCE
16 MAY 2023



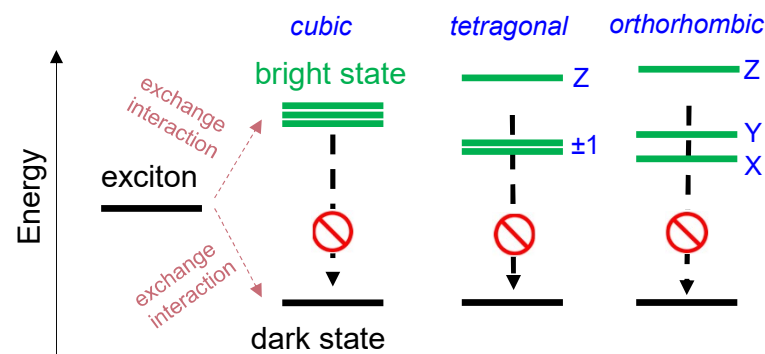
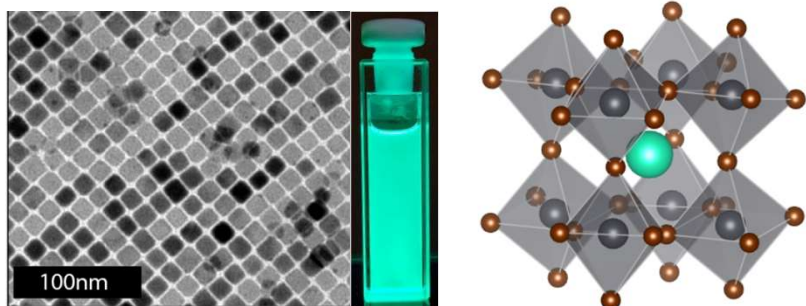
Colloidal Semiconductor Nanocrystals



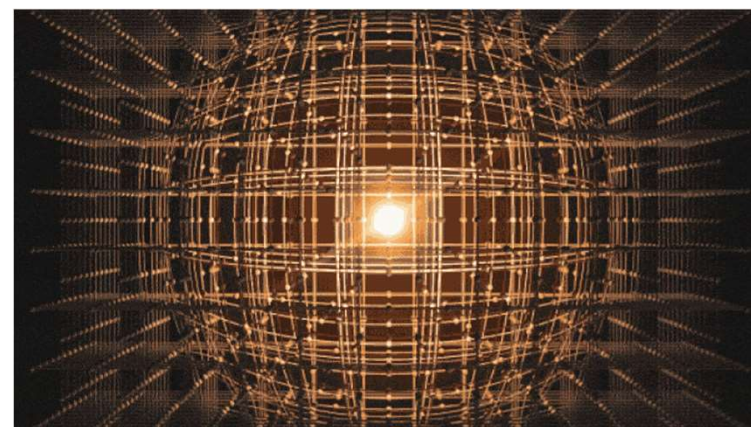
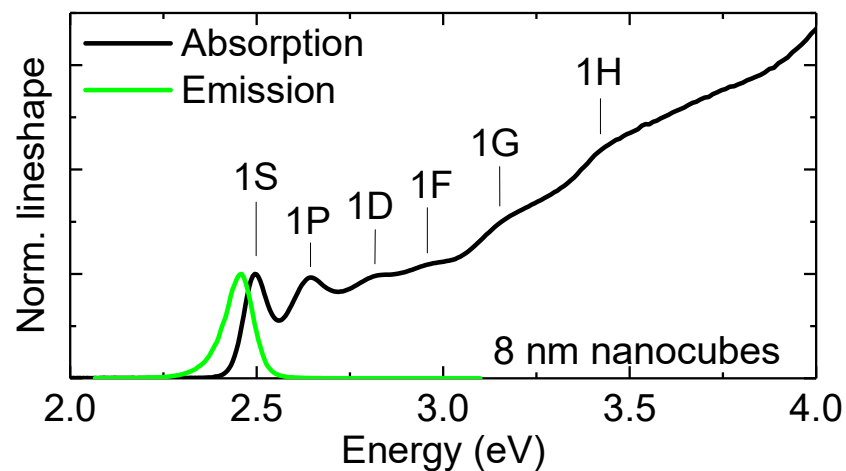
timeline adapted from:
 Owen & Brus, *JACS* **2017**, 139, 32, 10939-10943
 Protesescu et al., *Nano Lett.* **2015**, 15, 6, 3692-3696
 Rainò et al., *Nature* **2018**, 563, 7733, 671-675

Lead Halide Perovskite Nanocrystals

CsPbBr_3



Efros, Even, Lounis, Sercel, and others



Guzelturk et al, *Nat. Mater.*, 2021, 20, 618-623

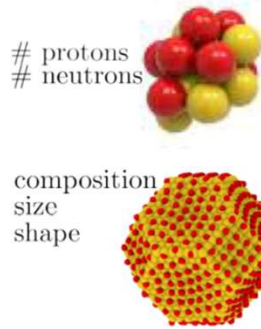


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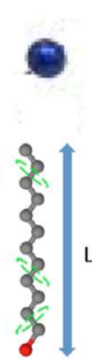
Artificial Atoms and Collective Effects

KITP
Nanoassembly23
conceptual
framework
Figure 1

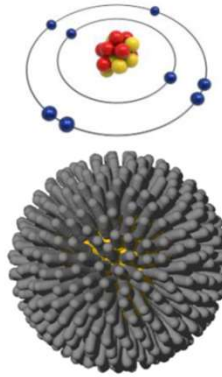
Nuclei



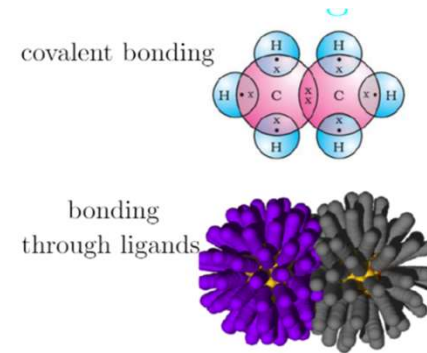
Electron



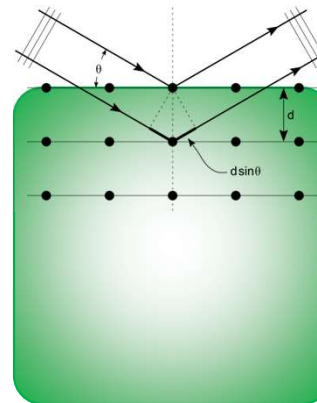
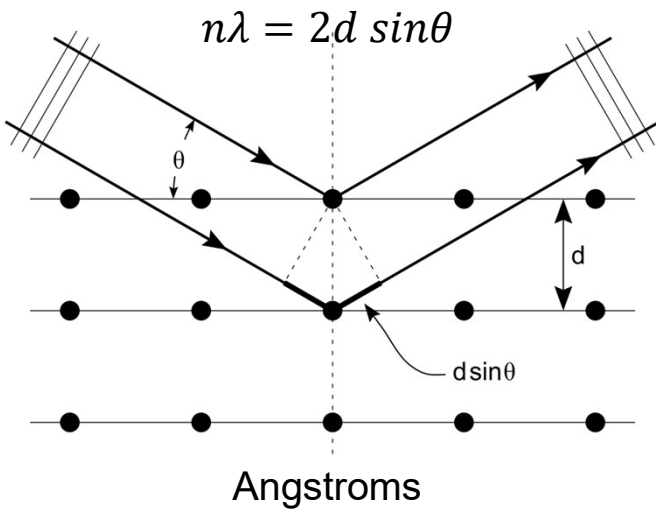
Atom



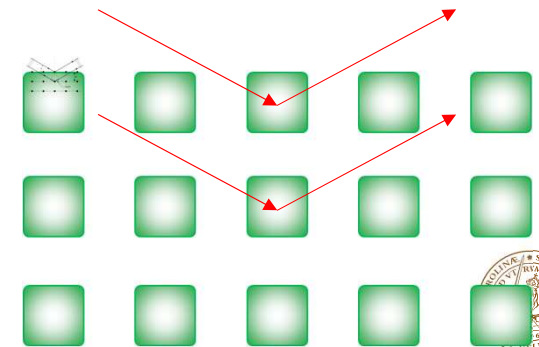
Bonding



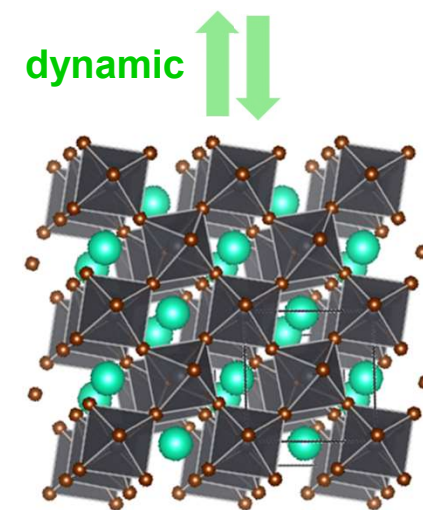
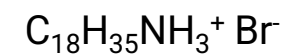
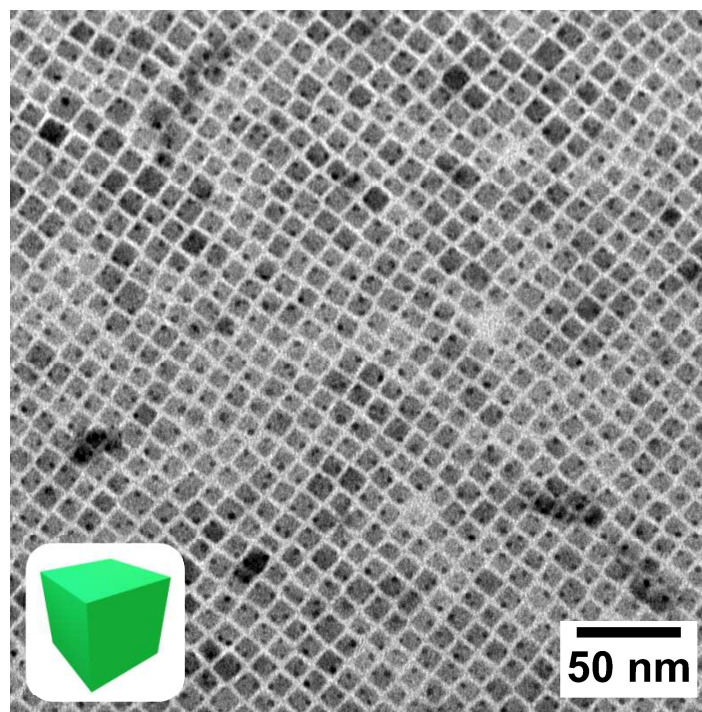
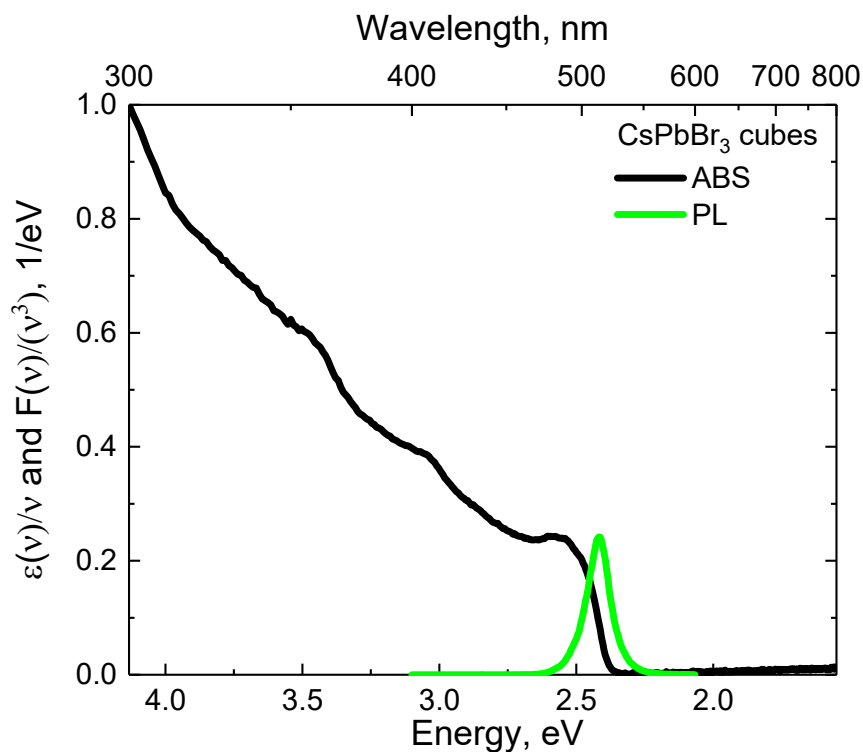
Material



Nanometers



8-10 nm CsPbBr₃ Nanocubes



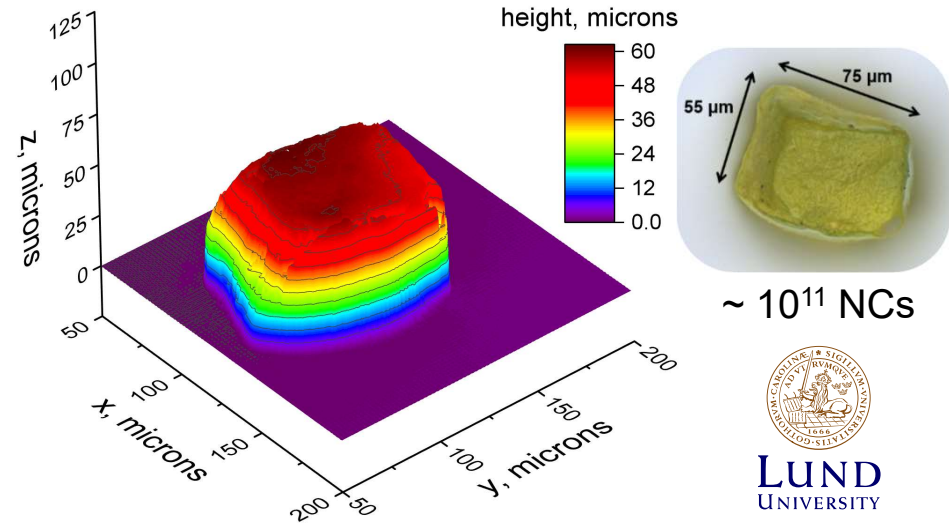
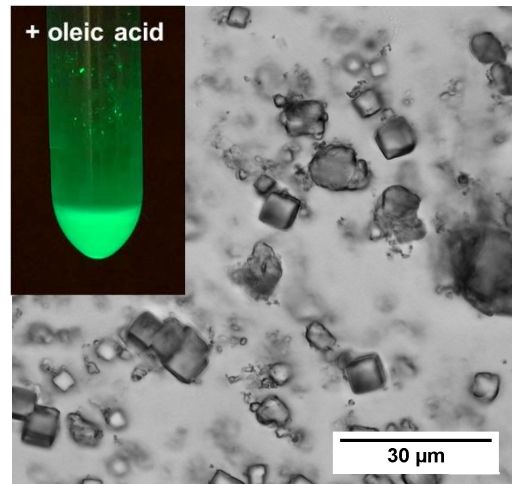
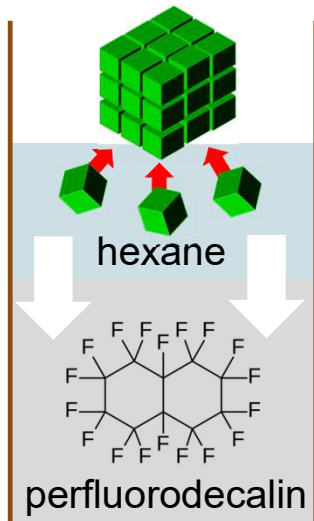
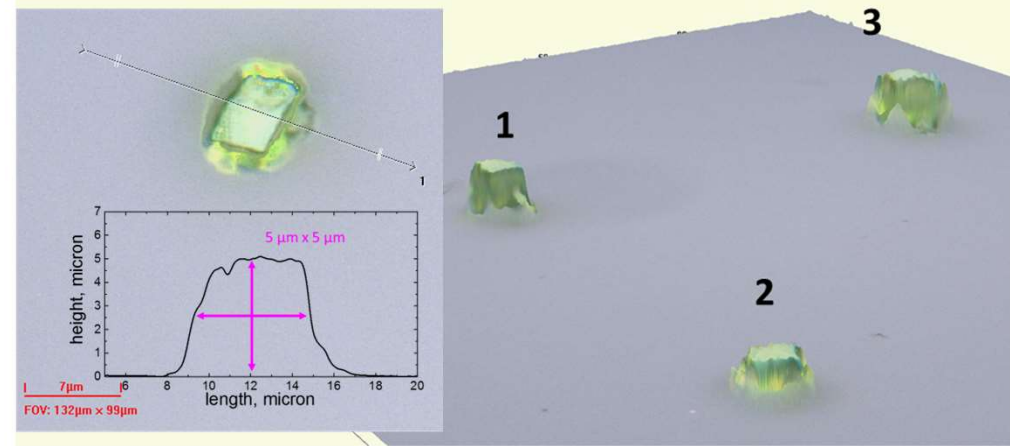
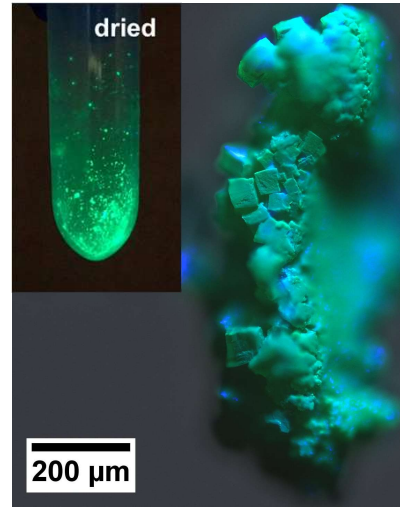
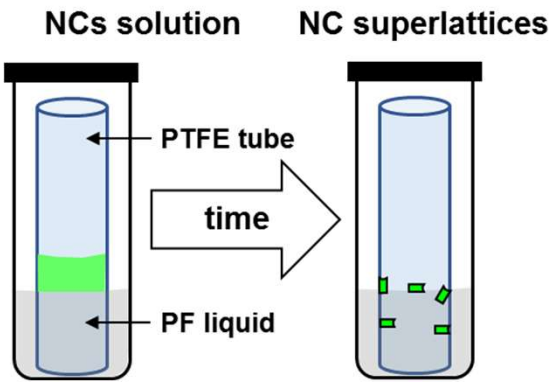
Protesescu et al., *Nano Lett*, 15 (6), 3692-3696, 2015

Almeida et al., *ACS Nano*, 12 (2), 1704-1711, 2018

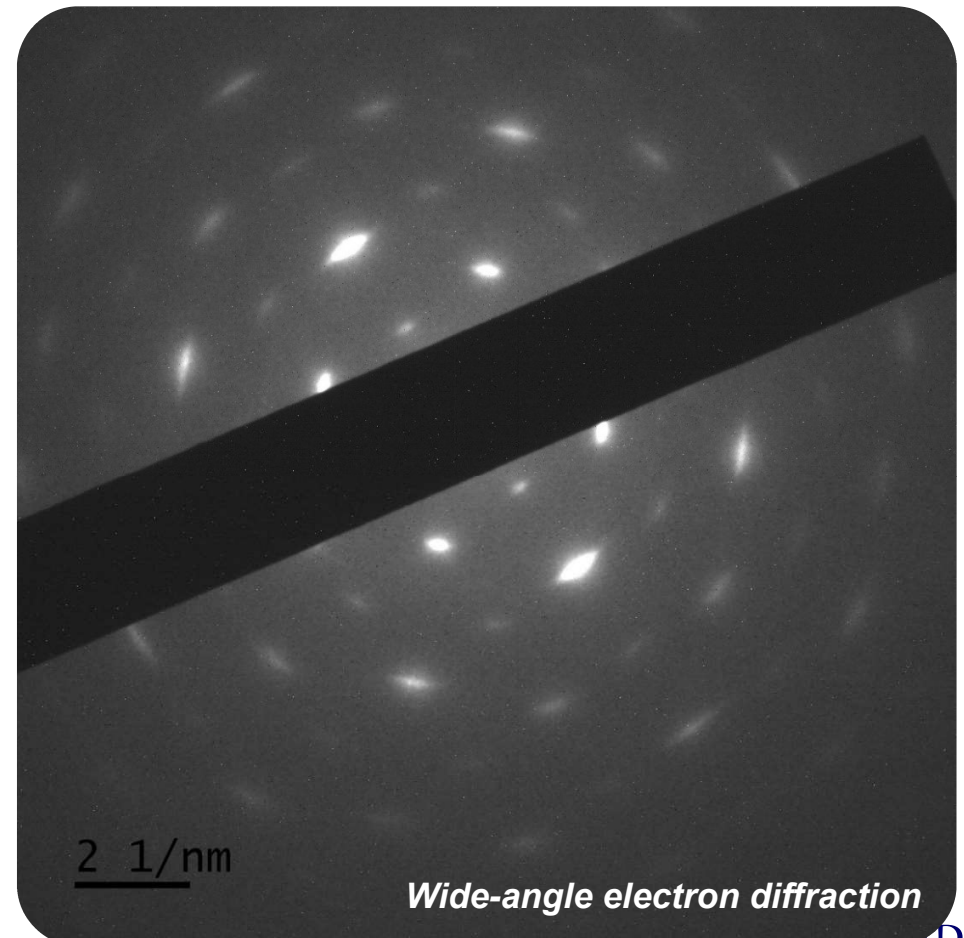
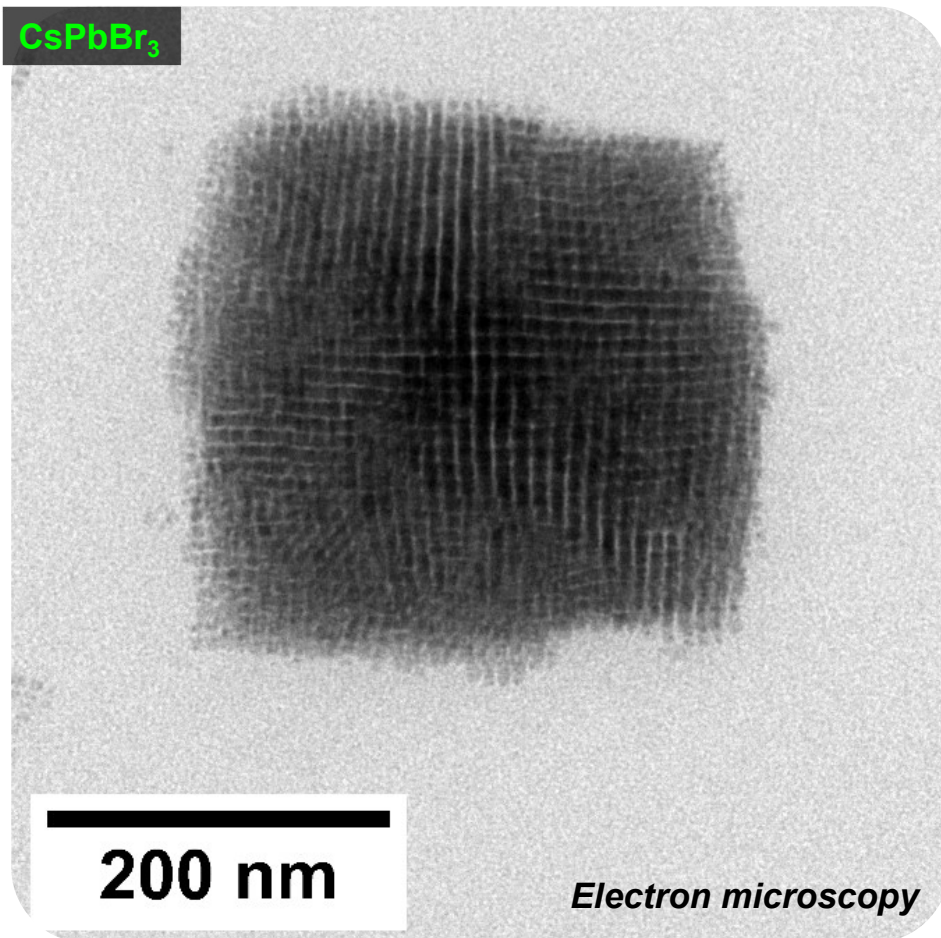


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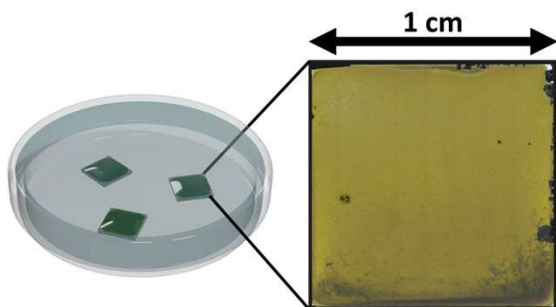
Superlattices by Solvent Removal



Well-ordered in Electron Diffraction



Assembly by Solvent Evaporation



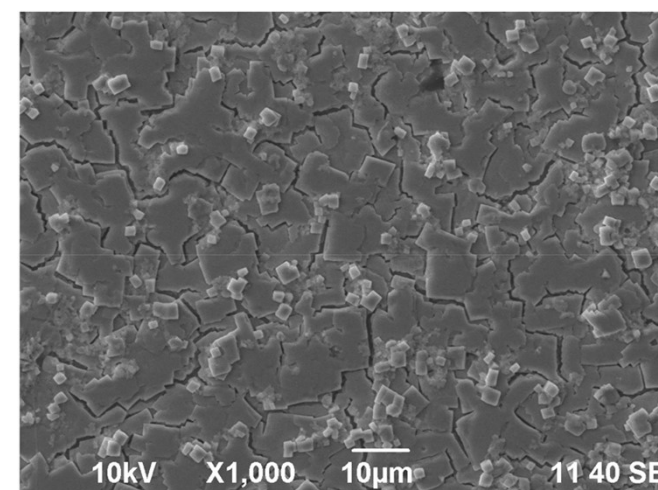
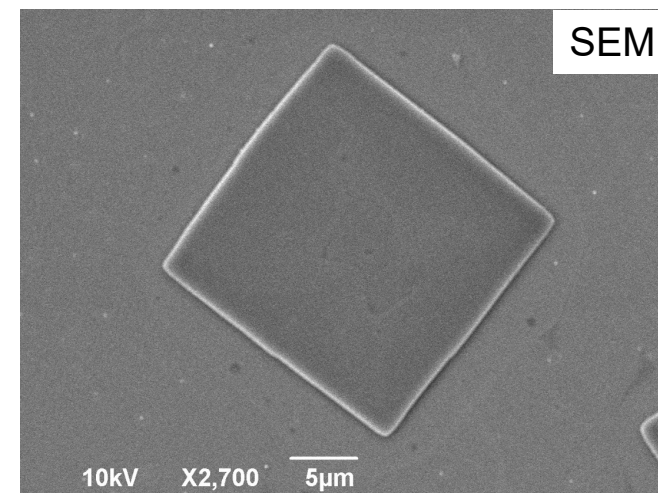
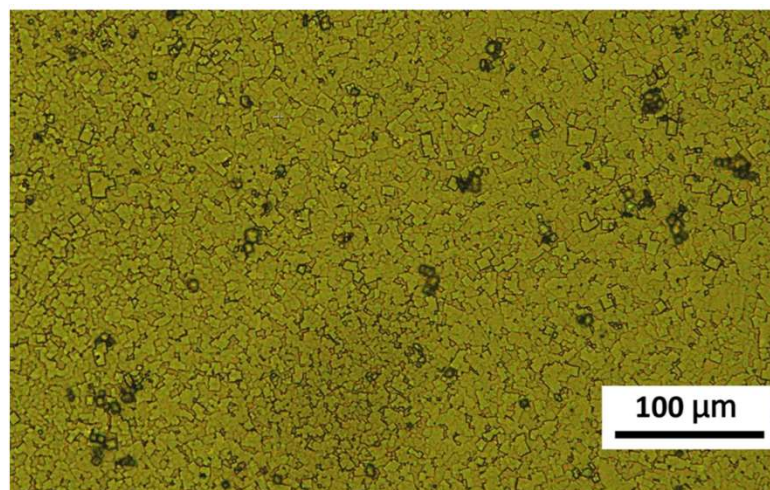
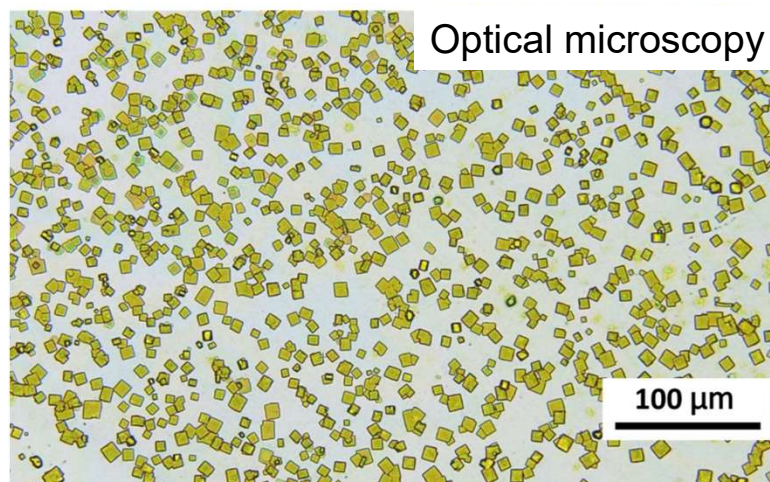
$[\text{CsPbBr}_3]_{\text{NC}} \approx 0.8\text{-}1 \mu\text{M}$

Solvents:

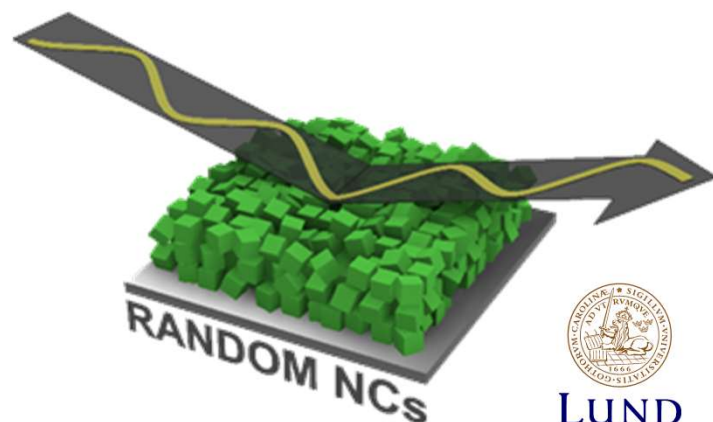
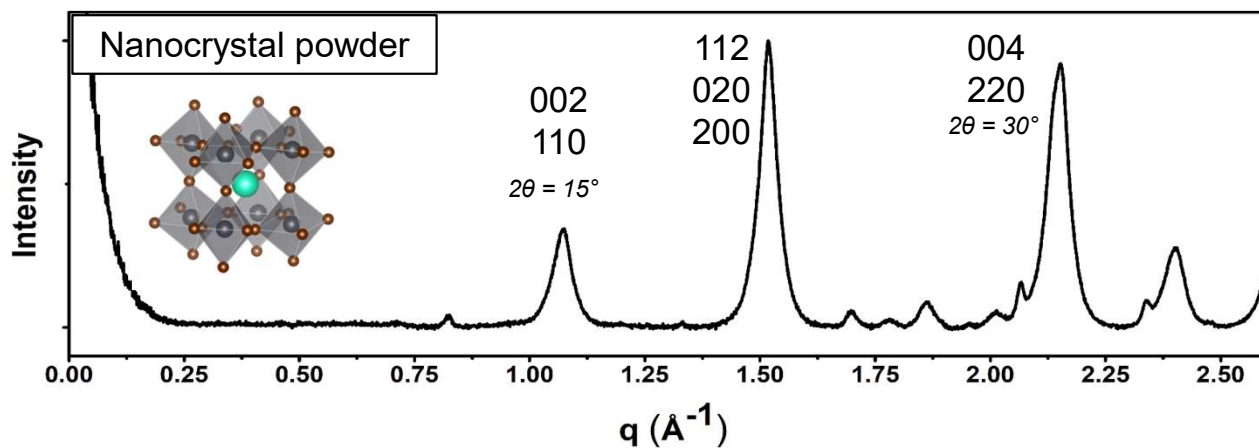
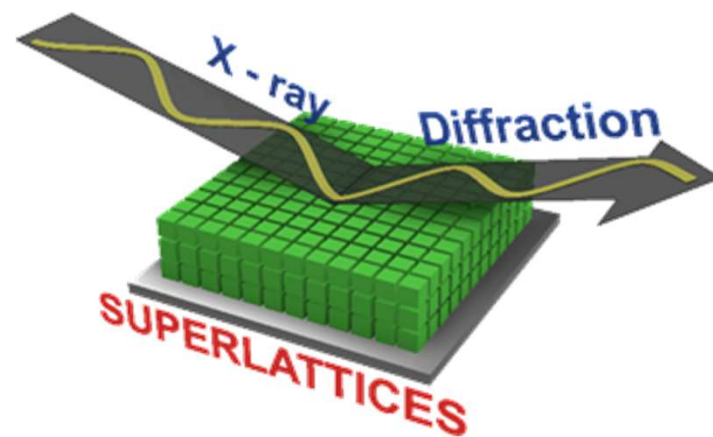
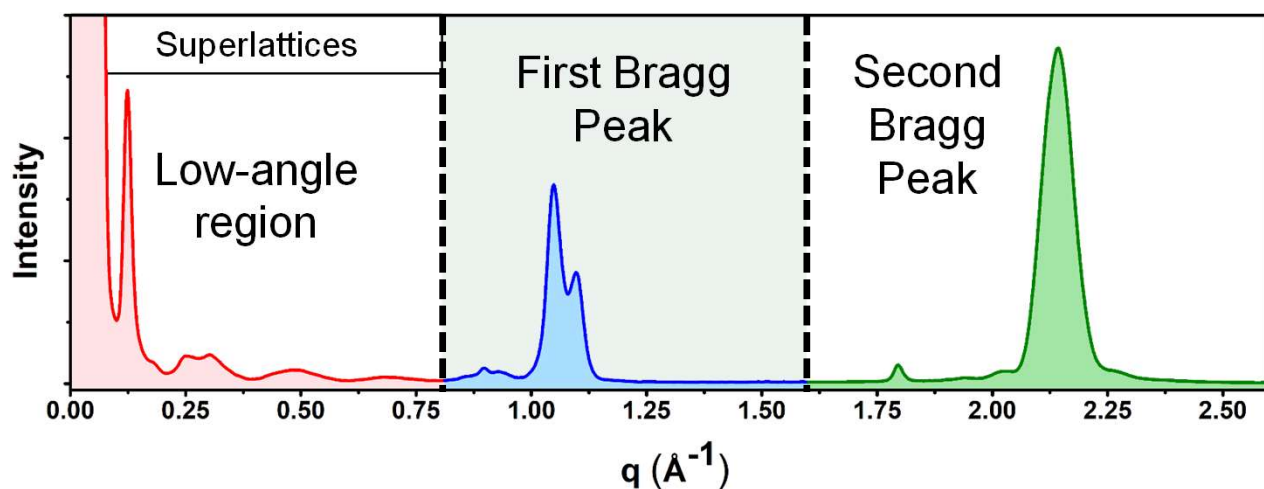
Toluene (*b.p.* 111 °C)

Tetrachloroethylene (*b.p.* 121 °C)

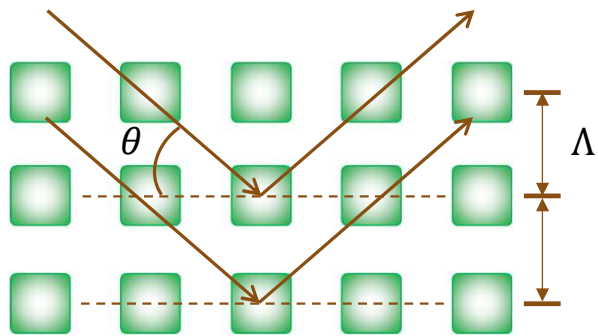
Evaporation takes 2-12 hours, depending on the amount of liquid.



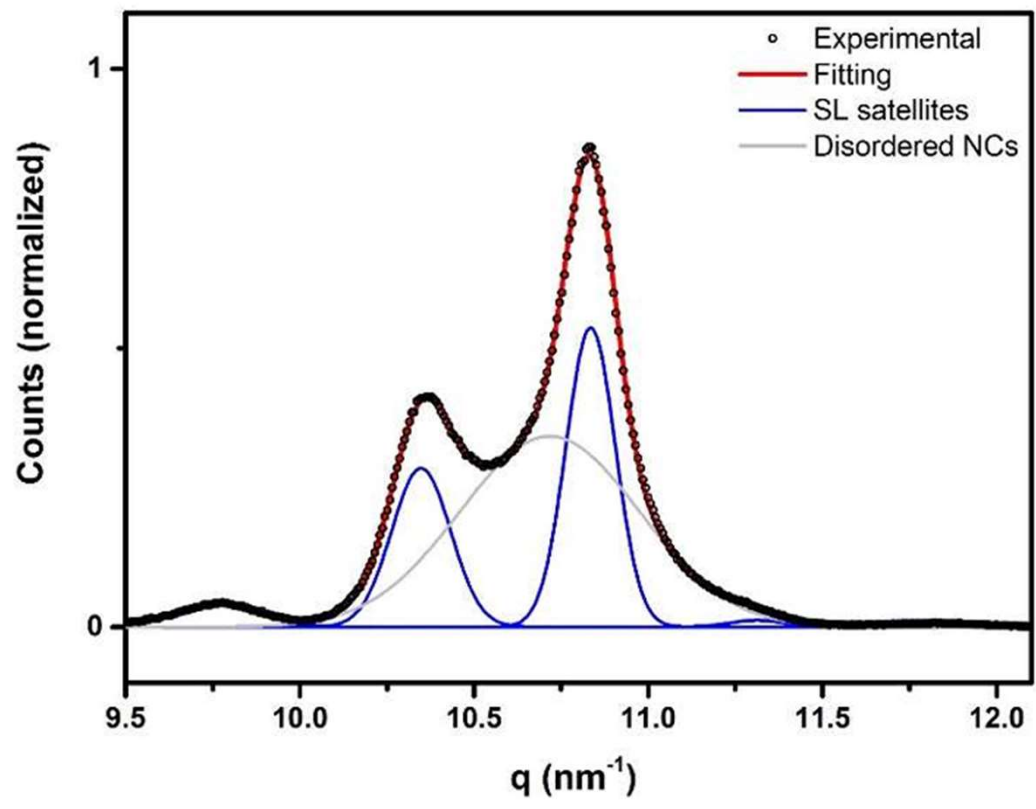
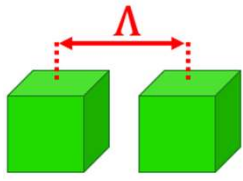
Wide-Angle X-Ray Diffraction



Superlattice Satellites of the 1st Bragg Peak

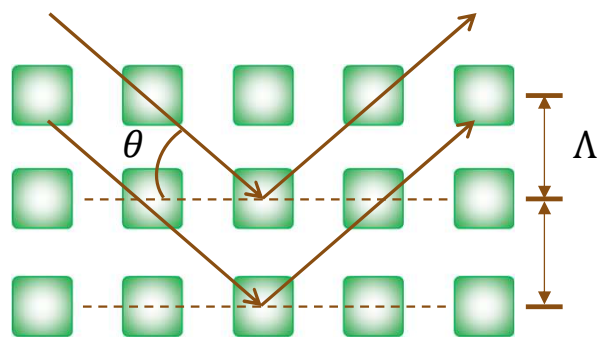


$$q_n = \frac{2\pi n}{\Lambda}$$

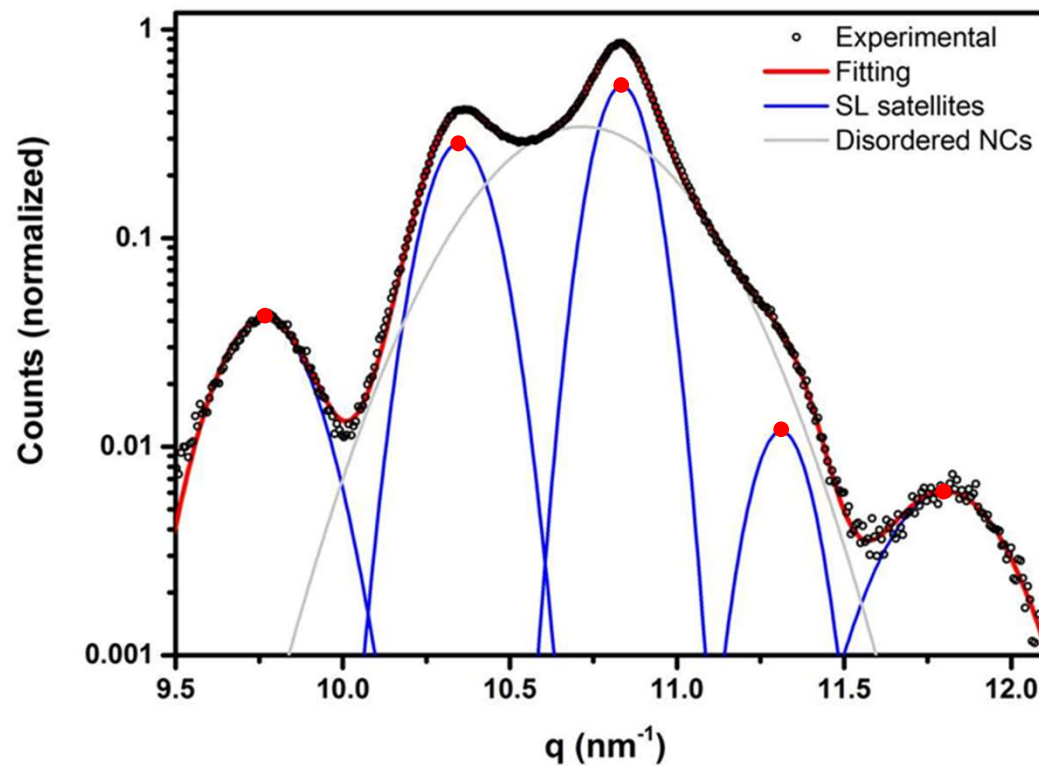
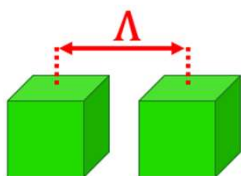


Schuller, *Phys. Rev. Lett.*, **1980**, 44, 24, 1597-1600
Toso, DB, Giannini, Manna, *ACS Mater. Lett.* **2019**, 1, 2, 272-276

Superlattice Satellites of the 1st Bragg Peak



$$q_n = \frac{2\pi n}{\Lambda}$$

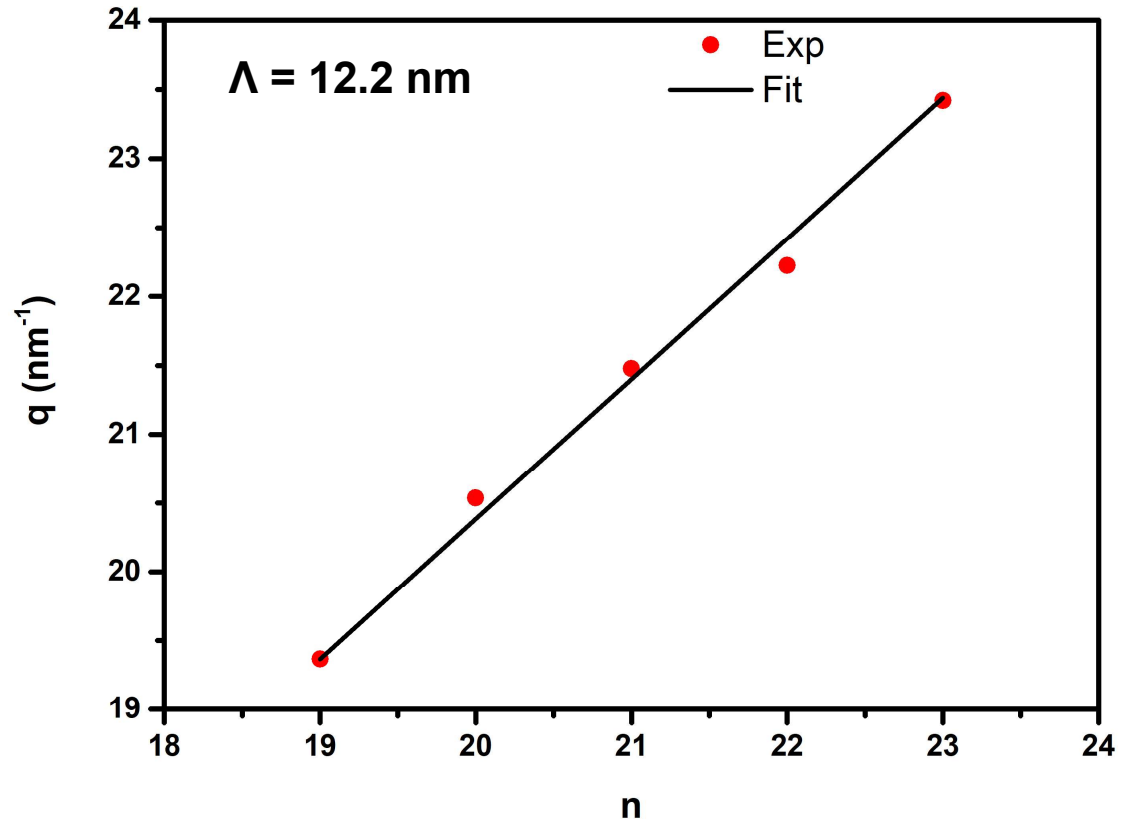
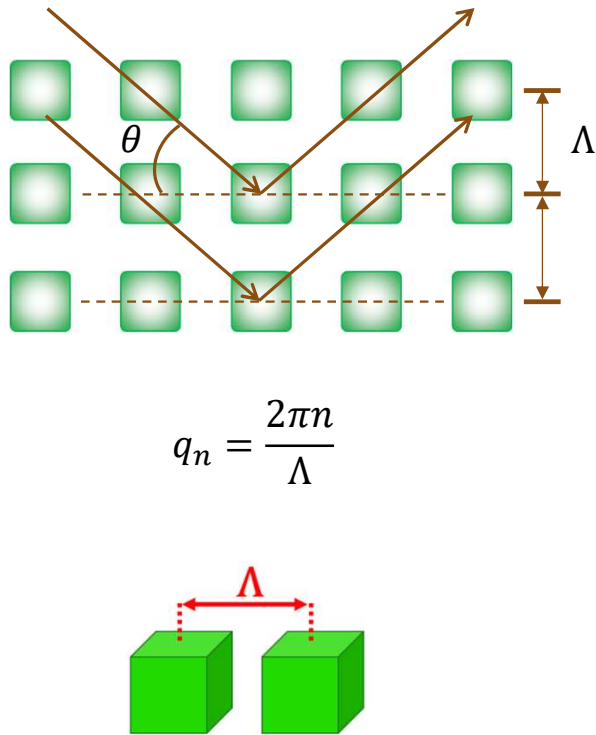


Schuller, *Phys. Rev. Lett.*, **1980**, 44, 24, 1597-1600
Toso, DB, Giannini, Manna, *ACS Mater. Lett.* **2019**, 1, 2, 272-276



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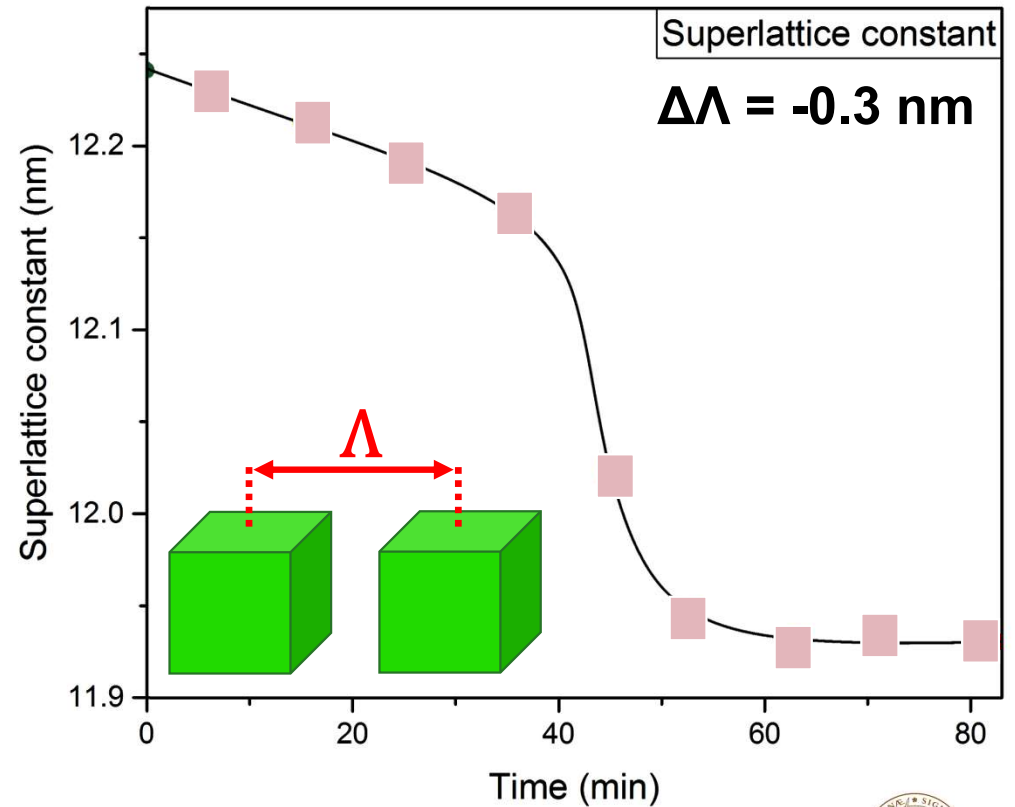
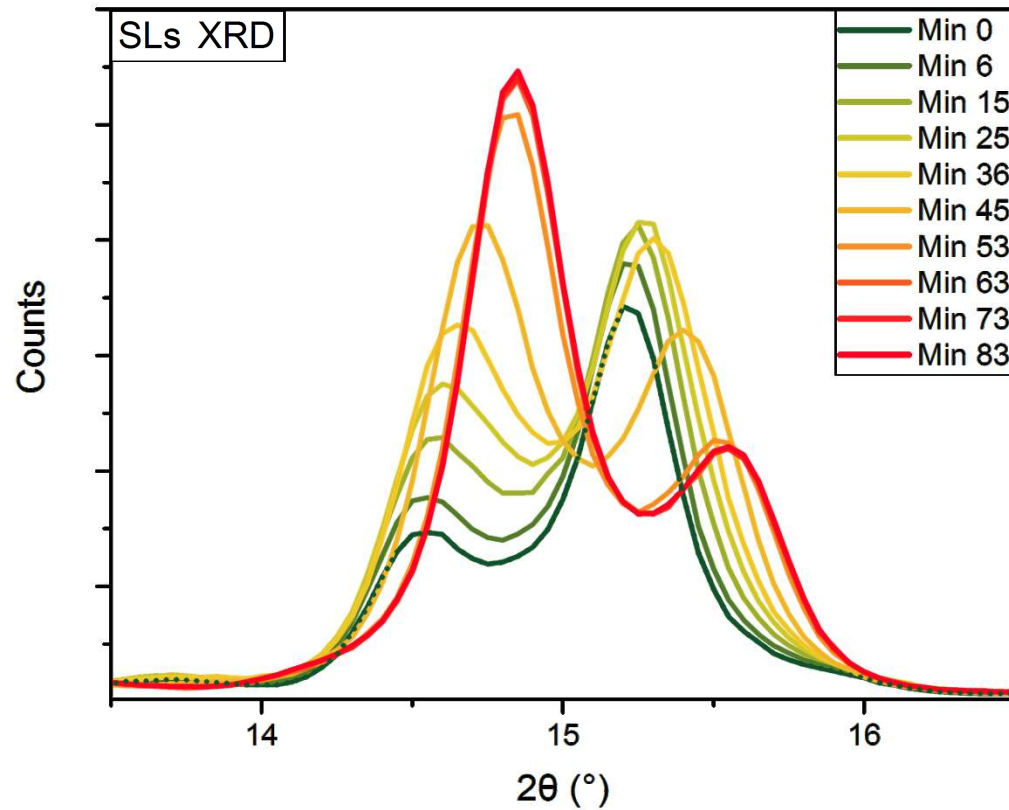
Superlattice Satellites of the 1st Bragg Peak



Schuller, *Phys. Rev. Lett.*, **1980**, 44, 24, 1597-1600
Toso, DB, Giannini, Manna, *ACS Mater. Lett.* **2019**, 1, 2, 272-276

From TEM of monolayers:
11.5-12.6 nm center-to-center distance

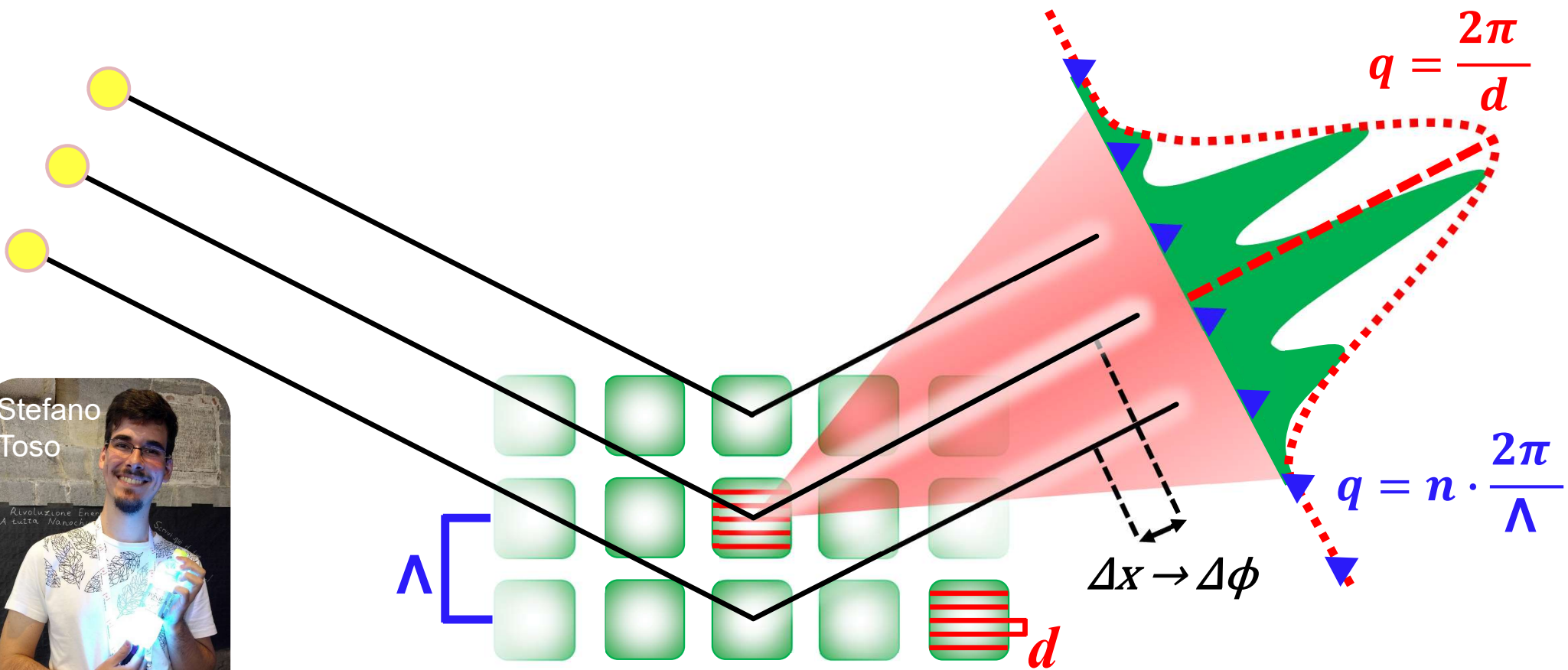
Drying Superlattice in Vacuum



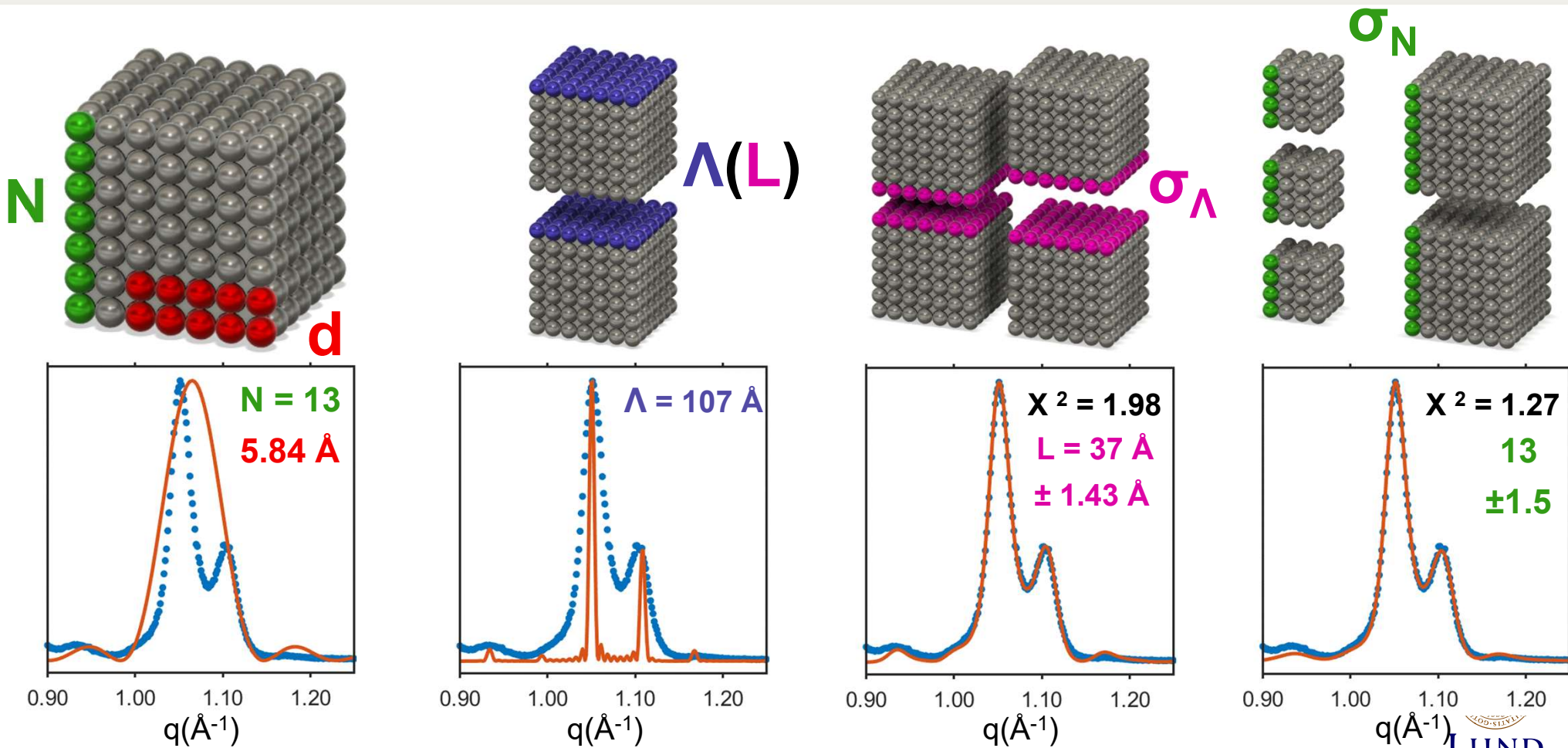
Toso, DB, Giannini, Manna, *ACS Mater. Lett.* **2019**, 1, 2, 272-276

0.3 nm = 3 Å is approx. the size of tetrachloroethylene (C_2Cl_4)

Physical Picture of Superlattice Diffraction



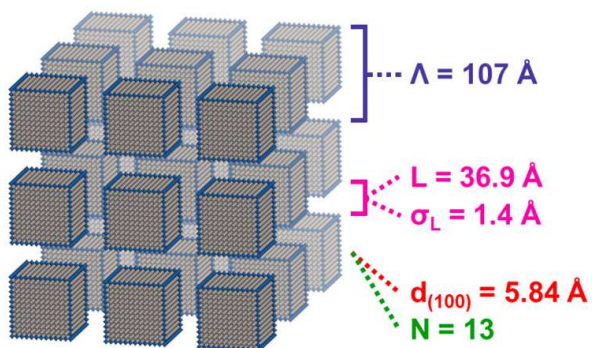
Quantitative Structural Refinement



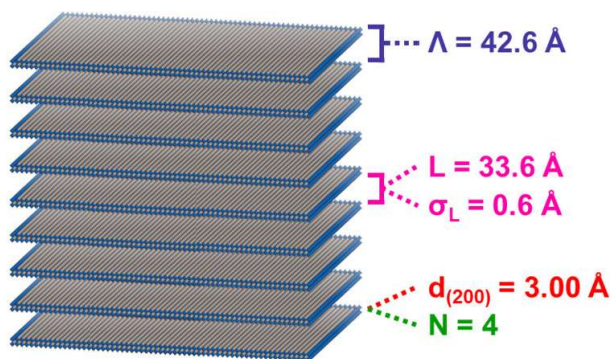
Toso, DB et al, *ACS Nano* **2021**, 15, 4, 6243-6256; model based on Fullerton et al., *Phys. Rev. B*, **1992**, 45, 16, 9292-9310

Quantification of Structural Parameters

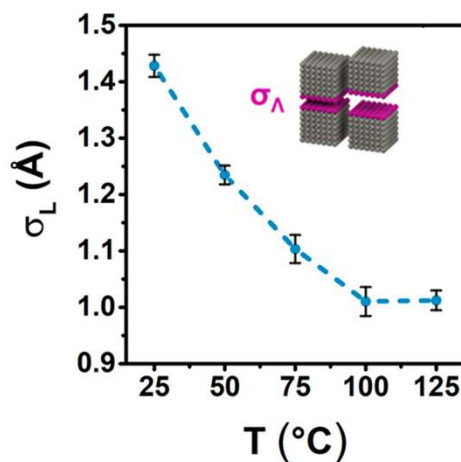
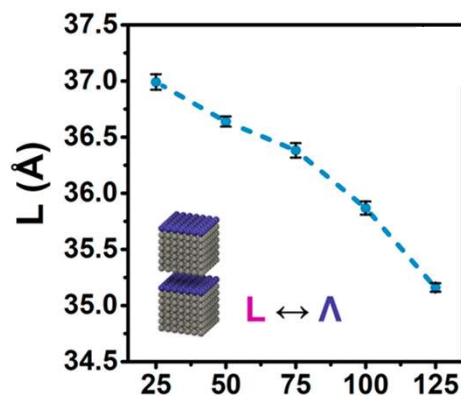
CsPbBr₃ nanocubes



CsPbBr₃ nanoplatelets



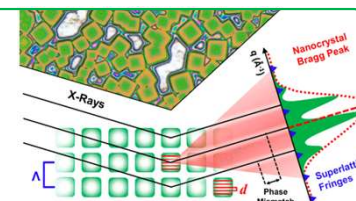
Tracking changes



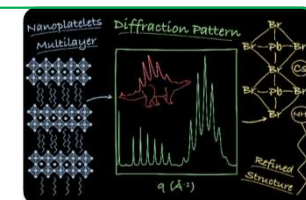
Open-source code (and exp. data)



Toso et al.
ACS Nano **2021**,
15, 4, 6243–6256



Toso et al.
ACS Nano **2021**,
15, 12, 20341–20352

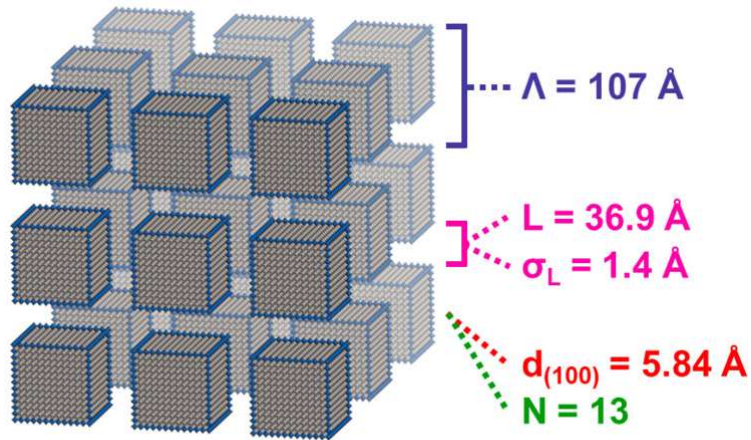


Toso, Baranov, Filippi,
Giannini, Manna
Acc. Chem. Res. **2023**,
56, 1, 66–76

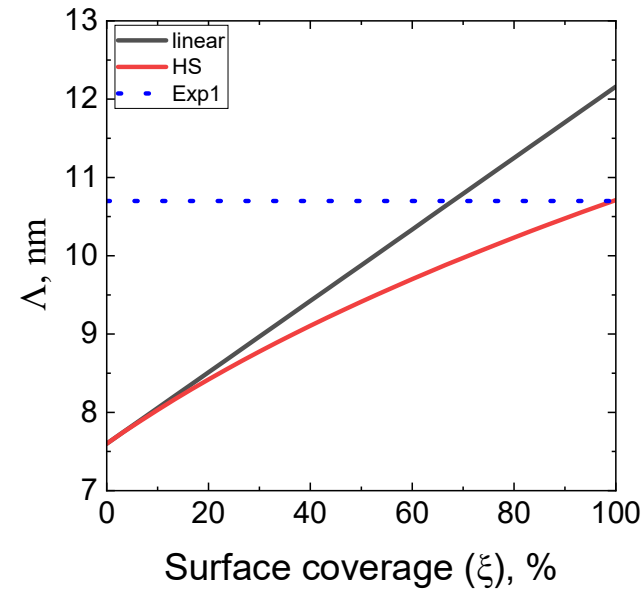
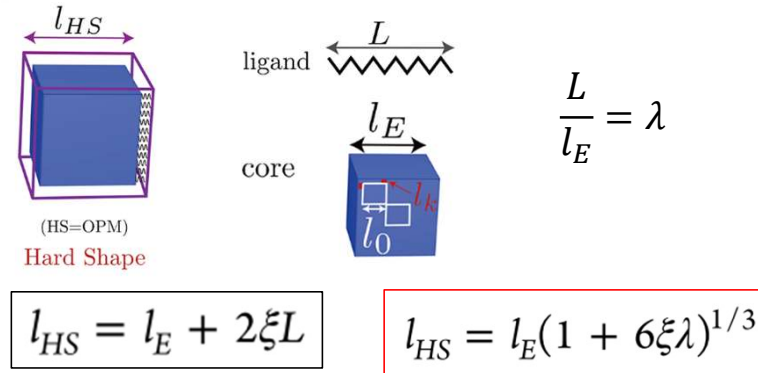


Comparison with Model

CsPbBr₃ nanocubes



Ligand: oleylamine (C18)
Surface coverage: unknown



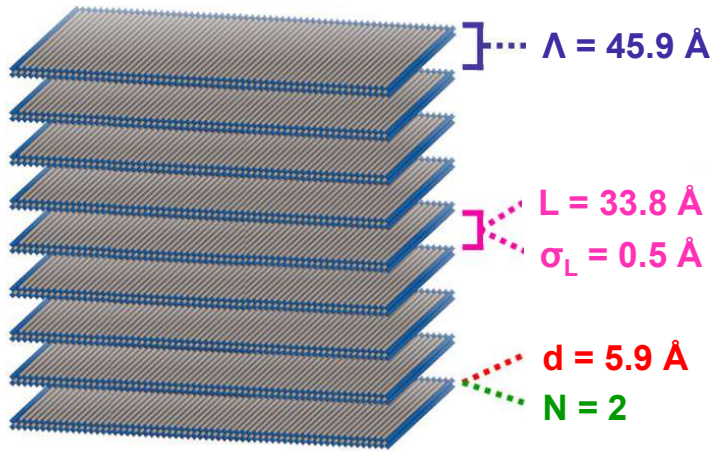
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Exp1: Toso et al., *ACS Nano* **2021**, 15, 4, 6243-6256

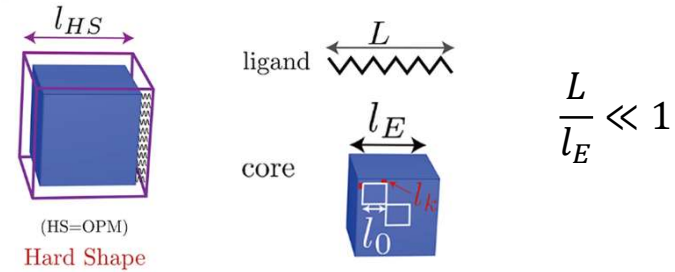
Hallstrom et al., *ACS Nano* **2023**, 17, 8, 7219-7228

Comparison with Model

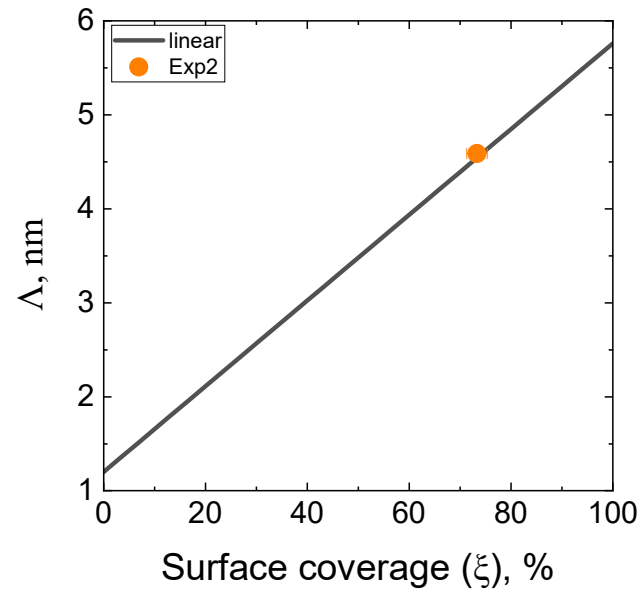
CsPbBr₃ nanoplatelets



Ligand: oleylamine (C18)
Surface coverage: $73 \pm 2\%$

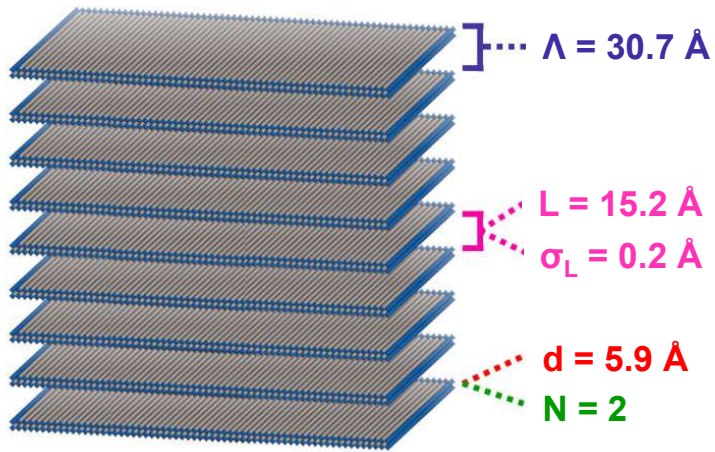


$$l_{HS} = l_E + 2\xi L$$

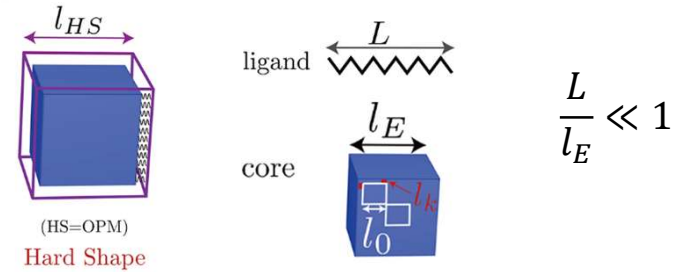


Comparison with Model

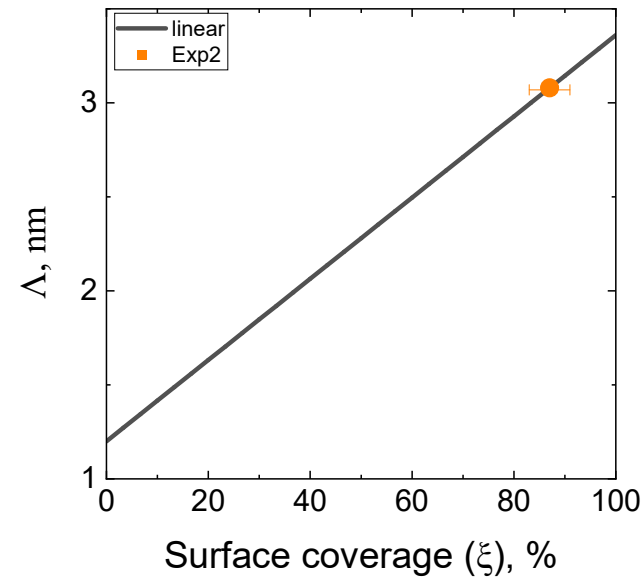
CsPbBr₃ nanoplatelets



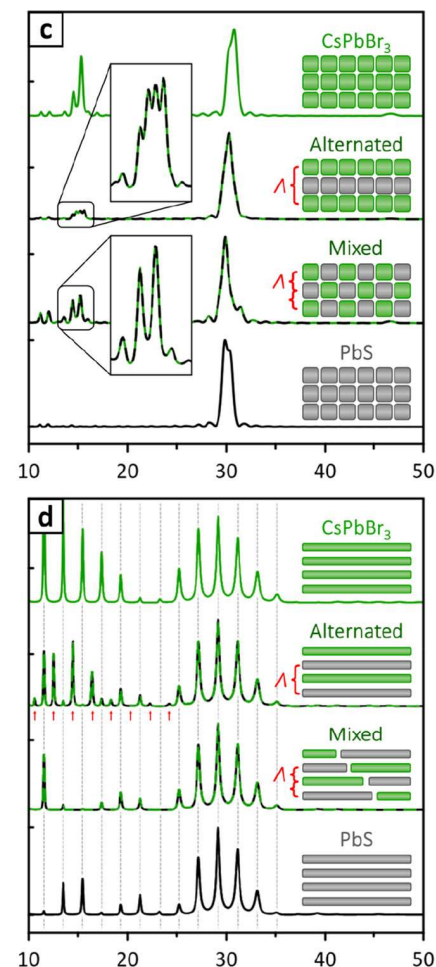
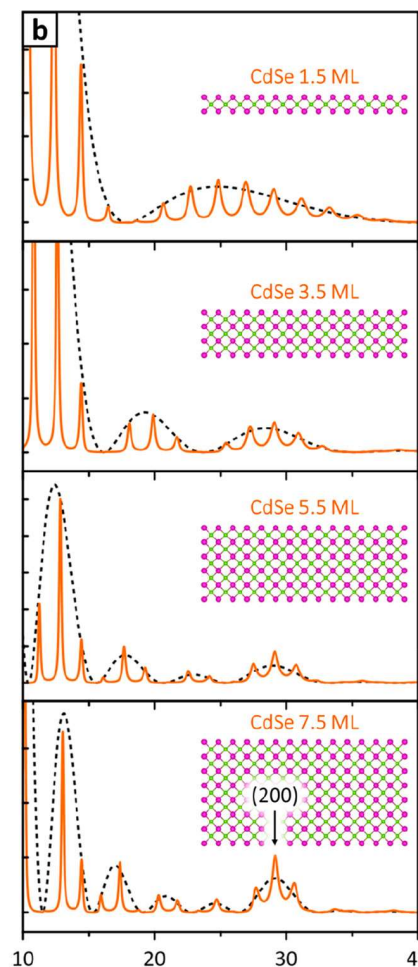
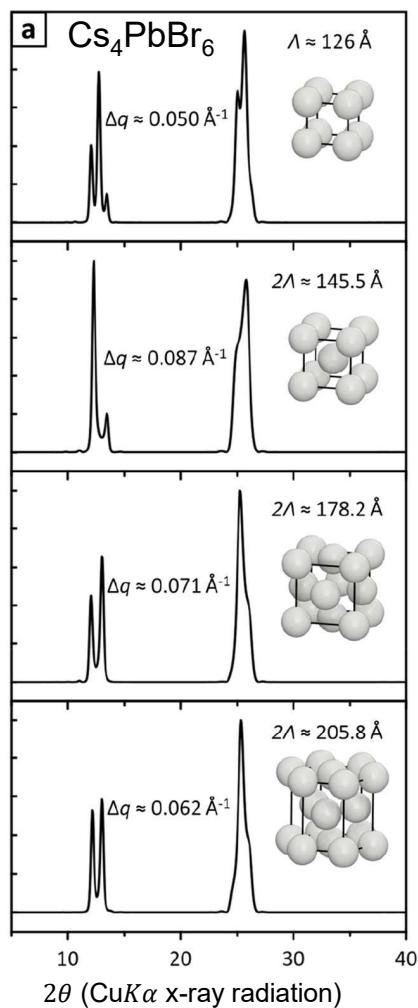
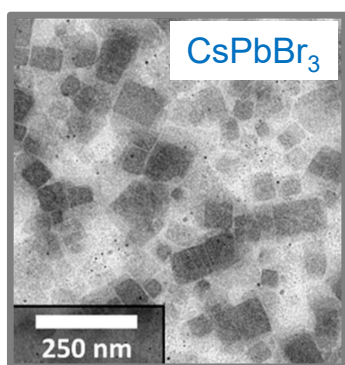
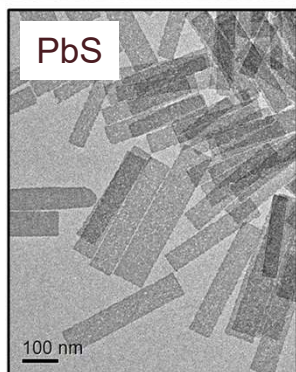
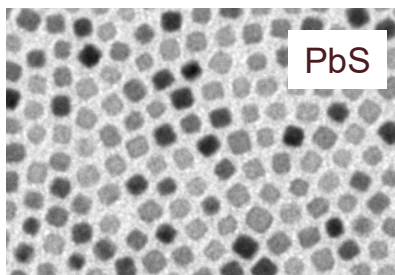
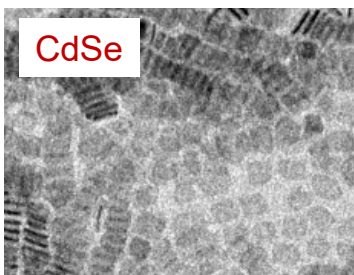
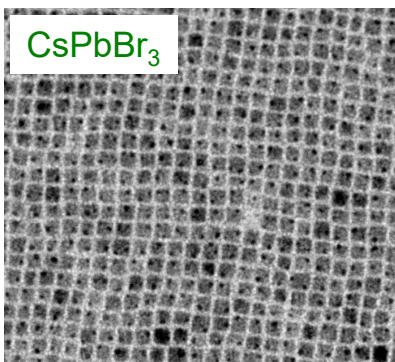
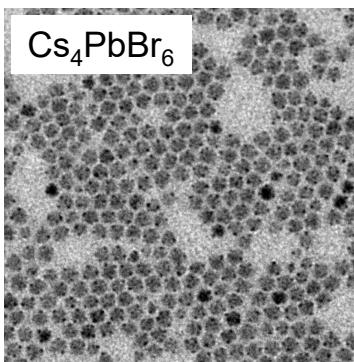
Ligand: octylamine (C8)
Surface coverage: $87 \pm 4\%$



$$l_{HS} = l_E + 2\xi L$$



Predicting Diffraction Patterns



Diffraction and Interference

$$R_1 = A_1 \cos(\omega_1 t + \phi_1)$$

$$R_2 = A_2 \cos(\omega_2 t + \phi_2)$$

...

$$R_n = A_n \cos(\omega_n t + \phi_n)$$

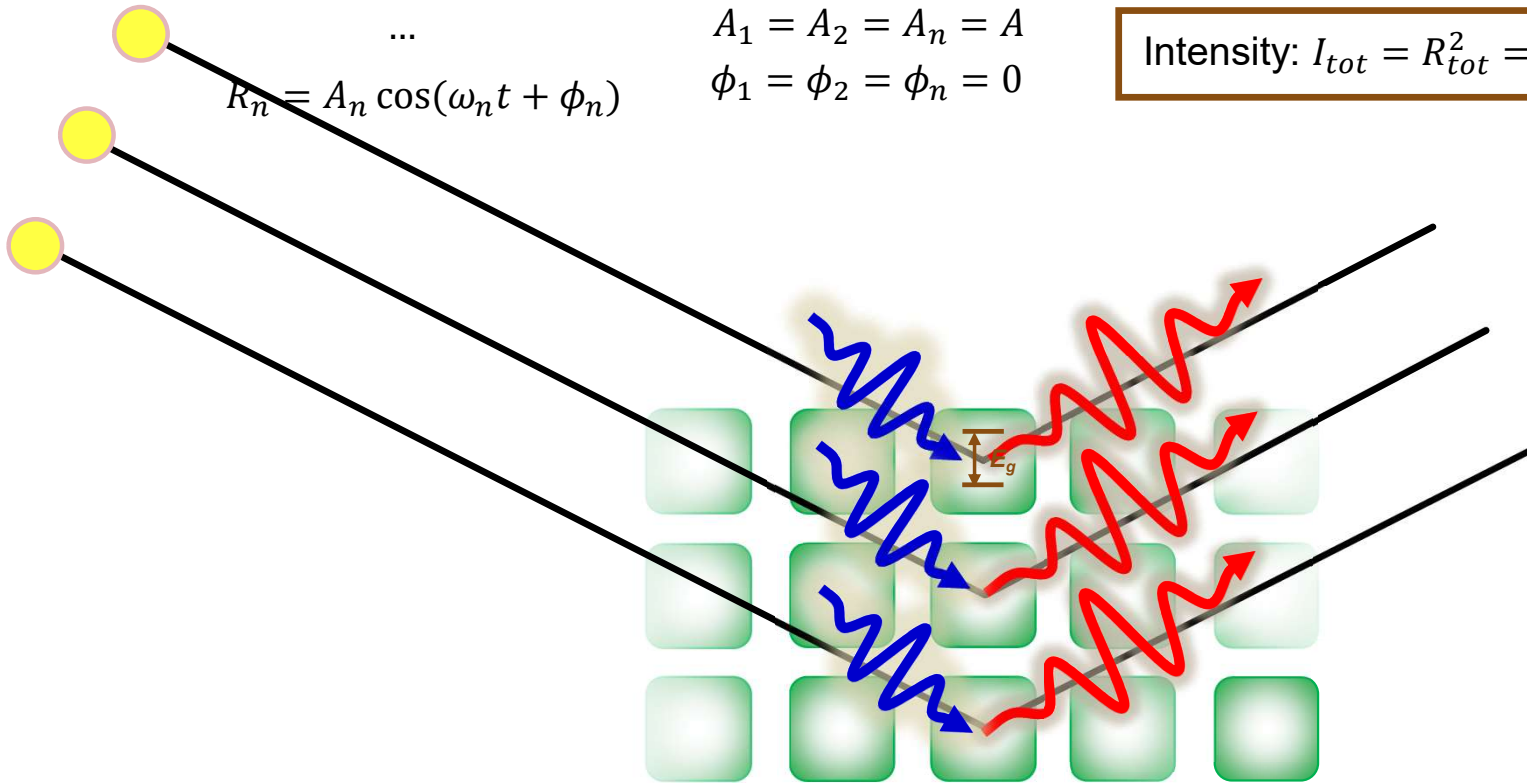
$$\omega_1 = \omega_2 = \omega_n = \omega$$

$$A_1 = A_2 = A_n = A$$

$$\phi_1 = \phi_2 = \phi_n = 0$$

$$\text{Amplitude: } R_{tot} = nA \cos(\omega t)$$

$$\text{Intensity: } I_{tot} = R_{tot}^2 = n^2 A^2 [\cos(\omega t)]^2$$



Experiments on Superfluorescence/-radiance in Perovskites

2018, Raino et al.
Nature, v563p671
 OLAC/OLAM CsPbBr₃,
 CsPbBr₂Cl NC SLs,
 Red-shifted SF, 4K

2020, Krieg et al.
ACS Central Sci., v7p135
 SF in CsPbBr₃@Sulfobetaine
 NC SLs, red-shifted SF, 4K

2021, Cherniukh et al.
Nature, v593p535
 BNSLs CsPbBr₃ NC
 Blue/red-shifted SF, 4K

2022, Cherniukh et al.
ACS Nano, v16p7210
 BNSLs CsPbBr₃ NC
 Collective states

2023, Boehme et al.
ACS Nano, v17p2089
 Rhomb SLs CsPbBr₃
 no SF, 4K

2020, Zhou et al.
Nat. Commun., v11p329
 Cavity-enh SF,
 aged CsPbBr₃ NC SLs

2021, Cherniukh et al.
ACS Nano, v15p16488
 BNSLs CsPbBr₃/LaF₃ disks
 Blue/red-shifted SF, 6K

2022, Zhong et al.
Adv. Opt. Mater., v10p2102290
 Cavity-enh SF,
 aged CsPbBr₃ NC SLs

2023, Be'er et al.
arXiv, 2301.01608
 SF by CL, CsPbBr₃ NCs, 4K

2020, Findik et al.
Nat. Photon., v15p676
 MAPI thin film, **77K**

2021, Wang et al.
Adv. Opt. Mater., v9p2100879
 Single CsPbBr₃ NC,
 superbunching

2022, Biliroglu et al.
Nat. Photon., v16p324
 PEA:CsPbBr₃ thin film, **300K**

2023, Miloch et al.
arXiv, 2303.08791
 Maybe collective state,
 Aged CsPbBr₃ NC SLs, 17K

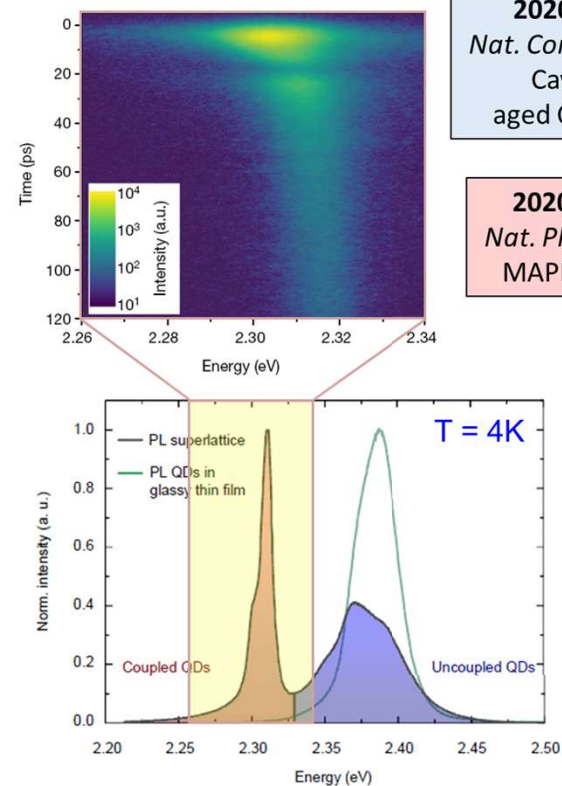
2021, Baranov et al.
ACS Nano, v15p650
 Fresh CsPbBr₃ NC SLs no SF,
 Aged – maybe, bulk effect?

2022, Blach et al.
Nano Lett., v22p7811
 CsPbBr₃ NC SLs, SR

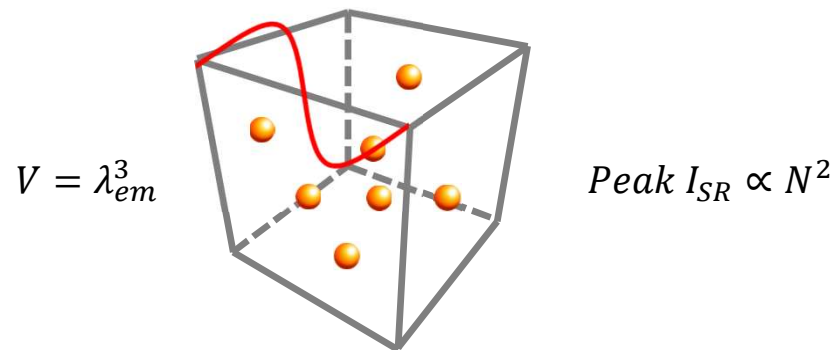
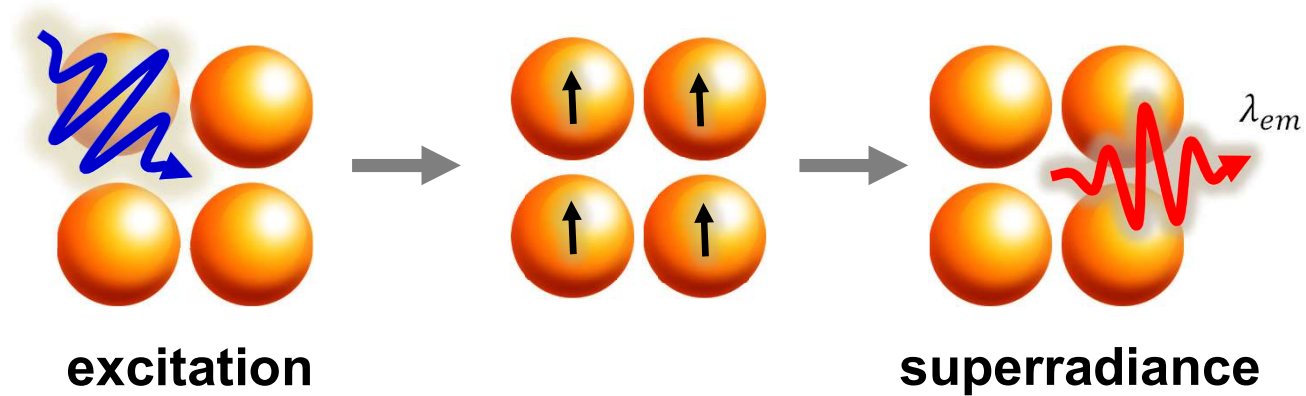
2023, He et al.
JPC Lett., v14p2627
 CsPbBr₃/NaCsYYbErF, 4K
 Magnetic field

2023, Chen et al.
Adv. Sci., 2301589
 Blue Cavity-enh SF,
 CsPbBr₂Cl NC SLs

- What is this effect?
- How to achieve it?

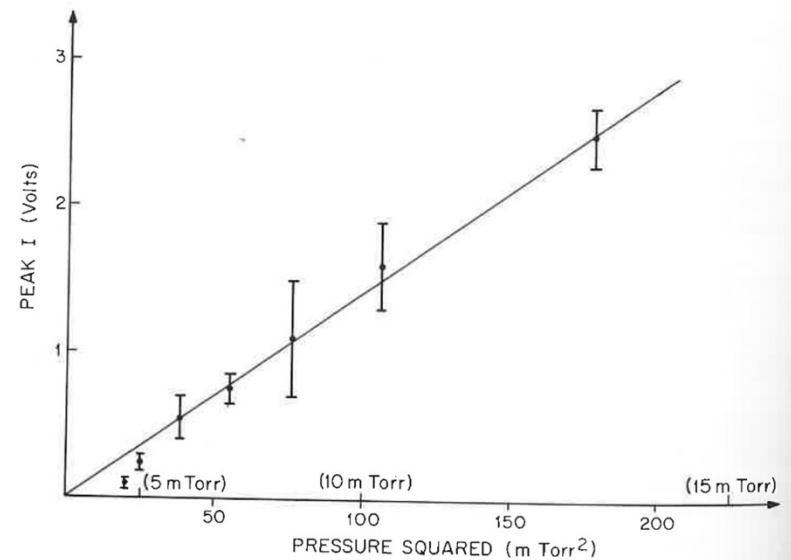
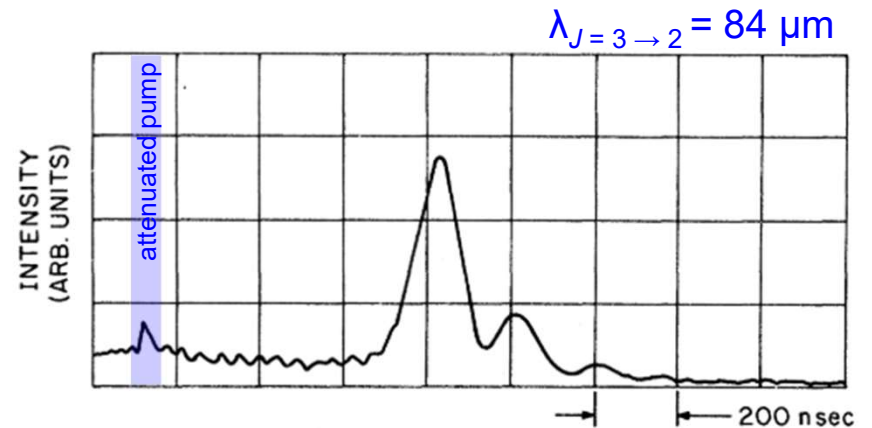
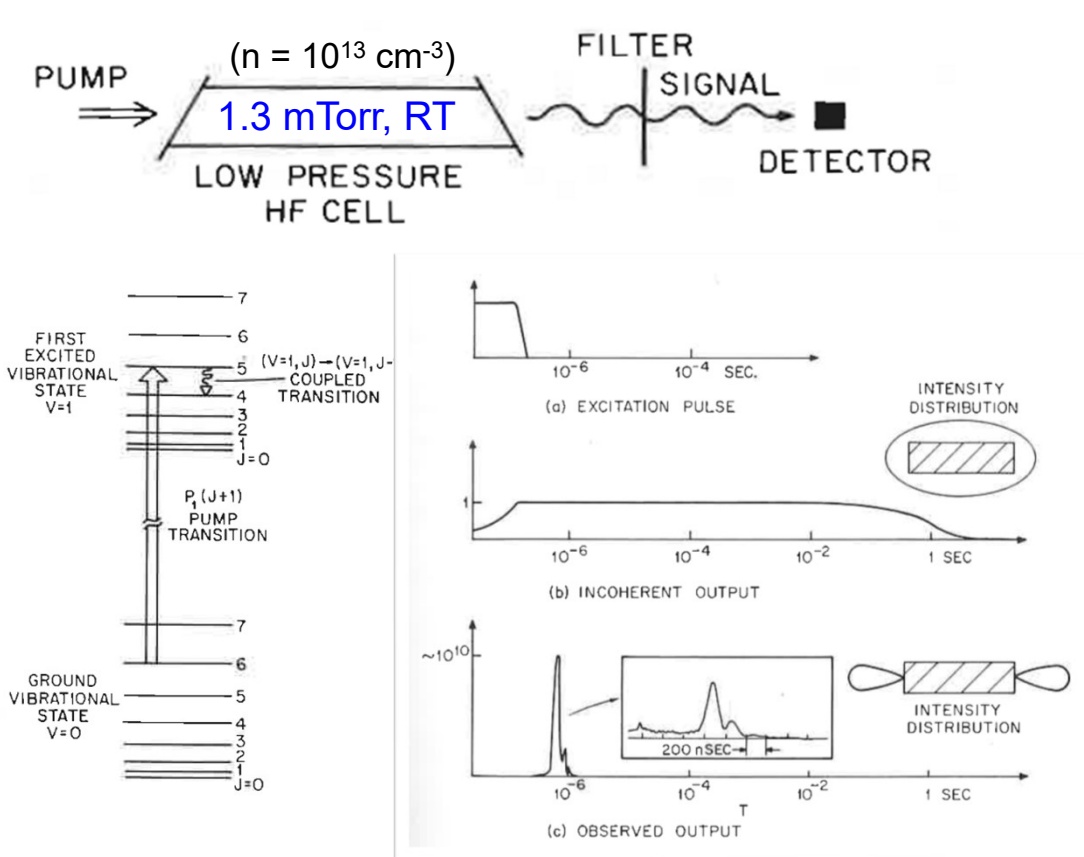


Dicke Superradiance, 1954



Dicke, *Phys. Rev.*, 93 (1), 1954

Experiment with HF gas, 1972

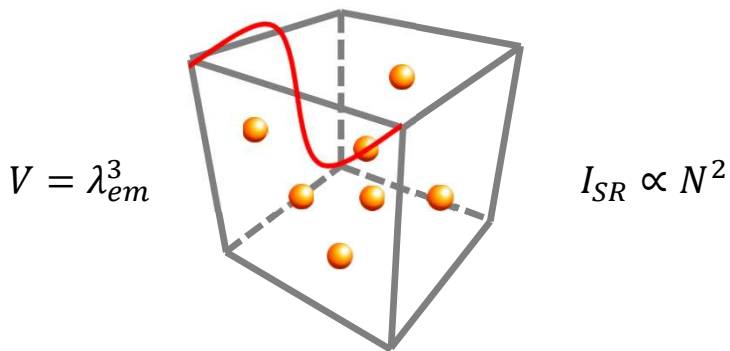
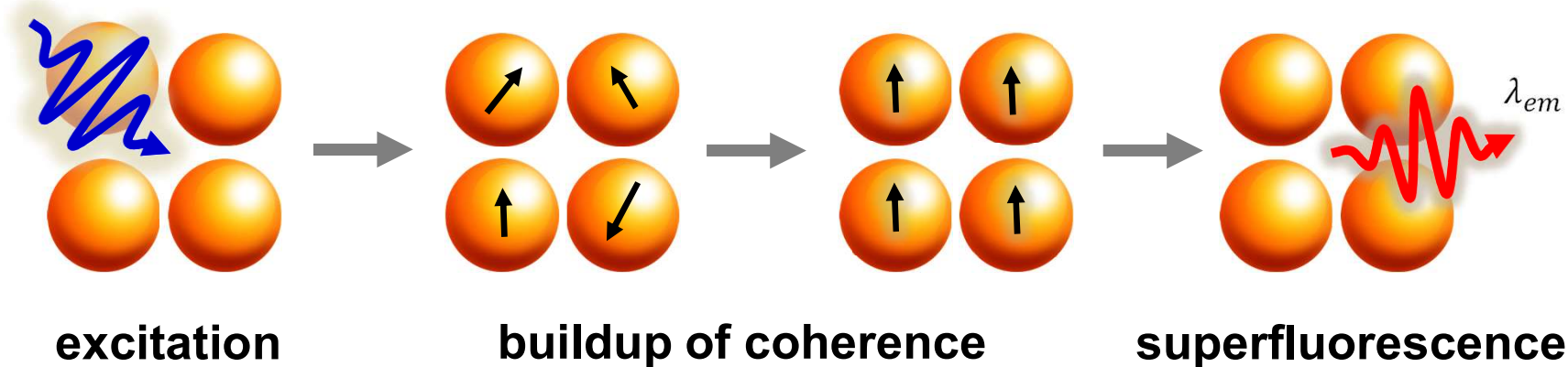


Skribanowitz et al., *Appl. Phys. Lett.* 20 (11), 428-431, 1972

Skribanowitz et al., *PRL* 30 (8), 309-312, 1973

Herman et al., in *Laser Spectroscopy*, 379-492, 1974

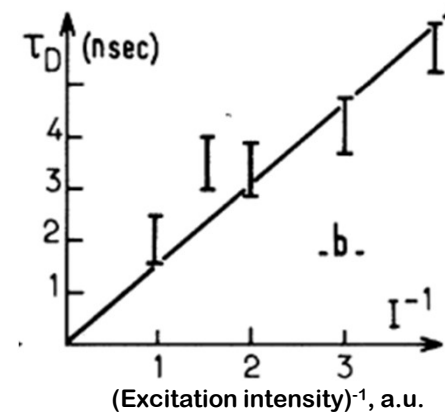
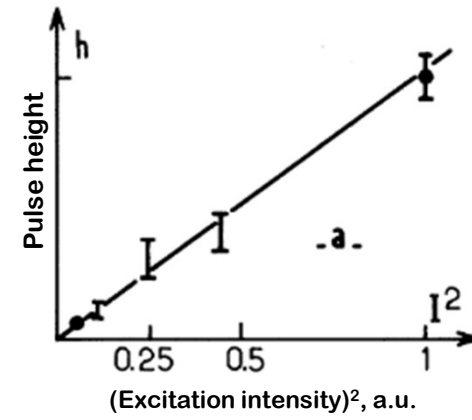
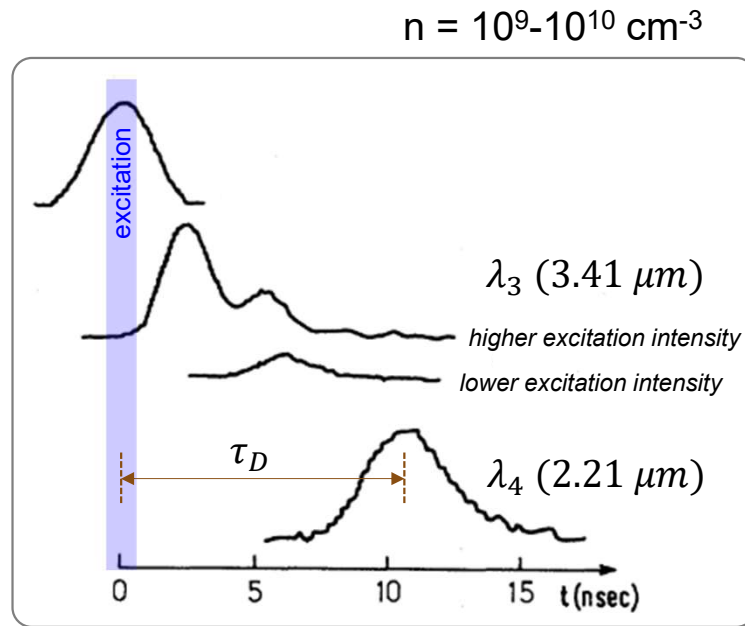
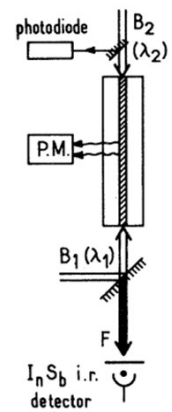
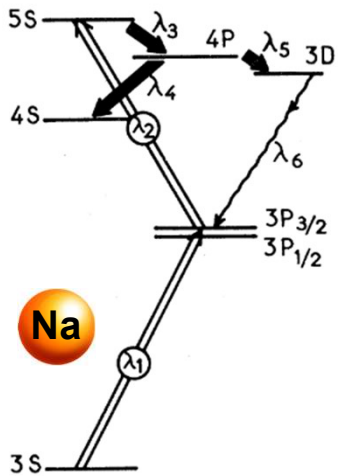
Superfluorescence, 1975



Bonifacio and Lugiato, *Phys. Rev. A*, 11 (5), 1975

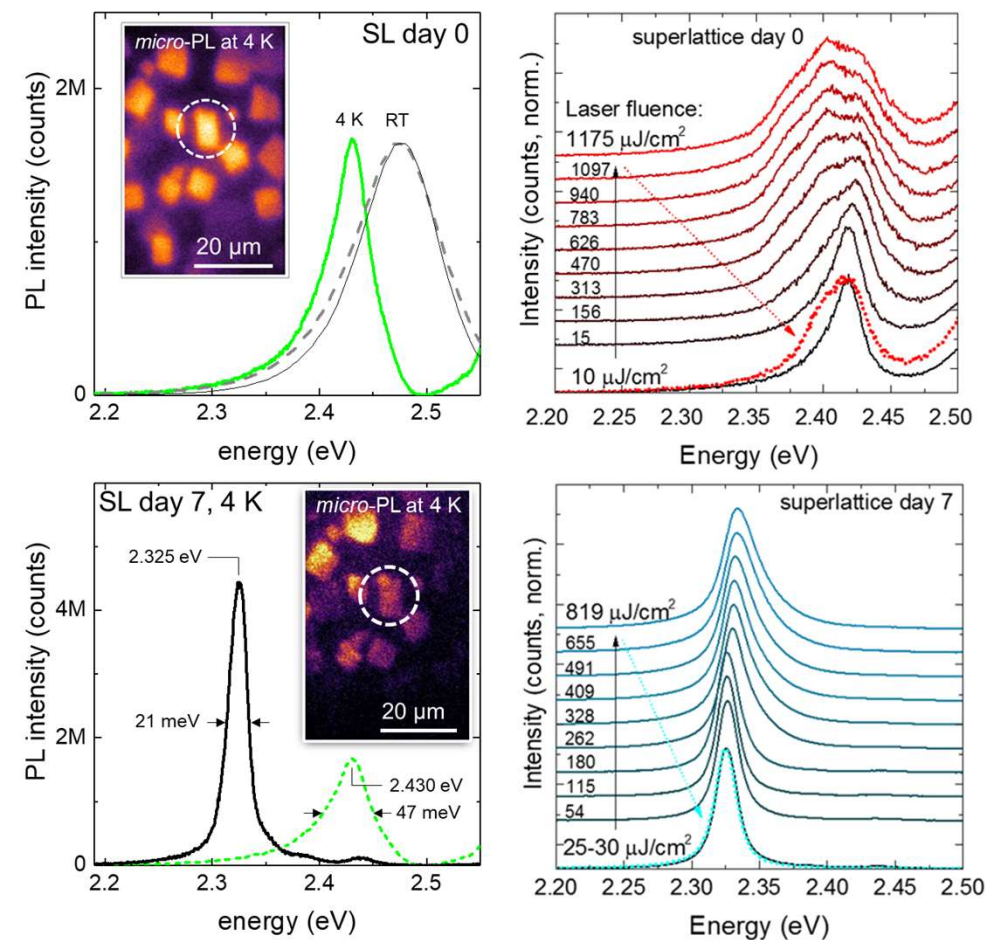


Experiment with Na vapor, 1976

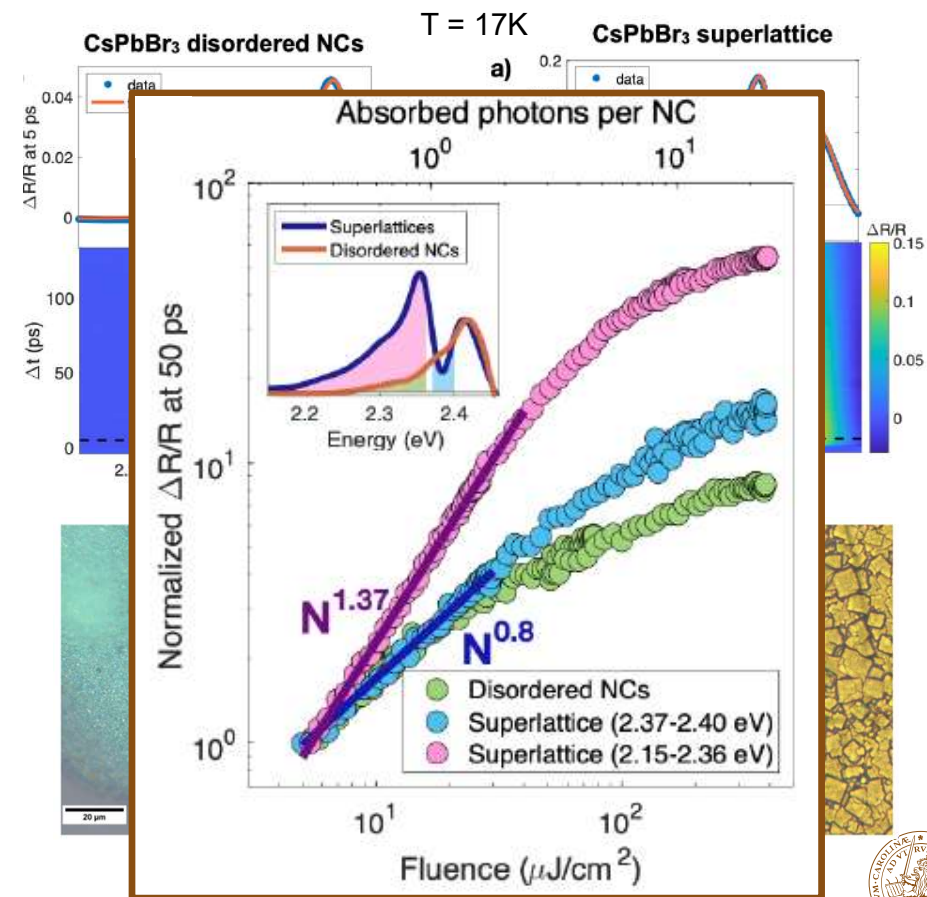


Gross et al., *Phys. Rev. Lett.*, 36 (17), 1035-1038, 1976

Looking for Superfluorescence



Baranov et al., *ACS Nano*, 15 (1), 650-664, **2021**



Miloch et al., *arXiv*, **2023**, 2303.08791



Important Open Questions

Timescales:

- Radiative vs. decoherence vs. cooperative

Disorder:

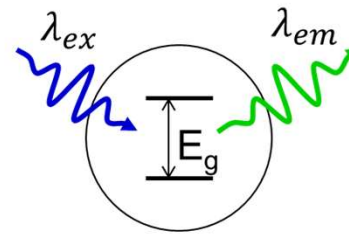
- Energy, position, and orientation

Interactions:

- Dipole-dipole, long-range

Structure:

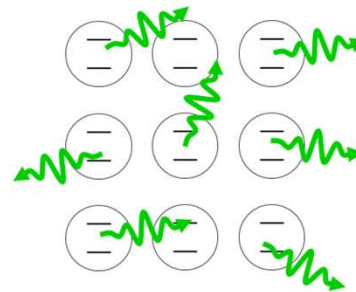
- intrinsic or necessary superlattice



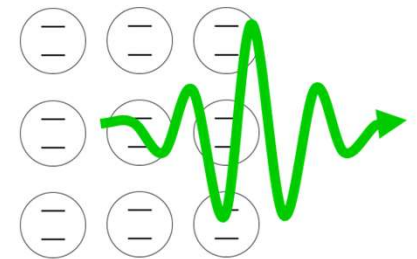
T_1 , radiative decay
 T_2 , decoherence time
 T_{SR} , characteristic time

$$T_1 > T_{SR} > T_2$$

$$T_1 > T_{SR} < T_2$$



$$I_{max}(t) \propto N$$



$$I_{max}(t) \propto N^2$$





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This research has received funding from the European Union's Horizon 2020 research and innovation programme under the MSCA grant agreement No 794560 (RETAIN), Nanochemistry Department @IIT, Horizon Europe ERC Starting Grant No. 101039683 (PROMETHEUS), and Faculty of Science, Lund University.